

1 **Title page**

2 **Microalgae-derived hydrochar application on rice paddy soil: Higher rice yield**
3 **but increased gaseous nitrogen loss**

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26 **Highlights**

27 ■ *Chlorella vulgaris* hydrochars (CVH) were fabricated by hydrothermal

28 carbonization

29 ■ CVH addition improved N use efficiency, sugar content, and grain yield of rice

30 ■ CVH addition stimulated NH₃ volatilization and N₂O emission from paddy soil

31 ■ Compared to direct addition of CV, CVH addition inhibited NH₃ volatilization

32 ■ Increasing gaseous N loss results from physiochemical and microbiological factors

33 **Abstract**

34 Hydrothermal carbonization represents a promising technique for transforming
35 microalgae into the hydrochar with abundant phytoavailable nutrients. However, the
36 effects of microalgae-derived hydrochars on the gaseous nitrogen (N) loss from
37 agricultural field are still unclear. *Chlorella vulgaris* powder (CVP) and two *Chlorella*
38 *vulgaris*-derived hydrochars that employ water (CVHW) or citrate acid solution
39 (CVHCA) as the reaction medium were applied to a soil column system grown with
40 rice. The temporal variations of nitrous oxide (N₂O) emissions and ammonia (NH₃)
41 volatilization were monitored during the whole rice-growing season. Results showed
42 that CVHW and CVHCA addition significantly increased the grain yield (by 13.5-26.8%
43 and 10.5-23.4%) compared with control and CVP group, while concomitantly
44 increasing the ammonia volatilization (by 53.8% and 72.9%) as well as N₂O emissions
45 (by 2.17- and 2.82-fold) from paddy soil compared to control. The microbial functional
46 genes (*AOA*, *AOB*, *nirK*, *nirS*, *nosZ*) in soil indicated that CVHW and CVHCA
47 treatment stimulated the nitrification and denitrification, and inhibited the N₂O
48 oxidation in soil. Notably, CVHW was recommended in the view of improving yield
49 and controlling NH₃ volatilization because no significant difference of the yield-scale
50 NH₃ volatilization was detected between control and CVHW treatment. This study for
51 the first time uncovered that *Chlorella vulgaris*-derived hydrochars have positive
52 effects on rice N utilization and growth but negative effects on the atmospheric
53 environment.

54 **Keywords:** Ammonia volatilization; *Chlorella vulgaris*; Hydrothermal carbonization;
55 Nitrogen use efficiency; Nitrous oxide emission; Non-point pollution

56 **1. Introduction**

57 Blue and green microalgae are predominant biological pollutants of harmful algal
58 blooms in eutrophicated waterbodies that have severe impacts on aquatic ecosystems
59 and human health (O'Neil et al., 2012; Zhang et al., 2016). These microalgae are
60 nutrients-rich, can store the inorganic nitrogen (N) and phosphorus (P) in excess within
61 the cells in the form of protein and polyphosphate (Solovchenko et al., 2016), and thus
62 own potential to be transformed from biowaste to biofertilizer (Ray et al., 2013;
63 Mukherjee et al., 2015; Santos and Pires, 2018). *Chlorella*, a green microalga, contains
64 significant quantities of N and P (up to 7%–12% and 1%–3% of their cell dry weight,
65 respectively) (Powell et al., 2009; Cabanelas et al., 2013; Zhu et al., 2015). However,
66 direct application of microalgae did not deliver significant difference on the growth of
67 wheat (Schreiber et al., 2018) or rice (Ray et al., 2013; Mukherjee et al., 2015), because
68 the dominant forms of the stored N and P in microalgae are proteins and polyphosphates
69 that are difficult to decompose in soil and unable to be directly utilized by plants.
70 Hydrothermal carbonization (HTC), which can transform the microalgae biomass into
71 hydrochars, is recommended because has been demonstrated to transform most
72 polyphosphates and proteins from lignocellulosic feedstock (Funke et al., 2013; Kruse
73 et al., 2016) and biosolids (Huang and Tang, 2015; Huang et al., 2017; Yu et al., 2019)

74 into orthophosphate and ammonium or nitrate. Also, HTC is cost-effective to avoid
75 dehydrating the microalgae collected from the wastewater.

76 Rice (*Oryza sativa* L.) is a primary food source for more than half of the world's
77 population (Khush, 2005). When applying microalgae-derived hydrochar to rice paddy
78 fields as a fertilizer, the effects on atmospheric environment should be considered,
79 because rice fields have been confirmed as a major emission source of ammonia (NH₃)
80 volatilization and nitrous oxide (N₂O) emissions (Bhattacharyya et al., 2013; Kim et al.,
81 2015; Wang et al., 2018). Ammonia (NH₃) volatilization from agricultural fields is a
82 major non-point pollution (Norse, 2005) because it is widely dispersed to the
83 atmospheric environment, leading to atmospheric pollution, such as forming particulate
84 matters (e.g., PM_{2.5}, with diameters ≤ 2.5 μm) (Zhao et al., 2017; Dubache et al., 2019).
85 In paddy fields, N is often applied in excess of plant demand. As a result, large
86 proportions of overused N fertilizers are lost via NH₃ volatilization (Sun et al., 2017;
87 Sun et al., 2019; Hayashi et al., 2008). NH₃ volatilization from agricultural fields
88 accounts for 10-60% of the total nitrogen input (Tilman et al., 2011). In addition, the
89 midseason and final drainage periods stimulate nitrification, thus caused substantial
90 nitrous oxide (N₂O) emissions from paddy fields (Bhattacharyya et al., 2013; Ali et al.,
91 2015; Zhou et al., 2018).

92 Generally, biochar application can reduce soil pH, and increase porosity, aeration,
93 and redox potential, thus reducing NH₃ volatilization or N₂O emission by altering soil
94 properties such as NH₄⁺-N adsorption and activity of nitrobacterium and

95 denitrobacterium (Nelissen et al., 2014; Feng et al., 2017; Mandal et al., 2019; Sha et
96 al., 2019). However, these controversial results also indicate that biochar application
97 can potentially promote NH₃ volatilization (Sun et al., 2017; Feng et al., 2018b; H. Sun
98 et al., 2019b) or N₂O emissions (Duan et al., 2018; Senbayram et al., 2019; Zhang et
99 al., 2019), depending on different properties of soil or feedstock of biochars.
100 Microalgae-derived biochars own a number of distinctive attributes compared with
101 biochars from other feedstocks. For instance, the surface area of *Chlorella vulgaris*-
102 derived biochar was markedly lower than biochar produced from lignocellulosic
103 biomass (Wang et al., 2013), which may reduce the adsorption capacity to ammonium
104 (NH₄⁺). The hydrolysis via HTC promotes the chain breakage in macromolecules,
105 which may provide more labile carbon (C) and N pools to nitrobacteria and
106 denitrobacteria. Therefore, it is necessary to investigate the effects of microalgae-
107 derived hydrochar application on gaseous N loss from rice paddy soils.

108 In this study, the primary objective was to investigate the response of NH₃
109 volatilization and N₂O emissions when applying microalgal hydrochar as fertilizers.
110 We also aimed to identify the key biogeochemical factors influencing gaseous N loss
111 in rice paddy soil. We hypothesize that the addition of microalgae-derived hydrochars
112 would increase N gasification loss from paddy soil because of introduced labile C and
113 N sources to soil.

114 **2. Materials and Methods**

115 **2.1. Hydrochar production**

116 *Chlorella vulgaris*, a common microalgal strain existing in eutrophication, was
117 employed in this study and acquired from Guangyu BioScience Limited, Shanghai. The
118 inoculum was maintained in 2 L borosilicate bioreactors using sterilized medium, 3N-
119 BBMV. The operational conditions were as follows: constant aeration using 2.5% CO₂
120 at 0.2 vvm, a photoperiod of 14:10 light : dark cycles, 150 μmol/m²/s of luminance and
121 temperature of 25 ± 1 °C. Microalgal cells were collected from a cultivation broth using
122 centrifugation at 8,000 g for 5 min at 4 °C, washed using distilled water and lyophilized.

123 HTC was conducted in a high-pressure (approximate 8 Mpa, auto-generated during
124 HTC) hydrothermal reactor, using a solid:liquid ratio of 1:10 (w/w). The structure of
125 high-pressure hydrothermal reactor is shown in **Fig S1**. The reactor was sealed and
126 heated at 260 °C for 1 h and then allowed to naturally cool down to room temperature
127 overnight. The solid hydrochars produced by HTC were collected by centrifugation,
128 and dried at 70 °C until no further weight loss. Two hydrochars were produced using
129 different reaction media: CVHW (employing deionized water) and CVHCA
130 (employing 1 wt.% citric acid). Citrate acid was added to increase the hydrochar yield
131 ([Heilmann et al., 2010](#)), reduce hydrochar pH and promote the degradation of proteins
132 and polyphosphates by acidic hydrolysis ([Huang et al., 2017](#)).

133 **2.2. Soil column experiment design**

134 Polyvinyl chloride sleeves (diameter: 30 cm diameter; height: 50 cm) were filled
135 with dry paddy soil each. The paddy soil, which was classified as hydroagric Stahnic
136 Anthrosol, was collected from a paddy field in Yixing, Jiangsu Province in China.
137 During the rice season of the paddy field in 2017, the local average temperature was
138 25.7 °C and total precipitation was 1082.5 mm. Soil-column experiments were
139 performed in the glasshouse of Jiangsu Academy of Agriculture Science, with natural
140 light and temperature. Each soil column was filled with 35 kg of paddy soil. The soil
141 was air-dried and ground to pass through a 2 mm sieve. The soil had the following
142 properties: pH 6.42 (soil : water, 1 : 2.5), organic matter content, 2.28%, total N, 1.56
143 g kg⁻¹, total P 0.96 g kg⁻¹, and total potassium 4.12 g kg⁻¹. Rice (*Oryza sativa* L.,
144 Nangeng 46) was grown in the soil-column with three plants. All experiments were
145 performed in triplicate.

146 Prior to sowing, the *Chlorella vulgaris* powder (CVP), CVHW or CVHCA were
147 mixed with paddy soil at an application rate of 1%, respectively. Rice plants were
148 transplanted on June 29, 2018 and harvested on November 9, 2018. All treatments were
149 applied with 240 kg urea-N ha⁻¹ throughout the rice-growing season. This amount of
150 N fertilizer was applied thrice: as a basal fertilizer (BF) prior to transplanting, as a first
151 supplementary fertilizer (SF1) after tillering and as a second supplementary N fertilizer
152 (SF2) after panicle formation at a proportion of 4:4:2. The BF was applied at a rate of
153 96 kg N ha⁻¹, 96 kg P₂O₅ ha⁻¹ and 192 kg K₂O ha⁻¹ in the form of urea, calcium
154 superphosphate, and potassium chloride. No hydrochar was applied to the control, but

155 chemical fertilizers were applied in the same rate as other treatments. All plants were
156 flooded to a water level of 3–5 cm and a mid-season drainage was done during the
157 period from 7-18 August, 2018.

158 ***2.3. Characterization of hydrochar***

159 The pH value of hydrochar samples was determined using a solid/Milli-Q water ratio
160 of 1:2.5 (w/v). Total C, N, hydrogen (H), and sulfur (S) contents of hydrochar were
161 determined by an Elemental Analyzer (EL III; Elementar Analysensysteme GmbH,
162 Germany). The surface morphologies of the hydrochars were visualized by scanning
163 electron microscopy (SEM, Quanta200, FEI, Netherlands) at 5000×. The surface
164 functional groups were characterized by Fourier transform infrared spectroscopy (FTIR)
165 on an Agilent Cary 660 FTIR Analyzer (California), as described in a previous study
166 ([Feng et al., 2018a](#)). The surface elements contents (C, O, N, P, Ca, Si) on the surface
167 of CVP, CVHW, and CVHCA were characterized using X-ray photoelectron
168 spectroscopy (XPS) technology (the details of analysis are shown in Supplementary
169 Materials). The specific surface area (SSA), porous diameter, and porous volume for
170 adsorption were measured using a NOVA 1200 analyzer, and the parameters were
171 calculated using the Brunauer–Emmett–Teller (BET) method, using a surface area
172 analyzer (Quadrasorb SI, America) ([Yu et al., 2019](#); [Chu et al., 2020](#)).

173 ***2.4. Measurement of rice growth index, grain N content and grain yield***

174 Rice plants were manually harvested from each pot at physiological maturity to
175 determine the grain N content and grain yield. The grain N content was determined
176 using the Kjeldahl method, as described in previous study (Chu et al., 2016b).
177 Furthermore, The N use efficiency was calculated as the percentage of applied fertilizer
178 N recovered in aboveground biomass minus that of the control treatment without N
179 fertilizer application (H. Sun et al., 2019a). The N content of aboveground biomass in
180 the control treatment without N fertilizer application was 1.18 g pot⁻¹.

181 *2.5. Monitoring NH₃ volatilization and N₂O emissions from paddy soil*

182 NH₃ volatilization was measured in parallel with floodwater sampling. The
183 continuous air-flow enclosure method was used to estimate daily NH₃ volatilization
184 fluxes, as described by Feng et al. (2017). In brief, volatilized NH₃ was captured in a
185 Plexiglas chamber with an inner diameter of 15 cm and 20 cm height using a mixture
186 of 80 mL 2% boric acid, an indicator of methyl red, bromocresol, and ethanol as NH₃
187 absorbent. The NH₃-containing solution was titrated against with 0.01 M H₂SO₄.

188 The cumulative volatilized NH₃ was calculated as the sum of daily NH₃ volatilization
189 amounts during the monitoring period (The NH₃ volatilization mainly derives from the
190 urea addition and thus the first 7 days after urea application at three fertilization dates
191 was applied) after BF, SF1, and SF2, respectively.

192 The monitoring of N₂O, as well as gas flux measurement, were conducted as
193 described in previous study (Zhou et al., 2018). N₂O concentrations was determined by

194 gas chromatography (Agilent 7890, USA). The yield-scale NH_3 volatilization or N_2O
195 emissions were calculated using the aforementioned cumulative NH_3 volatilization loss
196 or N_2O emission loss by dividing it, respectively, with the rice grain yield of the
197 corresponding treatment.

198 ***2.6. Analysis of soil N and microbial biomass C and N***

199 Fresh soil samples were collected from the top layer (0-20 cm) of soil at BF, SF1,
200 and SF2, respectively. The soil samples from each plot were mixed and homogenized,
201 grounded to < 2 mm, and then divided into two aliquots. The first section was frozen
202 immediately in liquid N_2 and stored at -80°C for molecular analysis; the remaining
203 section was stored at -20°C for nitrate (NO_3^- -N) and ammonium (NH_4^+ -N) analysis, as
204 well as microbial biomass C and N.

205 Soil pH was determined by the same method as described above. The NH_4^+ -N and
206 NO_3^- -N extracted from the soil by 2.0 M KCl were measured using a San++ Continuous
207 Flow Analyzer, as previously described (Chu et al., 2016a; Feng et al., 2017). Soil
208 microbial biomass C and N were measured by the chloroform-fumigation-extraction
209 method described by (Moore et al., 2000). All results were reported as the averages of
210 the duplicated analyses and were expressed on a moisture-free basis. Moisture was
211 determined after drying at 105°C for 48h.

212 ***2.7. DNA extraction and quantitative polymerase chain reaction (qPCR)***

213 Total DNA was isolated from soil samples (*ca.* 0.5 g) using the mericon DNA
214 Bacteria Kit and mericon Bacteria Plus Kit (Qiagen, Germany) according to the
215 manufacturer's instructions. PCR amplifications of genes of archaeal NH₃-oxidizers
216 (*AOA*), bacterial NH₃-oxidizers (*AOB*), nitrite reductase (*nirK* and *nirS*) and N₂O
217 reductase (*nosZ*) were performed for primers as shown in **Table S1**. The abundance of
218 these genes was quantified by qPCR using an AB17500 thermocycler (Applied
219 Biosystems Inc., USA) and was described as the gene copy number per gram of dry soil.
220 The details of this process are shown in the Supplementary Information.

221 **2.8. Statistical analyses**

222 Statistical analyses were performed using the SPSS version 18.0 (SPSS Inc. Chicago,
223 IL, USA). One-way analysis of variance (ANOVA) was used to evaluate the results at
224 $P < 0.05$ probability. Duncan's multiple range test was employed only when the
225 ANOVA F-test indicated significant treatment effects at the significance level ($P <$
226 0.05).

227 **3. Results**

228 **3.1. Characteristics of hydrochars**

229 Physiochemical characteristics of CVP, CVHW, and CVHCA are listed in **Table 1**,
230 where the major elements, including C, H, N, and S of the hydrochars are presented.
231 CVHCA exhibited an obviously lower pH than CVP or CVHW. Compared with CVP,

232 CVHW and CVHCA hydrochars had higher C and lower N and S content. The C/N and
233 H/C ratio established whether hydrochars were susceptible to mineralization and
234 degradation by soil microorganisms (Zimmermann et al., 2012; Mukome et al., 2013;
235 Senbayram et al., 2019). By employing HTC, CVHW and CVHCA improved C/N
236 ratios by 2.15- and 1.99-fold, respectively, and the H/N ratio indicated a 2.09- and 1.69-
237 fold improvement compared with CVP, respectively. In addition, the surface atomic
238 concentration was analyzed by XPS technology (Table S1). CVHW and CVHCA both
239 increased surface O concentration and decreased surface N and P concentration
240 compared to CVP, suggesting the possible variation on the according surface O-, N-,
241 and P-functional groups. Surface Ca concentration was only detected in CVHW.

242 High SSA and porous volume levels are extremely desirable for enhancing nutrients
243 retention in soil because these features facilitate high mass transfer fluxes and
244 adsorption loading. With respect to adsorption capacity, CVHW and CVHCA had
245 significantly larger SSA (6.71- and 7.76-fold, respectively), and porous diameter (4.33-
246 and 4.54-fold, respectively) than CVP. The porous volume of CVHW was 1.93-fold
247 larger than CVHCA, and that of CVHCA was 3.26-fold larger than CVP. Furthermore,
248 SEM structural image showed that more obvious crystal structures formed in the
249 CVHW and CVHCA hydrochars than in the CVP (Fig. S2). A larger porous diameter
250 and more abundant pores were observed in the SEM images of CVHW and CVHCA,
251 which was supported by analysis using the BET method.

252 The FTIR spectra of the hydrochars revealed the functional groups present on their
253 surfaces (**Fig. S3**). The differences between treatments were detected in the peak
254 intensity of the aliphatic C–H at 2923.5 cm⁻¹ and 2852.7 cm⁻¹ wave numbers, in the –
255 COOH peak at a wavenumber of 1700 cm⁻¹, in aliphatic C–O/C–O–C at 1028 cm⁻¹, in
256 aromatic C-H at 825 cm⁻¹, and phenolic hydroxyl group (phenolic –OH) at 3420 cm⁻¹.
257 The peak intensity for all these functional groups for the CVHW and CVHCA
258 hydrochars was markedly lower than those for CVP.

259 **3.2. Effects of CVP and hydrochars on the rice growth, quality and yield**

260 The effects of hydrochars application on grain yield (dry weight, [DW]) and quality
261 are listed in **Table 2**. Compared with the control, CVHW and CVHCA significantly
262 increased grain yield by 26.7% and 21.2%, respectively. A similar trend was also
263 detected for harvest index and soluble sugars. The application of CVHW and CVHCA
264 significantly improved grain N assimilation by 48.7% and 49.6% compared with
265 control, and by 23.9% compared with CVP addition, respectively. Conversely, the
266 application of CVHW and CVHCA significantly reduced straw N assimilation by 42.9%
267 and 59.1% compared to control. These results suggested that the addition of hydrochars
268 promoted the N partition to edible part of rice. Moreover, the application of CVHW
269 and CVHCA significantly improved N use efficiency by 51.5% and 38.6% compared
270 with control, and by 38.5% and 32.7% compared with CVP addition. Similar trend was

271 detected in grain soluble sugars content among treatments. No significant difference
272 was detected in crude fiber or starch levels.

273 **3.3. Effects of CVP and hydrochars on NH₃ volatilization**

274 The effects of CVP and hydrochars on NH₃ volatilization from rice paddy soil are
275 shown in **Fig. 1**. The trends of NH₃ volatilization flux from paddy soil were quite
276 similar among all treatments (**Fig. 1A**). After BF the peak flux appeared at the second,
277 third, or fourth day and declined shortly afterwards. After SF1 the peak flux only
278 appeared at the second day and declined shortly afterwards. After SF2 no peak flux was
279 detected. CVHCA treatment frequently resulted in the maximum peak value of NH₃
280 volatilization. Also, CVP and CVHW treatment increased NH₃ volatilization compared
281 to control. Following BF, CVHCA addition significantly increased cumulative NH₃
282 volatilization by 2.29-, 1.51-, and 1.30-fold compared with the control, CVP, and
283 CVHW, respectively (**Fig. 1B**). Following SF1, CVHW and CVHCA application
284 significantly increased cumulative NH₃ volatilization by 61.7% and 62.6% compared
285 with the control but reduced it by 35.7% and 34.9% compared with CVP application.
286 Following SF2, no significant difference was detected for cumulative NH₃
287 volatilization. Yield-scale NH₃ volatilization is the quantification of cumulative NH₃
288 volatilization over the entire growth stage of rice based on yield production. Yield-scale
289 NH₃ volatilization was comparable between the control and CVHW groups (**Fig. 1C**).

290 Compared with the control, CVP and CVHCA treatment significantly increased yield-
291 scale NH₃ volatilization by 51.5% and 42.4%, respectively.

292 **3.4. Effects of CVP and hydrochars on N₂O emissions**

293 The dynamics of N₂O emissions during rice-growing seasons under different
294 treatments are shown in **Fig. 2A**. Different treatments showed the similar temporal
295 trends in N₂O fluxes during the rice-growing season. Obvious emission peaks were
296 observed approximately 30, 45, 80, and 95 days following transplantation. The highest
297 emission peaks of CVHW and CVHCA were observed on the 45 days after
298 transplantation, and the emissions under CVHW and CVHCA were 3.08- and 2.78-fold,
299 and 4.61- and 4.17-fold higher than under control and CVP, respectively. The
300 cumulative N₂O emissions after three fertilization dates affected by different treatments
301 are shown in **Fig. 2B**. Compared to control, CVP, CVHW and CVHCA all significantly
302 increased cumulative N₂O emissions and the significantly highest N₂O emissions were
303 observed in CVHCA treatment; CVHCA application significantly increased
304 cumulative N₂O emissions by 3.20-, 1.46-, and 1.47-fold compared with control, CVP,
305 and CVHW, respectively. Moreover, CVP, CVHW and CVHCA all significantly
306 increased yield-scale N₂O emissions as well, however, no significant difference was
307 detected between CVP and CVHW, or between CVP and CVHCA (**Fig. 2C**).

308 **3.5. Effects of CVP and hydrochars on soil pH, NH₄⁺-N and NO₃⁻-N**

309 The effects of CVP and hydrochars application on soil $\text{NH}_4^+\text{-N}$ and $\text{NO}_3^-\text{-N}$
310 concentrations after BF, SF1 and SF2 are shown in **Fig 3**. Following BF and SF1, the
311 CVP, CVHW, and CVHCA all significantly improved the soil NH_4^+ concentration
312 compared with control (**Fig 3A**). Compared with CVP, CVHCA significantly reduced
313 soil NH_4^+ concentration by 25.1% and 22.8% after BF and SF1, respectively. However,
314 after SF2, soil NH_4^+ concentration was comparable among the control, CVHW, and
315 CVHCA groups; only CVP significantly improved the soil NH_4^+ concentration.
316 Moreover, CVHCA significantly increased soil $\text{NO}_3^-\text{-N}$ concentrations compared with
317 the control, CVP and CVHW, irrespective of fertilization stages. Furthermore,
318 following BF CVHW significantly increased soil $\text{NO}_3^-\text{-N}$ concentrations by 46.5%
319 compared with the control; however, after SF1, CVHW significantly reduced soil
320 $\text{NO}_3^-\text{-N}$ concentrations by 2.06-fold. In addition, the effects of CVP, CVHW, and
321 CVHCA application on soil pH were consistent with the pH of materials; CVP, CVHW,
322 and CVHCA all significantly increased soil pH after different fertilization stage and the
323 strongest effect was detected for CVP treatment (**Table. S3**).

324 **3.6. Effects of CVP and hydrochars on soil microbial C and N, and the abundance** 325 **of microbial functional genes controlling N_2O emission**

326 The effects of CVP and *Chlorella vulgaris* hydrochar application on soil microbial
327 C and N are shown in **Fig. 4**. CVHW and CVHCA application significantly reduced
328 microbial C content compared with CVP. Compared with the control, CVP, CVHW,

329 and CVHCA all significantly reduced microbial N content. The effects of CVP and
330 *Chlorella vulgaris* hydrochars application on the abundance of microbial functional
331 genes controlling nitrification (*AOA* and *AOB*), denitrification (*nirK* and *nirS*) and N₂O
332 oxidation (*nosZ*) are shown in **Fig. 5**. Compared with the control, only CVHCA
333 treatment significantly improved the abundance of *AOA*. Also, CVHCA treatment
334 significantly improved the abundance of *AOB* by 2.56-, 1.52-, and 4.26-fold compared
335 with the control, CVP, CVHW treatment. The *Chlorella vulgaris* hydrochars, CVHW
336 and CVHCA, significantly improved the abundance of *nirK* compared with CVP and
337 control. By contrast, the *Chlorella vulgaris*-derived hydrochars, CVHW and CVHCA,
338 significantly improved the abundance of *nirK* compared with CVP and control.
339 CVHCA treatment significantly ameliorated the abundance of *nirS* by 3.48-, 1.68, and
340 1.27-fold compared with the control, CVP, and CVHW treatment, respectively.
341 Compared with the control, CVP application significantly improved the abundance of
342 *nosZ* by 68.3%; however, CVHW and CVHCA application significantly reduced *nosZ*
343 levels by 44.8% and 56.1%, respectively. On the basis of these results, the application
344 of CVP and hydrochars had marked impacts on the activities of soil microorganisms
345 that are responsible for nitrification and denitrification.

346 **4. Discussions**

347 CVHW and CVHCA addition both significantly improved the grain yield of rice
348 (**Table. 2**), suggesting that HTC is an effective way to transform the *Chlorella vulgaris*

349 into a fertilizer. The important drivers of increased yield production resulting from
350 microalgae-based hydrochar addition is increased soil NH_4^+ -N concentration (**Fig. 3**)
351 and N use efficiency by rice plant (**Table. 2**). N is one of the most important mineral
352 nutrients for plant growth comprising 40-50% of dry matter of protoplasm, and is a
353 constituent of amino acids and chlorophyll, and the building blocks of protein
354 (Marschner, 2011; Chu et al., 2019). HTC was also shown to transform the bulk of
355 organic N compounds to inorganic N (Funke et al., 2013; Kruse et al., 2016; Yu et al.,
356 2019). The increased soil NH_4^+ concentration may stimulate rice growth because NH_4^+ -
357 N is the dominant N source for rice. Similar results of increased grain yield derived
358 from increased soil N retention and plant N use efficiency after hydrochar addition have
359 been reported previously (Sun et al., 2017; Yu et al., 2019; Chu et al., 2020).
360 Hydrochars can release an abundant amount of macro and micronutrients, most of
361 which are necessary and beneficial minerals for photosynthesis and plant growth (Feng
362 et al., 2018a; Joseph et al., 2018; Li et al., 2019). The larger SSA of CVHW and CVHCA
363 (8.31- and 9.36-fold higher than CVP, respectively) potentially plays an important role
364 in improving N and nutrients adsorption (Feng et al., 2017; Mandal et al., 2018). The
365 increased grain soluble sugar content (**Table. 2**) might be associated with the stimulated
366 photosynthesis resulting from mineral nutrients introduced by hydrochars, and this may
367 partly explain the increased yield (Chu et al., 2016a).

368 *Chlorella vulgaris*-derived hydrochars increased soil NH_4^+ concentration (**Fig. 3A**),
369 NH_3 volatilization (**Fig. 1**), N_2O emission (**Fig. 2**), and plant N use efficiency (**Table.**

370 2). Although the reduced soil microbial N can partly explain the N balance, the
371 increased N gasification and plant N uptake was more likely attributed to the increased
372 soil N mineralization. The N content in CVP and hydrochars was quite high (7.25-
373 12.29%, **Table 1**) and thus increased the N source for soil microorganisms and plants.
374 However, the low C/N ratio of CVP and hydrochars (3.92-8.44) were able to stimulate
375 soil microbial activity and further accelerated the decomposition of soil organic N
376 (Zimmermann et al., 2012; Mukome et al., 2013; Senbayram et al., 2019). Therefore,
377 the abundant N compounds in hydrochars may serve as a paradoxical aspect for
378 improving both plant N utilization and gaseous N loss to the atmospheric environment.
379 The increased NH₃ volatilization exacerbated the non-point pollution.

380 The introduction of more N and H sources into soil, together with an increase in pore
381 conductivity that allows for gas exchange, may be important factors for promoting NH₃
382 volatilization, as shown in previous studies on biochars (Sun et al., 2017; Feng et al.,
383 2018b; H. Sun et al., 2019b). Soil pH is a primary factor that affects NH₃ volatilization
384 because an acidic state may promote the negatively charged organic functional groups
385 to become the main adsorption sites for ammonium, thereby preventing NH₃
386 volatilization (Sha et al., 2019). CVP, which had the highest pH, caused the highest
387 volume of NH₃ volatilization loss after BF (**Fig. 1**). HTC reduced pH of hydrochars,
388 particularly in the citric acid medium (CVHCA), thus helping reduce NH₃ volatilization
389 loss at BF compared with CVP. Similar studies have also indicated an increase in NH₃
390 loss following biochar amendment to rice paddy and coastal saline soils due to the

391 increased soil pH (Schomberg et al., 2012; Feng et al., 2018b). Following SF1, the
392 short-term acidifying disturbance became weak and NH₃ volatilization loss became
393 stronger under CVHW and CVHCA treatments. Furthermore, CVHW and CVHCA
394 treatments both increased SSA compared with CVP, and thus were also expected to
395 increase the NH₃ retention rather than NH₃ volatilization. The larger porous volume
396 and porous diameter likely led to higher moisture retention in a bid to compete with
397 NH₄⁺-N adsorption (Sarkar et al., 2011; Mandal et al., 2019). Furthermore, the reduced
398 abundance of a carboxyl group in CVHW and CVHCA compared with CVP, as shown
399 in FTIR spectra (**Fig. S3**), may have resulted in the increased NH₃ volatilization
400 because NH₃ is an alkaline gas and the acidic surface functional groups on the
401 hydrochar, such as the carboxyl and carbonyl group, can protonate NH₃ gas to NH₄⁺
402 ions (Spokas et al., 2012; Mandal et al., 2018). Therefore, the increased pH and reduced
403 abundance of acidic surface groups impart *Chlorella vulgaris* hydrochars to increased
404 NH₃ volatilization loss. Additionally, due to the increased yield by CVHW addition, no
405 significant difference of the yield-scale NH₃ volatilization was detected between
406 control and CVHW treatment. Therefore, in view of improving yield and avoiding NH₃
407 volatilization, CVHW is a better recommendation.

408 Furthermore, CVP and *Chlorella vulgaris* hydrochars all significantly increased N₂O
409 emissions compared with the control group (**Fig. 2**). Obvious emission peaks were
410 observed at the mid-season drainage (30 and 45 days after transplantation) and close to
411 harvest (80 and 95 days after transplantation). The mid-season drainage changed the

412 soil aeration and thus stimulate nitrification, thus caused substantial N₂O emissions
413 from paddy soil (Bhattacharyya et al., 2013; Ali et al., 2015; Zhou et al., 2018). The
414 peaks of N₂O emissions close to harvest was possibly caused by the accumulated N
415 source from the slow decomposition of CVP or hydrochars. In addition, contradictory
416 results that are related to the mitigation effects of biochar application on N₂O emission
417 have been reported (Zhou et al., 2018; Borchard et al., 2019; Wu et al., 2020). The
418 liming effect of the CVP and hydrochars may also be responsible for increased N₂O
419 emission, because increased soil pH above a neutral state enables the NH₄⁺ from the
420 adsorption sites of soil minerals to further promote the nitrification (Mandal et al., 2018;
421 Senbayram et al., 2019). However, the inconsistent results between a lower pH in
422 CVHCA and increased N₂O emission suggest that other factors are involved. Compared
423 with CVP, CVHCA significantly reduced the soil NH₄⁺ concentration and increased
424 soil NO₃⁻ concentration, which is consistent with increased abundances of AOA and
425 AOB (Fig. 5A and B). Therefore, CVHCA application improved the nitrification and
426 potentially promoted the generation of more N₂O. Additionally, the increased
427 abundances of *nirK* and *nirS* by CVHCA and CVHW addition suggested the promotion
428 of nitrite reduction, as well as the production of N₂O (Fig. 5C and D). *Chlorella*
429 *vulgaris* hydrochars readily released the labile C and N pool to the soil, which possibly
430 stimulates the activity of nitrobacteria and denitrobacteria. Similar results were reported
431 in previous studies using other biochars (Duan et al., 2018; Senbayram et al., 2019;
432 Zhang et al., 2019). Furthermore, the expression of *nosZ* was increased by CVP but

433 was inhibited by CVHW and CVHCA (**Fig. 5E**) which reflects the inhibition of CVP
434 on the oxidation of N₂O ([Krause et al., 2018](#)). This difference may serve as an important
435 reason for the increased N₂O emission by CVHW and CVHCA compared with CVP.

436 **5. Conclusions**

437 The results of this study demonstrated that *Chlorella vulgaris*-derived hydrochars
438 that employ a water or citrate acid solution as a reaction medium in HTC improved the
439 yield and soluble sugars of rice grains while concomitantly increasing the NH₃
440 volatilization and N₂O emissions from rice paddy soil. The positive effects on rice
441 growth and yield indicate that *Chlorella vulgaris*-derived hydrochars have the potential
442 to serve as a fertilizer that can recycle nutrients from wastewater (enriched by
443 microalgae) into the crop-soil system; this approach is superior to the direct application
444 of CVP. However, the relatively lower C/N and H/C ratio of *Chlorella vulgaris*-derived
445 hydrochars may readily provide a labile C pool for soil microorganisms to stimulate
446 gaseous N loss. In future studies the additional modifications on *Chlorella vulgaris*
447 hydrochars must be attempted to find out a way to improve the plant nutrients utilization
448 without polluting the atmospheric environment.

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458 **Conflicts of Interest**

459 The authors declare no conflict of interest.

460 **Electronic supplementary material**

461 The online version of this article contains supplementary material, which is available
462 to authorized users.

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