

## Supporting Information

### Photo-responsive hydrogels with photoswitchable mechanical properties allow time-resolved analysis of cellular responses to matrix stiffening

I-Ning Lee <sup>a,b,1</sup>, Oana Dobre <sup>c,d,1</sup>, David Richards <sup>d</sup>, Christoph Ballestrem <sup>c,d</sup>, Judith M. Curran <sup>b</sup>, John A. Hunt <sup>e,f</sup>, Stephen M. Richardson <sup>d,\*</sup>, Joe Swift <sup>c,d,\*</sup>, Lu Shin Wong <sup>a,\*</sup>

<sup>a</sup> Manchester Institute of Biotechnology and School of Chemistry, University of Manchester, 131 Princess Street, Manchester M1 7DN, United Kingdom.

<sup>b</sup> School of Engineering, University of Liverpool, Harrison Hughes Building, Liverpool L69 3GH, United Kingdom.

<sup>c</sup> Wellcome Trust Centre for Cell-Matrix Research, University of Manchester, Oxford Road, Manchester M13 9PT, United Kingdom.

<sup>d</sup> Division of Cell Matrix Biology and Regenerative Medicine, School of Biological Sciences, Faculty of Biology, Medicine and Health, University of Manchester, Manchester M13 9PL, United Kingdom.

<sup>e</sup> School of Science and Technology, Nottingham Trent University, Nottingham NG11 8NS, United Kingdom.

<sup>f</sup> Current address: School of Science and Technology, Nottingham Trent University, Nottingham NG11 8NS, United Kingdom.

<sup>1</sup> These authors contributed equally to this work.

\* *E-mail address:* [s.richardson@manchester.ac.uk](mailto:s.richardson@manchester.ac.uk), [joe.swift@manchester.ac.uk](mailto:joe.swift@manchester.ac.uk),  
[l.s.wong@manchester.ac.uk](mailto:l.s.wong@manchester.ac.uk)

## 1. Tables of prepolymer formulations.

**Table S1.** Formulations with ratios of AM/BIS/AZO monomers in DMF and EtOH.

Formula	Monomer/Crosslinker			Solvent System					Heating <sup>(b)</sup>	Observation <sup>(c)</sup>
	AM (M)	BIS (mM)	AZO (mM)	EtOH ( $\mu$ L)	DMF ( $\mu$ L)	DMSO ( $\mu$ L)	H <sub>2</sub> O ( $\mu$ L)	PBS <sup>(a)</sup> ( $\mu$ L)		
<b>1</b>	1.65	26.26	10.12	--	500	--	500	225	--	x
<b>2</b>	1.72	27.37	--	--	225	--	725	225	--	x
<b>3</b>	1.65	26.26	--	--	500	--	500	225	--	x
<b>4</b>	1.84	29.22	--	--	150	--	725	225	--	hydrogel
<b>5</b>	1.84	29.22	--	225	150	--	500	225	--	hydrogel
<b>6</b>	1.84	29.22	11.26	--	150	--	725	225	--	precipitation
<b>7</b>	1.84	29.22	5.63	225	150	--	500	225	--	precipitation
<b>8</b>	1.92	30.60	2.95	225	100	--	500	225	--	precipitation
<b>9</b>	1.92	30.60	2.95	225	100	--	500	225	70°C	hydrogel
<b>10</b>	0.41	6.59	--	100	--	--	625	250	--	hydrogel
<b>11</b>	0.46	7.33	--	45	--	--	580	250	--	hydrogel
<b>12</b>	0.41	6.59	--	45	100	--	580	250	--	hydrogel
<b>13</b>	0.46	7.33	--	45	200	--	380	250	--	x
<b>14</b>	0.46	7.33	--	--	625	--	--	250	--	x
<b>15</b>	0.41	6.59	--	413	100	--	212	250	--	x

<sup>(a)</sup> PBS: phosphate buffered saline

<sup>(b)</sup> "--" in this column indicates reaction was performed at room temperature.

<sup>(c)</sup> "x" in this column indicates no gel formation observed, mixture remained liquid

**Table S2.** Formulations with minimal AM/BIS monomers in different solvent systems.

Formula	Monomer/Crosslinker			Solvent System						
	AM (M)	BIS (mM)	AZO (mM)	EtOH ( $\mu$ L)	DMF ( $\mu$ L)	DMSO ( $\mu$ L)	H <sub>2</sub> O ( $\mu$ L)	PBS <sup>(a)</sup> ( $\mu$ L)	Heating <sup>(b)</sup>	Observation <sup>(c)</sup>
<b>16</b>	0.46	7.33	--	625	--	--		250	--	x
<b>17</b>	0.46	7.33	--	--	--	--	625	250	--	x
<b>18</b>	2.81	0.14	--	--	--	--	765	225	--	hydrogel
<b>19</b>	2.81	0.11	--	--	--	--	765	225	--	hydrogel
<b>20</b>	2.81	0.07	--	--	--	--	765	225	--	hydrogel
<b>21</b>	2.79	0.14	--	225	--	--	550	225	--	hydrogel
<b>22</b>	2.79	0.03	--	225	--	--	550	225	--	x

<sup>(a)</sup> PBS: phosphate buffered saline

<sup>(b)</sup> "--" in this column indicates reaction was performed at room temperature.

<sup>(c)</sup> "x" in this column indicates no gel formation observed, mixture remained liquid

**Table S3.** Formulation of AM/BIS/AZO polymer heated during polymerisation.

Formula	Monomer/Crosslinker			Solvent System					Heating <sup>(b)</sup>	Observation <sup>(c)</sup>
	AM (M)	BIS (mM)	AZO (mM)	EtOH ( $\mu$ L)	DMF ( $\mu$ L)	DMSO ( $\mu$ L)	H <sub>2</sub> O ( $\mu$ L)	PBS <sup>(a)</sup> ( $\mu$ L)		
<b>23</b>	2.87	45.68	4.40	100	--	--	600	--	70°C	hydrogel
<b>24</b>	0.80	12.72	6.13	200	--	--	300	--	70°C	hydrogel
<b>25</b>	1.32	20.92	10.08	100	--	--	200	--	70°C	hydrogel
<b>26</b>	0.45	7.13	3.43	400	--	--	500	--	70°C	x
<b>27</b>	0.45	7.13	3.43	--	--	--	900	--	70°C	precipitate
<b>28</b>	0.45	7.13	3.43	--	--	200	700	--	70°C	precipitate
<b>29</b>	0.45	7.13	3.43	--	--	200	700	--	70°C	precipitate
<b>30</b>	0.45	7.13	3.43	--	--	200	700	--	70°C	precipitate
<b>31</b>	0.45	7.13	3.43	--	--	600	300	--	70°C	precipitate
<b>32</b>	0.45	7.13	3.43	--	--	600	300	--	70°C	precipitate
<b>33</b>	0.40	6.36	3.06	100	--	--	100	800	70°C	precipitate

<sup>(a)</sup> PBS: phosphate buffered saline

<sup>(b)</sup> "--" in this column indicates reaction was performed at room temperature.

<sup>(c)</sup> "x" in this column indicates no gel formation observed, mixture remained liquid

**Table S4.** Formulations with high concentrations of monomers.

Formula	Monomer/Crosslinker			Solvent System					Heating <sup>(b)</sup>	Observation <sup>(c)</sup>
	AM (M)	BIS (mM)	AZO (mM)	EtOH ( $\mu$ L)	DMF ( $\mu$ L)	DMSO ( $\mu$ L)	H <sub>2</sub> O ( $\mu$ L)	PBS <sup>(a)</sup> ( $\mu$ L)		
<b>34</b>	0.50	--	0.35	--	--	--	880	--	--	precipitate
<b>35</b>	0.80	0.28	4.46	--	--	--	650	--	--	precipitate
<b>36</b>	0.66	0.23	3.68	--	--	350	450	--	--	x
<b>37</b>	0.66	0.23	3.68	--	--	250	550	--	--	x
<b>38</b>	0.66	0.23	3.68	250	--	--	550	--	--	x
<b>39</b>	2.56	0.08	3.55	225	--	--	420	225	--	precipitate
<b>40</b>	2.30	0.07	3.19	325	--	--	420	225	--	precipitate
<b>41</b>	1.72	0.07	3.19	425	--	--	320	225	--	precipitate
<b>42</b>	1.15	0.07	3.19	525	--	--	220	225	--	precipitate
<b>43</b>	1.76	--	3.26	425	--	--	300	225	--	precipitate
<b>44</b>	1.76	--	3.26	--	--	225	500	225	--	precipitate
<b>45</b>	1.72	0.07	3.19	--	--	225	520	225	--	x
<b>46</b>	0.59	0.07	--	--	--	225	500	225	--	x
<b>47</b>	1.15	0.07	3.19	--	--	225	520	225	--	x
<b>48</b>	0.57	0.07	3.19	--	--	325	420	225	--	x

<sup>(a)</sup> PBS: phosphate buffered saline

<sup>(b)</sup> "--" in this column indicates reaction was performed at room temperature.

<sup>(c)</sup> "x" in this column indicates no gel formation observed, mixture remained liquid

**Table S5.** Formulations of AM/BIS/AZO monomers with various solvents.

Formula	Monomer/Crosslinker			Solvent System					Heating <sup>(b)</sup>	Observation <sup>(c)</sup>
	AM (M)	BIS (mM)	AZO (mM)	EtOH ( $\mu$ L)	DMF ( $\mu$ L)	DMSO ( $\mu$ L)	H <sub>2</sub> O ( $\mu$ L)	PBS <sup>(a)</sup> ( $\mu$ L)		
<b>49</b>	2.30	0.07	1.59	325	--	--	420	225	--	hydrogel
<b>50</b>	1.72	0.07	1.59	425	--	--	320	225	--	hydrogel
<b>51</b>	1.15	0.07	1.59	525	--	--	220	225	--	hydrogel
<b>52</b>	1.76	--	3.26	--	--	225	500	225	--	hydrogel
<b>53</b>	1.76	0.04	1.63	--	--	225	500	225	--	hydrogel
<b>54</b>	1.17	0.04	1.63	--	--	125	600	225	--	precipitate
<b>55</b>	1.17	0.04	1.63	--	--	100	625	225	--	hydrogel
<b>56</b>	1.76	--	1.63	--	--	225	300	425	--	hydrogel
<b>57</b>	1.06	--	1.47	--	--	100	200	750	--	precipitate
<b>58</b>	1.06	--	1.47	--	--	100	725	225	--	precipitate
<b>59</b>	1.17	--	1.63	--	--	225	200	525	--	precipitate
<b>60</b>	1.06	--	1.47	--	--	100	725	225	--	hydrogel
<b>61</b>	0.59	--	1.63	--	--	100	625	225	--	x
<b>62</b>	1.17	0.04	1.63	--	--	225	500	225	--	hydrogel
<b>63</b>	0.59	--	1.63	--	--	225	100	625	--	x
<b>64</b>	1.76	--	0.81	--	--	225	300	425	--	hydrogel
<b>65</b>	1.17	--	0.41	--	--	113	200	637	--	hydrogel
<b>66</b>	1.13	--	0.16	--	--	45	200	737	--	hydrogel
<b>67</b>	1.17	--	0.81	--	--	225	200	525	--	hydrogel
<b>68</b>	1.17	0.04	0.41	--	--	113	200	637	--	hydrogel
<b>69</b>	1.17	0.04	0.16	--	--	45	200	705	--	hydrogel

<sup>(a)</sup> PBS: phosphate buffered saline<sup>(b)</sup> "--" in this column indicates reaction was performed at room temperature.<sup>(c)</sup> "x" in this column indicates no gel formation observed, mixture remained liquid

**Table S5.** Continued.

Formula	Monomer/Crosslinker			Solvent System						
	AM (M)	BIS (mM)	AZO (mM)	EtOH ( $\mu$ L)	DMF ( $\mu$ L)	DMSO ( $\mu$ L)	H <sub>2</sub> O ( $\mu$ L)	PBS <sup>(a)</sup> ( $\mu$ L)	Heating <sup>(b)</sup>	Observation <sup>(c)</sup>
<b>70</b>	1.76	--	8.14	200	--	225	300	225	--	hydrogel
<b>71</b>	1.59	--	7.37	225	--	100	300	425	--	precipitate
<b>72</b>	1.56	--	7.23	120	--	220	300	430	--	precipitate
<b>73</b>	1.56	--	7.23	220	--	220	300	330	--	precipitate
<b>74</b>	2.45	--	7.27	225	--	100	375	150	--	precipitate
<b>75</b>	2.20	--	3.40	225	--	100	360	225	--	hydrogel

<sup>(a)</sup> PBS: phosphate buffered saline

<sup>(b)</sup> "--" in this column indicates reaction was performed at room temperature.

<sup>(c)</sup> "x" in this column indicates no gel formation observed, mixture remained liquid

**Table S6.** Formulations of AM/AZO polymer using improved conditions based on previous formulations.

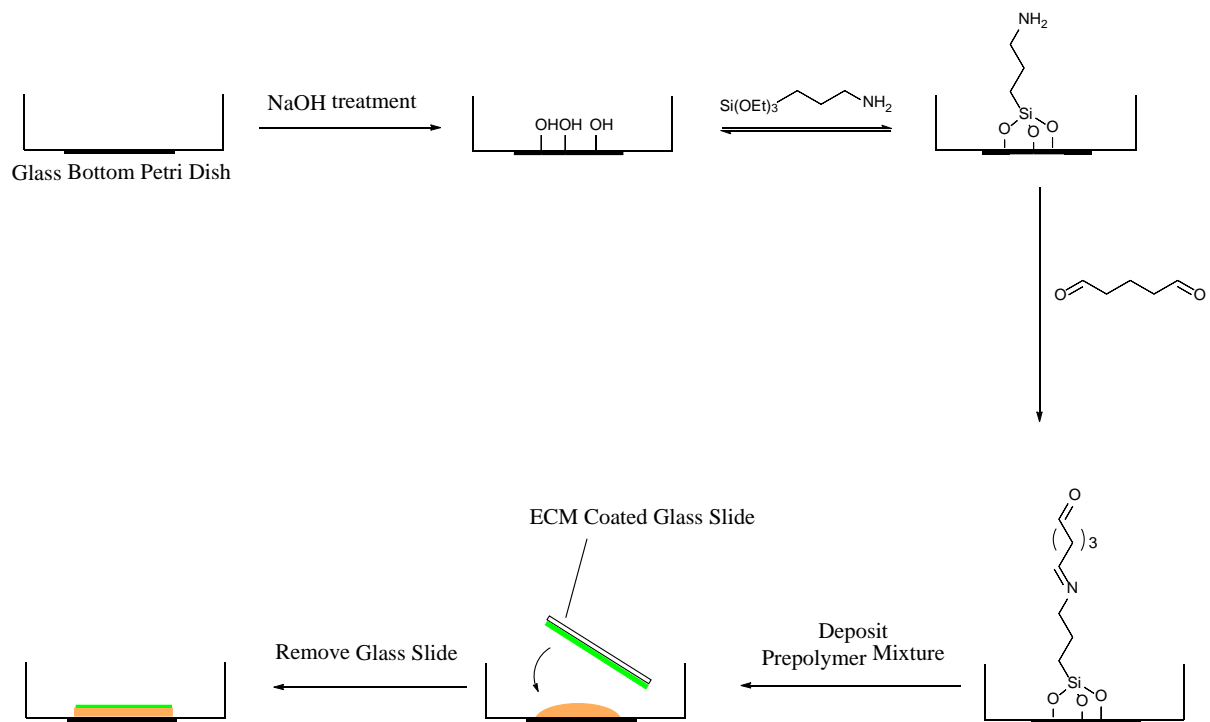
Formula	Monomer/Crosslinker			Solvent System					Heating <sup>(b)</sup>	Observation <sup>(c)</sup>
	AM (M)	BIS (mM)	AZO (mM)	EtOH (μL)	DMF (μL)	DMSO (μL)	H <sub>2</sub> O (μL)	PBS <sup>(a)</sup> (μL)		
<b>76</b>	2.20	--	6.79	225	--	100	360	225	--	hydrogel with precipitated particulates
<b>77</b>	2.20	--	5.10	225	--	100	360	225	--	hydrogel with precipitated particulates
<b>78</b>	2.20	--	3.40	180	--	60	360	310	--	hydrogel
<b>79</b>	2.20	--	6.79	180	--	60	360	310	--	hydrogel with precipitated particulates
<b>80</b>	2.20	--	3.40	100	--	60	360	390	--	hydrogel with precipitated particulates
<b>81</b>	2.20	--	3.40	100	--	80	360	370	--	hydrogel
<b>82</b>	2.03	--	3.13	100	--	80	360	450	--	hydrogel with precipitated particulates
<b>83</b>	4.05	--	3.13	100	--	80	720	90	--	hydrogel
<b>84</b>	1.01	--	3.13	100	--	80	180	630	--	precipitation
<b>85</b>	4.05	--	3.13	100	--	160	720	10	--	x
<b>86</b>	4.40	--	6.79	100	--	80	720	10	--	precipitation

<sup>(a)</sup> PBS: phosphate buffered saline

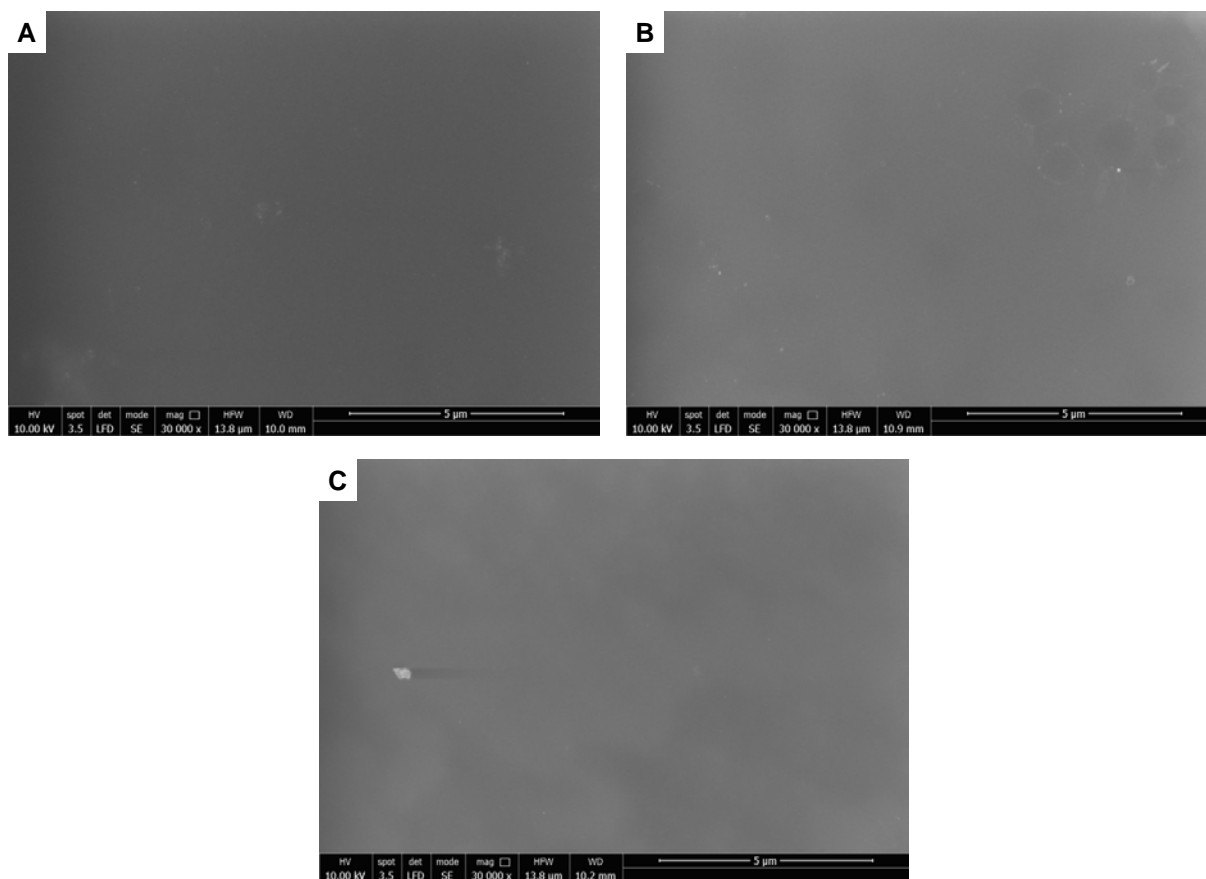
<sup>(b)</sup> "--" in this column indicates reaction was performed at room temperature.

<sup>(c)</sup> "x" in this column indicates no gel formation observed, mixture remained liquid

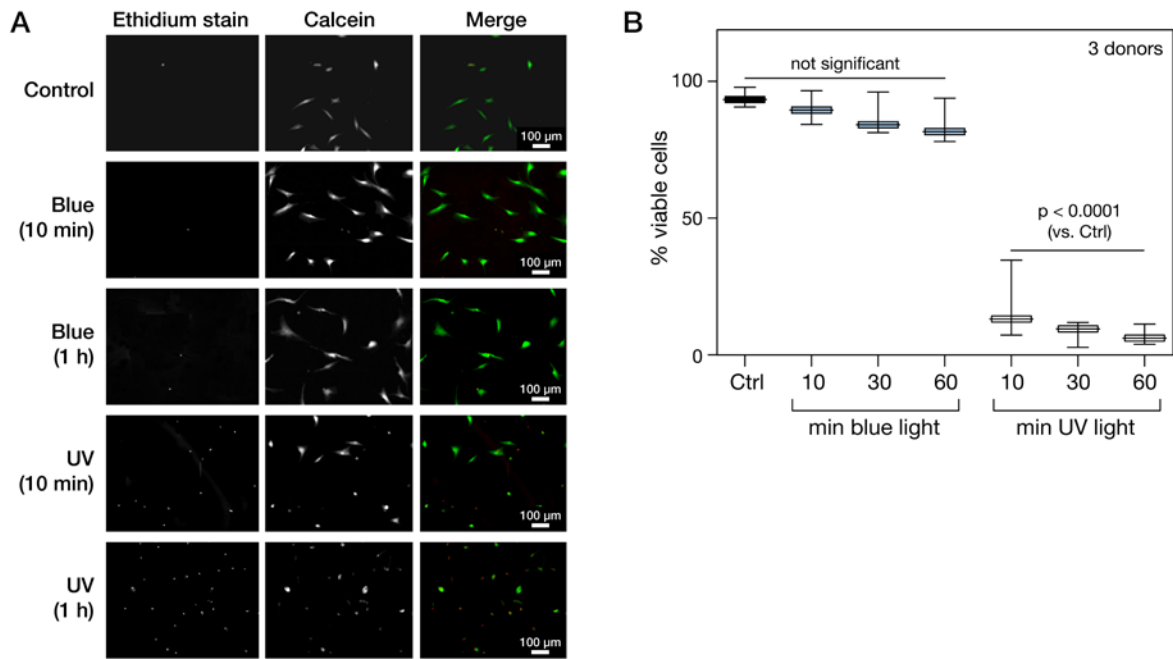




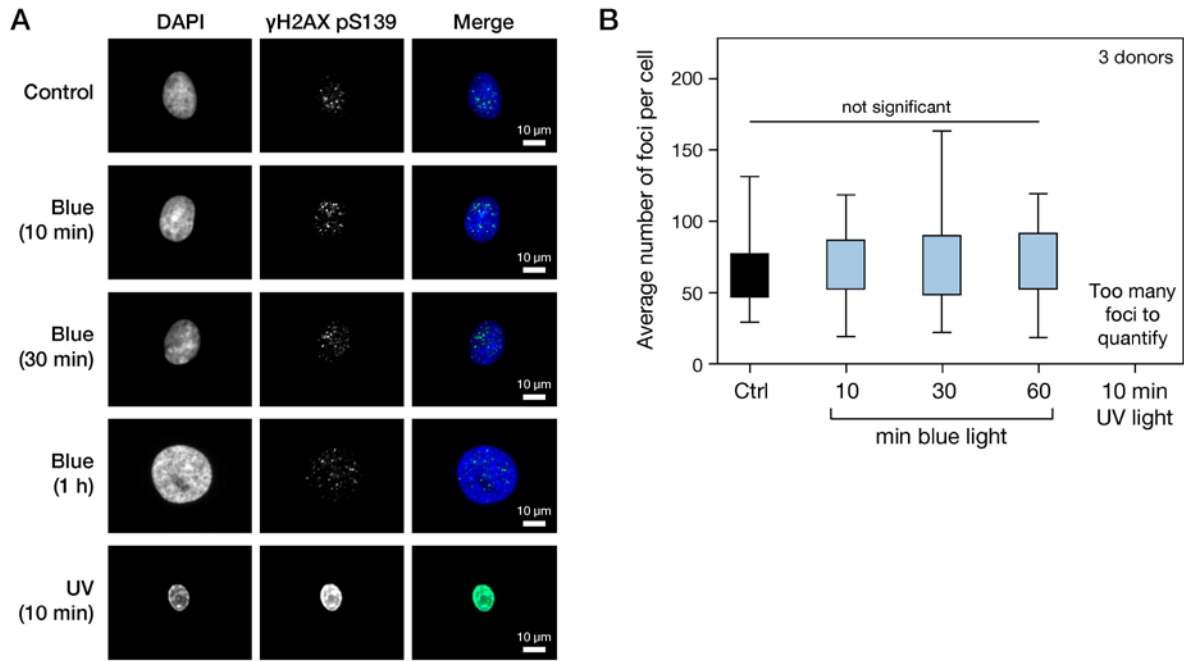
**Figure S1.** Schematic diagram of the substrate preparation for cell culture.



**Figure S2.** Low magnification ESEM micrographs of photo-responsive hydrogels before and after irradiation. The images were recorded following the treatments indicated in Figure 4A at the point of “stiffness measurement”. (A) Control (“ctrl”) sample: AZO hydrogels not subjected to irradiation. (B) “(-) blue” sample: hydrogel treated with UV (365 nm) irradiation. (C) “(+) blue” sample: hydrogel treated with UV (365 nm) and subsequent blue light (490 nm) irradiation.



**Figure S3. Cell viability following UV or blue light irradiation.** Live cells are indicated by green fluorescence due to the generation of free calcein by intracellular esterases, while dead cells exhibit red fluorescence from ethidium, following loss of cell membrane integrity. (A) Primary human MSCs were subjected to different durations of UV (365 nm) or blue light (490 nm) irradiation. Cell viability was assayed by ethidium and calcein staining (scale bar = 100  $\mu\text{m}$ ). (B) Quantitative analysis of cell viability images. Exposure to blue light for up to 1 hour did not significantly reduce cell viability. Exposure to 10 minutes of UV irradiation significantly reduced cell viability to  $18 \pm 8\%$  ( $\pm$  S.E.M.;  $n > 177$  measurements, cells from three donors;  $p$ -values indicated from ANOVA testing).



**Figure S4. DNA damage following UV or blue light irradiation.** (A) Primary human MSCs subjected to UV or blue light irradiation were imaged with the nuclei DAPI stained and immuno-stained against  $\gamma$ H<sub>2</sub>AX pS139 (scale bar = 10  $\mu$ m). (B) Quantification of the number of  $\gamma$ H<sub>2</sub>AX pS139 foci per nucleus. Blue light exposure for up to 1 hour did not significantly increase the number of foci, but UV exposure of just 10 minutes created more foci than could be resolved with the microscope ( $n > 57$  measurements, cells from three donors; significance from ANOVA and *post hoc* testing).