

Supporting Information

Structural and electronic properties of 3,3'-Azothiophene photo-switching systems.

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PDB file for *anti(trans)*-*trans* conformer of 3,3'-azothiophene

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PDB file for *anti(trans)*-*cis* conformer of 3,3'-azothiophene

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PDB file for *syn(cis)*-*cis* conformer of 3,3'-azothiophene

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PDB file for *syn(cis)*-*trans* conformer of 3,3'-azothiophene

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PDB file for *anti(trans)*-*trans* conformer of 3,3'-azothiophene-methoxycarbonyl

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PDB file for *anti(trans)*-*cis* conformer of 3,3'-azothiophene-methoxycarbonyl

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PDB file for *syn(cis)*-*cis* conformer of 3,3'-azothiophene-methoxycarbonyl

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PDB file for *syn(cis)*-*trans* conformer of 3,3'-azothiophene-methoxycarbonyl

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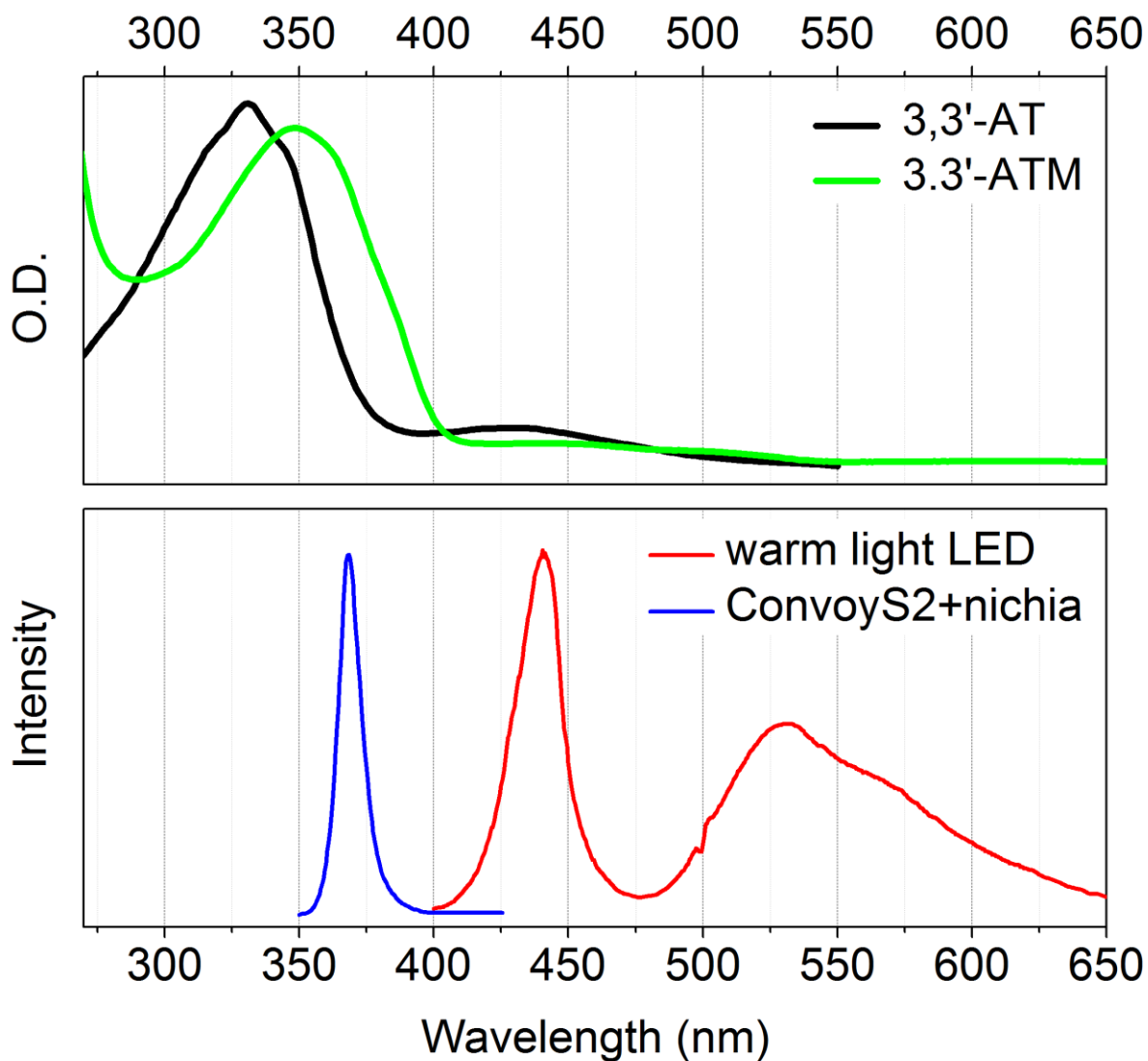


Figure S1. Upper panel: UV-VIS spectra of 3,3'-AT (black line) and 3,3'-ATM (green line) at room temperature under thermal equilibrium. Lower panel: output spectra of warm light LED and Convoy S2+ nichia LED used for photo-switching of azothiophene molecular systems.

Table S1. Contributions of Canonical Molecular Orbitals (CMO) into the most intense UV-VIS optical transitions for *anti(trans)*-cis conformer of 3,3' Azothiophene.

N	λ (nm)	Strength	CMO involved
1	448.4	0	H-1->L (99%)
2	360.8	0.7039	H->L (98%)
3	320	0.115	H-2->L (95%)
4	315.2	0.0606	H-3->L (97%)
7	243.2	0.1299	H->L+1 (83%)
8	232.7	0.0869	H->L+2 (90%)
13	213.3	0.1268	H-2->L+1 (97%)
14	210.3	0.1562	H-3->L+1 (85%)

Table S2. Contributions of CMO into the most intense UV-VIS optical transitions for *anti(trans)*-trans conformer of 3,3' Azothiophene.

N	λ (nm)	Strength	CMO involved
1	469.9	0	H-1->L (99%)
2	380.3	0.5852	H->L (99%)
3	334.9	0	H-2->L (98%)
4	322.9	0.239	H-3->L (98%)
5	266.6	0	H-4->L (93%)
6	245	0	H-1->L+1 (99%)
8	228.6	0.1497	H->L+2 (93%)
13	212.1	0.348	H-2->L+1 (92%)

Table S3. Contributions of CMO into the most intense UV-VIS optical transitions for *syn(cis)*-trans conformer of 3,3' Azothiophene.

N	λ (nm)	Strength	CMO involved
1	471.4	0.0854	H->L (88%), H-3->L (11%)
2	326.2	0.0032	H-1->L (98%)
3	315.4	0.0036	H-2->L (95%)
4	308.2	0.3579	H-3->L (87%), H->L (10%)
5	280.1	0.0457	H-4->L (96%)
6	276	0.0805	H->L+1 (95%)
11	222.9	0.1608	H-1->L+1 (61%), H-2->L+1 (17%), H->L+4 (10%)
12	222.9	0.0621	H-2->L+1 (64%), H-1->L+1 (17%), H-1->L+2 (11%)
13	219.4	0.0248	H-3->L+1 (90%)

Table S4. Contributions of CMO into the most intense UV-VIS optical transitions for *syn(cis)*-cis conformer of 3,3' Azothiophene.

N	λ (nm)	Strength	CMO involved
1	472.7	0.0694	H->L (89%)
2	336.7	0.0221	H-1->L (88%)
3	315.2	0.0855	H-2->L (79%)
4	298.8	0.131	H-3->L (81%), H-2->L (11%)
5	289.4	0.0338	H-4->L (80%), H->L+1 (15%)
6	272.4	0.1385	H-4->L (14%), H->L+1 (75%)
7	252.3	0.0375	H->L+2 (92%)
9	230.8	0.1384	H-1->L+1 (77%), H->L+3 (11%)
12	223.2	0.0567	H-2->L+1 (71%)

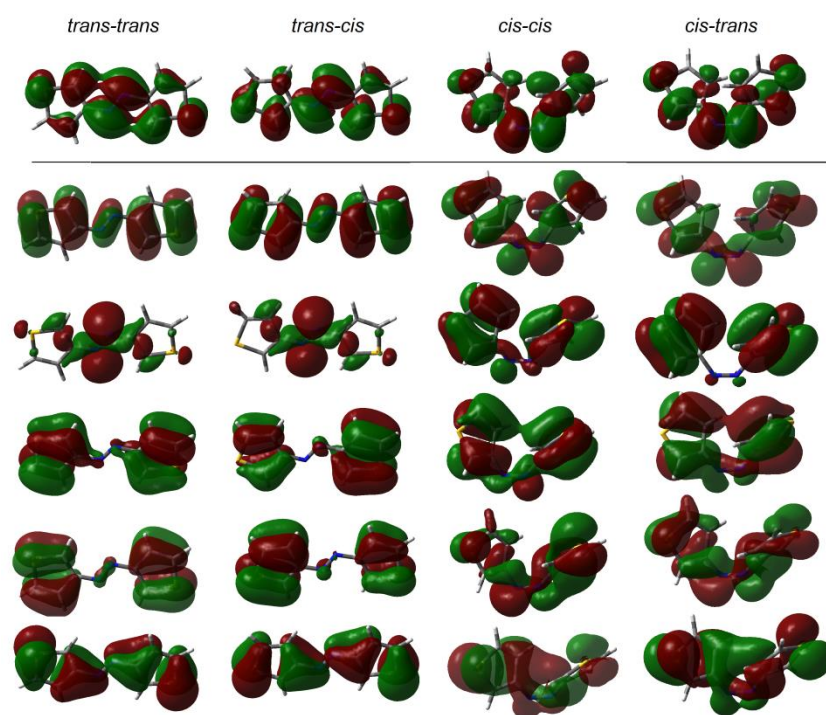


Figure S2. CMO for the conformers of 3,3' Azothiophene: HOMO at the top, LUMO and the lower states are presented below in descending order

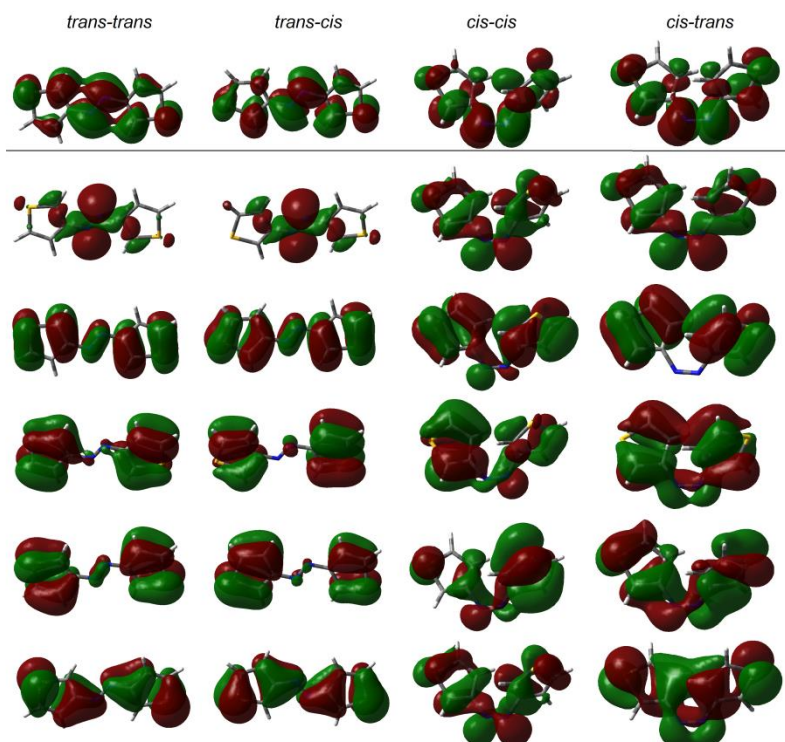


Figure S3. Natural Transition Orbitals (NTO) for the conformers of 3,3' Azothiophene: ELECTRON (receiving) states are shown at the top, HOLE states are presented below in descending order according to the energy gap of the corresponding electron-hole pairs.

Table S5. Contributions of CMO into the most intense UV-VIS optical transitions for *anti(trans)*-cis conformer of 3,3'-Azothiophene-Methoxycarbonyl.

N	λ (nm)	Strength	MO involved
1	510.3	0.0517	H->L (73%), H-1->L (17%)
2	371.9	0.2586	H-1->L (79%), H->L (18%)
3	343.9	0.2447	H-2->L (87%)
4	341.1	0.1421	H-3->L (94%)
10	259.7	0.2072	H-1->L+1 (68%), H->L+1 (15%)
11	248.9	0.0606	H-2->L+1 (56%), H-7->L (15%), H-3->L+1 (12%)
15	238.5	0.1555	H-1->L+2 (70%), H->L+2 (11%)
19	221.2	0.0885	H-3->L+2 (64%), H-2->L+2 (16%)

Table S6. Contributions of CMO into the most intense UV-VIS optical transitions for *anti(trans)*-trans conformer of 3,3'-Azothiophene-Methoxycarbonyl.

N	λ (nm)	Strength	CMO involved
1	468.2	0	H->L (93%)
2	372.1	0.3311	H-1->L (98%)
3	342	0	H-2->L (97%)
4	330.1	0.1904	H-3->L (98%)
5	297.8	0.0793	H->L+1 (93%)
11	253.7	0.1479	H-2->L+1 (87%)
13	241.8	0.3153	H-1->L+2 (85%)

Table S7. Contributions of CMO into the most intense UV-VIS optical transitions for *syn(cis)*-trans conformer of 3,3'-Azothiophene-Methoxycarbonyl.

N	λ (nm)	Strength	MO involved
1	477.9	0.0905	H->L (90%)
2	350.1	0.0038	H-1->L (93%)
3	336.9	0.0389	H-2->L (97%)
4	314.8	0.3128	H-4->L (85%)
6	295.5	0.054	H->L+2 (89%)
10	253.3	0.245	H-1->L+1 (95%)
12	245.1	0.1168	H-1->L+2 (78%), H-2->L+1 (12%)
13	241.2	0.0507	H-2->L+2 (93%)
18	227.3	0.0936	H-4->L+2 (85%)

Table S8. Contributions of CMO into the most intense UV-VIS optical transitions for *syn(cis)*-cis conformer of 3,3'-Azothiophene-Methoxycarbonyl.

N	λ (nm)	Strength	CMO involved
1	483.9	0.0489	H->L (91%)
2	349.1	0.0148	H-1->L (85%)
3	333	0.0162	H-2->L (95%)
4	317.7	0.0663	H->L+1 (63%), H-3->L (17%), H-4->L (10%)
5	311.2	0.0793	H-4->L (53%), H-3->L (36%)
6	299.2	0.1631	H->L+2 (37%), H->L+1 (11%), H-3->L (17%), H-4->L (11%)
11	259.7	0.1859	H-2->L+1 (80%), H-1->L+1 (10%)
13	244.8	0.0712	H-1->L+2 (46%), H-4->L+1 (40%)
14	240.9	0.1182	H-4->L+1 (39%), H-1->L+2 (21%), H-7->L (19%)

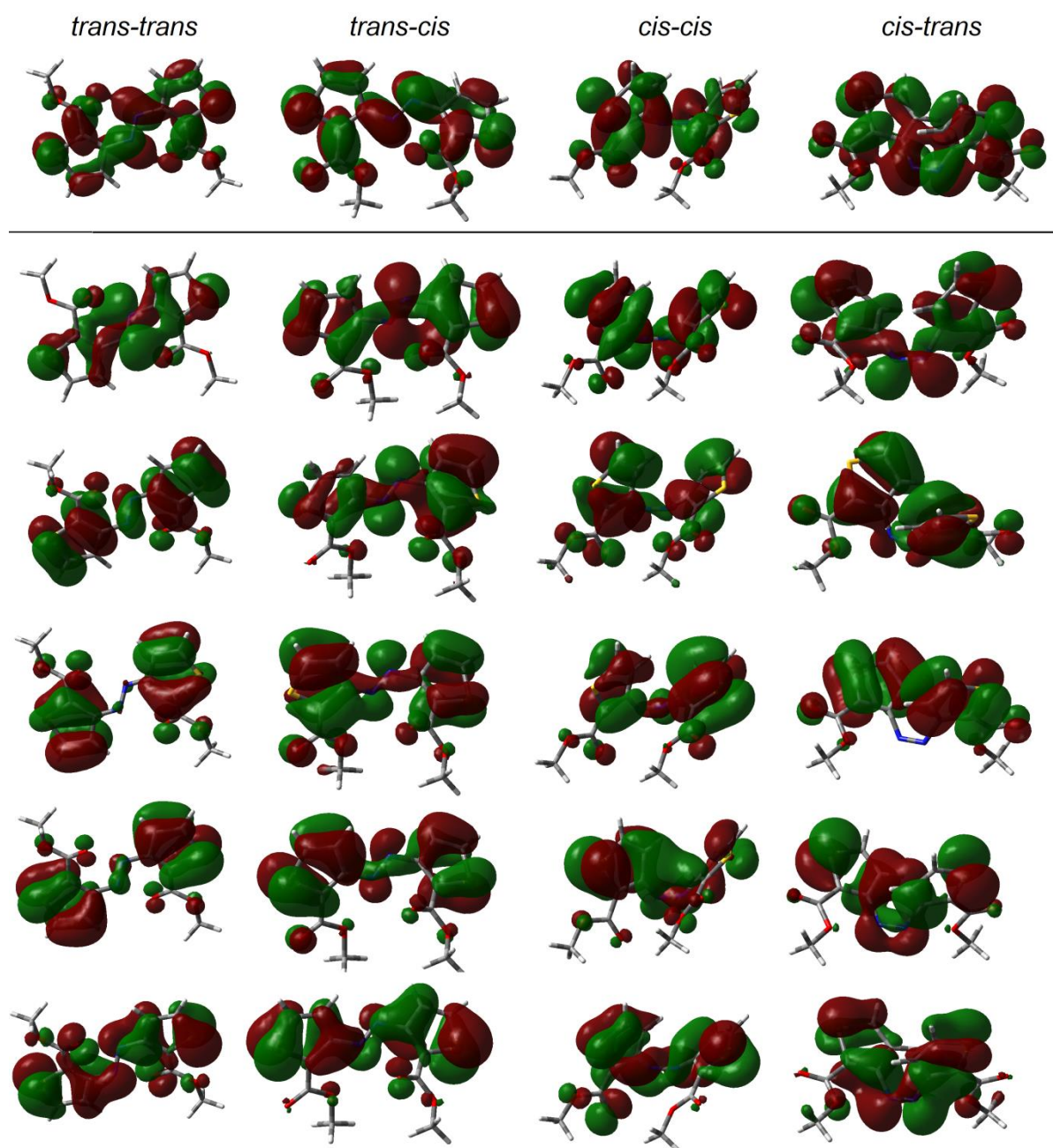


Figure S4. CMO for the conformers of 3,3'-Azothiophene-Methoxycarbonyl: HOMO at the top, LUMO and lower states are presented below in descending order

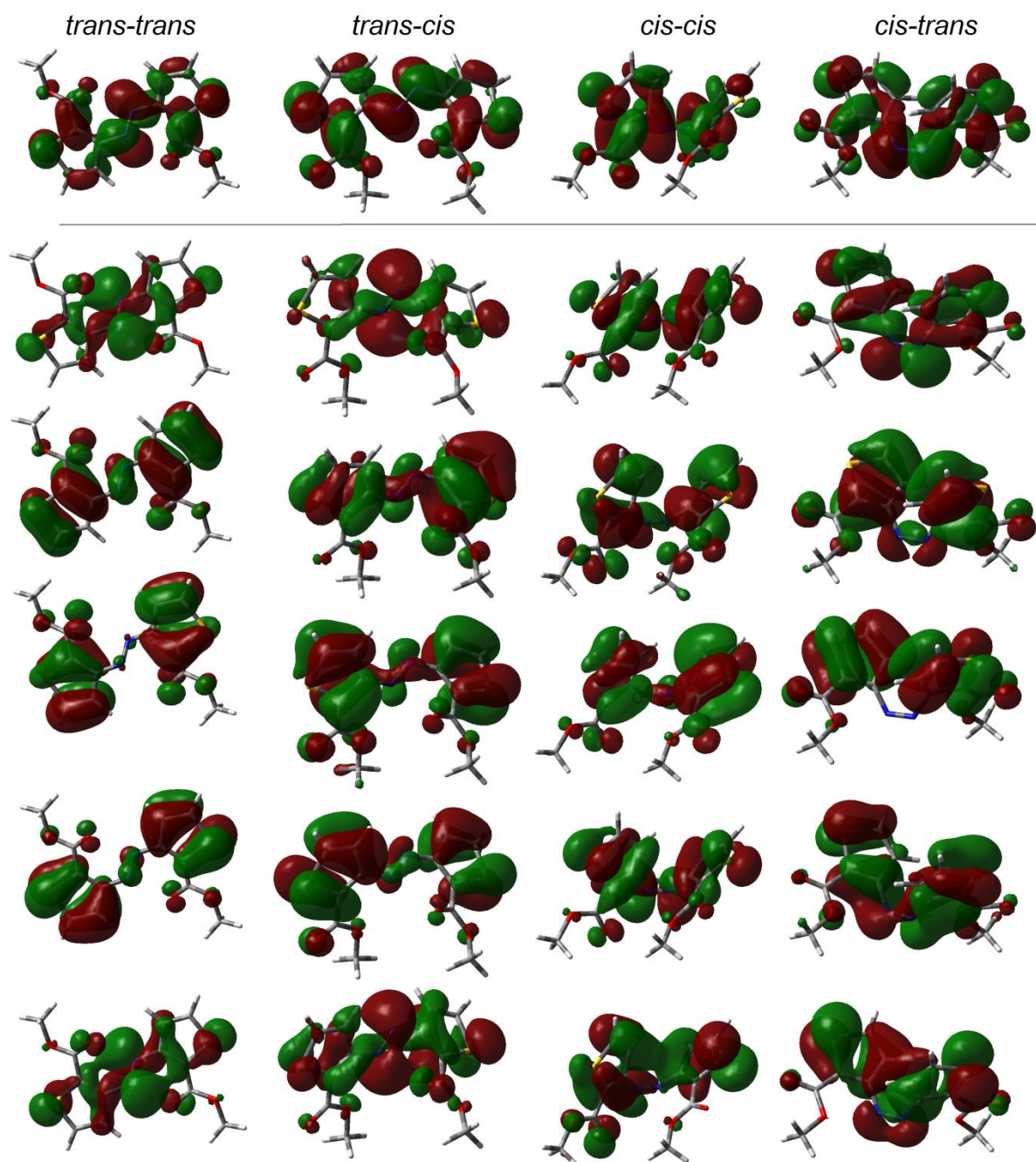


Figure S5. NTO for the conformers of 3,3'-Azothiophene-Methoxycarbonyl: ELECTRON (receiving) states are shown at the top, HOLE states are presented below in descending order according to the energy gap of the corresponding electron-hole pairs.

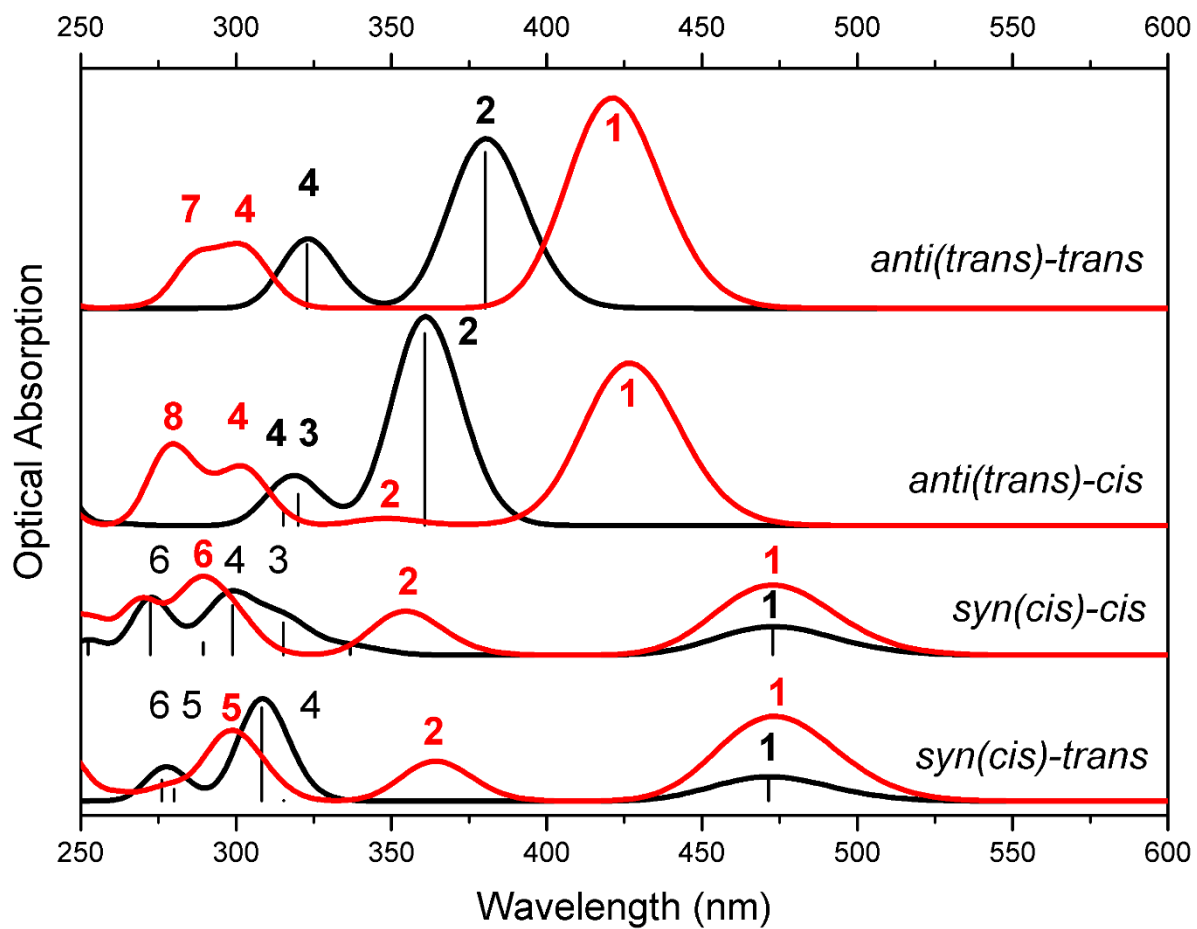


Figure S6. Calculated UV-VIS spectra for the four lowest energy conformers of 3,3'-AT (black lines) and 3,3'-AT-NH₂ (red lines).

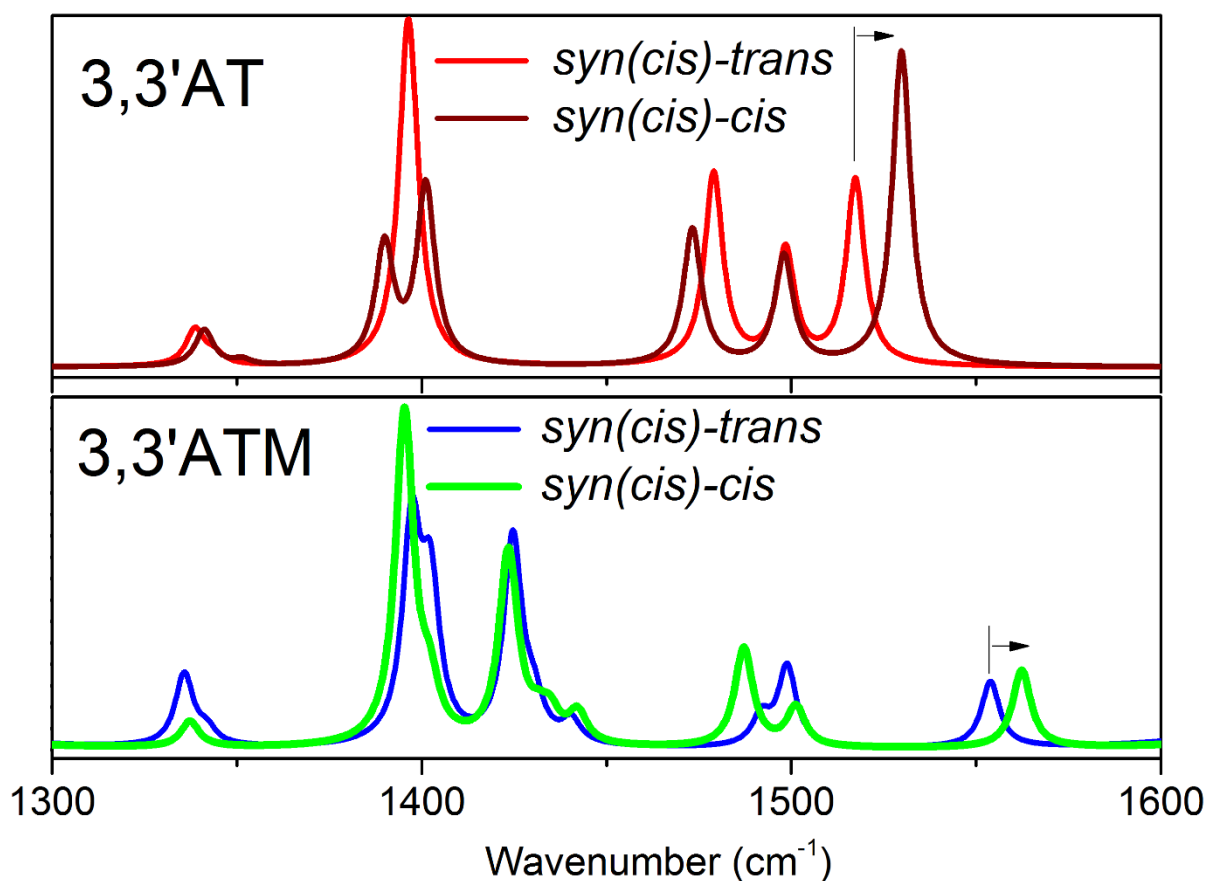


Figure S7. Comparison of DFT calculated infrared spectra of *syn(cis)-trans* and of *syn(cis)-cis* for 3,3'-AT and 3,3'-ATM as indicated. Frequency and intensity of mode 42 of *syn(cis)-cis* 3,3'-AT are higher and stronger comparing to the frequency and intensity of the same mode 42 of *syn(cis)-trans* 3,3'-AT. At the same time, in Figure 3b in the main text the photo-induced difference FTIR spectrum indicated no strong positive spectral signature at and about 1529 cm^{-1} . This lends additional support, beside the probabilities (see values in brackets in Figure 1 of the main text) according to the energies of *syn(cis)-trans* and of *syn(cis)-cis* conformers, that the *syn(cis)-trans* conformer of 3,3'-AT is a dominant conformer of the photoproduct state for 3,3'-AT.

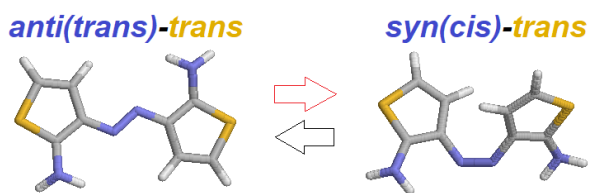


Figure S8. NH_2 substituted 3,3'-azothiophene (3,3'-AT- NH_2), as the NH_2 side groups are of electron donating character.

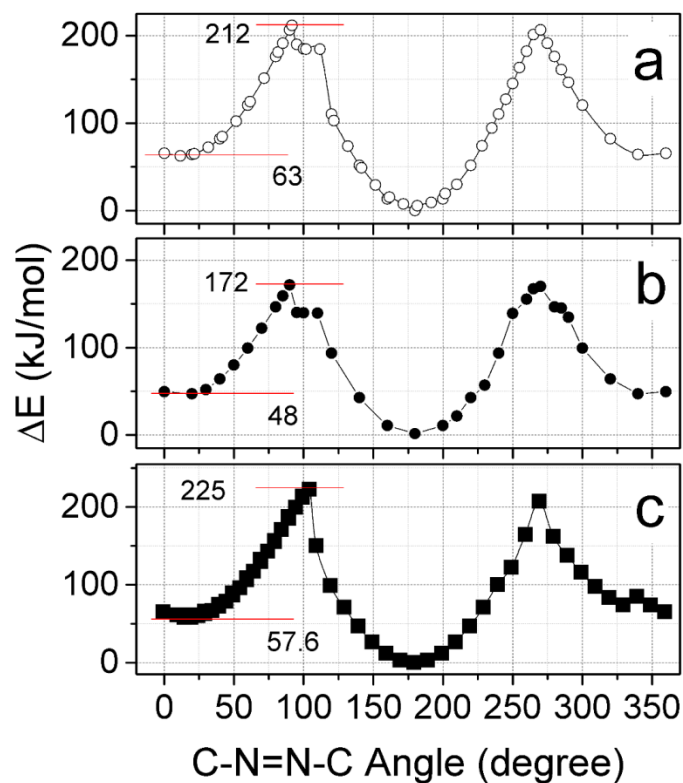


Figure S9. Thermodynamic properties. a-c: Potential energies for optimized conformers of 3,3'-AT, 3,3'-ATM and 3,3'-AT- NH_2 , respectively, sampled for a series of fixed $\text{C}_3\text{NN}'\text{C}'_3$ dihedral angles along the path from *syn* to *anti* geometry of the $-\text{N}=\text{N}-$ moiety.

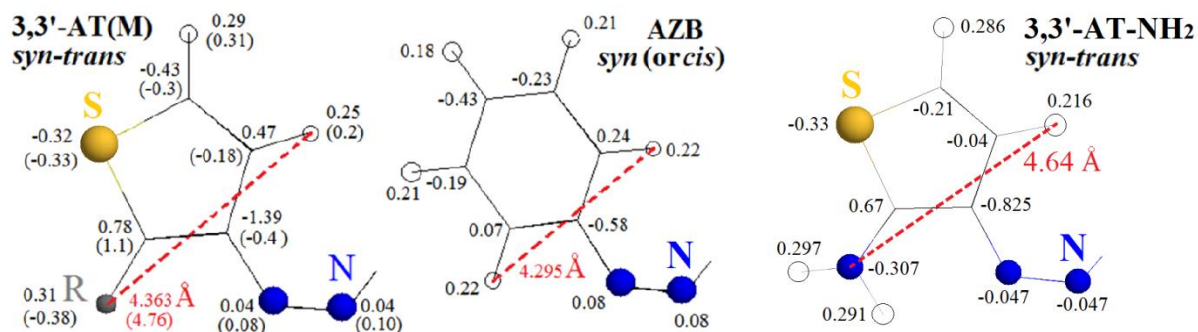


Figure S10. Mulliken atomic partial charges and geometric properties for the structural moieties of *syn(cis)-trans* 3,3'-AT, 3,3'-ATM, *cis(syn)* azobenzene and *syn(cis)-trans* 3,3'-AT- NH_2 . R is either hydrogen or the side group of 3,3'-AT, 3,3'-ATM, respectively. Properties of 3,3'-ATM are provided in brackets.