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1 Laser induced ultrafast combustion synthesis of solution-based AlO_x for thin 2 film transistors

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16 17 Abstract

18 Solution processing of amorphous metal oxides using excimer laser annealing (ELA) has been
19 lately used as a viable option to implement large-area electronics, offering high quality
20 materials at a reduced associated cost and process time. However, the research has been focused
21 on semiconductor and transparent conductive oxide layers rather than on the insulator layer. In
22 this work we present amorphous aluminum oxide (AlO_x) thin films produced at low temperature
23 (≤ 150 °C) via combustion synthesis triggered by ELA, for oxide thin film transistors (TFTs)
24 suitable for manufacturing flexible electronics. The study showed that combining ELA and
25 combustion synthesis leads to an improvement in the dielectric thin film's densification in a
26 shorter time (≤ 15 min). Optimized dielectric layers were obtained combining a short drying
27 cycle at 150 °C followed by ELA treatment. High breakdown voltage ($4\text{MV}\cdot\text{cm}^{-1}$) and optimal
28 dielectric constant (9) was attained. In general, TFT devices comprising the AlO_x fabricated
29 with a drying cycle of 15 min followed by ELA presented great TFT properties, a high
30 saturation mobility ($20.4\pm 0.9\text{ cm}^2\cdot\text{V}^{-1}\cdot\text{s}^{-1}$), a small subthreshold slope ($0.10\pm 0.01\text{ V}\cdot\text{dec}^{-1}$) and
31 a turn-on voltage ~ 0 V. ELA is shown to provide excellent quality solution-based high- κ AlO_x
32 dielectric, that surpass other methods, like hot plate annealing and deep ultraviolet (DUV)
33 curing. The results achieved are promising and expected to be of high value to the printed
34 electronic industry due to the ultra-fast film densification and the surface/area selective nature
35 of ELA.

36 **Keywords:** Low temperature processing, excimer laser annealing, solution combustion
37 synthesis, oxide thin film transistors, solution based high- κ dielectric, printed electronics

1. INTRODUCTION

Over the last decade, solution-based metal oxide (MO) materials have become excellent candidates for thin film transistors (TFTs), offering conductors, semiconductors and high- κ dielectrics applicable to flexible and large area electronics.¹⁻⁶ Several processing techniques have been used to fabricate the constituent layers for the TFTs, like inkjet-printing, spray-coating, flexographic printing, screen printing, dip-coating and the most widely utilised spin-coating.^{2,7-9} These techniques allow to reduce the associated fabrication costs (compared to vacuum technology) and at the same time can provide a high carrier mobility, excellent uniformity and operational stability. The high processing temperatures required to process most semiconductors and dielectric layers has been surpassed with the use of solution combustion synthesis (SCS) and deep-ultraviolet (DUV) photochemical activation. SCS provides additional thermal energy, via an exothermic reaction, resulting in a reduction of the external heating required (both in duration and temperature) for the film formation; i.e. the removal of organic solvents and film densification.^{10,11} The DUV concurrent treatment additionally allows for the cleavage of alkoxy groups, active metals, and oxygen atoms to promote the metal-oxide-metal (M-O-M) network formation.¹²⁻¹⁵ Overall, the combinatorial utilisation of SCS and DUV improves condensation and densification of the resultant metal oxide thin film, at processing temperatures ≤ 150 °C.^{2,12,16-18} However, these methods require long processing times (≥ 30 min) that are not suitable for large-scale roll-to-roll (R2R) processes.^{12,16,19} In a R2R processing line, annealing times are restricted by the length of the in-line curing ovens. For example, for a moderate web speed of 1 m/min and a typical oven length of 5 m, the curing time for each processed layer is limited to 5 min.^{12,20,21} Consequently, the industry demands alternative methods that can deliver a rapid annealing process.^{22,23} Lately, excimer laser annealing (ELA) has been introduced as a favourable alternative in the fabrication of solution-based MO thin films. This method offers ultrafast

1 processing (in the nanosecond regime), with precise and selective energy delivery, both in
2 depth via critical photon energy absorption as well as spatially (over a well-defined location),
3 ^{24,25} resulting in a significant improvement of the electronic properties of MO thin films, that
4 can outperform conventional annealing.^{25,26} Nevertheless most effort has concentrated in semi-
5 conductive and conductive thin films ^{23–32}, with a notable lack of work on dielectric materials,
6 despite its importance in the robust operation of TFTs.¹⁸ Laser annealing of sol-gel processed
7 HfO₂ has been reported by Teodorescu et al.³³ demonstrating the applicability of XeCl (308
8 nm) in the densification of amorphous hafnium oxide structure while achieving a high dielectric
9 constant. However, a double coated layer was required (to eliminate nanoporosity, as well as a
10 very high number of pulses (10000) and a 30 min annealing at 150 °C prior to ELA. Anodization
11 is another promising solution process for the production of high quality dielectric thin films for
12 application in TFTs.^{34–38}

13 The promise of future state-of-the-art electronics (like wearable electronics) can only be
14 enabled by low cost, easy to process, high quality dielectrics, that will allow low power
15 consumption, as well as good mechanical and operational stability.^{19,20,25} High-κ dielectric
16 materials such as ZrO₂, HfO₂, Ta₂O₅, SrO_x, ZrGdO_x or Al₂O₃ increase the areal capacitance of
17 the gate dielectric allowing low operation voltage of TFTs.^{2,39,40} Among these dielectrics,
18 amorphous Al₂O₃ is one of the most promising due to its characteristic properties such as its
19 high dielectric constant (~9) combined with a large band gap (8.9 eV), low interfacial trap
20 density with semiconductors, high breakdown electric field.¹⁸ At the same time, Al₂O₃ can be
21 obtained from aluminium which is one of the most abundant materials on earth and hence it
22 meets current demands for use of non-critical materials in the life-cycle assessment.⁷

23 In this work we demonstrate the production of amorphous aluminium oxide (AlO_x) thin films
24 from solution and via a combination of SCS and ELA, and we report the performance of TFTs
25 comprising these thin films. We applied a short drying cycle (15 min or 1 min) at low

1 temperature (≤ 150 °C) after spin coating of the solution. Subsequently the samples were
2 irradiated with a few pulses (1 to 3) from a KrF (248 nm) excimer laser, varying the laser fluence.
3 The dielectric quality of the AlO_x layers was evaluated via metal insulator semiconductor (MIS)
4 devices, prior to implementing them in low operation voltage TFTs.

5 2. EXPERIMENTAL SECTION

6 2.1. Precursor solution preparation

7 Aluminum nitrate nonahydrate ($\text{Al}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$, Fluka, 98%) was dissolved in 2-
8 methoxyethanol (2-ME, $\text{C}_3\text{H}_8\text{O}_2$, ACROS Organics, 99%), to yield a solution with an Al^{3+} ion
9 concentration of 0.1 M. Urea ($\text{CO}(\text{NH}_2)_2$, Sigma, 98%) was then added to the prepared solution
10 was maintained under constant stirring (430 rpm) for at least 1 h after dissolving the precursors.
11 The urea to aluminum nitrate molar proportion was 2.5:1, to guarantee the redox stoichiometry
12 of the reaction.⁴¹ The precursor solution was filtered through a 0.20 μm hydrophilic filter before
13 use.

14 2.2. Dielectric deposition, processing and characterization

15 Prior to deposition all substrates (p-type silicon wafers, of 1-10 $\Omega \cdot \text{cm}$, with size of 2.5cm \times 2.5
16 cm) were cleaned in an ultrasonic bath at 60 °C in acetone for 10 min, then in 2-isopropanol for
17 10 min and dried under nitrogen (N_2); followed by a 10 min UV/Ozone surface activation step
18 in a PSD-UV Novascan system. Thin films were deposited by spin coating a single layer of the
19 AlO_x precursor solution for 35 s at 2000 rpm (Laurell Technologies) followed immediately by
20 a short drying cycle (hot plate annealing) at 150 °C with two different durations, 15 min or 1
21 min. Then, the photonic processing was performed, on different areas of the sample and with
22 various laser conditions (fluence and number of pulses). The photonic processing station
23 comprises a KrF excimer laser (Lambda Physik 305i, $\lambda = 248$ nm, 25 ns pulse duration) and a
24 beam delivery system that provides the sample with a top hat profile laser spot (3x3 mm² in
25 size) with fluence (energy density, mJ/cm²) uniformity better than 2 % across the area of the

1 laser spot. The photonic processing station is described in detail elsewhere.^{42,43} Finally,
2 reference layers thermally annealed on a hotplate for 1h at 300 °C (without any photonic
3 processing) were also fabricated, for comparison.

4 The films' structure was assessed by glancing angle X-ray diffraction (GAXRD) performed on
5 an X'Pert PRO PANalytical powder diffractometer using the Cu K α line radiation ($\lambda =$
6 1.540598 Å) with an angle of incidence fixed at 0.9°. The surface morphology was investigated
7 by atomic force microscopy (AFM, Asylum MFP3D). Spectroscopic ellipsometry
8 measurements of the thin films deposited on silicon substrates were made over an energy range
9 of 1.5–6.0 eV, at an incident angle of 70°, with a Jobin Yvon Uvisel system, in order to access
10 the films' thickness.

11 X-Ray Reflectivity (XRR) measurements were performed in a Rigaku Ultima IV diffractometer
12 equipped with a multilayer X-ray mirror, using the Cu K α line. The X-ray generator was
13 operated at 40 kV and 40 mA and the reflected intensity was recorded by a scintillator detector
14 in the range of 0.1 to 5 degrees 2 theta.

15 Fourier Transform Infra-Red (FTIR) spectroscopy characterization of the thin films was
16 performed using an Attenuated Total Reflectance (ATR) sampling accessory (Smart iTR)
17 equipped with a single bounce diamond crystal on a Thermo Nicolet 6700 Spectrometer. The
18 spectra were acquired at a 45° incident angle in the range of 4500–540 cm⁻¹ and with a 4 cm⁻¹
19 resolution.

20 **2.3. Electronic device fabrication and characterization**

21 Two type of devices were fabricated for the evaluation of the quality of the AlO_x thin films
22 produced: Metal-Insulator-Semiconductor (MIS) capacitors and Thin-Film-Transistor (TFTs).
23 MIS capacitors were produced by depositing an AlO_x single layer onto p-type silicon substrates
24 as described above. Aluminum electrodes with an area of 7.85×10^{-3} cm² (dots of 1 mm in
25 diameter 100 nm thickness), were deposited by thermal evaporation via a shadow mask on top

1 of the insulators, with similar but unpatterned aluminium electrodes being also deposited on the
2 back of the silicon wafer (after removal of the native SiO₂ by scratching with a diamond tip
3 pen). Electrical characterization was performed measuring both the capacitance–voltage and
4 capacitance-frequency characteristics in the range of 1 kHz to 1 MHz, using a semiconductor
5 parameter analyser (Keysight B1500A).

6 TFTs were produced in a staggered bottom-gate architecture, by depositing the single layer
7 AlO_x onto p-type silicon substrates to act as gate dielectric.

8 The indium-gallium-zinc oxide (IGZO) semiconductor film (30 nm thick) was sputtered onto
9 the dielectric thin film via a shadow mask. The deposition was performed using a commercial
10 IGZO ceramic target (2:1:2 In:Ga:Zn atomic ratio) by RF magnetron sputtering in an Ar+O₂
11 (14 sccm and 2 sccm respectively) atmosphere and without intentional substrate heating in an
12 AJA 1300-F system.⁴⁴

13 Finally, source and drain aluminum electrodes (80 nm thick) were deposited by thermal
14 evaporation via a shadow mask, forming TFTs with a channel length (L) of 90 μm and width
15 (W) of 1000 μm, as shown in the inset of Figure 6 (a). Thereafter the IGZO TFTs were annealed
16 at 150 °C, for 1 h in air. A back contact of 80 nm thick aluminum film was also deposited on
17 the back of the silicon wafer to improve the ohmic contact.

18 The current–voltage characteristics of the devices were obtained in double sweep mode in
19 ambient conditions using a semiconductor parameter analyser (Agilent 4155C) attached to a
20 microprobe station (Cascade M150) inside a dark box.

21 The saturation mobility (μ_{SAT}) was determined in the saturation region of the I_{DS}^{1/2} vs V_{GS} plot
22 (with a linear fitting) from the following equation:

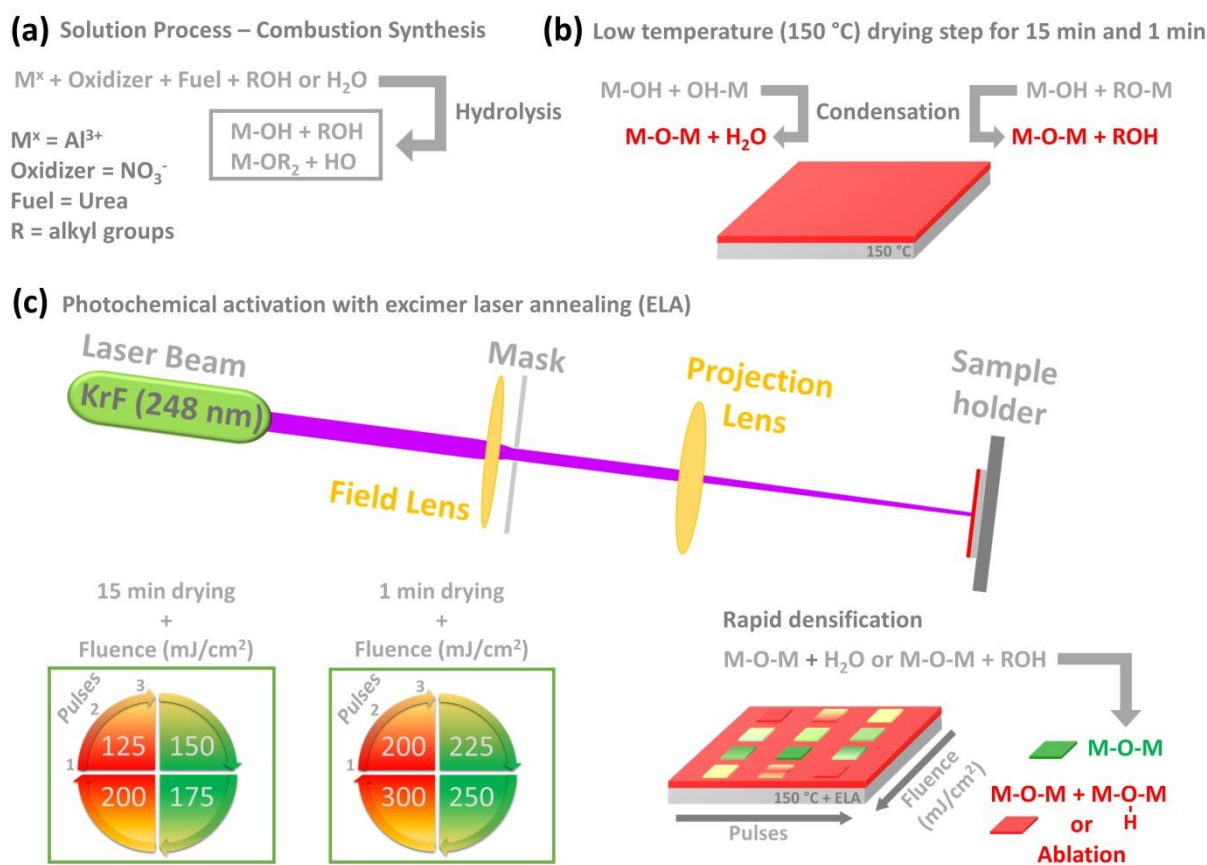
$$23 \quad I_{DS} = \left(\frac{C_{ox} W \mu_{SAT}}{2L} \right) (V_{GS} - V_T)^2$$

1 where I_{DS} is the drain current, C_{ox} is the gate dielectric capacitance per unit area, L and W are
2 the channel length and width respectively, V_{GS} is the gate voltage and V_T is the threshold
3 voltage.

4 To assess device shelf life stability tests were performed after 1 year of storage in atmospheric
5 conditions. Positive gate bias stress tests were performed on the TFTs using a semiconductor
6 parameter analyzer (Keysight 4200SCS) and probe station (Janis ST-500) under atmospheric
7 conditions by applying a constant gate voltage (corresponding to 0.5 MV/cm electric field) for
8 one hour. The transfer characteristics were measured at fixed time intervals during the bias-
9 stress process.

3. RESULTS AND DISCUSSION

The combinatorial methodology (SCS and ELA) implemented in this work, to achieve ultra-rapid high-quality solution-based M-O-M thin film formation, is depicted in Figure 1. At the outset, the metal precursor ($\text{Al}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$) is dissolved in 2-methoxyethanol (2-ME) and stirred for more than 12 h at room temperature (20–28 °C), where a ligand exchange reaction occurs from nitrate to hydroxide (Figure 1 (a)). Then the as-spun films are subjected to a short drying cycle (15 min or 1 min) at 150 °C (Figure 1 (b)). The choice of 150 °C, besides allowing the use of flexible substrates, was also made in order to sit roughly in the middle of the range defined by the boiling point of the solvent utilised (126 °C for 2-ME and hence some solvent removal is achieved), but below the threshold for combustion (180 °C and hence no combustion is initiated).¹⁹



12

13 **Figure 1.** Illustration of the SCS and ELA combinatorial approach showing the (a) hydrolysis
14 and (b) condensation mechanism of metal-oxide precursors; (c) Schematic representation of the

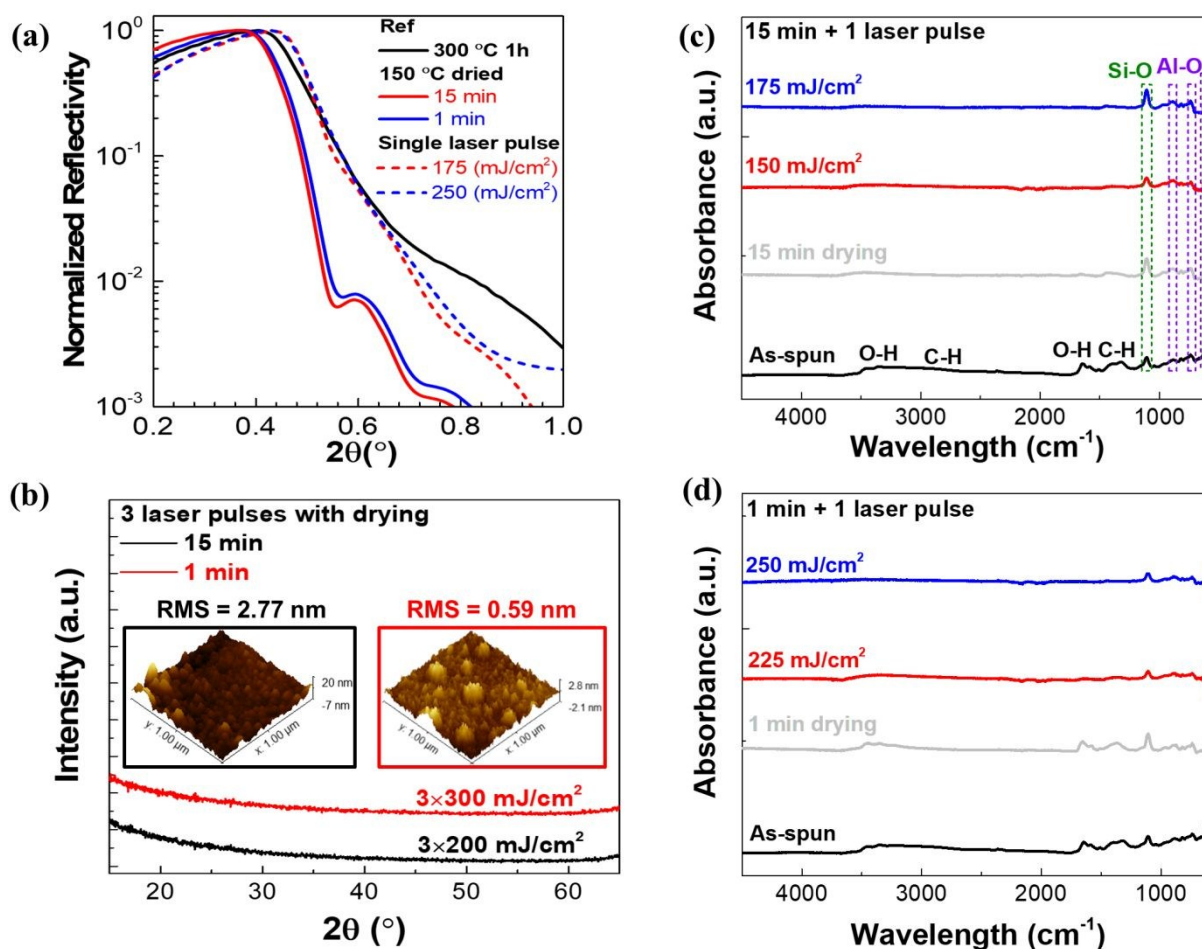
1 photonic processing for AlO_x thin films with different fluences and number of pulses, after 15
2 min or 1 min of hotplate drying at 150 °C immediately after spin coating.

3
4 Previous work¹⁰ has revealed that the precursor solution displays an absorption peak well in the
5 vicinity of a KrF excimer laser emission (248 nm). Naturally, this wavelength enables the
6 photochemical cleavage of alkoxy groups, activating oxygen and metal ions, for the facilitation
7 of the M-O-M network formation.¹² This solid-state chemical reaction is extremely fast and
8 rapidly activates the reaction between the oxidizer (aluminium nitrate) and the fuel (urea).
9 Additionally, the ultra-fast character of ELA enables the utilization of flexible substrates,⁴⁵
10 along with reducing the processing time drastically (as compared to DUV lamp irradiation
11 typically requiring 30 min).¹⁰ In Figure 1 (c), the circular schematics depict (in a clockwise
12 manner) the increasing fluence and number of pulses applied on samples of each drying cycle.
13 The coloured coded regions indicate AlO_x thin films obtained at different conditions. The colour
14 intermixing suggests conditions where the photonic processing was either not enough to cause
15 densification or was excessive and caused damage (red regions), and where it was suitable for
16 the formation of AlO_x thin films (green regions).

17 **3.1. Thin films structural characterization**

18 The degree of densification can be revealed by the critical angle of an X-ray Reflectivity (XRR)
19 measurement. Figure 2 (a) shows the XRR curves around the critical angle for 5 characteristic
20 cases: the reference thin film (hotplate treated for 1h at 300 °C), the as spun film after drying at
21 150 °C for 15 min and for 1 min, as well as the same two dried films photonicly processed
22 with one laser pulse at 175 mJ/cm² and 250 mJ/cm² respectively. By qualitatively comparing
23 the critical angles of the dried films, one can notice that both samples have the same critical
24 angle and thus density, independently of the drying time. The additional photonic processing
25 has resulted in a considerable densification compared to the dried thin films and notably present
26 a similar density to the reference sample (1h at 300 °C). Therefore, it immediately educes that

1 the photonic processing approach can provide AlO_x thin films of similar density to those
 2 produced by a long duration higher temperature thermal (hotplate) annealing, but at a
 3 considerable fraction of the time and in a macroscopically cold process. The XRR curves were
 4 fitted with a three layer model (AlO_x film/ SiO_2/Si) using Parrat's formalism.⁴⁶ The density
 5 extracted for the dried samples was $1.6 \pm 0.1 \text{ g/cm}^3$, irrespective of the drying time. On the
 6 other hand, the density of the reference film was found to be $2.6 \pm 0.1 \text{ g/cm}^3$, which is in
 7 agreement to values reported in the literature ($\sim 2.53 \text{ g/cm}^3$).⁴⁷ The photonic processed
 8 samples were found to have densities of $2.4 \pm 0.1 \text{ g/cm}^3$ (for the 15 min + 175 mJ/cm^2) and 1.7
 9 $\pm 0.1 \text{ g/cm}^3$ (for the 1 min + 250 mJ/cm^2) respectively.



10
 11 **Figure 2.** (a) XRR measurements of 5 selected samples showing the critical angle, (b) XRD
 12 diffractograms and AFM images ($1 \times 1 \mu\text{m}^2$) of two samples dried at 15 min and 1 min,
 13 combined with photonic processing at the highest fluence values in each case (200 mJ/cm^2 and
 14 300 mJ/cm^2 respectively) and maximum pulses (three pulses), (c) and (d) FTIR spectra of AlO_x
 15 thin films before and after single laser pulse photonic processing of the 15 min and 1 min dried
 16 (at 150 $^\circ\text{C}$) sample respectively.

1 The microstructure of the solution-based AlO_x thin films was investigated by grazing angle X-
2 ray diffraction (GAXRD). The absence of any diffraction peaks in Figure 2 (b) reveals that the
3 thin films are amorphous, for both drying cycles combined with the photonic processing, even
4 at the highest fluences and number of pulses. The insets in Figure 2 (b) are show the films'
5 surface morphology obtained by atomic force microscopy (AFM) measurements for the highest
6 fluences. A surface roughness (≤ 2.8 nm) was observed for both drying cycles combined with
7 photonic processing at the highest fluence and number of pulses. However, the surface
8 morphology was not highly uniform, with some peak to valley values at 27.2 nm and 4.9 nm,
9 for 15 min and 1 min drying cycle, respectively. For lower fluences the surface roughness was
10 below 2 nm which guarantee a more uniform interface between the dielectric and the
11 semiconductor.

12 In order to evaluate the extent of solvent removal, the AlO_x thin films were characterized by
13 attenuated total reflection Fourier-transform infrared spectroscopy (ATR-FTIR), as shown in
14 Figure 2 (c) and (d), for the two drying cycles respectively. All samples show an absorbance
15 peak attributed to Si-O vibration (1107 cm^{-1}), due to the presence of native oxide on the Si
16 wafer. Absorption bands related to O-H stretching vibrations ($3700 - 3000\text{ cm}^{-1}$; 1630 cm^{-1} ;
17 1575 cm^{-1}) associated to water absorption and C-H ($3090 - 2800\text{ cm}^{-1}$; $1500 - 1300\text{ cm}^{-1}$) due
18 to the organic solvent, can be observed on the as-spun samples (prior to drying) and after both
19 drying cycles at $150\text{ }^\circ\text{C}$ (for both 15 or 1 min, prior to ELA). Evidently, the residuals are more
20 prominent after the shorter drying cycle. A subsequent ELA treatment even with a single pulse
21 seems to eliminate the residual solvent, as demonstrated by the diminishing relevant bands in
22 the FTIR spectra, with increasing the laser fluence. For a full set of FTIR spectra for all the
23 samples see Figure S1 and S2 in Supplementary Information. Finally, the absorption peaks
24 observed at, 889, 739–748 and $611-601\text{ cm}^{-1}$, are attributed to the presence of Al-O chemical
25 bonds.⁴⁸

3.2. Dielectric thin film electrical characterization

The electrical performance of the dielectric was evaluated by measuring the capacitance-voltage (C-V), capacitance-frequency (C-f) and breakdown field (E) of metal-insulator-semiconductor (MIS) structures. A summary of these results is shown in Figure 3, while the full measurements analytically can be found in Figures S3, S4, S5 and S6, as well as Tables S1 and S2 in the Supplementary Information. Each MIS device was replicated three times for reproducibility purposes.

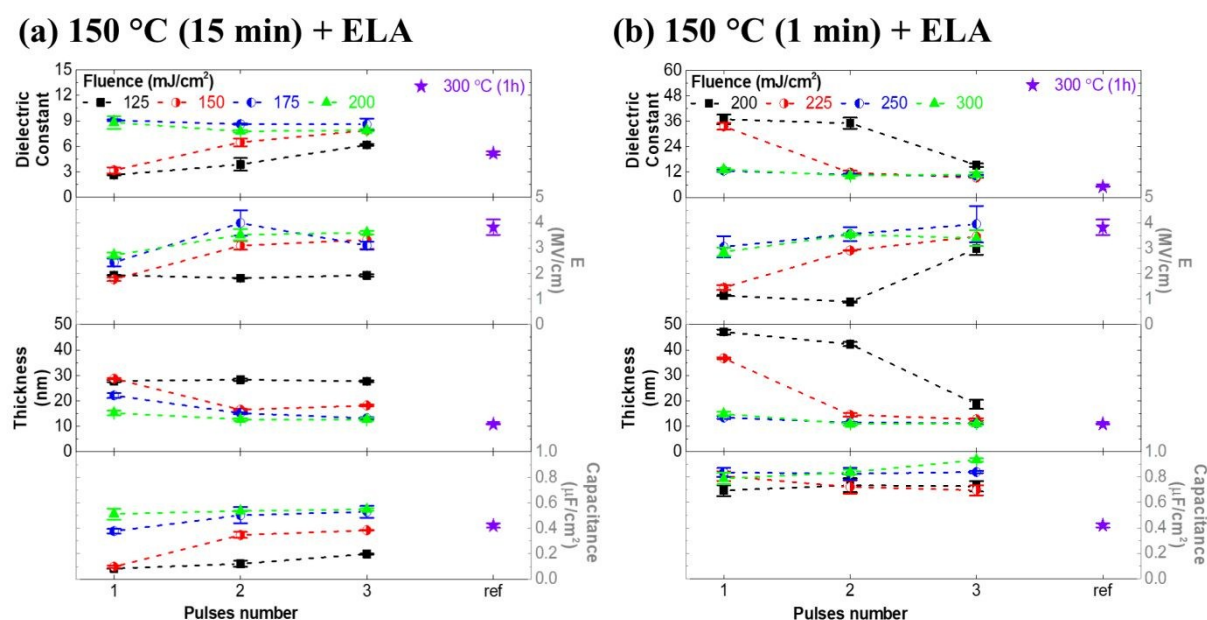


Figure 3. Statistical analysis of AlO_x thin films' performance: dielectric constant, breakdown voltage (E), thickness and capacitance measured at 1 kHz, for all fluences and number of pulses applied on films with initial drying of (a) 15 min and (b) 1 min. For comparison the equivalent characteristic of the reference device (thermally annealed at 300 °C for 1h) is also shown.

The films dried for 15 min (Figure 3a), show a considerable improvement of the dielectric constant as the fluence and the number of pulses is increased. For fluences above 150 mJ/cm^2 , a dielectric constant of ~ 9 (calculated at 1 kHz) is achieved, which is typical for Al_2O_3 achieved via PVD techniques.⁴⁹ Interestingly, the dielectric constant of the reference MIS device (annealed at 300 °C for 1 h) was surpassed even with the lower fluence of 125 mJ/cm^2 using 3 pulses. These results corroborate the XRR findings, suggesting that ELA enhances condensation and densification of the AlO_x thin film, providing an improved dielectric constant

1 in an ultra-rapid fashion and at lower processing temperatures. Conversely when the drying
2 cycle is shorter (1min, Figure 3b), higher fluences were required in order to achieve reasonable
3 values for the dielectric constant. The unrealistic values (>10) for the dielectric constant
4 achieved at lower fluences are explained by the presence of ionic movement (see Figure S4 in
5 Supplementary Information).

6 Another important factor is the durability and stability of the AlO_x dielectric thin films, as
7 expressed by the breakdown voltage. ELA provided films with breakdown voltages up to 4.0
8 MV/cm, closely matching the breakdown voltage demonstrated by the reference sample (3.8
9 MV/cm). The breakdown voltage values were found to show a local maximum within the range
10 of fluences investigated, therefore the optimum annealing conditions have been identified.

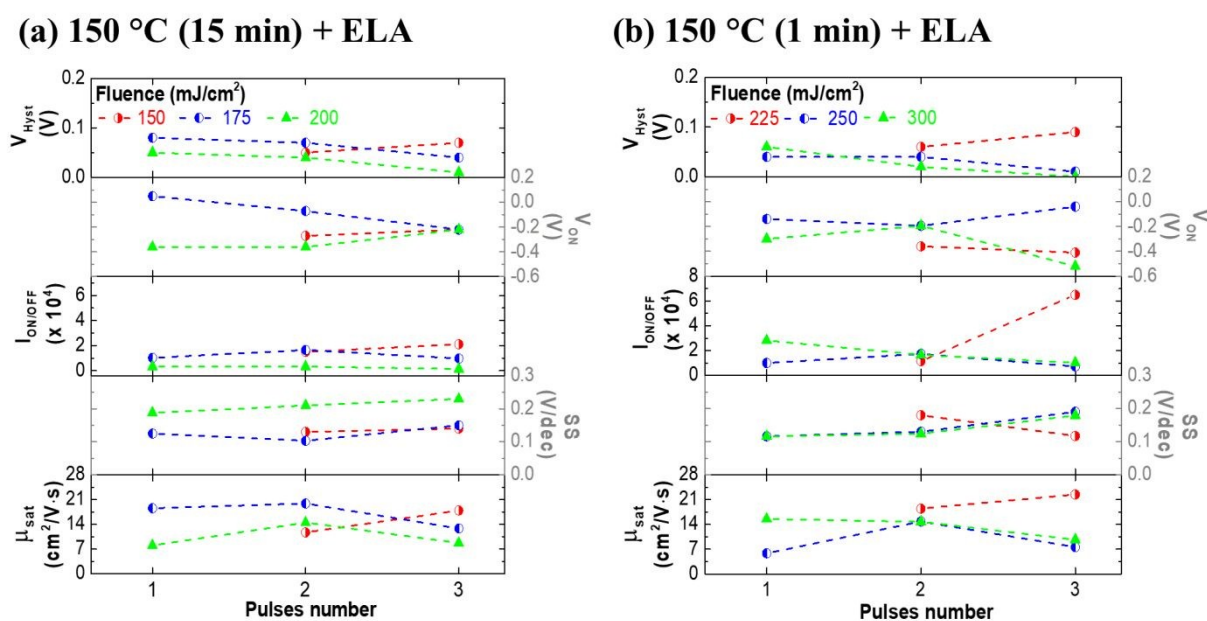
11 These are two pulses of 175 mJ/cm^2 and three pulses of 250 mJ/cm^2 for the 15 min and 1 min
12 drying cycles respectively, presenting breakdown voltages of $4.0 \pm 0.5 \text{ MV/cm}$ and 3.9 ± 0.7
13 MV/cm, whilst the reference device presented a breakdown voltage of $3.82 \pm 0.03 \text{ MV/cm}$. The
14 thickness required in order to calculate the breakdown voltage and the dielectric constant, was
15 determined by UV-VIS ellipsometry, for each device. Besides Figure 3, the full results for all
16 devices fabricated can be seen in the Supplementary Information in Tables S1 and S2 as well
17 as Figures S5 and S6.

18 The devices' aerial capacitance values at 1 kHz are shown in the bottom panel of Figure 3 (a)
19 and (b) for the two drying cycles. The majority of the ELA devices present a higher aerial
20 capacitance than the reference device ($0.42 \pm 0.02 \mu\text{F/cm}^2$). Full C-V characteristic curves at
21 100 kHz and a summary of aerial capacitance vs frequency is presented in the Supplementary
22 Information in Figures S3 and S4, respectively. The drying cycles result in a considerable
23 difference in the dispersive character of capacitance, with the shorter drying cycle being more
24 adversely impacted (i.e. more dispersive) potentially due to the probable incomplete conversion
25 and removal of the precursor.

1 In conclusion, the combination of short drying cycles (15 or 1 min) at relatively low
2 temperatures (150 °C) and ELA delivers in an ultra-fast manner AlO_x thin films with similar or
3 enhanced dielectric properties compared to those achieved via high temperature (300 °C)
4 prolonged (1h) thermal annealing.

5 3.3. Low operation voltage IGZO/AlO_x TFTs

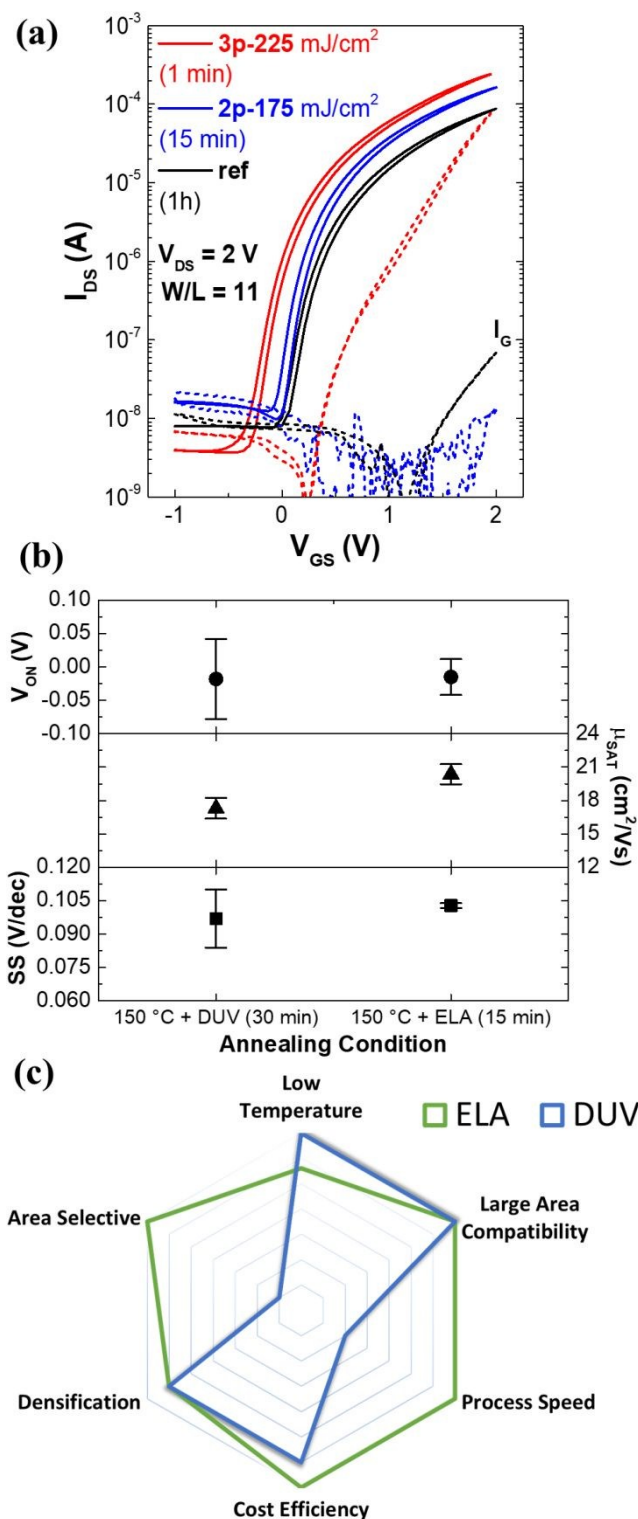
6 Informed by the results presented above (namely the dielectric constant and breakdown voltage)
7 a suitable selection of AlO_x thin films were tested in TFT devices, with IGZO as the
8 semiconductor. A summary of the electrical characteristics of these devices is shown in Figure
9 4 (a) and (b). The full set of the respective transfer curves for each device is shown in Figures
10 S7 and S8 of the Supplementary Information. Device performance was assessed through the
11 measurement of the hysteresis (V_{Hyst}), turn-on voltage (V_{ON} , the on-set voltage at which the
12 drain current starts to increase), drain current on-off ratio ($I_{\text{ON}}/I_{\text{OFF}}$), subthreshold slope (SS),
13 and saturation mobility (μ_{SAT}) which was calculated using the dielectric capacitance measured
14 in AlO_x MIS devices at a frequency of 1 kHz (see Table S3 and S4 in Supplementary
15 Information).



16
17 **Figure 4.** TFT parameters: hysteresis (V_{Hyst}), turn-on voltage (V_{ON}), current on-off ratio
18 ($I_{\text{ON/OFF}}$), subthreshold slope (SS) and saturation mobility (μ_{SAT}), for the two drying cycles: (a)
19 15 min and (b) 1 min followed by ELA treatment.

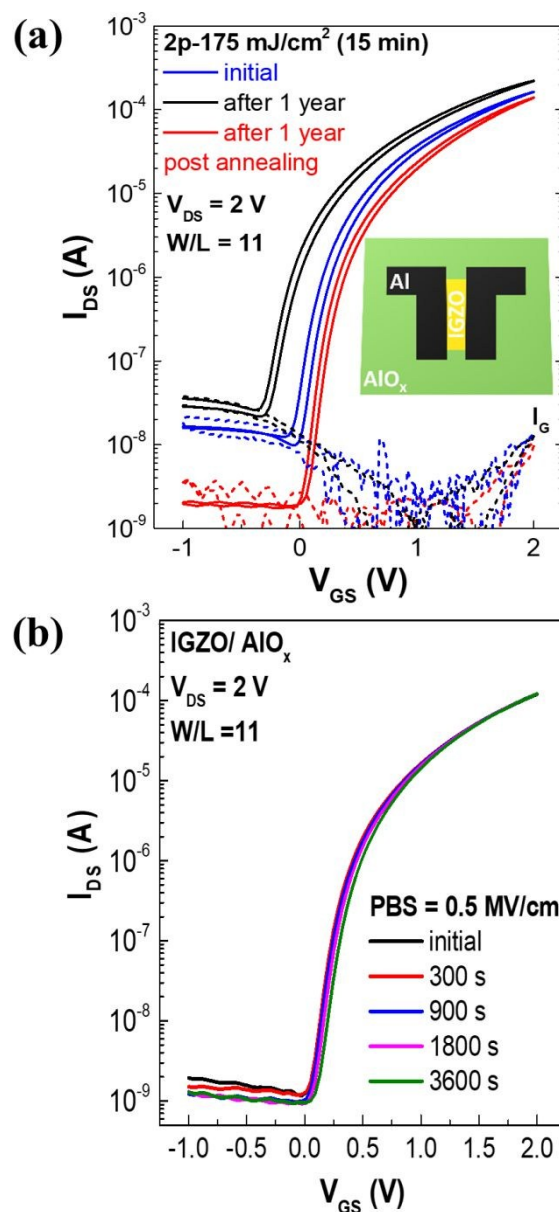
1 All the devices exhibited low hysteresis (less than 0.15 V). A moderate hysteresis reduction
2 with the increase of number of pulses and fluence was observed. Equally all of the device
3 presented a current on-off ration of the order of 10^4 . In the case of the 15 min drying cycle
4 followed by ELA, the turn-on voltage presents an optimum value (closer to 0 V) at a fluence of
5 175 mJ/cm^2 , whereas higher fluences and number of pulses result in a negative turn-on voltage.
6 Below this optimum fluence some ionic species could remain in the dielectric, whereas above
7 surface damage may occur compromising the quality of the semiconductor-dielectric interface.
8 This is further supported by the deterioration of both the subthreshold slope and the saturation
9 mobility. Surface damage and high density of generated defects is likely at higher fluences,
10 since the solution combustion synthesis (SCS) is an exothermic reaction that occurs in the thin
11 film in an ultra-rapid fashion (due to the ELA pulse lasting only a few nanoseconds). In the
12 case of the 1 min drying cycle followed by ELA, a similar trend was found with an optimum
13 fluence value of 250 mJ/cm^2 . The results above are summarised in Figure 5 (a) were the best
14 performing ELA devices, for each drying cycle, are presented alongside the reference device
15 ($300 \text{ }^\circ\text{C}$ for 1h).

16 .



1
2 **Figure 5.** (a) Transfer characteristics of IGZO TFT devices fabricated with solution-based
3 aluminium oxide (AlO_x) insulator thin films for different drying times (15 min and 1 min)
4 followed by ELA treatment. The reference device (300 °C for 1h) is also shown for comparison.
5 (b) Summary of TFT device characteristics using deep ultraviolet (DUV) for 30 min (previous
6 report¹⁰) and this work using the 15 min drying followed by ELA. (c) Comparison of quality
7 characteristics between ELA and DUV treatment for the production of solution-based oxides
8 for the printed electronics industry.

1 Both drying cycles provide TFTs with comparable or better properties than the reference device,
2 however, the 1 min drying followed by ELA present a higher gate leakage current. Furthermore,
3 we present a comparison of the best performing ELA device (15 min drying cycle and 2x175
4 mJ/cm²) to devices with AlO_x fabricated with DUV curing at low temperature¹⁰ (the IGZO layer
5 is identical in both cases). Figure 5 (b) shows a statistical analysis revealing that ELA provides
6 solution processed AlO_x dielectric thin films for TFTs with similar electrical characteristics to
7 the DUV devices, but with an improved saturation mobility (see also Table S5 in
8 Supplementary Information). In general, we can conclude that the combination of ELA
9 treatment with a short drying cycle, immediately after spin coating, can result in AlO_x thin films
10 with quality comparable, if not better, than the standard approach of high temperature thermal
11 annealing (hot plate). The implications of this approach are considerable for the industry of
12 flexible and plastic electronics, as it lends itself to R2R production, for a variety of reasons
13 including: - it enables the use of plastic substrates (maximum processing temperature 150 °C)
14 as it is a macroscopically cold process; - it is rapid enough to allow inline processing while
15 maintaining a reasonable web speed; - it is area selective via laser writing, removing the need
16 for extra fabrication steps for patterning; - it offers oxide thin films of high quality from solution
17 based materials. As such, this approach could provide a significant boost to the printed
18 electronics industry. In comparison to DUV curing (see Figure 5(c)) ELA presents similarly
19 high performance in densification and in large area processability but offers increased overall
20 efficiency through its excellent area selectivity and its massively faster processing time as can
21 be seen in the Supplementary Information in Table S6. The temperature rise in the metal oxide
22 material is far more extreme with ELA, but at the same time considerably faster, resulting in a
23 completely macroscopically cold process, unlike DUV processing. To investigate the effect of
24 the ambient atmosphere on the IGZO/AlO_x TFTs were performed some ageing and stress
25 measurements as depicted in Figure 6.



1
2 **Figure 6.** Stability tests for the best condition achieved with ELA treatment: (a) Ageing effects
3 of IGZO/AIO_x TFTs after 1 year exposed in air environment without passivation and after a
4 post annealing at 120 °C; inset shows the top view TFT schematic (each electrode had an area
5 of 11.9×10^{-3} cm²). (b) Positive gate-bias stress (PBS) measurements of 0.5 MV/cm on
6 IGZO/AIO_x TFTs for 1 h in air environment.
7 Figure 6 (a) shows transfer characteristics of the same device taken a few days after full
8 fabrication and after 1 year of storage in atmospheric conditions, to determine device ageing.
9 In general, the device performance was preserved, however the turn-on voltage displayed a
10 slight negative shift a year later, most probably associated to the increase of carrier density.⁵⁰
11 Also, a small increase of off-current was observed, most probably due to the humidity present

1 in atmospheric environment (which can be surpassed by device passivation).^{51,52} To determine
2 the device behaviour after one year but without the influence of surface adsorption of water
3 molecules, a post annealing at 120 °C (30 min) was performed. As expected, the device's on
4 voltage recovered fully and furthermore the off current was improved by an order of magnitude
5 compared to the initial fabrication. Following this, the device was subjected to a positive gate-
6 bias stress (PBS) of 0.5 MV/cm for 3600 s as depicted in Figure 6 (b). The IGZO/ AlO_x TFTs
7 under PBS remained stable showing only a slight threshold voltage shift to a positive value of
8 just 0.02 V, meaning that a reduced quantity of trap sites exists in the dielectric/semiconductor
9 interface. As no deterioration was observed in the subthreshold slope under bias stress, the
10 mechanism present is the charge trapping model.⁵³ Optimization of production parameters is
11 ongoing, nonetheless these tests already prove the potential of ELA treatment for solution-based
12 oxide TFTs.

4. CONCLUSIONS

In conclusion we have demonstrated the combustion solution synthesis alliance with ELA of AlO_x dielectric thin films at low temperatures in a short drying cycle (15 min or 1 min) and their implementation in TFTs. The AlO_x dielectric films revealed an amorphous nature, low roughness, high- κ (9), low leakage currents (10^{-6} A/cm² at 1 MV/cm) and high breakdown voltages ~ 4 MV/cm. These results indicate that ELA treatment is a viable platform for the fabrication of metal-oxide dielectric thin films at low processing temperatures. Subsequently ELA treatment in combination with combustion synthesis, can be the answer to obtain high performance printed metal oxide TFTs in the R2R industry.

1

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10

11 **AUTHOR CONTRIBUTIONS**

12 E.C., S.D., N.K., D.C.K. and R.B designed and planned the project. E.C., S.D. and N.K.
13 fabricated and characterised the samples. L.K. performed the XRR measurements and analysis.
14 E.C., D.C.K and R.B. analysed the data, prepared the figures and wrote the paper. All authors
15 discussed the results and commented on the manuscript.

16

17 **COMPETING INTERESTS**

18 The Authors declare no Competing Financial or Non-Financial Interest.

19

20 **SUPPLEMENTARY INFORMATION**

21 The Supplementary Information contains the full data sets related to the production of AlO_x
22 dielectric thin films from solution, by excimer laser annealing (ELA) treatment inducing
23 combustion synthesis, and their application in thin film transistors (TFTs). Table S1 and S2
24 shows the statistical parameters of dielectric properties obtained from MIS devices. Figure S1
25 and S2 depicts FTIR spectra for all ELA conditions and for both drying durations, 15 min and

1 1 min, respectively. Figure S3, S4 and S5-S6 show the capacitance–voltage, capacitance–
2 frequency and breakdown voltage (E) characteristics of Al/p-type Si/(AlO_x)/Al MIS capacitors
3 for both drying durations. Tables S3 and S4 depicts the electrical characteristics summary
4 obtained from IGZO TFTs comprising the sol-gel AlO_x fabricated by all the different ELA
5 treatments, while Figures S7 and S8 show the transfer curves of the same IGZO TFTs devices.
6 Finally, Figure S9 depicts the typical output curves of two test TFT devices and the reference.
7 Table S5 shows the statistical TFT parameters from a previous report using deep ultraviolet
8 (DUV) curing for 30 min¹⁰, alongside those of the optimum devices produced in this work
9 (using the 15 min drying cycle followed by ELA). Table S6 depicts relevant low temperature
10 solution-based AlO_x layers processed by different curing methods applied in TFTs reported
11 previously in the literature.

12

13 **DATA AVAILABILITY**

14 The authors declare that all data supporting the findings of this study are available within the
15 paper and its supplementary information files.

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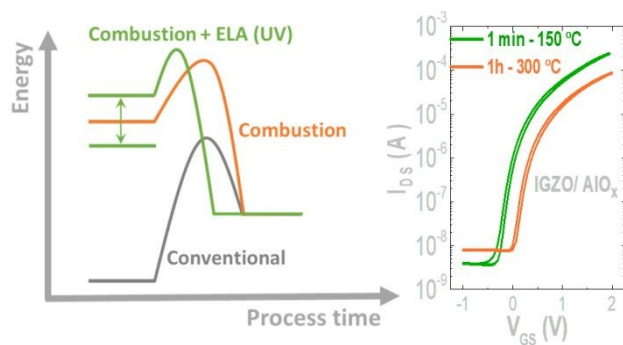
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19

1 Table of Contents/Abstract Graphic (max 20 words)

2 Excimer laser annealing (ELA) combined with combustion synthesis leads to high quality metal
3 oxide TFTs in a short processing time.

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