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#### QUALITY OF URBAN STORMWATER RUNOFF:

#### AN APPRAISAL OF THE ROLE OF GULLY POT LIQUORS

IAN JOHN FLETCHER

# VOL I

Submitted to the Council for National Academic Awards in partial fulfilment of the requirements for the degree of Doctor of Philosophy.

Sponsoring Establishment: Trent Polytechnic, Nottingham. Date of Submission: March, 1981. BhD 1.6 SEE SI/FIEE Rep.

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VOLUME 1

In memory of Eric Stanley Fletcher.

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### LIST OF SYMBOLS

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A	=	area	
В	=	surface breadth	
c	=	concentration	
C	=	specific concentrations	
$C_1, C_2, C_3$	=	Muskingum coefficients	
d	=	pipe diameter	
g	=	acceleration due to gravity	
i	=	inflow	
I	=	rainfall intensity	+
К	=	rate of release of settled sediments	
ĸs	=	roughness coefficient	
L	=	pipe length	
М	=	mass	
Р	=	percentage mixed	
q	=	outflow	
Q	=	flow rate, discharge	
r	=	pipe radius	,
R	=	retention time	
s <sub>o</sub>	=	channel slope	
s <sub>f</sub>	=	friction slope	
S	=	storage	
t	=	time	
v	=	volume	
W	=	wave speed	
x	=	downstream distance	
У	=	water depth	
~	=	storage routing coefficient	
3	=	Muskingum parameter	
β	=	Muskingum parameter	
Σ	=	the sum of	·
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#### ACKNOWLEDGEMENTS

1. 1. 1.

My sincere thanks to the following people for their advice, assistance and encouragement throughout this research project.

Dr. C.J. Pratt - Director of Studies Dr. G.E.P. Elliott - Second Supervisor

Mr. J. Brown (Head of Department) and other academic staff at the Department of Civil and Structural Engineering, Trent Polytechnic. the state of the second st

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Mr. F.J. Wright (Principal Technician) and all technical staff of the Department of Civil and Structural Engineering. Staff of the Computing Department at Trent Polytechnic. Dr. C.H.R. Kidd - (Institute of Hydrology) Dr. G. Mance - (Water Research Centre) Mrs. V. Farren - Typing

Finally, to my wife Heather, for her support and patience.

#### DECLARATION

- 1) While registered as a candidate for the degree for which submission is made, the author has not been a registered candidate for any other award of the CNAA or of a University.
- 2) The candidate has completed three postgraduate courses:
  - a) Course leading to Part 1 of the Diploma of the Institute of Water Pollution Control.
  - b) Course in computing methods.
  - c) Course in statistical techniques.
- 3) The candidate has participated in the following colloquia:
  - a) First International Conference on Urban Storm Drainage, Southampton, April 1978.
  - b) Symposium of the International Workshop on Rainfall-Runoff Processes Over Urban Surfaces, Institute of Hydrology, April, 1978.

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#### ABSTRACT

The quantity and quality of stormwater runoff from a small separately sewered residential suburban catchment has been monitored for 24 storm events over an 18 month period. An automatically operated continuous injection lithium dosing apparatus in conjunction with an automatic sampling machine was used as a means of flow determination and sampling for quality measurements. The samples were analysed for several pollutants and the results enabled an estimate of the annual pollutant loads discharged from the catchment to be calculated. Using multiple linear regression the relationship of pollutant loads and concentrations with meteorological parameters was investigated. Conclusions have been drawn on the relative importance of parameters in defining pollutant loads, the mode of removal of pollutants, pollutant interrelationships and the significance of sources of pollution.

The quality of water stored in roadside gully pots on the same catchment was investigated over a two year period. This study and additional laboratory investigations have enabled detailed conclusions to be drawn on the impact of sources of pollution on the chemistry of the stored water and the effect of rainfall, dry periods, seasonal variations and human activity upon the quality of the stored water.

The removal of dissolved material and settled sediments from gully pots for different flow rates was investigated in the laboratory using purpose-built apparatus. The results obtained were used to develop and calibrate mathematical expressions describing the pollutant transport processes. In order that the equations could be applied to real catchments and storm events, it was necessary to develop a mathematical simulation of the rainfall-runoff process. The model developed employs a single linear reservoir for the prediction of an inlet hydrograph to the sewer system from a net rainfall hyetograph and the fixed parameter Muskingum-Cunge technique for routing the inlet hydrographs through a pipe network to form the outlet hydrographs. Inclusion in this model of the equations, describing the removal of water stored in roadside gully pots prior to the rainfall event, enabled the contribution of gully liquors to storm flow to be calculated.

Contemporaneous monitoring of rainfall and runoff and contemporaneous examination of the quality of stormwater runoff and the quality of water stored in gully pots provided the data necessary for utilising the model to determine the pollutant loads in stormwater runoff that are attributable to pollutants stored in gully pots.

#### 1. INTRODUCTION

Only within the last decade has urban stormwater run-off been identified as an important source of pollution of natural water courses. All waste water was originally conveyed to sewage treatment works through what has been termed the combined sewerage system, that is, domestic waste, industrial effluent and stormwater run-off are transported through the same pipe network. However, during periods of heavy rainfall, when flows exceeded the capacity of the treatment works, sewage was discharged without treatment into adjacent water courses and resulted in cases of severe river pollution. To alleviate this pollution the adoption of separate sewer systems for new urban developments was advocated. In a separate system stormwater run-off is carried through an independent pipe network to that used for domestic waste and is discharged without treatment into a convenient water course, thereby relieving the river of polluting discharges of storm sewage overflow. Whereas in the combined system the quality of stormwater was obscured, in the separate system the true polluting potential of this run-off was revealed.

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Originally, pollution of surface waters was largely governed by the discharge of poor quality sewage and industrial effluents. The advent of secondary and even tertiary treatment of these point sources of pollution has now rendered them less significant in determining river pollution. Therefore, the assessment of the quality of non-point or diffuse polluting sources such as rural and urban stormwater run-off has become of growing importance. The need to accurately characterise the extent of pollution arising from these sources can be seen to be of particular importance when the growing number of new urban developments drained by separate systems is considered.

Additionally, the last decade has seen an increase in public awareness and concern over pollution of the environment. This concern, combined with advances in our appreciation of the deleterious effects of water pollution on the environment and on public health, has led to the recognition of water as being a national resource of great value. The result has been a trend towards an integrated approach to the management of the Nation's water resources. To assist in this approach mathematical simulation has been increasingly utilised as a management tool to predict the quantity and behaviour of pollutants in the

environment. Such techniques require an in-depth quantitative knowledge of pollutant concentrations and the processes involved in their accumulation, formation and entry into water-courses. Consequently, the need for detailed information on these polluting processes is as great as ever and previously neglected sources of pollution, such as urban run-off, have assumed some considerable importance.

#### 1.1. THE POLLUTION OF STORMWATER RUN-OFF

The pollution of stormwater results from the contamination of rainwater from the time of its origin in the atmosphere until the moment of its discharge into a receiving body of water. The variation in quality of the run-off may be brought about by such a wide variety of different events and processes that the complex changes in water chemistry that result are difficult to predict. In broad outline pollution of stormwater run-off may be attributed to the following principal sources:-

- a) Open land contributes wind blown dust and soil particles which accumulate in roadside gutters. Additionally it acts as a source of nutrients and micro-organisms. In heavy storms run-off can occur directly off open land resulting in high solids loads.
- b) Natural vegetation results in the accumulation of leaves in the Autumn and other decaying vegetable matter during the rest of the year.
- c) Roof and road surfaces are sources of inorganic solids, cement, sand, bitumen and other erroded road material. Additionally, they act as collecting areas for other pollutants.
- d) Motor traffic is a source of oil, petrol, exhaust gases, lead and other particles from car bodies and tyres.
- e) Human activity results in the accumulation of litter, detergents from car washing, garden fertilisers, grass-cuttings, herbicides, pesticides and salt used for de-icing in the winter.
- f) Animal activity leads to the deposit of faeces and urine on impervious surfaces.

g) Factories and chimneys - sources of dust and gaseous emissions in the atmosphere that subsequently settle or are washed down in rainfall.

The inter-relation of the sources and the processes at work in the accumulation of pollutants on impervious surfaces is illustrated diagrammatically in Figure 1. The relative importance of these sources at any given location is governed by the type of land use (urban, industrial, commercial) and by individual catchment characteristics such as impervious area, slope and local soil type.

For any given storm event the degree of pollution is determined by the following factors:-

- a) The characteristics of the storm event, e.g. volume, intensity and duration.
- b) The antecedent weather conditions. The length of the preceding dry period determines the mass of pollutants accumulated on impervious surfaces and the quality of water stored in gully pots. Temperature can affect the degree of degradation of organic matter by micro-organisms and in winter the processof road surface erosion by frost action. The wind direction can influence the degree of atmospheric pollution to be expected from adjacent industries.
- c) The time of year. This defines the type of antecedent weather conditions to be expected, the type of human activity to be expected (road salting in winter, grass cutting in summer) and the type of biological activity expected (leaf fall in autumn).

d) The effectiveness of council cleansing operations.

#### 1.2. LITERATURE REVIEW

#### 1.2.1. Introduction

Although, prior to 1970, considerable research effort had been expended on the efficient design of stormwater systems, little attention had been paid to the quality of stormwater run-off and the polluting impact of these discharges on water courses. However, recognition of the severe magnitude of pollution derived from combined sewer overflows and the introduction of separate stormwater systems initiated a flood of



research projects on the problems of urban stormwater pollution. By the middle of the decade several countries had identified problem areas, formulated national priorities and initiated research projects (UNESCO, 1974). Research was particularly active in the U.S.A., where, under the Environmental Protection Agency's Storm and Combined Sewer Technology program, over 90 million dollars was awarded to some 140 projects (Field and Gardner, 1978). In comparison, the research effort in the U.K. has been minimal and is considered separately in Section 1.2.2. All published work in stormwater research (system design, stormwater quality, management practices, etc.) has been reviewed annually since 1973 in the Journal of the Water Pollution Control Federation (Field et al, 1973 - 1979). In the field of stormwater quality researchers have addressed a wide variety of problem areas, these have been conveniently considered under the following headings:-

New Section

the reference of the start in

a) Pollutant concentrations and loadings.

- b) Sources of pollution
- c) Impact of stormwater runoff on watercourses.
- d) Modelling and predicting the quality of stormwater runoff
- e) Control of pollution.

#### 1.2.2. <u>Summary of Research into the Quality of Stormwater Runoff in</u> the United Kingdom.

The earliest study on the quality of runoff from a separate system of drainage was conducted by Wilkinson (1956) on a small suburban housing estate at Oxhey, Hertfordshire. The problems of urban stormwater pollution uncovered by his study received no further attention until 1970 when the Technical Committee on Storm Overflows and the Disposal of Storm Sewage, whilst acknowledging that high pollutant loads could occur from industrial catchments, advocated the use of separate sewer systems for new urban developments. The magnitude of runoff pollution from industrial areas was soon confirmed by a study of the River Tame which drains a substantial portion of industrialised Birmingham (Lester et al 1971). A discrepancy between pollutant loads

in the river and the loads predicted from all known sources was noted by Garland and Hart (1971), the excess load was attributed to polluted runoff from urban surfaces. Horner et al (1975) considered the problem of stormwater pollution from London. They concluded that the massive improvement in quality of the River Thames resulting from improved sewage treatment facilities left stormwater discharges as the major deleterious effect on water quality. Tucker (1975) reported results on the discharge of suspended solids from two suburban catchments in Nottingham and confirmed many of Wilkinson's earlier observations. Additionally, he presented some measurements on the quality of waters stored in roadside gully pots. The results indicated that gully pots could harbour highly polluting liquors which could be rapidly discharged at the start of a storm. These preliminary observations were further investigated by this research project which was initiated in Autumn 1975. The influence of stormwater runoff on an urban streamway in Hendon, Greater London, was investigated by Ellis (1976, 1977). He paid particular attention to the nature of particulate material and stressed its significance in determining the polluting potential of runoff. The quality of runoff from a busy urban motorway in Birmingham was the attention of a study by Hedley and Lockley (1976). They recorded extremely high pollutant loadings over the year and reported winter road salting and motor vehicles to be the most significant pollution sources. In 1977 the Working Party on Storm Sewage (Scotland) (Nicholl and McGillivray) recommended a return to the construction of combined sewer systems in Scotland. Significant financial savings and the poor quality of stormwater runoff influenced their decision. The results of a project funded by the Department of the Environment and conducted at the Water Research Centre were presented in two papers (Water Research Centre (1977)), Mance and Harman (1978)). Storm runoff from the whole of Stevenage New Town and from a purely residential sub-catchment were monitored over a period of three years. In addition to concentration and loading data, the effect of stormwater discharges on receiving streams, some approaches to the control of pollution and a comparison with a hypothetical combined system were also considered. The preliminary findings of the research presented in this thesis on the quality of water stored in roadside gully pots were published in 1978 (Fletcher et al). Also in 1978 the quality of runoff from a predominantly industrial catchment at Newcastle was the attention of short study by Adams (1978). Significant pollutant loads were recorded.

#### 1.2.3. Pollutant Concentrations and Loads in Stormwater Runoff

The earliest studies on pollution of stormwater runoff were confined to qualitative aspects alone. Palmer (1950) gathered grab samples of runoff at catch basins in Detroit, U.S.A., and found B.O.D's up to 234 mg/l, total solids up to 914 mg/l and coliform MPN's/100 ml of 930,000. Similarly, Shigorin (1956) evaluating street runoff in Moscow and Leningrad found B.O.D's of 285 mg/l and suspended solids of 14,541 mg/l. Akerlindh (1950) looked at runoff from streets and parks in Stockholm and recorded COD's up to 3,100 mg/l, total solids of 3,000 mg/l, BOD's of 80 mg/l and coliforms up to 200,000 per 100 ml. These early studies served to indicate the possible magnitude of pollution of stormwater runoff, however, it became apparent that, to meaningfully evaluate the significance of stormwater runoff, it was necessary to monitor not only the quality of runoff, but also the quantity and, thereby, to compute the actual mass of pollutants discharged to watercourses. Wilkinson (1956) was the first to measure stormwater flow at the same time as sampling for quality determination and his studies pointed the way for future work. There is now a large bank of published work on pollutant concentrations and loadings from a variety of different catchment types and different storm events. Some of the more important studies, in chronological order, are:-Weibel et al, (1964); Pravoshinsky and Gatillo, (1966); Burm, (1978); De Filippi and Shih, (1971); Soderlund and Lehtinen, (1972); Bryan, (1972); Waller, (1972); Whipple et al, (1974); Kluesener and Lee, (1974); Droste and Hartt, (1975); Tucker, (1975): Hedley, (1976); Holbrook et al, (1976); Whipple and Hunter, (1977); Cordery, (1977); Mills, (1977); Water Research Centre, (1977); Roberts et al, (1977); Randall et al, (1977); Malmquist and Svensson, (1977); Wanielista, (1977); Ellis, (1977); Fletcher et al, (1978); Lindholm and Balmer, (1978); Mance and Harman, (1978); Bedient et al, (1978); Rimer and Reynolds, (1978); Helsel et al, (1979).

The wide variability in concentration of pollutants that can occur is illustrated in Tables 1 and 2, where mean and maximum concentrations for selected studies have been listed. Comparison of pollutant loading rates from different studies is not straightforward and is frequently hindered by the lack of uniformity and, in some instances, the unreliability of the methods employed to obtain the data. Some authors have used an average loading rate per unit of rainfall or runoff to compute the annual pollutant load discharged from the catchments. Again, methods used to

Mean and Maximum concentrations of pollutants in stormwater runoff from residential areas (in mg/l) TABLE 1

CONSTITUENT	OXHEY Wilkins	U.K. on 1956	CINCINAT Weibe <sup>1</sup>	rti, USA L, 1964	ANN ARB Burm,	or, USA 1968	DURHA Bryar	M, USA 1, 1972	Soderla	KHOLM nd, 1972
	Mean	Max.	Mean	Max.	Mean	Max.	Mean	Max.	Mean	Max.
Suspended solids	376	2045	210	1200	2080	11,100	3930	13,900	122	500
Volatile solids	27%	20%	53	290	218	570	426	1,100	80	360
Dissolved solids									489	2000
COD			66	610			224	660		
BOD	20	100	19	84	28	62	31.3	232	13	25
Nitrate nitrogen			0.4	1.5	1.5	3.6				
Ammoniacal nitrogen			0.6	1.9	1.0	2.0				90 <b>9</b> 94   1999
Phospate			0.8	4.3	5.0	16.4	0.55	2.5	0.09	0.3
Chloride			12	35			18	390	168	320
Lead							0.67	1.85	1.0	3.0

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ZURICK, CHR Roberts, 1977	Mean	59			37	6	1.1	4.0	1.8	0.22		0.14	0.16	0.014		
OSLO, NORWAY Lindholm, 1978	Mean	929	72		44				2.3	0.3		0.07	0.10	0.03		
ATLANTA, USA Holbrook, 1976	Mean	428			11	б	0.38	0.15		0.35		0.13			5	
GE, U.K. , 1978	Max.	1386		390			6.9	6.0	7.6		1008	3.7	1.63	0.35		
STEVENA( Mance	Mean	112		258			1.7	0.28	2.0		49	0.205	0.271	0.028		
CANADA 1975	Max.	,1230	392			78.4	4.7	0.71			1585					
WINDSOR, Droste,	Mean	279	112			20.5	1.2	0.05			72					
CONSTITUENT		Suspended solids	Volatile solids	Dissolved solids	COD	BOD	Nitrate Nitrogen	Ammoniacal nitrogen	Total nitrogen	Phosphate	Chloride	Lead	Zinc	Copper		

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TABLE 1 (continued)

Lindholm, 1978 OSLO, NORWAY Commercial 4.2 1.2 0.82 1.73 0.52 Mean 1038 189 244 Holbrook, 1976 Industrial/Res. ATLANTA USA 63 22 Mean 0.24 0.26 0.35 323 0.57 0.27 Industrial/Res. CRAMLINGTON, UK 2.5 0.54 Adams, 1978 2070 1630 1520 Max. 0.19 Mean 225 196 0.53 124 70,000 BIRMINGHAM U.K. Hedley, 1976 Urban Motorway Max. 11.5 2.5 14.5 5700 1400 3.999 28.2% 29.5 2592 2.08 3.2 0.68 Mean STOCKHOLM, SWEDEN Soderland, 1972 4.0 0.3 360 1000 600 2800 8 100 Max. Highway 132 2.1 0.08 261 Mean 282 855 36 10 Ammoniacal nitrogen Volatile S. Solids Nitrate nitrogen Suspended solids Dissolved Solids CONSTITUENT Total nitrogen Phosphate Chloride Copper Zinc Lead BOD COD

TABLE 2 Mean and maximum concentrations of pollutants in stormwater runoff from other catchment types (in mg/l)

TABLE 3 Estimated annual pollutant loads discharged from urban catchments calculated using impervious area and annual rainfall (kg/ha/vr)

CONSTITUENT	OXHEY. UK Wilkinson 1956	HUCKNALL UK. Tucker, 1975	RISE PARK UK. Tucker 1975	B'HAM U.K. Hedley 1976	ZURICH, CHR Roberts 1977	SHEPHALL, UK. Mance, 1978	CRAMLINGTON UK Adams 1978
	Residential	Residential	Residential	Motorway	Residential	Residential	Industrial
Suspended solids	580	342	692	6289	09ħ	148	946
Volatile S. solids	157			1556			
Dissolved solids						168	581
BOD	23			173	17		
COD					280		451
Nitrate nitrogen				~	8.5	1.2	
Ammoniacal nitrogen					3.0	0.96	
Phosphate	M79/8				1.7		2.1
Chloride				11611		28	-
Lead				12.9	0.89	0.041	0.33
Zinc				19	1.1	0.066	
Copper				3.7	0.11	0.016	
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TABLE 4 Estimated annual pollutant loads discharged from urban catchments where the method of calculation is not known (kg/ha/yr)

	WINDSOR CANADA	DURHAM, USA	STOCKHOLM, SWEDEN	STOCKHOLM, SWEDEN	CINCINATTI USA	TWIN RIVERS	TWIN RIVERS II	SYDNEY AUSTRALIA
Constituent	Droste 1975	Bryan, 1972	Soderlund 1972	Soderlund, 1972	Weibel, 1964	Whipple 1978	USA Whipple 1978	Cordery, 1978
	Residential	Mixed	Highway	Residential	Residential	Residential	Residential	Residential
Suspended solids	609	18000	1185	620	803	816	1479	286.3
Volatile S. solids	198	9634	860	06ħ	166			
Dissolved solids			2040	630				
BOD	24.1	94.1	100	tt3	37	90.1	32	17.2
COD		1165	•		264	44.2	193	
Nitrate nitrogen	1.25					6.4	1.5	
Ammoniacal "	0.67					4.j	6.2	2.29
Total nitrogen			5.1	3.45	9.8			
Phosphate		4.09	0.2	0.16	2.75	7.6	95	2.86
Chloride	75.4	81.8						
Lead Zinc		2.5			a.	1.2	0.38 185	
Copper						0.58	0.38	

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estimate these values have varied considerably and have not always been reported in the literature. Some of the factors which may lead to variation in the estimation of annual loading rates are:-

- a) Storm sampling. For the computation of storm loads sampling should ideally be conducted at short time intervals throughout each event, however, some reported loading data has been obtained using only a few grab samples per storm.
- b) The number of storms used to estimate the annual load. When only a limited number of events are available to compute the annual load, the representative nature of these events must be questioned. As an example, Tucker (1978) quotes two widely different estimated annual loads for the Rise Park catchment, 692 and 325 kg/ha/annum, the first figure includes the results of one particularly heavy thunderstorm, the second figure excludes this one storm.
- c) The use of total or impervious catchment area.
- d) The annual rainfall used will vary from catchment to catchment.
- e) Use of discharge volume instead of rainfall to compute the average load per unit of run off.

Bearing in mind the limitations outlined above, Table 3 presents some estimated annual pollutant loads, where the impervious area was used in their calculation. Table 4 presents other figures from the literature where the method of calculation is not known. Again the figures show a wide variability in the pollutant loads discharged.

1.2.3.1. Shape of the Pollutant Runoff Curve

The initial stages of storm runoff are frequently characterised by the occurrence of high concentrations of pollutants. This effect has been termed the "first flush" and was first observed by Wilkinson (1956) and has since been confirmed by other workers. Tucker (1975) considered the first flush to be the removal, independent of rainfall intensity, of fine solids accumulated on the catchment area since previous storms. The duration of the flush and the recorded concentrations
were considered to be dependent upon the flow rate, (Figure 2). Another explanation for the first flush was tentatively provided by Ellis (1976), who observed the occurrence of peak solids concentration prior to the flood peak for the Silk Stream catchment in Greater London. He explained the phenomenon firstly by the settlement of solids on the antecedent storm recession that are rapidly flushed out on the rising limb of the storm wave, and secondly, by mats of fungus and slime in the sewer that act as filters trapping large amounts of fine particulates, these mats are subsequently broken up and loosened by the initial phases of runoff (Figure 3). Another source of first flush material is roadside gully pots or catch basins (American Public Works Association, 1969). Poor quality super natant liquors stored in the basins can be rapidly flushed out during the early stages of runoff.

The first flush is primarily a concentration effect. The actual mass of pollutant transported, in grams per second, is strongly related to flow rate. The occurrence of high pollutant concentrations at low flow rates, a situation that is typical for the first flush, can be of less significance when overall storm loads are considered. This is illustrated in Figure 4 where the removal of phosphate shows a first flush effect in concentration that is masked when the mass transported is considered (Cordery, 1977).

The first flush is not a constant feature of all storm events and its duration or occurrence has not been defined quantitatively. Wilkinson (1956) empirically considered it to be the first 30 minutes of runoff, however, he noted the variability of the effect:-

"in many storms runoff water remained fairly dirty for a considerable time and water discharged throughout the storm was stronger than the first flush of other storms"

Mance and Harman (1978) similarly defined the first flush as being the occurrence of the peak concentration within the first 40 minutes of runoff and observed this to occur for 80% of the events monitored. A dimensionless cumulative plot has been employed by Bedient et al (1978) to give a representation of the first flush effect (Figure 5). A steeply rising curve at the start of an event with a slope greater than 1 is indicative of a first flush.

In subsequent runoff, solids concentrations are normally related to discharge, occurring, either coincident or more typically, lagging just

Fig. 2 THE FIRST FLUSH, FLOW INDEPENDENT, REMOVAL OF SUSPENDED SOLIDS (SS) FOR A STORM AT RISE PARK, NOTTINGHAM (AFTER TUCKER, 1975)







Fig. 4 FIRST FLUSH AND FLOW RELATED REMOVAL OF SUSPENDED SOLIDS (AFTER ELLIS, 1976)



sat. 17 may 1975





behind the flow peaks. This effect has been tentatively explained by different vertical velocity and concentration profiles that are developed in the sewer. The position of mean solids concentration in a cross-section may be below the position of mean flow velocity and as a result the velocity of solids transport is less than that of the fluid. This results in a lag between flow and concentration peaks. In storms of long duration or of high rainfall intensity, the catchment may be stripped of all particulate material that is available This exhaustion of material results in low solids for release. concentrations in the latter stages of a storm that are unrelated to flow rate. Other parameters that are usually associated with solids, such as BOD, COD, heavy metals and phosphates, have been shown to follow a similar pattern of runoff to that outlined above (Cordery, 1977). However, there is a tendancy for the ratio of oxygen demand to solids concentration to decrease during a storm. This can be ascribed to the removal of the lighter organic fractions that exert a greater oxygen demand in the earlier phases of runoff.

The concentration of dissolved solids and other soluble species in runoff, such as nitrates, ammonia, etc., show little relationship with flow rate. Their concentrations tend to steadily decrease throughout a storm event with the highest concentrations occurring in the first flush phase (Figure 6).

No hard and fast rules can be applied to the shape of a pollutant runoff curve in relation to the discharge hydrograph. Although regression analysis has demonstrated that the concentration of parameters can be linearly related to the rate of discharge, a wide variability exists between storm events (Mance and Harman, 1978). The prediction of pollutant concentrations from hydrological data is therefore unreliable.

1.2.3.2. Relationship of Pollutant Loads with Meteorological Parameters

Recent studies employing multiple regression analysis have demonstrated that the characteristics of the current rainfall-runoff event are the dominant factors determining the pollutant loads discharged from a given catchment. The results of Mance and Harman (1978) indicate that, with the exception of chloride and ammonia, 60% to 90% of the variance is explained by the total volume discharged, the length of the antecedent dry period and the magnitude of the previous runoff event: the antecedent dry period and the magnitude of the previous event

THE REMOVAL OF NITRATE (NO3) AND AMMONIUM (NH4) SHOWING NO RELATIONSHIP TO DISCHARGE (FLETCHER, 1981)







Fig. 6

Fig. 8 RELATIONSHIP BETWEEN THE ANTECEDENT DRY PERIOD AND THE AVERAGE BOD CONCENTRATION FOR THE FIRST FLUSH (AFTER WILKINSON, 1956)







accounting for a maximum of only 6%. A similar analysis by Adams (1978) showed that the amount of rainfall and its average intensity were the most significant paramters in determining the pollutant load. The duration of the rainfall event and the antecedent dry period were of less significance. A strong correlation between pollutant load and total runoff volume was also noted by Bedient et al (1978) for suspended solids, COD, soluble organic carbon, Kjeldahl nitrogen and phosphorus. They used the derived relationship to predict loads from other storms with some degree of success. Similarly, Weibel et al (1964) demonstrated a good correlation between total rainfall and solids and BOD load (Figure 7).

The significance of the antecedent dry period has been a matter of conjecture. Wilkinson (1956) observed that the concentration of BOD in the first flush increased with an increase in the period of drought, approximately 3 mg/l for every 24 hours (Figure 8). However, he acknowledged that the curves were ill defined, due to wide variations that occurred for short dry periods and the limited number of results that were available for longer periods of drought. A tentative linear relationship between the first flush solids load and the preceding dry period was indicated by the results of Tucker (1975). However, other workers (Weibel et al, (1964); Bryan, (1972); and Waller, (1972)), could find no statistical relationship between the two. Although, logically the antecedent dry period will influence the amount of material available for transport, in reality the characteristics of the current event are of more significance in determining the actual pollutant load discharged.

1.2.3.3. Relationship of Pollutant Loads with Catchment Characteristics

It is to be expected that the physical characteristics of a catchment, e.g. impervious area, slope, land usage (urban, industrial, commercial), will strongly influence the amount of pollutant available and load discharged from the sewer. Bradford (1977) collated all available data on pollutant loadings from published work in the U.S.A. up to 1972 and attempted to relate loading rates with a variety of catchment characteristics including land use, population density, impervious area and road surface type. However, multiple regression analysis yielded only limited success and that with data from one area only. A relationship between land use and pollution loading could not be demonstrated. Other workers have reached similar conclusions,

pollutant loading rates are dependent upon so many variables that are specific to single catchments only, that the prediction of pollutant loads by the simplistic approach of 'labelling' by catchment type and utilising generalised statistical expressions is unsatisfactory.

It has been generally stated that pollutant loads tend to increase with increased urbanization of a catchment and are usually highest for industrial areas, however, there is much conflicting evidence. Holbrook et al (1976) monitored runoff from four small suburban watersheds and a central business area in Atlanta, U.S.A. They noted, with the exception of suspended solids, that all concentrations for the business area were higher than the mean concentration for the suburban area. However, Rimer and Reynolds (1978), characterising runoff from a number of different land uses found that levels of suspended solids, COD, phosphorus amd lead were a function of the degree of development and the percentage of area impervious of the catchment with the exception of the central business area (Table 5). Suspended solids levels for the business area were approximately half those observed at the 'high activity' residential areas. This was explained by the lesser degree of land disturbing activity and more regular municipal street sweeping. Bradford (1977) also noted, with the exception of BOD, organic nitrogen and lead, that loading rates for commercial areas were generally below average. Similar results were obtained for a study of four catchments of differing land uses in Houston (Bedient et al, 1978). Nutrient loadings, in particular phosphorus, were found to be greater at the two urban catchments than the two background forested watersheds, with higher loads being recorded on the catchment containing some industrial development. Suspended solids followed a similar pattern, but no differences were observed for total COD, due probably to the leaching of organic material at the forested sites. Helsel et al (1979) concentrated their attention on the influence of land use on the quality of runoff for seven metals (lead, zinc, copper, chromium, iron, manganese and cadmium). Their study was conducted using 19 small sub-basins of seven different land use categories in Washington, D.C. Although significant differences in runoff quality occurred between sub-catchments for all of the metals, they concluded that valid comparisons between broad land use categories could not be made, because of large observed variations between stations within the same category. They considered individual land use characteristics, such as traffic and ground slope

TABLE 5 Comparison of the quality of urban runoff for different land users (after Rimer and Reynolds 1978)

Land Use	Percentage	Number of	Av	erage Peak Con	centration	ŭ
	Lmpervious	storms	S. Solids	COD	Phosphate	Lead
Rural (low activity)	2.7	5	284	11	1.40	0.1
Rural (high activity)	5.1	2	416	58	0.76	0.1
Conmercial (low activity)	12	6	. 575	87	0.83	0.2
Residential (low activity)	16	6	664	166	0.84	0.6
Residential (high activity)	32	8	1199	194	1.03	2.1
Commercial (high activity)	35	8	1082	320	1.08	1.7
Central Business District	80	11	528	336	0.66	0.9

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to be better indicators of stormwater quality. Urban catchments were found to give significantly higher levels of lead, zinc and copper and good non-linear correlations for these metals were found against traffic and percentage impervious cover. The results were interpreted as suggesting that motor vehicles were a major source of these metals and that impervious surfaces delivered them more efficiently to storm drainage systems. The importance of the percentage of impervious cover was also noted by Lindholm and Balmer (1978). They found good correlations with COD, total phosphorus, total nitrogen, lead and suspended solids for seven urban catchments in Norway. With the exception of one catchment where land development gave rise to larger solids loads, there was a tendency for loadings from the city centre, commercial catchment to be greater than those from the residential catchments.

## 1.2.3.4 Comparison of Combined and Separate Sewerage Systems

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Several workers have attempted to compare the advantages and the disadvantages of the two types of system in terms of pollutant loads that might be discharged to watercourses. Unfortunately, a direct comparison is hampered by the obvious fact that no catchment has been served by both systems! Additionally, there are other problems of comparison, for example, should loading rates be compared for the time period of a runoff event or over a whole year; should stormwater loads be compared against raw sewage, primary sewage effluent, secondary sewage effluent, secondary sewage effluent or combined sewage overflow; and should loads or concentrations be compared? Different criteria will apply to each catchment and, as a result, there is much conflicting opinion in the literature. However, the fundamental pros and cons can be simply listed as follows:-Combined system advantages:-

- a) reduced cost of a single pipe network.
- b) when no overflow occurs all sewage is treated at the treatment works.
- c) some measure of protection against pollution by accidental spillage of toxic chemicals, e.g. road tanker accidents.

#### Disadvantages:-

a) in storm conditions, when the system overflow limit is exceeded, storm sewage is discharged directly into a watercourse.

b) the whole polluting load is discharged at one point.

Separate system advantages:-

- a) alleviates storm sewage overflow.
- b) the river receives its due flow from the natural catchment.
- c) enables more efficient design of sewage treatment plants by eliminating wide volume and quality variations and eliminating the need for storm tanks.

Disadvantages:-

- a) Stormwater is discharged untreated and may also be polluting.
- b) the impervious nature of urban areas produces runoff more quickly than natural catchments resulting in rapid river flow variations with the possible disruption of stream ecology.
- c) accidental spillages pass directly into the river without treatment.

d) increased capital costs of installing two pipe networks.

Wilkinson (1956) compared the quality of storm runoff from a residential catchment with that of a typical secondary sewage effluent. He concluded that the separate system reduced polluting loads when judged by BOD, but increased them in terms of suspended matter by 6 to 7 times. However, he noted that even in the first flush, BOD values rarely exceeded the 20 mg/l value proposed by the Royal Commission on Sewage Disposal and that the percentage of organic matter in the suspended solids was substantially less than that in sewage effluent. Weibel et al (1964) compared annual storm runoff loads for a residential catchment in Cincinatti to annual raw domestic sewage loads calculated from average sewage concentrations and typical sewage flow rates for a population of 9 persons/acre (Table 6). Suspended solids were again found to be greater than those in domestic sewage, even on an annual basis. During storm runoff events, greater loads of all paramters were considered to be discharged than in raw sewage over a similar time period. Similar

TABLE 6 Comparison of stormwater runoff loads and domestic sewage loads over one year (after Weibel et al, 1964)

Percentage	0†1	1117	25	9	6	11		
Domestic Sewage . (lbs/acre/yr)	240	360	096	540	27	81		
Storm Runoff Load (lbs/acre/yr)	730	160	240	33	2.5	6.8		
Constituent	Suspended Solids	Volatile S. Solids	COD	BOD	Phosphate	Total Nitrogen		

TABLE 7 Comparison of combined sewer overflow and storm runoff loads over one year (after Burm et al., 1968)

Percentage	505	199	- 74 767	TC TC	- 4	0 uc	0		534 21	67				
Ann Arbor, Storm Runoff load (lb/acre/yr)	1010	185	780	13/		б <b>.</b> С	ω •	0.7	0.8	0.4				
Detroit, Combined sewer overflow (lb/acre/yr)	200	93	185	83	06	5.6	11.0	6.2	0.15	1.6			-	
Constituent	Suspended solids	Volatile S. Solids	Settled solids	Volatile settled solids	BOD	Soluble Phosphate	Total Phosphate	Ammoniacal - Nitrogen	Nitrate - Nitrogen	Organic - Nitrogen				

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conclusions were reached by Bryan (1972) when comparing runoff from a residential estate in Durham with raw domestic sewage. Pollutant loads from a combined system in Detroit were compared to those from a separate system in Ann Arbor by Burm et al (1968). Their results showed that for all parameters, except solids, the normalised load discharged was greater from the combined sewer overflow than from the storm sewer (Table 7). The solids load was about 5 times greater than the storm sewer, however, the organic (volatile) portion of this load was only 2 times greater. A comparison of the bacteriological quality of runoff discharged from the two systems was also conducted (Burm and Total coliforms in combined sewer overflow were found Vaughan, 1966). to approach those in raw sewage and were 10 times greater than levels found in both types of runoff. Soderlund and Lehtinen (1972) compared the quality of urban runoff, combined sewer overflow and treated sewage for catchments in Stockholm, and concluded that the level of contamination of storm runoff was similar to that of combined sewer overflow. The Water Research Centre (1977) compared polluting loads released from a storm sewer serving a small residential catchment in Stevenage to loads that might have been discharged from a combined system draining the same catchment. The comparison was achieved through the real time combination of flow and quality data for the existing foul and storm sewer systems at a variety of overflow settings for the hypothetical combined system. At a setting of 6 times D.W.F. (dry weather flow), a value commonly employed in combined system design, the predicted mass discharged for all determinands, with the exception of ammoniacal nitrogen, was marginally less than that observed to be discharged from the storm sewer. However, the mean concentrations of storm sewer discharges were lower than those from the combined system overflow. To achieve a significant reduction in the mass of ammoniacal nitrogen discharged, it was estimated that an overflow setting of 20 D.W.F. was necessary. It was concluded that the choice of system for individual locations would largely be governed by local circumstances, such as the presence of an existing system, distance to the point of discharge, the condition of the river and the acceptability of any discharge of urban runoff.

#### 1.2.4. Sources of Pollution

#### 1.2.4.1 Impervious Surfaces

The natural decay and erosion of roads, roofs, footpaths and driveways is a source of significant quanitities of particulate matter in

# Fig.10(a) ACCUMULATION OF SOLIDS ON ROAD SURFACES FOR DIFFERENT LAND USE CATEGORIES



Constituent	Mass (kg per km of road)
Total Solids	397
Oxygen demand.	
BOD COD Volatile Solids	3.8 27 28.3
Nutrients.	
Phosphates Nitrates Kjeldahl Nitrogen	0.31 0.027 0.63
Heavy Metals.	
Zinc Copper Lead Nickel Mercury Chromium	0.18 0.057 0.16 0.014 0.021 0.031
Pesticides	
P.p DDD P.p DDT Dieldrin PCB <u>Bacteriological</u>	$19 \times 10 \stackrel{-6}{-6} \\ 17 \times 10 \stackrel{-6}{-6} \\ 6.8 \times 10 \stackrel{-6}{-6} \\ 310 \times 10 \stackrel{-6}{-6} $
Total Coliform Fecal Coliform	$28 \times 10^9$ 1.6 x 10 9

TABLE 8 Quantity and characteristics of contaminants found on street surfaces (after Sartor et al., 1974)

Contaminant	FRACTION	OF TOTAL MASS	(%)
	43m	43-246	246m
Total solids	5.9	37.5	56.5
BOD	24.3	32.5	43.2
COD	22.7	57.4	19.9
Volatile Solids	25.6	34.0	40.4
Phosphates	56.2	36.0	7.8
Nitrates	31.9	45.1	23.0
Kjeldahl Nitrogen	18.7	39.8	41.5
Heavy Metals	5	2	48.7
Pesticides	7	3	27
PCB	3	4	66_

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# TABLE 9 Pollutant fractions associated with particle sizes (after Sartor et al. 1974)

TABLE 10 Loading of solid material on street surfaces for different land uses (after Sartor et al., 1974)

Land Use	Loading (kg/km)
Residential:-	240 ) 252 ) 122 ) 270 340 ) 397 )
Industrial:- Light Medium Heavy	735 ) 252 ) 659 990 )
Commercial:-	82 82

storm runoff: additionally, and more importantly, these impervious surfaces act as collecting areas for pollutants from other sources, (atmospheric dustfall, wind blown material), that can then be rapidly and easily transported by overland flow to the sewer system. The accumulation and composition of contaminants on street surfaces has been the attention of studies by Sartor, Boyd and Agardy (1974) and Pitt and Amy (1974), after an initial survey by the American Public Works Association (1969) had shown the accumulation of appreciable quantities of street litter at 18 test sites.

A study of 12 U.S. cities, Sartor et al (1974), showed the major constituent of street surface contaminants was consistently inorganic, mineral-like matter similar to common sand and silt. This finding has been confirmed in the U.K. by Ellis (1976), who estimated that street surface sediments consisted of an inorganic mineral fraction of 50% -80% of the total, being composed of substantial amounts of brick, glass and concrete. He found that organic materials, which usually comprised only a small fraction of the total, but could contribute up to 30%, were comprised primarily of rubber and bitumen fractions, with lesser amounts of humic materials, plant debris and other complex organics associated with vehicular pollution. Average values for the accumulation of pollutants per kilometre of roadway are shown in Table 8.

The largest proportion of the polluting potential of street sediments has been shown to be associated with the fine, primarily organic, solids fraction. Sartor et al found that material below 43 microns, although comprising only 6% of the total solids accounted for 25% of the oxygen demand, up to 50% of the nutrients and heavy metals and 75% of the pesticides (Table 9). The mass of material that accumulates on a given street surface shows great variability from site to site but is generally dependent upon four factors:-

- a) Type of land use (Table 10). Industrial locations usually have higher loading rates than urban sites. Commercial sites probably have high loading rates, however, they are also swept more frequently and therefore the net recorded loading rates are lower.
- b) Surface type. Asphalt surfaces have been shown to result in 80% heavier loads than concrete surfaces.

- c) Age of the surfaces. The rate of decay of old surfaces can be up to 2 times greater than a new surface in good condition.
- d) Time elapsed since the surface was last cleaned, either by municipal sweeping or by rainfall. There is a tendency for a maximum loading to be reached after approximately 10 days, when natural accumulating and removal processes are balanced, (Figure 10).

Roadside channels and kerbs form barriers in and against which contaminants accumulate. Sartor et al recorded that some 80% of the material on the road surface was located within 15 cm of the kerb, (Figure 10).

Decay of roofed surfaces will also contribute to pollutant loadings in runoff, however, this source is less significant than material derived from road surfaces. The contribution from surface decay was quantified by the Water Research Centre (1977) at approximately 3% of the total mass discharged as compared to a contribution of 72% from roads and pavements.

## 1.2.4.2 Aerial Deposition

Aerial deposition results in the accumulation of pollutant material on urban surfaces by three processes:-

- a) Settlement of particulate matter over dry periods, commonly known as dustfall or dry deposition.
- b) Rainout. The removal of particles by condensation processes in clouds (Rasool, 1973).

c) Washout. The removal of particles by falling raindrops.

Several workers whilst investigating the quality of storm runoff have also made measurements of aerial deposition rates. However, the methods of measurement used have often differed, some workers considering just dustfall, others measuring total aerial deposition. Partly as a result of this variability in measurement technique, evidence on the significance of insoluble (particulate) aerial deposition in determining

TABLE 11 Insoluble (particulate) deposition rates for several studies

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Catchment	Reference	Deposition Rate (kg/ha/annum)
Hamilton, Canada	Waller, 1972	169 - 300
Cincinatti, USA	Weibel, 1964	570
Stevenage, UK	Mance, 1978	95
Birmingham, UK	Hedley, 1976	1160
Styrrup, UK	Cawse, 1972	28 - 250
Blyth, UK	Warren Spring Lab. 1971	370 - 890
Chicago, USA	U.S. Dept. of Interior 1969	<b>1560 –</b> 2320

TABLE 12 Rates of aerial deposition and the percentage mass of pollutants discharged that is derived from aerial deposition (after Mance and Harman, 1978)

Constituent	Rate of Deposition (mg/m <sup>2</sup> /day)	Percentage Contribution to runoff
Suspended solids	26.1	23
Soluble solids	77.1	· 30
Chloride	14.6	28
Nitrate - N	1.2	68
Ammoniacal - N	1.2	417
Nitrite - N	0.026	50
Copper	0.051	96
Manganese	0.051	49
Lead	0.128	54
Zinc	0.160	38

the quality of storm runoff is conflicting. A large variance in insoluble deposition rates for different catchments has been recorded (Table 11). In a Chicago study, Heaney and Sullivan (1971) estimated that approximately 70% of material on street surfaces was attributable to dustfall whereas Hedley and Lockley (1976) found that only 3 tonnes per annum (1160 kg/ha/ annum), or 1.2% of the total of 250 tonnes input to an urban motorway catchment in Birmingham, U.K., could be ascribed to aerial deposition. Even this figure is higher than that obtained for a survey of six U.K. sites (Cause, 1974) where a variation in insoluble deposition rates between 28 and 250 kg/ha/annum was noted, the maximum occurring for the one urban catchment monitored at Styrrup, Notts.

Other workers have attempted to assess the contribution of aerial deposition to the mass of solids discharged in runoff. Waller (1972) considered that dustfall accounted for all of the solids in roof runoff for a number of small urban catchments in Hamilton, Canada. His findings were in agreement with those of Weibel et al (1964) who calculated that a dustfall rate of 570 kg/ha/annum accounted for some 70% of the 820 kg/ ha/annum of solids discharged from an urban catchment in Cincinatti, U.S.A. However, for a residential catchment in the U.K. Mance and Harman (1978) estimated that total aerial deposition accounted for only 25% of the mass exported from the catchment. Their measurements of total aerial deposition also enabled them to estimate the contribution to runoff from other constituents (Table 12). Aerial deposition was found to be the principal source of inorganic nitrogen and a major contributor of heavy metals. These findings were in agreement with those of Kluesener and Lee (1974), who found that nitrate in rainfall accounted for 20 - 90% of the nitrate loadings in runoff and Fisher (1968) who noted that nitrate input in rainfall exceeded the output in stream flow. The discrepancy in ammonia input and output recorded by Mance and Harman were also observed by Cause (1974) and Kluesener and Lee (1974), and have been tentatively explained by undetected contamination from bird faeces. Work by Barkdoll et al (1977) indicated that dustfall was the major source of COD, chloride, mercury, arsenic and phosphate in stormwater runoff.

The U.K. Atomic Energy Authority conducted a comprehensive two year study of trace element concentrations in air and rainfall and trace element deposition and washout rates at seven sites in the U.K. (Cause, 1974). Additionally, element ratios in the atmosphere were compared to those occurring naturally in the earth's crust and conclusions were

drawn on the causes of variability. A 10 to 40 fold increase in the concentrations was noted for areas influenced by population and industry. These increases were primarily ascribed to the following sources:-

a) combustion of coal, oil and petrol with the release of fly ash.

b) smelting of metals with the production of 'fume', e.g. Zn<sup>0</sup>

c) mining and crushing of minerals.

d) land cultivation, earth moving and building operation.

Evidence for the importance of fossil fuel combustion in influencing atmospheric pollution was manifold. The ratio of lead to bromide was near to the ratio found in ethyl fluid added to petrol and the ratio of nickel to vanadium agreed with the ratio of particulate emission from coal combustion. Local industry and domestic coal burning was considered to contribute sulphur, ammonia and the oxides of nitrogen in addition to releasing fly ash. The internal combustion engine was also considered to be an important source of the oxides of nitrogen. Hallsworth and Adams (1973) have also recorded the contribution of significant quantities of fly ash to the atmosphere, often associated with high lead levels, from combustion at power stations in the East Midlands.

Pronounced seasonal changes were noted for the majority of elements with higher values occurring in the winter rather than the summer months. This has been explained by reduced atmospheric dispersion under conditions of persistent low inversion layers and the combustion of additional fuels for heating purposes. Johnson et al (1966) also observed a seasonal variation for dustfall composition at Seattle, U.S.A. with maximum values occurring over March, April and May. concentrations remaining approximately constant for the remainder of the year. Johnson also recorded the significance of industrial contamination of the atmosphere, with the exception of lead and phosphate, greater accumulations of all constituents were found near industrialised areas. Lead was found to correlate well with traffic flow and phosphates were greatest in residential and suburban areas.

Huff (1976) investigated the effect of atmospheric effluents from a large urban - industrial area at St. Louis, U.S.A. on water quality in two small basins that were downwind of the city. In spite of considerable atmospheric contamination by industry he concluded that no strong relationship between streamwater quality and weather conditions could be found. With the exception of nitrate and zinc which were believed to be derived from an industrial source, total atmospheric deposition accounted for only a small fraction of the total stream load.

### 1.2.4.3 Motor Vehicles

That vehicles are a major contributor to the accumulation of pollutants on roadways is hardly surprising. Vehicle contamination of the environment arises from the following sources:-

a) expelation of exhaust gases and particulates.

b) oil and petrol leaks and spillages.

c) tyre wear

d) bodywork corrosion.

There is little doubt that vehicle exhaust is the most significant source of lead in the environment. The resultant levels of lead in runoff have been the attention of several workers, Newton et al (1974), Oliver et al (1974), Solomon (1977). The sources of lead and the literature pertaining to the complex processes involved in its transportation and physiochemical interactions in the environment have been reviewed by Laxen and Harrison (1977).

Lead is added to petrol as an anti-knock agent in the form of tetra alkyl lead compounds at a concentration of between 0.45 and 0.55 g/l. However, the largest proportion of lead in exhaust is emitted as particulate halides and oxyhalides due to reaction with other petrol additives such as ethylene dibromide and chloride. The quantity of lead emitted has been shown to be dependant upon engine speed and driving mode (Ter Haar et al, 1972) varying from 5% of the mass added up to 2000% during rapid acceleration. This results in an average lead emission rate of between 0.06 and 0.2 g/km (Solomon, 1977). The subsequent deposition of the lead is largely determined by the size of the particulates with which it is associated. Fine particulates may be

swept into the atmosphere and deposited many miles from their point of emission, but the majority of lead is deposited in a strip approximately 30 metres on either side of the highway. Lead falling on open land is considered to be effectively immobilised in the surface soil layer and causes an insignificant contribution to water pollution. On the road surface lead is almost exclusively insoluble and is associated, primarily by adsorption, with particulates. Levels of lead ranging from 1000 to 2000 mg/l have been recorded for urban street dusts (Day et al, 1975; Turner, 1971) that are substantially higher than background levels for rural soils of 50 to 100 mg/l. The speciation of lead in water is the subject of some debate, however, the majority of research indicates that the metal remains closely associated with suspended solids resulting in total concentrations in water of the order of 0.01 to 10 mg/l. Levels of lead in road runoff are usually considerably higher than background levels and are considered to pose a significant threat to receiving water ecosystems depending upon its speciation.

The presence of black stains on roadways due to oil spillages and leaks is a common occurrence and many investigators have observed oil in stormwater discharges as a blue/green sheen present on the surface of the water. However, little quantitative information is available on the mass of oil reaching surface waters due to technical problems involved in obtaining representative samples, nevertheless, the problem is clearly one of some magnitude. In addition to the oxygen demand that oils and petrols exert on surface waters the occurrence of complex and toxic trace organic compounds in runoff that are associated with oils and petrols, including known carcinogens such as polycyclic aromatic hydrocarbons, may present further problems.

Tyre wear has been observed by Hedley and Lockley (1976) to result in the accumulation of particulate rubber in sedimentation chambers that collected drainage from an urban motorway. They estimated that approximately 1.5 kg/ha/annum were discharged from the catchment. Additionally tyres contain significant quantities of zinc (circa 1% of the total mass) and may be the cause of the high levels of this metal observed in runoff. Bourcier and Hindu (1979) recorded an average zinc level of 19.1 mg/l in runoff from a highway viaduct in Washington, D.C. where vehicles were virtually the only possible source of pollution. Hedley and Lockley's study found values ranging from 0.6 to 8.0 mg/l for their similar catchment.

Catchment	Reference	Maximum Chloride Concentration (mg/l)
Motorway, Birmingham UK Housing Est. Stevenage UK	Hedley (1975) Mance (1978)	70,000 2,724
Expressway, Chicago USA	Amercian Public Works Association (1969)	25,100
Highway, Wisconsin, USA	Schraufnagel (1967)	10,250
Street " "	11	3,275
Snow pile " "	11	1,130

TABLE 13 Chloride levels in runoff after application of rock salt as a deicer. Sector Part

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TABLE 14 The composition of a typical rock salt used for deicing (after Hedley, 1975)

Constituent	Concentration (mg/kg)	
Sodium Chloride	906,000	
Solids	92,000	
Iron	1,550	
Nickel	12	
Lead	8.7	
Zinc	6	
Chromium	4.7	
Cyanide	5.7	
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Corrosion of car bodywork gives rise to the accumulation of metal, paint, glass and mud on road surfaces. Bishop (1968) noted that these processes were accelerated in periods of winter road salting and were responsible for a considerable loss of metal from vehicles. He found this salt induced corrosion to be particularly active over concentration ranges of 0.05% to 10%, levels that are typically encountered during winter road salting.

#### 1.2.4.4 Winter Road Salting

The use of rock salt as a deicer for roadways in winter is now common practice. In the U.K. the Department of the Environment have recommended that salt is applied at rates of 14 to 56  $g/m^2$  depending upon conditions at the time of application. In America application rates of 110 to 330 kg/km have been reported (Field et al, 1974). During periods of snow melt and subsequent runoff events the majority of this salt is rapidly washed off the roads and results in extreme high concentrations of chloride and other associated parameters in streams draining urban areas (Table 13). Not only may runoff of such large quantities of salt adversely affect river ecosystems but also trace elements and other additives in the rock salt may present a more serious threat to the environment. Ferric, sodium and potassium ferrocyanides are often added to prevent caking of salt stocks. The sodium form has been shown to generate cyanide in the presence of sunlight, a concentration of 3.8 mg/l cyanide have been produced from an initial 15.5 mg/l of the sodium salt (Hanes et al, 1970). Chromate and phosphate are sometimes added to salt stocks as corrosion inhibitors and levels of 24 mg/l sodium chromate, 1.7 mg/l hexavalent chromium and 3.9 mg/l total chromium have been reported in Minneapolis, U.S.A. (Field et al, 1974). Analysis of a typical rock salt applied in Birmingham, U.K. (Hedley and Lockley, 1975) revealed the presence of substantial amounts of the heavy metals, nickel, lead, zinc and chromium (Table 14). The application of this salt was shown to be the major source of chloride, nickel and chromium in runoff when the annual total loadings were considered.

The environmental implications of the large scale application of salt are further considered in Section 1.2.5.

1.2.4.5 Human and Animal Activity

The important influence of human and animal behaviour in polluting the urban environment has been noted by many workers. However, there is little quantitative information available on the masses of pollutants involved and their significance compared to pollutants derived from other sources. The main activities which can influence the polluting potential of runoff and have not been previously considered in other categories include the following:-

a) the use of garden fertilisers, pesticides and herbicides.

b) grass cutting

c) littering, careless disposal of paper, cigarette ends, metal cans and bottles.

d) car washing.

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e) urination and defaecation by dogs and birds.

The relative importance of these activities will vary considerably from catchment to catchment. For example in commercial areas littering may be the most significant activity but in residential housing estates car washing and grass cutting are likely to be of more importance.

The American Public Works Association (1969) collected information on the accumulation of streetlitter and the use of pesticides, herbicides and fertilisers at a number of cities. Contributions of paper, metal and glass ranging from 0 to 40% of the total mass of material on the impervious surface was noted and the use of pesticides, etc. was reported at 90% of the cities. The presence of pesticides in runoff has been monitored by Weibel et al (1964), Bryan (1972) and Murphy and Carleo (1978). Murphy and Carleo recorded a significant loading of chlorinated organics in runoff from a predominantly residential area of 0.034 kg/ha/annum, concentrations were found to be higher for the summer months presumably due to the increased application of insecticides and herbicides.

The accumulation of dog faeces on a residential housing estate has been conservatively estimated by the Water Research Centre (1977) to be about 17 g/m<sup>2</sup>/year. The faeces were reported as being a potential source

of harmful pathogens, viable ova of which have been isolated from areas inundated by stormwater.

## 1.2.4.6 Roadside Gully Pots

Gully pots or catch basins are small chambers located beneath roadside channels to act as an inlet point for overland runoff to the sewer system. The purpose of their location at this point is two-fold. Firstly, they remove particulate material transported by runoff that may subsequently cause sewer blockage, and secondly, in combined systems they serve as a water seal to prevent the release of bad odours from the sewer.

The accumulation and subsequent putrefaction of organic matter washed into gully pots was first noted as early as 1900 by Folwell, additionally, he drew attention to the irregularity and inefficiency of cleaning procedures employed at the time. The role of gully pots as a source of pollution of stormwater received no further attention until 1969 when the American Public Works Association measured BOD's of the stored liquors. Concentrations between 35 and 225 mg/l were recorded after a period of several days without rainfall and they concluded for an average BDD of 60 mg/l the loading possible from gully pot liquors was  $7\frac{1}{2}$  times that contained in runoff water due to street liquor contact. Their observations indicated that between runoff events the supernatant liquors tended to become septic and the trapped solids took on the general characteristics of an anaerobic sludge, this resulted in the degradation of water quality and the release of foul odours. Sartor and Boyd (1974) also noted that the decomposition of accumulated litter, leaves and engine oil in gully pots presented a threat to receiving water quality. They considered that whilst gully pots were effective in removing coarse inorganic solids, they were ineffective in trapping the more polluting fine solids and organic matter. Tucker (1975) collected 55 samples from 10 pairs of gully pots at different locations in the city of Nottingham. His results (Table 15) showed great variability between pots at the same and at different locations which he ascribed to the influence of human activity. He also noted the anaerobic decomposition of organic matter and stressed the poor quality of water that could be encountered when pots became clogged with fallen leaves in autumn. Mance and Harman (1978) recorded that even after flushing by heavy rain, pots became anoxic within 24 hours. They observed the formation of thick

Catchment	STEVEN	AGE, U.K.	CHICAGO, U.S.A.	NOTTINGHAM U.K.	-
Reference	MANCE	(1978)	Amer. Public W'ks Ann. (1969)	Tucker (1975)	
	Mean	Max.	Range	Range	
ed solids	7.06	1032		0 - 4031	
ed solids	282	5860			
			35 - 225	2.3 - 350	
				422 - 965	
organic carbon	28.9	255			
cal - N	0.7	4.5			
N -	0.08	0.34			
N	0.8	3•0			
	0.02	0.04			
Se	0.2	0.62			
	0.17	0.60			
	0.11	0.25			
U	76.4	1829.0		6.4 - 31525.0	
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TABLE 15 The quality of water stored in roadside gully posts

bacterial scums that were considered to possibly represent a significant portion of the first flush of BOD and suspended solids. Dry periods were noted to result in a slow deterioration in quality of the stored water with increasing concentrations of soluble compounds including chloride, soluble metals and ammoniacal nitrogen. Increases in ammoniacal nitrogen were ascribed to the action of denitrifying bacteria reducing nitrate nitrogen. They concluded that gully pots contributed significantly to the pollutant load of runoff.

## 1.2.5 The Polluting Impact of Stormwater Discharges on Watercourses

The polluting potential of stormwater discharges on watercourses is dependent not only upon the quality of the runoff but also upon the size of the discharge and the diluting capacity of the watercourse into which runoff occurs. The manner in which stormwater discharges occurs, that is, short duration but rapidly flowing slugs of polluted water, is also important in defining the polluting potential of runoff. Therefore, to survive in this environment, a species must be able to withstand the toxic properties of the runoff, the hydraulic force of the discharge and an environment where thick sediments and organic rich sludges are common. The result is a decrease in the diversity of species in streams receiving runoff from developed areas. In Greenfield, Massachusetts (Whipple et al, 1978) the benthic macroinvertebrate community was shown to progressively decrease in species diversity as the stream passed through an urban setting. The bottom sediments and the bodies of these species were found to contain large quantities of heavy metals, far above concentrations that are normally considered toxic. High concentrations of heavy metals have also been recorded in sediments of the R. Beane, downstream of the storm sewer outfall draining Stevenage New Town (Water Research Centre, 1977): the levels of metals decreasing the greater the distance from the outfall (Figure 11). The fauna and flora of the river were found to be contaminated with heavy metals and limited to a few species only. The diversity of biota recovered only after the river had passed over two weirs. Data from Maryland, U.S.A. (Whipple et al, 1976) indicated that there was a general reduction in fish species diversity as watersheds are developed. In the R. Beane, caged trout were found to be able to survive exposure to short summer storm discharges.

One of the principal reasons for the decrease in species diversity is the considerable oxygen demand, primarily from the organic solids load, that urban runoff exerts upon the receiving water. This can result in the

depletion of dissolved oxygen to levels where fish life is threatened. An example of how serious this effect can be is the R. Tame, Staffordshire, which drains a substantial area of the industrialised West Midlands (Lester et al, 1971; Garland and Hart, 1971). In storm conditions the river may rise four-fold in a short space of time, the resuspension of river bed sediments and the scouring of sediments from urban surfaces creates a demand for oxygen that has resulted in the river becoming anoxic and devoid of fish life. Horner et al (1977) monitored dissolved oxygen depletion due to stormwater runoff in the R. Thames. They noted that the impact of runoff on oxygen levels was dependent not only on the volume of discharge, its quality and the point of discharge, but also on the antecedent conditions in the main river. When oxygen levels were high, very large discharges could be absorbed with little harmful effect, when oxygen levels are low small discharges could have a considerable impact. Ellis (1976) recorded dissolved oxygen sag curves for the Silk Stream in North London (Figure 12), a catchment that is heavily urbanised and receives large discharges of urban runoff. He noted the accumulation of large quantities of sediment and decomposing benthal sludges in the stream. The resuspension of these fine sludges during turbulent storm surges was considered to bring the deeper anaerobic layers up towards the surface aerobic zone and thus create a high demand for additional oxygen for oxidation. In addition to organic rich sediments, oil in runoff can also exert a considerable oxygen demand. Oil has the additional deleterious property of spreading out as a thin film across the surface of receiving waters. This can, particularly in still waters, effectively hamper the natural process of dissolved oxygen exchange with the atmosphere and so further the creation of anoxic conditions. Oils can also affect river species directly by smearing their surfaces and clogging their respiratory organs.

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Ellis (1976) has stressed the importance of stormwater solids as a sink for heavy metals. Cadmium and lead extracts up to 10 mg/l were recorded for particles below 50 microns diameter in the Silk Stream sludges. Although toxic metals show a preference for association with particulate matter and are therefore rendered less harmful (levels in solids may be up to  $10^{2}$  times higher than in solution), in certain circumstances their release may be affected. One such circumstance is winter road salting. The input of high concentrations of the alkali metals, sodium and calcium, can effect the release of significant concentrations of toxic metals through ion exchange processes or complexation with chloride ions and so suddenly render the stream toxic for many forms of life.







Fig. 12 DISSOLVED OXYGEN SAG CURVE FOR THE SILK STREAM, LONDON (AFTER ELLIS, 1976)



Winter road salting can result in other deleterious effects upon water bodies. In the U.S.A., salt concentrations in lakes and waterways have been observed to increase five-fold over the last 20 years and to show a strong seasonal dependency (Field et al, 1974). Runoff of salt into Irondequo t Bay, Rocester, U.S.A. (Bubeck et al, 1971) created a density stratification of chlorides in the lake that was sufficient to prevent complete vertical mixing of the bay during the spring and therefore, the transfer of oxygen to the bottom sediments. The same effect was reported for a small lake receiving urban runoff at Ann Arbor, U.S.A. (Judd, 1970). The storage of salt for deicing has been reported as resulting in the contamination of ground water drinking supplies to levels well in excess of the recommended limits (Stevens, 1973; Hawkins, 1976). In Connecticut, U.S.A., tastes and odours in water supplies were ascribed to contamination from chloride and sodium ferrocyanide from a nearby salt storage area (Scheidt, 1967).

Nutrients in storm runoff may present a further threat to lake and river water quality, namely eutrophication. Kluesener and Lee (1974) monitored the nutrient loading from a separate storm sewer in Madison, Wisconsin. Their results indicated that approximately 80% of the total nitrogen influent to Lake Wingra arose from urban runoff. Phosphorus was considered to be primarily derived from accumulated litter and car exhaust, whilst rainfall was found to be the major source of inorganic nitrogen in runoff. The availability of the nutrients nitrogen and phosphorus to algae was investigated by Cowen and Lee (1976) and Cowen, Sirisinha and Lee (1976). Bacterial activity was found to increase the algal available nitrogen by approximately ten-fold in runoff. However, physico chemical processes were considered to be more important in the release of particulate inorganic phosphorus to solution. Ellis (1976) monitored a large build up of nutrients in the Silk Stream with the consequent growth of algae and waterblooms during low flow summer periods. Break up of these growths by autumn rains, followed by their settlement and decomposition with the release of mineral and organic nutrients, was seen as a secondary source of stream pollution and a means by which eutrophication was enhanced at the commencement of the next growing season.

## 1.2.6 Control of Pollution from Urban Runoff

The previous sections have illustrated that non-point pollution, in particular runoff from urban areas, has become recognised as a major contributor to water quality degradation. Previous efforts to improve

river water quality have concentrated upon the treatment of point sources of pollution, such as sewage and industrial effluents. The reduction of pollution from these sources through the introduction of improved treatment methods has revealed the need to control pollution from non-point sources in order that current river quality objectives may be achieved. The approaches available to control pollution from urban runoff fall into three principal categories:-

a) Control at source

b) Control within the sewer system

c) Treatment at the sewer outfall.

It is beyond the scope of this project to conduct a detailed review of all the literature available on this broad subject area but to indicate, within each category, the more important alternatives that are available. A review has been published by Field and Lager (1975).

1.2.6.1 Source Control

The theory behind source control of pollution is the limitation of the supply of contaminants and their accumulation on urban surfaces. Examples of this type of approach include municipal street sweeping, and gully emptying, restrictions on the use of chemicals (deicing compounds, pesticides, etc.) and public environmental education. The efficiency of current street sweeping methods has been questioned by Sartor and Boyd They found the removal of the more polluting dust and dirt (1974). fraction to be typically only 50%, whereas the removal of litter and other debris was 95% efficient. The effort required to achieve a greater removal effectiveness of this fraction was considered to be several times the effort normally expended. The American Public Works Association (1969) recommended that greater effort be devoted to public education programs to stem the careless scattering of litter and other polluting practices by the public. They proposed not only regulatory measures and anti-littering campaigns, but also government action in the form of purchase and placement of convenient litter containers with prompt collection of their contents.

### 1.2.6.2 Sewer System Controls

Examples of this type of control include sewer flushing, the use of regulator/concentrator devices, inflow/infiltration control and in-system storage.

When the possible importance of solid material deposited in sewers in influencing the quality of the first flush is considered it is surprising that little research has been conducted on sewer flushing as an effective means of removing these solids. An alternative approach has been conceived in the use of regulator/concentrator devices, such as the swirl regulator (Field, 1974; American Public Works Association, 1972), for the effective separation of solids over wide ranges of storm flow. The separated solids can then be diverted to a convenient combined system, when this is not feasible the ultimate disposal of the solids could present problems. The Water Research Centre (1977) have tentatively proposed the use of porous gully pots, these would serve the dual purpose of eliminating the storage of anoxic polluting waters and providing a large volume for the storing of the first flush of runoff. The use of porous pavements as a means of reducing the volume and rapidity of runoff is another possibility that has been investigated (Field, 1974).

## 1.2.6.3 Treatment of Storm Runoff

Many of the conventional treatment processes currently employed for the treatment of sewage may be applied to storm runoff discharges. However, two principal criteria must be satisfied, firstly, the process must be amenable to the rapid and variable flow rates associated with storm runoff and, secondly, for economic reasons treatment should be Since the most significant pollutant in urban runoff is simple. suspended solids and many other pollutants are also solid related, the removal of solids is of prime importance. Simple, physical treatment through sedimentation satisfies all these criteria and thus lends itself to the treatment of storm runoff and has been investigated by several workers. Weibel et al (1966) looked at the settling characteristics of storm runoff samples and found that settling periods of one hour or more were necessary for the removal of over 50% of the suspended solids. Shorter settlement times of 10 or 20 minutes did not remove suspended solids and BOD effectively. However, Cordery (1977) performing similar jar tests found a rapid sedimentation of suspended solids with

80% removal being effected in the first 4 minutes. Wanielista (1978) investigated the use of a percolation pond for capturing and treating the first flush of storm runoff. The pollutant load was substantially reduced over the year, however, percolation resulted in some groundwater contamination. Nightingale (1975) investigating the use of urban storm runoff detention basins found levels of heavy metal contamination of soil in the basin that could present health problems.

## 1.2.7. Modelling and Predicting the Quality of Stormwater Runoff

Methods for predicting the quality of stormwater runoff have arisen as tools for objective guidance in decision making for water resources management. Through prediction of pollutant loads from nonpoint sources, these models allow comparison with other sources of pollution in order that the best, cost-effective decision for the reduction of river pollution can be reached. Such models may also be used for assessment of the effectiveness of different forms of treatment and control of urban runoff, the relative merits of combined or separate systems for a given circumstance and for the day-to-day, year-to-year prediction of pollutant loading from urban areas. The accuracy and reliability of such models is hindered by the great variability and complexity of the contributing processes and the shortage of good quality data for model calibration. Ideally any process or factor that influences the accumulation and generation of pollutant loads on urban surfaces and their subsequent transport to a receiving water should be taken into account when modelling. Inevitably a large number of factors may be significant in defining the pollutant loading rate including land usage, catchment characteristics, meteorological characteristics, municipal street cleaning practices, etc. Consequently, models of widely differing complexity have been conceived, ranging from simple linear regression to highly complex deterministic models. Litwin and Donigian (1978) have catalogued the types of model available under four headings:-

#### a) Non-Analytical methods

The non-analytical approach, typically, utilised local data and land use inventories to formulate pollutant load/runoff rates for various pollutants to the areas of various land uses, e.g. Singh (1975). When calibrated from good local data the method can yield satisfactory results, however, when there is no data available tables for similar land used must be applied and the technique cannot be considered as reliable.
## b) Statistical Techniques

Linear, multiple linear and non-linear regression have all been employed to relate pollutant loadings to selected variables such as meteorological characteristics, topographical characteristics, etc. The resultant equations have often been applied with some success to single catchments from which the data for the regression has been gathered, e.g. Bedient et al (1978). However, the derivation of a generalised equation for application to any catchment has met with little success (Bradford, 1977). Statistical models fall short of the requirements since they do not account directly for the processes involved. The models represent average conditions and characteristics which are not applicable to the wide variability of situations that can occur.

## c) Simulation Models

Simulation models were originally conceived for the prediction of stormwater flows for the more efficient design of sewer systems. Such models were based on mathematical algorithms representing individual hydrological and hydraulic processes which were then linked together in a logical fashion to produce the desired output. Subsequently these models were extended to include expressions for pollutant accumulation and washoff processes, the hydrological and hydraulic portions of the model being used as 'vehicles' for calculating pollutant transport, loads and stormwater quality. A wide variety of design and simulation methods are available for evaluating stormwater flows and it is not proposed to mention them here. The state of the art has been comprehensively reviewed by Colyer and Pethick (1976) and Brandstetter (1976). Of the simulation models that can also be used for the prediction of stormwater quality, two have been most widely employed, the Storm Water Managment Model (SWMM) Metcalf and Eddy Inc. 1971) and the Storage Treatment and Overflow Model (STORM), (U.S. Army Corps of Engineers, 1975). Other models have also been proposed, notably the Hydrocomp Simulation Program (HSP) (Hydrocomp International Inc., 1975) and two models for the prediction of suspended solids (Price and Mance, 1978; Tucker and Mortimer, 1978). Simulation models fall into two main categories, models that deal with single storm events, of which SWMM is an example and models that perform continuous simulation of which STORM is an example.

The methodology available for the simulation of pollutant accumulation and removal is still in its infancy and this is reflected by the empirical nature of the mathematical expressions that are utilised. In both SWMM and STORM the expressions employed are nearly identical, for the calculation of the accumulation of dust and dirt between storm events this is:-

P<sub>1</sub> = DDFACT x QFACT x ADD P<sub>1</sub> = mass of pollutant DDFACT = dust and dirt accumulation rate QFACT = fraction of dust and dirt that is pollutant ADD = number of antecedent dry days.

The daily accumulation rate, DDFACT, is variable and is a function of land use, street cleaning activities and the season of the year. Having accumulated the pollutants they are then washed off by simulated runoff using an expression first proposed by Sartor and Boyd (1974):-

 $P_2 = M (1 - e^{-KR\Delta t})$ 

P2	=	mass of pollutant washed off in time increment ${\tt \Delta} t$
М	8	mass of pollutant available for washoff
К	=	decay coefficient
R	=	runoff rate

The parameters DDFACT, QFACT and K are all optimised using local data, however, when no such data is available the program reverts to preset default values. The use of the uncalibrated version of the model has been shown to result in gross errors in pollutant prediction (Jewell, 1978). Jewell has questioned both the pollutant accumulation and washoff functions, calibration for a catchment in Greenfield, Massachusetts, indicated that pollutant accumulation rates were not linear as the model purports but were in fact non-linear and reached a maximum value after about 10 days. In spite of these deficiencies both models have now seen several applications, Rimer and Reynolds (1978) used SWMM for seven different land use types and Hajas et al (1978) and Holbrook et al (1976) employed STORM for the prediction

of pollutant loads for large catchments comprising a variety of land use types. All authors indicated the need to calibrate the models with good quality local data.

Both SWMM and STORM use, what have been termed 'black box' or 'lumped parameter' expressions for the washoff of pollutants, that is, the functions utilised do not mathematically describe the actual processes involved in soil erosion and pollutant transport. A more deterministic approach to the problem has been presented by Price and Mance (1978). They derived mathematical expressions for the removal of particulate material from impermeable surfaces where the rate of removal was considered to be dependent upon two factors: firstly, the disturbance of particles by the impact of raindrops and, secondly, the entrainment of particles by the flow of water over the surface. In the first instance, removal was considered to be directly proportional to some power of the rainfall intensity and in the second instance removal was proportional to the excess of the shear stress generated by the flowing water over some critical stress. Due to the lack of laboratory studies on these phenomena a number of parameters had to be calibrated from field data. The model yielded promising results for the one residential catchment studied but, to date, has received no further application.

It can be seen that the chief weakness of the simulation models mentioned is the complex data requirements and the need for careful calibration to achieve reliable results.

## d) Intermediate Methods

These methods provide a compromise between the simplicity of the non-analytical techniques and the complexity of the linked process simulation models. The models are based, as before, on empirical mathematical representation of pollutant accumulation and washoff and the simpler versions commonly employ a design hydrograph as a means of representing the surface runoff (Singh, 1975). One weakness of the simpler models of this nature is that the continuous dynamic nature of the polluting processes is ignored. This situation has been rectified by the introduction of continuous non-point pollution simulation models that can provide for daily,

monthly and yearly statistical summaries of polluting loading data in addition to single event simulation. Examples of this type of model include the Nonpoint Source model (NPS) (Litwin and Donigian, 1978), the Management of Urban Nonpoint Pollution model (MUNP) (Sutherland and McCuen, 1978) and the Quality/Quantity Simulation (QQS) (Geiger, 1975). The NPS model computes sediment accumulation rates for different land uses on a daily basis and rainfall events are simulated in 15 minute loading rates of other pollutants are defined by what have been termed "pollutant potency factors", in reality this means that the pollutant load is considered to be a fixed percentage of the solids load. The MUNP model similarly uses pollutant potency factors to estimate the accumulation and removal of eight pollutants on urban streets. The model is capable of reflecting variation in a number of factors including the physical and chemical characteristics of accumulated pollutants, land use characteristics, rainfall characteristics, street sweeper characteristics, roadway characteristics and traffic conditions. The model employs data at hourly intervals to continuously simulate the accumulation and removal of sediment. Removal is calculated over 6 particle size ranges using the modified Yalin equation where sediment motion is effected when the lift force of flow exceeds a critical lift force. The QQS model calculated runoff and its pollution from different land use types using the unit hydrograph method modified for the calculation of water quality. The program provides for both statistical analysis at monthly and yearly intervals as well as single event simulation over 5 minute increments.

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## 1.3. PROJECT INTRODUCTION

1.3.1. Aims

Previous work at Trent Polytechnic (Tucker, 1975) and other studies (see Section 1.2.4.6.) had indicated that roadside gully pots acted as a reservoir for poor quality water that was rapidly flushed out in the early phases of a storm event. Several researchers concluded that this stored water was highly significant in influencing the quality of stormwater runoff. However, a detailed examination of the quality of water stored in gully pots and the role of gully pots in removing sediment and concentrating polluting matter had not formed the core of any major study.

Due to the sparsity of information on this important aspect of urban stormwater pollution the broad emphasis in this project was on classifying the pollutants in gully pot liquors and their significance in determining the quality of urban stormwater runoff. To achieve these ends the following approach was adopted:-

- Assessment of the quality of stored liquors over a period of one year for selected gully pots to establish the influence of seasonal effects, the importance of different sources of pollution and inter-relationships between pollution parameters.
- b) Assessment of the changes in quality of gully pot liquors occurring over dry periods and during storm events.
- c) Assessment of the variance of stored water quality across the catchment for any one day, in order to test the relevance of small sets of samples in representing the mean quality of the stored water.
- Assessment of the quality of stormwater runoff and the pollutant loads discharged from the whole catchment in order that comparisons with gully pot liquor quality may be drawn.
- e) Mathematical modelling of the removal of polluting material from gully pots under runoff conditions.
- f) Integration of the gully pot pollutant removal model into a rainfall/runoff model for the whole catchment in order that the quantity of stored water and the mass of pollutants contributed from gully pots at the sewer system outfall can be predicted.

## 1.3.2. Selection of a Catchment

The selection of a suitable catchment for study was dictated by two factors: firstly, the aims of the project and, secondly, practical considerations of monitoring and installing instrumentation. Apart from the obvious criterion that the catchment should have a separate sewer system, there is a choice of four principal land use types, that is, urban, industrial, commercial and rural. The quality of runoff at industrial locations is likely to be strongly influenced by the specific nature of





Fig.14 MAP OF CLIFTON GROVE ESTATE, SHOWING THE STORM SEWER SYSTEM, GULLY POT SAMPLING POINTS AND STORM RUNOFF SAMPLING POINT

TABLE 16 Characteristics of the Clifton Grove Catchment.

•

Roofed Area	1.78 hectares
Roads and Footpaths	1.93 hectares
Impervious Area	3.71 hectares
Pervious Area	6.9 hectares
Total Area	10.61 hectares
Average catchment slope	4% - 5%
No. of houses	265
Housing density	25 houses per hectare
Population density	75 - 100 persons per hectare
Traffic density	Low

1.20

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the industries located on the catchment. Hence, results obtained would only be of limited application to other industrial catchments in the U.K. Commercial locations would present problems with the installation of equipment and, due to their age, are rarely served by separate sewer systems. Rural locations are usually less prone to pollution and are, therefore, of less interest. The majority of catchments served by separate systems in the U.K. are residential urban developments built within the last 10 to 20 years. Such catchments are often of similar construction, polluting activities and sources are of a similar nature and, therefore, results obtained would be more applicable to other urban developments in the U.K. on similar soil and environment.

Runoff from two residential catchments in the Nottingham region had previously been monitored for suspended solids concentrations (Tucker, 1975). In order that results could be compared it was desirous that a new catchment be found. Only the Clifton Grove estate was found to fit all the criteria mentioned and it was therefore adopted as the study site.

Clifton Grove is a small suburban housing estate covering an area of some 10.6 hectares of which 3.7 hectares are impervious roof and road surfaces. The estate is comprised entirely of modern "detached" private housing being built between 1973 and 1976. The development is arranged with areas of grassed open land, designated "amenity areas", separating areas of housing arranged around culs-de-sac. In accordance with modern practice the estate is sewered on a separate system with stormwater runoff draining directly into the R. Trent (Figure 14). The principal characteristics of the catchment are listed in Table 16.

## 1.3.3. Selection of Gully Pots for Prolonged Study

The following subcatchment characteristics were taken into consideration in the selection of gully pots:-

a) Impervious area.

- b) Subcatchment slope.
- c) Number of houses in the subcatchment, possibly of importance in defining the extent of human activity within the subcatchment.

TABLE 17 Characteristics of gully pots selected for study.

1					
Situation	Susidiary Road.	Cul-de-sac Feeder Road	Cul-de-sac		
Open Land	No	Yes Possible	Little		
No. of Houses	0	tr 5°2	N		
Imp. Area	140 m <sup>2</sup>	680 <sup>н -</sup> 513 <sup>н 2</sup>	270 m <sup>2</sup>		
Slope	ti ti 0 •	•038 •040	.047		
Gully Pot	F	. 2	ω		

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## Fig.15 SUBCATCHMENT MAP - GULLY POT 6

# Scale 1/500



Impervious area contributing





Fig. 16 SUBCATCHMENT MAP - GULLY POT 1



Fig.17 SUBCATCHMENT MAP - GULLY POT 8





- d) Presence of open land which might contribute to the runoff entering the pot.
- e) Situation on the estate, e.g. cul-de-sac, main feeder road.

The Clifton Grove sewer system contains 108 gully pots. Each pot normally holds about 95 litres of water, resulting in a total volume of stored water of some 10,200 litres of water, there being no other storage of water in the system. Four gully pots were selected for detailed examination of pollution parameters over an extended period according to the physical criteria listed above, their subcatchment characteristics are listed in Table 17. Figures 15, 16, 17 and 18 are catchment maps for each of the gully pots. These pots were selected to represent the diverse situations that could be encountered on the estate: the occasional sampling of other gully pots was based on random selection using random number tables.

## 1.3.4. Selection of Pollution Parameters for Analysis

The main parameters of importance in characterising the polluting nature of a water include the following:-

- a) Solids content:- settleable solids, suspended solids (SS), volatile suspended solids (VSS), dissolved solids (DS).
- b) Oxygen demand:- biochemical oxygen demand (BOD), chemical oxygen demand (COD),total organic carbon (TOC).
- c) Dissolved oxygen (DO)
- d) pH

e) Nutrients:- nitrogen species (ammonia, nitrite, nitrate) phosphorus species (phosphate).

f) Anions:- chloride, sulphate, etc.

g) Heavy metals:- lead, zinc, cadmium, mercury

h) Toxic organics:- herbicides, pesticides, etc.

i) 0il and other hydrocarbons.

The relative importance of these parameters in characterising the polluting nature of a water is dependent upon the origin of the water and the polluting processes with which it comes in contact. Selection of parameters was a compromise between aiming for as wide a coverage as possible and the limited time available for analytical procedures. The parameters that were analysed on a regular basis were SS, VSS, DS, BOD, COD, DO, pH, ammonium, nitrate, chloride, sodium and calcium. The heavy metals lead, zinc, copper and cadmium were analysed less regularly. The presence of oil and detergents was noted but quantitative determination was not undertaken.

## 2. INSTRUMENTATION, SAMPLING AND DATA COLLECTION

For a project of this nature the data requirements and instrumentation needed for its collection are necessarily diverse and can be seen primarily as four fold:-

a) Water quality sampling from gully pots and sewers

b) Flow determination in sewers

- c) Meteorological measurements
- d) Other pollutant input and output measurements, i.e.
   rainfall quality, atmospheric dustfall, council cleaning practices, human/animal behaviour, factory emissions, etc.

## 2.1. WATER QUALITY SAMPLING

Sampling is a frequently disregarded aspect of water quality determinations, whereas errors in analytical methods can be reported accurately, the nature and relevance of errors from sampling techniques is rarely considered. Two principal criteria should be applied in sampling waters. Firstly, the samples should be collected in such a manner as to be representative of the mean quality of the water body under consideration and secondly, sampling should be reproducible. Both criteria are hard to satisfy and difficult to verify. The procedure adopted in this research was to perform sampling in a standardised manner and in that way results within the project were directly comparable.

## 2.1.1. Sampling from Gully Pots

Sampling of gully pot waters was performed using a purpose built 'S' stick (Figure 19). This arrangement allowed bottles of either 1 litre or 500 ml capacity to be immersed to a known depth, determined by a mark on the support rod and for samples to be taken without preferential inclusion of surface film. The procedure ensured that the samples collected from different gully pots or the same gully pot over time were similarly obtained and, therefore, comparable.

#### 2.1.2 Sampling from the Sewer

Sampling was achieved using a Rock and Taylor multipurpose automatic sampling machine (Figure 20), which could receive up to a maximum of 48 samples in 500 ml plastic bottles. The machine was housed on a specially constructed wooden platform in the final manhole before the sewer outfall and was powered by a 12 volt battery.

Sampling was initiated by a Flygt ENHIO mercury float switch when flow rate in the sewer rose to approximately 8 litres/second. Pumping was performed by a twin channel peristaltic pump with a 3-stage adjustment for varying sample line velocities, i.e. sample volume. The cycle for each sample was comprised of forward pumping for sample collection and reverse pumping for emptying the inlet tube prior to the next sample. The duration of both pumping modes was controlled by adjustable timers. With the peristaltic pump adjusted to maximum pumping rate a line volocity greater than 50 mm/second was recorded which was sufficient to gain a representative suspended solids sample (Wood and Stanbridge, 1968).

The sampling arrangement on the bed of the sewer is illustrated in Figure 21. A metal 'T' piece was employed to secure the sampling tube and protect it from buffeting by objects washed down the sewer. The 'T' piece was constructed from 90° angled iron with the angle being pointed into the flow to minimise disturbance to the hydraulic regime. However, for low flow situations a minor 'stilling' effect could be observed at the sampling orifice. The orifice itself was 15 mm above the bed of the sewer and was protected from blocking by leaves or cellophane with a wire guard. Colston (1975) has reported a vertical variation in suspended solids dispersion in storm sewers, however, for this study the position of the sampling orifice was constant for all storms. This was believed to result in minimal errors since, for many of the low intensity storms monitored at Clifton Grove, the depth of flow at the sampling point rarely exceeded 200 mm, for such low depths of flow vertical variation was considered to be insignificant. The sampling arrangement described above was constant throughout the period of monitoring and was considered to represent the best compromise between effectiveness of operation and reproducibility of sampling.



LAYOUT OF THE SAMPLING MACHINE IN A MANHOLE Fig.20



Fig. 19

a) top view







## 2.2. FLOW DETERMINATION

Flow in sewers is conventionally determined either by the use of a calibrated section with a known depth to discharge relationship or by chemical tracer dilution (dilution gauging). The latter method had been employed in previous studies at Trent Polytechnic and, because of the availability of instrumentation was selected for this study.

Dilution gauging is normally performed by one of two methods:-

a) Constant rate tracer injection

b) Gulp tracer injection

Both methods rely upon the same principle, a known mass of tracer in solution is added to the flow and its concentration is subsequently determined downstream after cross-sectional mixing is complete. In gulp injection a fixed volume of solution containing a known mass of tracer is introducted to the flow. Downstream the complete passage of this pulse of tracer is sampled continuously, the pipe flow is then calculated from the equation:-

$$Q = \left(\frac{M}{M - M}\right) \quad q_s$$

where

Q = flow
q<sub>s</sub> = flow sampled
M = mass of tracer added
M<sub>s</sub> = mass of tracer sampled
M<sub>o</sub> = mass attributable to background levels.

This technique was used by Tucker (1975) forstudies on two residential estates in Nottingham, but was found to be unfeasible for the Clifton Grove catchment due to the low pipe gradient (1 in 300) in the dilution reach. Complete sampling of the tracer pulse in this situation would require inordinately long sampling times and result in unrepresentative sampling of suspended solids due to the low sampling line velocities required. Additionally, the long sampling times would result in poor resolution of the runoff hydrograph.

The constant rate injection method does not suffer from these limitations, is considered to be superior and was adopted in this study. A tracer solution of known concentration  $(C_1)$  is injected into the flow (Q) using a constant head apparatus to achieve a constant rate of dosage (q). A concentration of the tracer  $(C_2)$  will result downstream and, assuming background tracer levels to be negligible, then:-

$$Q = \frac{C_1}{C_2} q$$

This equation is only truly valid for steady state conditions, if flow rates are rapidly varying then a systematic error will be caused by the change in discharge with time. Gilman (1975,1977) has proposed a residence-time model which can be used to estimate errors occurring in dilution gauging throughout a storm hydrograph. The model relies upon the determination of a residence-time distribution from the response noted for an instantaneous input of tracer and the manner in which this distribution varies with flow rate. Similarly Price (1976) has proposed a diffusion model based on dispersion theory which can also be used to estimate errors in discharge calculations. Both models rely on the changes in flow rate being smooth and 'slowly varying'. The models have been utilised by Harvey et al (1977) in the assessment of prototype dilution gauging instrumentation, an example of their findings are shown in Figure 22, where errors of up to 5% are indicated.

## 2.2.1 Practical Criteria of Dilution Gauging

Before dilution gauging can be successfully implemented certain practical criteria must also be satisfied.

Between the point of tracer injection and the sampling point downstream, it is necessary for the tracer to undergo complete crosssectional mixing. Fortunately, the turbulent nature of flow in storm sewers provides very good mixing. In foul sewers an elementary rule has been established that the length of pipe needed to achieve crosssectional mixing should be 100 times its diameter. For the Clifton Grove catchment for the reach of sewer under consideration, the pipe diameter was 0.45 metre and its length was 100 metres, clearly satisfying this condition.

Fig.22DILUTION GAUGING RESULTS (AFTER HARVEY ET AL, 1977) ILLUSTRATING THE THEORETICAL ERROR IN DISCHARGE MEASUREMENT ESTIMATED BY THE DIFFUSION MODEL (PRICE, 1976) AND THE RESIDENCE MODEL (GILMAN, 1976)



There should be no additional input or abstraction from the flow between the injection and sampling points. To achieve this situation it was found necessary to divert a roof runoff input pipe at the sampling manhole to a point one metre downstream of the sampling position, where it was considered to satisfactorily bypass the sampling head and have no influence upon the sampling regime.

There should be no occurrence of still water at the sampling point as this will result in erroneous flow measurements. This problem arose on a few occasions when the River Trent in flood conditions 'backed up' from the sewer outfall and no flow measurements were possible.

## 2.2.2 Application of Dilution Gauging

#### 2.2.2.1 Instrumentation

The instrumentation developed at Trent Polytechnic for constant rate tracer injection is illustrated diagrammatically in Figure 23; the general layout of the apparatus in the sewer system is shown in Figure 24. Dosing of tracer solution was initiated by adaptation of the Rock and Taylor sampling machine to send an impulse upstream by cable at the same instant that the float switch activated sampling. The impulse was used to trigger a relay which brought into operation a peristaltic pump. The pump, powered from a separate 12 volt battery, transferred tracer solution from a 25 litre container to a constant head vessel which discharged the solution through a calibrated glass capillary tube into the flow. The capillary tube was calibrated in the laboratory to a dose rate of 1.025 ml/second (averaged over 50 observations). The injection rate allowed up to 31 hours dosing and was consistent with the amount of sampling time available on the sampling machine. Dosing rates were verified regularly during routine maintenance in the field when the capillary tube was flushed clean with distilled water and, whenever possible, prior to storm events.

The doser required approximately  $6\frac{1}{2}$  minutes before an equilibrium dose rate was achieved (Figure 25). This delay was created by the time needed for the constant head to reach equilibrium behind the discharge capillary tube. As a result of this delay it was necessary to apply a





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correction factor for the determination of flow in the second and third samples (Section 4.2.1.). Theoretically the problem could have been alleviated by the use of a two way valve located in the discharge line. However, preliminary investigations with such a valve were not successful. Some leakage occurred continuously when the machine was not in operation and inconsistent operation resulted in variations in dose rate and consequently unreliable flow determination.

## 2.2.2.2 Selection of Tracer

For a chemical tracer to be suitable for the determination of flow in storm sewers it must satisfy three criteria. Firstly, its background concentration in the flow must be negligible, secondly, it should not be significantly adsorbed from solution onto particulate matter and, lastly, it is preferable that the tracer is environmentally safe. Lithium fulfills all these requirements and has been successfully employed for several years, (Blakey, 1969; Gizzard and Harms, 1974). Neal and Jordan (1978), examining the use of iodide and lithium as tracers, concluded that whilst lithium was satisfactory, iodide had considerable limitations because of significant adsorption onto particulate matter. Lithium, as the chloride salt, had been used previously at Trent Polytechnic (Tucker, 1975) and was employed in this work.

## 2.2.2.3 Tracer Preparation

A suitable concentration for the lithium chloride tracer solution was determined by consideration of the postulated maximum flow rate and the detection limits of the analytical technique used. Using the EEL 240 atomic adsorption spectrometer lithium values in the range 0.01 to 8.0 mg/l can be readily determined. With an assumed maximum flow rate of 300 litres/second, a desired lithium concentration in the flow rate of 0.1 mg/litre and a doser injection rate of 1.0 ml/second, then from the equation:-

Concentration in tracer = 
$$\frac{C_2 Q}{q}$$
  
=  $\frac{0.1 \times 300}{1}$  mg/ml  
= 30 g/l

In practice a tracer concentration of 40 - 45 g/l lithium was used and with measured flow rates rarely exceeding 100 l/s, the concentration of lithium in the flow (circa 0.4 mg/l) was sufficient to make background concentrations insignificant.

3.5 kg of lithium chloride were used to prepare 15 1 of tracer in the laboratory. The solution was then transported in bulk and emptied into the container in the field, care being taken to ensure that no spillages occurred that would subsequently adversely affect flow determination.

## 2.2.2.4 Sampling Duration

In gulp injection dilution gauging the sampling duration is fixed by the time necessary to sample the whole of the tracer pulse under minimum flow conditions. With constant rate injection no such constraints on sampling times exist. Sampling times were minimised as much as possible in order that:-

- a) the velocity in the sampling line was sufficient to gain a representative sample of suspended solids;
- a close approximation to an instantaneous flow measurement was made; and

c) a better resolution of the storm hydrograph was achieved.

In practice the minimum sampling time was limited by the need to obtain a sample volume large enough for analysis. This was in turn dependent upon the maximum pumping capacity of the peristaltic pump. For the majority of events a sampling interval of 3 minutes was selected, this included a sampling period of 2 minutes and one minute of reverse pumping between samples. However, the pumping times were not accurate and these settings resulted in a mean sampling period of 2.06 minutes and an interval of 3.04 minutes between samples, with an average volume of 400 ml collected. Sampling times for the first pumping cycle were found to be slightly longer than the subsequent sampling times above and this had to be taken into account in calculations. A sampling interval of 3.37 minutes with a sampling period of 2.39 minutes were recorded for the first cycle.

## 2.2.2.5 Timing of Events

The commencement of sampling was recorded through the further adaptation of the Rock and Taylor sampling machine to switch off a purpose built 12V. quartz clock. The clock, powered by the 12V. battery, was set to Standard Time when the apparatus was commissioned in preparation for a storm event. When the float switch initiated sampling an impulse from the machine was used to operate a solenoid and so break the clock circuit. The clock was found to be accurate to a few seconds over several days.

## 2.3. METEOROLOGICAL MEASUREMENTS

Rainfall intensities for individual storm events were measured by an 8" Rimco tipping bucket raingauge. The gauge has an operating range of zero to 400 mm of rainfall per hour with an accuracy of  $\pm$  1% with 0.5 mm per bucket tip. The raingauge was calibrated in the laboratory using measured flow rates and recording the number of tips per second to obtain the true rainfall volume per bucket tip, this was found to be 0.432 mm. The sensitivity of the gauge was increased for use in the field to 0.113 mm per bucket tip by the use of four funnels to increase the collecting surface.

The accurate timing of storm events and raingauge tips was achieved using a battery operated data logging system (Microdata Ltd.). The data being recorded on standard tape cassettes which were subsequently translated at the Institute of Hydrology. Tapes and batteries were normally replaced at 3 to 4 day intervals. The raingauge was positioned on the flat roof of Clifton College, approximately 10 metres above ground level and 100 metres from the Clifton Grove catchment. Although not fulfilling all the standard conditions for siting raingauges, the placement was advantageous in that no buildings or trees overshadowed the site and vandalism of the equipment was avoided.

Daily meteorological data, including total rainfall, maximum and minimum temperatures, wind direction and speed, were available from Meteorological Office raingauge 116959 sited at Beeston sewage works approximately 1 mile to the north west of the catchment. If this station was inoperative data was also available from stations at Nottingham University or Nottingham Castle. Additionally, hourly weather data could be obtained from the regional weather centre at Watnall, approximately

7 miles to the north of the catchment.

#### 2.4. OTHER MEASUREMENTS

A large variety of other measurements for classifying and quantifying pollutant inputs and outputs are possible. Some important aspects of the urban pollutant generation system such as human and animal behaviour defy quantification. Although door to door surveys could yield useful information on car washing frequency, application of garden fertilisers and dog ownership, etc., a study of this kind was deemed to be outside the scope of the project and visual observations were considered to be sufficient.

Information on land use, sewerage systems and council cleaning practices was available, courtesy of Nottingham City Council. Although the dates of council gully emptying on the catchment were available, it was found impractical to estimate the actual mass of material removed.

Atmospheric dustfall measurements were made using three large plastic funnels which were exposed to the atmosphere for periods of one, two and three weeks. The particulate matter and rainfall falling on the funnels was collected in a litre flask, the mass of solids collected was then determined by filtration, care being taken to ensure that all solids adhering to the funnel were washed into the flask prior to analysis. The apparatus was positioned on Clifton College roof. Measurements were hampered by flies and insects that found their way into the containers, and had to be removed during filtation. Consequently, it was found necessary to ignore some 50% of the samples and only reliable data has been presented.

Rainfall for quality determinations was collected as runoff from a greenhouse roof, located at Clifton College. Difficulties were again encountered with flies and insects being present in the sample. However, lichen washed off wooden greenhouse supports, sediments accumulating in gutters and bird droppings presented additional more serious problems. These factors may have resulted in some high values of COD and ammonia (similar observations have been reported by Mance (1977)).

## 3. CHEMICAL ANALYSIS

With the generation of large numbers of samples for analysis of most importance in the selection of analytical techniques, after accuracy, was simplicity and rapidity. To this end it was necessary to employ methods that involved the minimum of sample preparation in preference to techniques that, although more accurate, were also more time consuming. As an example, selective ion electrodes were used in the analysis of nitrate and ammonia in place of the lengthier colorimetric procedures that are normally recommended. This approach may have been beneficial since sample deterioration and chemical changes induced in natural waters by, for example, bacterial activity makes it essential that samples are analysed within 24 hours of collection. For analyses where degradation was not a problem, samples were stored in a fridge at  $4^{\circ}$ C with the addition of relevant preserving agents when necessary, e.g. concentrated nitric acid for heavy metals.

Of prime importance in all analytical work is the thorough cleaning of glassware prior to use. A routine of scrubbing, soaking in dilute hydrochloric acid and rinsing several times with distilled water was rigidly followed to avoid sample contamination. For the analysis of heavy metals stricter routines were employed involving the soaking of glassware in concentrated nitric acid, washing with Decon 90 and rinsing with distilled water.

All the chemical analytical methods employed routinely in the research are outlined in this chapter. Theoretical considerations are included where this is applicable in explaining the choice of technique and the problems encountered. The performance of the methods and the specific problems of relevance in the analysis of stormwater samples are discussed.

As a check on the precision of the methods employed a four litre sample, separated into eight equal portions, was analysed for each of the parameters in turn. The mean value, standard deviation and coefficient of variation were calculated and are presented in Table 18.

### 3.1. SOLIDS

The classification of solids in urban stormwater runoff is of particular importance in characterising the polluting nature of the waste. High levels of suspended solids have been reported in all urban runoff studies and have been shown to be related to high levels of several other pollutants, either by adsorption or by the nature of the solid itself, e.g. BOD, COD, heavy metals.

The methods used in the determination of solids are necessarily empirical, the constituents being largely defined by the procedures employed. For example, the recorded value of suspended solids is dependent upon the porosity of the filter paper, loss of volatile compounds on evaporation and the drying rate of oils. As a consequence of these errors 'Standard Methods' (American Public Health Association, 1976) recommends the use of the terms non-filterable residue for dissolved solids and filterable residue for suspended solids. In this research the classical terminology has been retained owing to its common usage. Although the measure obtained may not be the definitive solids value, the use of standardised procedures ensures that results within this study are comparable and also comparable to other studies.

Analyses were performed routinely for suspended solids (SS) and dissolved solids (DS) and occasionally for volatile suspended solids (VSS) and turbidity. Originally turbidities were determined to assess the viability of a SS/turbidity relationship for the estimation of SS concentration. Turbidity being advantageous in that it is simply and rapidly determined. Although a fair correlation was found to exist, (Figure 26), the variance was too great for a useful estimation of SS to be possible. For example, at a turbidity of 30 f.t.u. (formazin turbidity units) the recorded SS concentrations lay between 50 mg/l and 150 mg/l. Nonetheless, turbidity was usefully employed as an aid to the interpolation of SS values between known concentrations in storm hydrographs.

The specific methods employed are outlined below:-

## 3.1.1 Suspended Solids and Volatile Suspended Solids

A glass fibre filter paper (Whatman GF/C) previously washed with distilled water was dried in an oven at  $105^{\circ}C$  for 1 hour. If volatile

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TABLE 18 Precision of Analysis





CORRELATION BETWEEN SUSPENDED SOLIDS AND TURBIDITY

matter was to be determined the paper was transferred to a muffle furnace and heated to  $550^{\circ}$ C for 15 minutes. The paper was allowed to cool to room temperature in a dessicator and weighed (W<sub>1</sub>).

100 mls of shaken sample were measured in a graduated cylinder and filtered through the paper using vacuum on a Hartley filter/Buchner flask apparatus. The cylinder was washed out with 3 x 10 mls of distilled water. The filter paper was now dried at  $105^{\circ}C$  for 1 hour, cooled to room temperature in a dessicator and weighed ( $W_2$ ). If volatiles were to be determined the paper was ignited in a muffle furnace for 15 mins at 550°C, transferred to a dessicator, cooled and weighed ( $W_2$ ).

> SS =  $\frac{(W_2 - W_1)}{V}$  x 1000 mg/l VSS =  $\frac{(W_2 - W_3)}{V}$  x 1000 mg/l

where V = Volume of sample

The coefficient of variation of 7.2% reported in this study agree favourably with a precision of  $\pm$  5.2 mg/l (33%) at 15 mg/l quoted in 'Standard Methods'.

## 3.1.2 Dissolved Solids

A porcelain evaporating dish was heated in an oven for 1 hour at  $105^{\circ}$ C, allowed to cool to room temperature in a dessicator and weighed (W<sub>1</sub>). 50 mls of filtered sample were added to the dish, evaporated in an oven at  $105^{\circ}$ C for 1 hour, cooled and weighed (W<sub>2</sub>).

$$DS = \frac{(W_2 - W_1)}{V} \times 1000 \text{ mg/l}$$

V = Volume of sample

Reported precisions of  $\pm$  5% compare to a coefficient of variation of 9.3% for this study. The discrepancy is ascribed to the lower sample volume used in determinations.

## 3.1.3 Turbidity

Measurements were made on a Hach Model 2100 Turbidimeter calibrated with standards of 10 f.t.u. (formazin turbidity units) and 100 f.t.u.

## 3.2. OXYGEN DEMAND

Oxygen demand tests were originally conceived as a means of quantitatively representing the polluting effect of a complex mixture of pollutants in waste waters, upon a receiving watercourse. Usually the tests make some assessment of the amount of oxygen required to oxidise, either biologically or chemically, the organic matter in a sample. As such, they give a measure of the presence of organic matter which is assumed to be a reasonable indication of the polluting capacity of the water. Many tests are available, the ones most commonly employed are:-

- a) Permanganate value (PV)
- b) Biochemical Oxygen Demand (BOD)
- c) Chemical Oxygen Demand (COD)
- d) Total Oxygen Demand (TOD)
- e) Total Organic Carbon (TOC)

BOD and COD are the parameters used most frequently and were deemed to be the most suitable determination for this research.

Although the BOD test has recently been criticised for its variability and unreliability it still remains widely used and is one of the more important pollution parameters in use today. The test measures the quantity of dissolved oxygen (DO) consumed by aerobic bacterial oxidation of a water sample under specified conditions. It therefore provides an assessment of the likely impact of an effluent on stream dissolved oxygen levels, bearing in mind that the conditions under which the test is made are very different to those in the streamway.

DO is determined either chemically by the Winkler titration or by the use of a dissolved oxygen sensing probe. The DO probe was used in this study and has the advantage of simplifying the procedure. The
difficulty in obtaining reproducible results is due principally to the biochemical nature of the test. Reproducibility is dependant upon the type of bacteria present, acclimatization time, the biodegradability of the sample and the presence of toxic substances. Relative standard deviations between 5% and 15% are normal (Water Research Centre, 1978), a value of 7.7% being recorded in this study.

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The test is made more precise by the measurement of the BOD attributable to carbonaceous material only. The degradation of carbonaceous matter normally precedes the nitrification of any ammonia in the sample. If this is not the case and a significant amount of ammonia is contained in the sample, greatly inflated BOD values can occur. Nitrification can be suppressed and the reproducibility consequently improved by the addition of allyl thiourea. The BOD test is normally performed over a period of 5 days, however, in this research an interval of 7 days was found to be more convenient, making the suppression of nitrification even more important.

The COD test uses an acid dichromate oxidising solution which ensures almost complete oxidation of the majority of organic matter present. The technique is faster and more quantitative than the BOD determination and is a better measure of the total organic load. In the absence of interferences it gives the value to which BOD would approach, if incubation was continued until all organic material had been degraded.

Interference can occur in the presence of chloride ions and when oxidation of amino-type compounds to nitrogen occurs. Chloride interference is minimised by the addition of mercuric sulphate, but when concentrations exceed 20,000 mg/l, as was evident in a few samples affected by winter road salting, interference can be excessive and precipitation with subsequent removal is the best solution. Silver sulphate must be employed as a catalyst to ensure complete oxidation of straight chain aliphatics and certain aromatic hydrocarbons, although this can inhibit the effectiveness of mercuric sulphate.

A standard method now exists and was used in this study. The method is applicable for COD values in the range 100 to 1000 mg/l, for samples containing less than 100 mg/l COD a similar procedure was followed but using reagents with revised concentrations. In later determinations(1978 onwards) use was made of the block digester sealed

tube technique (Hach Chemical Co.), based on work by Jurka and Carter (1975). Loss of volatiles is claimed to be overcome by the use of the sealed tube and the method has the advantage of being simpler and much more rapid for large numbers of samples. Guarded approval for the technique has been given by the Water Research Centre (1978) provided the usual 1 : 1 acid to dichromate reagent was used.

#### 3.2.1 Dissolved Oxygen

A Delta Model 2110 meter complete with a dissolved oxygen sensing probe and automatic stirrer was used for all DO determinations. The meter was air calibrated before use for both field and laboratory measurements.

# 3.2.2 B.O.D.

Reagents:-

Phosphate buffer solution:

8.5g potassium dihydrogen phosphate  $KH_2 PO_4$ 21.75g dipotassium hydrogen phosphate  $K_2 HPO_4$ 33.4g disodium hydrogen phosphate  $Na_2 HPO_4 7H_2O$ 1.7g ammonium chloride  $NH_4CL$ 

1 litre distilled water

Magnesium sulphate solution22.5g $MgSO_4$  $7H_2O/1itre$ Calcium chloride solution27.5g $CaC1_2/1itre$ Ferric chloride solution0.25g $FeC1_3$  $6H_2O/1itre$ Allyl thiourea0.5 g/1

#### Procedure:-

The dilution water was prepared by the addition of 1 ml of each of the above reagents to each litre of distilled water. 50 ml of sample were added to a graduated cylinder and made up to 250 ml with dilution water (a 1.5 dilution was found to be adequate for most determinations, if higher BOD's were anticipated a 1.25 dilution was also prepared). The sample and dilution water were mixed well and added to a BOD bottle. The initial dissolved oxygen ( $DO_1$ ) was determined using the DO probe and the

bottle was topped up with dilution water. Any remaining air bubbles were shaken out by placing the bottle in an ultrasonic bath for 5 minutes. The bottle was stoppered tightly avoiding entrainment of air and incubated for 7 days at  $20^{\circ}$ C. The same procedure was repeated for a blank of dilution water (DO<sub>1B</sub>). After incubation the final dissolved oxygen of the sample (DO<sub>2</sub>) and the blank (DO<sub>2B</sub>) were determined using the DO probe.

$$BOD = \frac{(DO_1 - DO_2) - (DO_{1B} - DO_{2B}) \times V_2 \text{ mg/l}}{V_1}$$

where  $V_1 =$  volume of sample

 $V_2$  = volume of BOD bottle

3.2.3. <u>C.O.D.</u>

3.2.3.1. C.O.D. Concentrations 100 to 1000 mg/1

Reagents:-

Potassium dichromate solution (0.042M) (PD) - 12.259 g/l  $K_2 Cr_2 O_7$ 

Silver sulphate solution - 22g  $Ag_2 SO_4$  in 1 litre of  $H_2SO_4$  (cone)

Ferrous ammonium sulphate (0.25M) - (FAS)

- 98g Fe  $(NH_4)_2$   $(SO_4)_2$ , 20 mls  $H_2SO_4$  in 1 litre of distilled water

Ferroin indicator Mercuric sulphate

Standardisation of ferrous ammonium sulphate:-

10 ml of potassium dichromate were added to a beaker and made up to 100 ml with distilled water. 30 ml of sulphuric acid were added and the solution was allowed to cool. The mixture was titrated against ferrous ammonium sulphate using ferroin as an indicator and the volume required to change the colour from violet to red was noted:-

 $T = \frac{Volume of PD}{Volume of FAS} \times 0.25$ 

Procedure:-

20 ml of sample, 0.4g mercuric sulphate, 5 ml of silver sulphate solution and several glass beads were added to a 250 ml reflux flask and mixed thoroughly. 10 ml of potassium dichromate and 25 ml of silver sulphate/sulphuric acid were introduced to the flask, the mixture was refluxed for 2 hours and then allowed to cool to room temperature. The mixture was diluted to 140 ml and titrated against ferrous ammonium sulphate using ferroin as an indicator. The operation was repeated for a blank of distilled water.

 $COD = \frac{8000 (V_2 - V_1) T}{V} mg.1$ 

 $V_1$  = Volume of FAS in titration  $V_2$  = Volume of blank in titration T = titre of FAS  $V_{\sigma}$  = Volume of sample

3.2.3.2. C.O.D. Concentrations 0 to 200 mg/1

The same procedure as for the determination of concentrated COD values was followed except a 10-fold dilution of potassium dichromate (0.0042M) was used for the reflux and titration was against a 5 fold dilution of ferrous ammonium sulphate (0.01M).

3.2.3.3. C.O.D. by Hach Procedure

Reagents:-

Dilute Reagent (0 - 200 mg/l C.O.D.) 1 part  $K_2 Cr_2 O_7$  (0.0084M, 2.452 g/l) 2 part conc.  $H_2SO_4$ 

Concentrated reagent (100 - 1000 mg/l COD)

1 part  $K_2 Cr_2 O_7$  (0.042 M, 12.26 g/l) 2 part conc.  $H_2SO_4$ 

Mixed Catalyst

1 part mercuric sulphate

1 part silver sulphate

Mixed and ground in a mortar and pestle.

Dilute ferrous ammonium sulphate (0.0025M, 0.98 g/l) Concentrated ferrous ammonium sulphate (0.0125M, 4.9 g/l) Ferroin indicator.

#### Procedure:-

3 ml of reagent, 2 ml of sample and 0.05g of catalyst were added to a glass vial. The tube was sealed, inverted until mixed, placed in the heater block and left for 2 hours at  $150^{\circ}$ C. After cooling, the contents of the vial were washed out into a conical flask with 4 x 5 ml of distilled water and the mixture was titrated against ferrous ammonium sulphate, 0.0025M or 0.0125M, dependent upon whether the concentrated or the dilute reagent was used for digestion. The operation was repeated for a blank of distilled water. The standardization of ferrous ammonium sulphate was checked for each set of samples.

# 3.3. SELECTIVE ION ELECTRODES

The use of selective ion electrodes is a relatively recent development in the field of water analysis. Electrodes do not measure the concentration of an ion but rather the activity of the ion in solution, that is the effective chemical reactivity of the ion. The potential of the electrode varies with the concentration of the ion and is determined by the Nernst equation:-

 $E = Ea + 2.3 \frac{RT}{F} \log A_{ion}$  E = Measured potential

Ea = portion of the potential due to the choice of reference electrode

2.3RT/F = Nernst factor

 $A_{ion}$  = activity of the ion being measured.

The relationship of activity and concentration to potential for the nitrate probe employed in this study is shown in Figure 27. The activity of an ion is also a function of the total ionic strength of the solution. Thus, samples that vary widely in total ionic strength need the addition of an ionic strength adjustor to both samples and standards.









Electrodes are subject to interferences when they are unable to distinguish between the desired ion and another ion or ions in solution. In this situation, which occurs to a greater or lesser degree for all electrodes, the potential developed is a function of the activity of the ion being measured and the activity of other interfering ions, and the Nernst equation can be written as:-

 $E = Ea + 2.3 \frac{RT}{F} \log (A_{ion} + K_{int} (A_{int})^{1/n})$   $A_{int} = activity of the interfering ion$   $K_{int} = selectivity constant for the interfering ion$  n = charge

Electrode manufacturers supply values for the selectivity constants of interfering ions enabling the user to calculate when levels of interfering ions will significantly affect a determination.

The main advantages of selective ion electrodes is that their response is usually very rapid, unless interferences are present samples need little or no pre-treatment and they give a linear response over a wide concentration range. Sommerfeldt et al (1971), using the nitrate probe for natural waters, reported that the range of detection without dilution was considerably greater than the phenoldisulphonic acid method and sample colouration and soluble salts did not interfere with the determination. They concluded that the accuracy of the electrode technique was sufficient but not as good as the more rigorous methods. Similar conclusions were reached in this study but substantial interference by high chloride concentrations was found difficult to remove. Bremner et al (1972) using the ammonia probe for Kjeldahl nitrogen determination in place of the customary distillation and titration found close agreement between the methods. He commented that the probe was rapid, simple and precise. Thomas and Booth (1973) compared the ammonia probe to the indophenol blue method for natural waters and found results differed by 1.2% and concluded that the accuracy was comparable to accepted methods.

The use of selective ion electrodes for the analysis of inorganic water pollutants has been generally reviewed by Riseman (1969) and Weber (1971).

## 3.3.1. <u>Nitrate</u>

An Orion nitrate probe, Model 92-07, was used in conjunction with a 1M sodium sulphate/calomel reference electrode and an E.I.L. 7050 meter. Between usage the electrode was stored upright in air, the membrane and filling solution being replaced on average every 6 months. Standard nitrate solutions of 2 mg/l and 20 mg/l, prepared freshly by dilution from a 1000 mg/l stock solution were used to calibrate the meter in the concentration mode. The stock solution had 1 ml of 10 mg/l pheny.mercuric acetate added as a preservative but was replaced on a regular basis. During a measurement the samples were stirred magnetically, calibration was checked after every 3 or 4 samples.

During operation the probe was found to respond rapidly but was subject to some drift and noise, this was particularly evident as the membrane became due for replacement and hence there was a need for regular calibration. High chloride levels in samples affected by winter road salting were found to interfere. Chloride interference can be prevented by precipitation with silver sulphate and removal or by mixing the sample with beads of silver ion exchange resin. The subsequent values were still subject to some doubt and a spectrometric technique was usually employed in preference. Relevant selectivity constants are presented below:-

HS<sup>-</sup> - 4 x 10<sup>-2</sup> HC0<sub>3</sub><sup>-</sup> - 9 x 10<sup>-3</sup> C0<sub>3</sub><sup>2-</sup> - 2 x 10<sup>-4</sup> F<sup>-</sup> - 6 x 10<sup>-5</sup> NO<sub>2</sub><sup>-</sup> - 4 x 10<sup>-2</sup> C1<sup>-</sup> - 4 x 10<sup>-3</sup> PO<sub>4</sub><sup>3-</sup> - 1 x 10<sup>-4</sup> SO<sub>4</sub><sup>2-</sup> - 3 x 10<sup>-5</sup>

## 3.3.2. Ammonium

Measurements were taken with an E.I.L. ammonia probe, model 8002-8, using an E.I.L. 7050 meter. Between usage the electrode was stored in a 0.1M ammonium chloride solution and rinsed thoroughly with distilled water prior to use; the electrode was renovated at 6 monthly intervals. Ammonium standards of 0.2 mg/l and 2 mg/l were used to standardise the meter and were prepared freshly by dilution from a 1000 mg/l stock solution. The ammonium ions were converted to free ammonia before measurement with the probe by the addition of 1M sodium hydroxide solution (5 mls to 50 mls of sample). During operation samples and standards were stirred magnetically and the calibration was checked after every 3 or 4 samples. Solution carryover was avoided by blotting the electrode with

#### absorbent tissue.

The response time for the probe was between 3 and 5 minutes for the range 0.2 - 2 mg/l. Below 0.2 mg/l response times considerably lengthened, up to 20 minutes, causing the accuracy of the determination to be lessened. The probe was free of interferences for the waters examined.

# 3.3.3 Chloride

An Activion type 003- 15 - 001 electrode was used in conjunction with a 1M sodium sulphate reference electrode and an E.I.L. 7050 meter. Chloride standards of 10 mg/l and 100 mg.l were prepared freshly by dilution from a 1000 mg/l stock solution and were used to standardise the meter. During operation samples and standards were stirred magnetically and the calibration checked after every 3 or 4 samples.

The probe was found to respond rapidly and, although noisy at low concentrations, no significant drift was noted. The probe gave comparable results to those obtained by the mercuric nitrate and silver nitrate titrimetric procedures. Relevant selectivity constants for the probes are:-

OH - 1.25 x  $10^{-2}$  Br - 3 x  $10^{2}$  1 - 2 x  $10^{6}$ NH<sub>4</sub> - 8.3 S<sup>2</sup> - may be present only in traces.

# 3.3.4 pH

Measurements were taken with an Orion pH combination electrode using the expanded pH scale on the E.I.L. 7050 meter. Buffers of pH7 and pH9 were used to calibrate the meter, calibration usually remaining constant throughout a series of determinations. A response time of 1 to 2 minutes was noted.

#### 3.4. HEAVY METALS

Two types of instrumentation were evaluated for the determination of heavy metals at trace levels:-

a) atomic absorption spectrometry (AAS)

b) anodic stripping voltammetry (ASV)

#### 3.4.1 Atomic Absorption Spectrometry

The use of AAS has become recognised as a standard procedure for the determination of low levels of heavy metals in water analysis. The technique has the advantage of speed and specificity over other analytical methods but since only one element can be determined at a time the method is not suitable for scanning simultaneously for a range of metals.

Sample Pre-treatment:-

Prior to analysis samples need some form of pre-treatment. For the determination of soluble metals filtration and acidification, as soon as possible after sample collection, is essential. If metals are only present at low levels a concentration step may be necessary either by evaporation or by chelation and extraction into an organic solvent. Evaporation is the most commonly employed method but chelation using ammonium pyrolidene dithiocarbamate (APDC) with subsequent extraction into methyl isobutyl ketone (MIBK) has gained favour (Brooks, 1967). One of the benefits of this procedure is the use of an organic solvent which has been reported as increasing the signal by 3 - 5 times depending upon the solvent used (Mulford, 1966). However, potential interferences of a chemical nature have been reported in the extraction stage (Lloyd, 1974) which can cause a lowering of extraction efficiency by the chelating agent. The use of evaporation as a concentration procedure can lead to problems of matrix interference or viscosity effects by the creation of high levels of dissolved solids. Hemsley (1971) recorded the interference of chromium on lead and recommended the use of lanthanum chloride as a releasing agent, however, he concluded that the technique was more accurate than colorimetry. For the determination of total metal concentrations digestion with some mix of inorganic acids (nitric, sulphuric, hydrochloric or perchloric acids) prior to concentration is the normal procedure. A number of different recipes have been documented depending upon the nature of the material to be analysed.

A number of sample preparation procedures were evaluated. These were:-

 Digestion with 2M nitric acid: 5 ml of nitric acid were added to 200 ml of sample, heated and evaporated down to a desired volume (usually 20 ml).

- 2. Digestion with concentrated sulphuric acid and potassium permanganate: 6 ml of 5% potassium permanganate, 1 ml of sulphuric acid were added to 200 ml of sample. The mixture was heated for 3½ hours, filtered and made up to the required volume.
- 3. Mixed acid digestion: 1 ml of 60% perchloric acid, 5 ml of conc. nitric acid and 0.5 ml of conc. sulphuric acid were added to 200 ml of sample. The mixture was digested slowly for 30 minutes (until the appearance of white fumes), filtered and made up to the required volume.
- 4. Ultraviolet irradiation in the presence of hydrogen peroxide:
  1 ml of 20 vol. hydrogen peroxide was added to 30 ml of sample and the mixture was irradiated overnight with UV.
- 5. APDC/MIBK concentration/extraction: 200 ml ofdigested sample were titrated to pH3 with sodium hydroxide and poured into a 500 ml stoppered separating flask. 5 ml of freshly prepared 5% APDC and 10 ml of redistilled MIBK were added to the flask and the mixture was shaken vigorously for 2 minutes. The layers were allowed to separate for 10 20 minutes and the bottom (water) layer was run off into a second separating flask, the MIBK layer being collected in a stoppered tube. The procedure was repeated for the water layer with a further 10 ml of MIBK.

Of all the procedures examined digestion with 2M nitric acid followed by a 10 fold increase in concentration by evaporation was found to be the simplest most reliable method and gave the best reproducibility. Chromium interference on lead was eliminated by the use of a 1% solution of lanthanum chloride. The other digestion procedures were found to yield high and variable blank values with similar variation being found in sample concentrations. Extraction with APDC/MIBK proved to be too time consuming for the effective analysis of large numbers of samples.

#### Procedure:-

An EEL 240 spectrometer was used for the majority of analyses. An air/acetylene flame was utilised with the appropriate instrument settings (wavelength, lamp current, slit width etc.) for the metal being determined as recommended. It was usually found necessary to use

10 -

the instrument in the integrating mode which gave an average value for absorption over 15 seconds. The spectrometer is a single beam instrument and, at full scale expansion, top standard drift was considerable and re-calibration between every sample was found to be necessary.

# 3.4.2 Anodic Stripping Voltammetry (ASV)

In recent years ASV has received widespread attention as a convenient technique for the determination of certain heavy metals at trace concentrations (Gardiner, 1974; Allen, 1970; Sinko, 1970). The main advantages of the technique are that high sensitivity is easily obtained making sample pre-concentration unnecessary and certain metals can be determined simultaneously, (principally lead, zinc, cadmium and copper).

#### Sample Pre-treatment:-

For the determination of total metals some digestion procedure is needed. The methods outlined for AAS were all evaluated and again digestion with nitric acid was found to give the best results. For ASV determinations a constant pH is necessary so digestion was followed by titration to pH 6.5 - 7.0 and addition of 1 ml of acetate buffer solution.

# Procedure:-

A Princeton Applied Research 17A polarographic analyser was used in conjunction with a hanging mercury drop electrode. Work by Chau and Chan (1974) had shown that for stripping the differential pulse mode gave great sensitivity for minimum plating times, 3 to 5 minutes compared to 40 - 70 minutes for other procedures, and was adopted for this research. To avoid contamination at the trace levels encountered a rigorous cleaning procedure was essential. The voltammetric cell was washed with nitric acid and several times with distilled water prior to analysis and between samples. When not in use the cell was filled with distilled water.

For determinations, 50 ml of sample were introduced into the cell and nitrogen was bubbled through for a period of 15 minutes. This was essential to completely deoxygenate the sample, the presence of oxygen giving rise to a sloping base line and additional current peaks at -0.8v

Fig.29 CADMIUM CALIBRATION CURVE FOR DETERMINATION BY ANODIC STRIPPING VOLTAMMETRY



Fig.30 LEAD CALIBRATION CURVE FOR DETERMINATION BY ANODIC STRIPPING VOLTAMMETRY



Fig.31 DIFFERENTIAL PULSE VOLTAMMAGRAM OF A STANDARD METAL SOLUTION.



and -0.05v. A mercury drop was prepared by turning the micrometer through 5 divisions and the metals were plated onto the electrode for 3 minutes at a constant voltage of -1100 mv, stirring being kept steady throughout deposition. The stirrer was then switched off and the metals were stripped from the electrode using a voltage scan of 5mv per second. When the potential reached a value at which the metal was electro-oxidised to its ion and returned into solution an electric current flowed which was proportional to the concentration of the metal plated. Current ranges of 2, 5 and 10 ua were used, dependent upon the concentration of metal present.

Sample concentrations were determined either from calibration curves prepared using standard solutions or by 'spiking' the sample with known concentrations of the metals to be determined. Figures 29 and 30 illustrate typical calibration curves for lead and cadmium, Figure 31 is a typical voltammagram for a standard solution.

## Interferences:-

The presence of complexing agents has been reported as leading to low estimates of total trace metals concentration by preventing complete reduction of the metal ion during the plating step (Gardiner, 1974). This factor was found to be one of the major complications in the analysis of gully pot waters where the presence of complex organic material was believed to be the offender. Complex formation resulted in peak shifts and peak broadening (Figure 32) sometimes to the extent of near obliteration, none of the digestion procedures employed were completely successful in eliminating this interference. The addition of small quantities of Teepol to standard solutions was found to produce a similar masking and obliteration of peaks. Thus surfactant material from car wash runoff present in the gully pot may have been responsible for interference in the samples. The presence of this complication, with no easy solution, resulted in AAS being the preferred technique.

## 3.5. THE ALKALI METALS

The metals sodium, calcium and potassium were measured rapidly and easily using an EEL flame photometer. Calibration curves were prepared for the three metals and standards of 5 mg/l for sodium, 5 mg/l for potassium and 50 mg/l for calcium were used to standardise the photometer.

Samples lying outside these ranges were diluted with distilled water. High levels of calcium can interfere with potassium determinations and were suppressed using aluminium nitrate.

# 3.6. LITHIUM

Lithium determinations were performed on an EEL 240 atomic absorption spectrometer at a wavelength of 670.8 nm. Instrument settings used were lamp current 7.5 mA, slit width 2 or 3 and fuel air/ acetylene ratio 3.7.

Sample preparation for analysis was minimal since filtration was found to be unnecessary. Careful decantation of the supernatant liquid from a settled sample made no significant difference to the results when compared to the normal procedure of analysis of the filtrate (Standard Error = 0.077). Although lithium sorption experiments have indicated that no significant uptake by sediments occurred within 7 days of addition for storm sewer waters (Neal and Jordan, 1977) samples stored for periods longer than 2 days were treated with 1 ml of conc. nitric acid.

For each run calibration standards from 0.25 mg/l to 8 mg/l were prepared from a stock solution of 1000 mg/l and a calibration curve was plotted (Figure 33) for the normal absorption mode operation. Samples were analysed 3 times in any one run, those with a concentration less than 2 mg/l were re-analysed on an expanded scale (2 mg/l equivalent to full scale deflection) using the integrating mode which gave a mean value over 15 seconds (Figure 34). Calibration was checked after every 4 or 5 samples.

Calcium has been reported as an interferent in lithium determination by A.A.S. (Water Pollution Research Laboratory, 1969) but at the levels found in stormwater samples no such interference was observed. Neal and Jordan (1977) found that matrix effects for two separate storm sewer systems were significantly different but were not significantly different for individual localities over a single storm event. They reported background response to be up to 6% at 1 mg/l and concluded that, in general, a bulked background water may be taken for use in calibration of storm sewer waters and to minimise any matrix effects. In this study it was found impracticable to gain a background sample for each event, however, isolated background samples taken during the course of the study





Fig. 34 LITHIUM CALIBRATION FOR DETERMINATION BY ATOMIC ABSORPTION (EXPANDED SCALE)



showed lithium background concentration to be insignificant (<1%) when compared to concentrations attributable to dosing.

## 3.7. OTHER PROCEDURES

#### 3.7.1. Chloride

Initial chloride determinations were made using the mercuric nitrate method. The method was subsequently used as a check on the reliability of the chloride probe and on occasions when the probe did not function.

Reagents:-

Standard sodium chloride solution - 1.648 g/l (1.0 ml = 1 mgCl<sup>-</sup>)

Nitric acid - 0.1M

Mercuric nitrate solution - 5.04g Hg  $(NO_3)_2$  H<sub>2</sub>O in 50 ml of distilled water containing 0.5 ml nitric acid and diluted to 1 litre.

Indicator - 0.5 g diphenylcarbazone and 0.05g bromophenol blue in 100 ml alchohol.

Procedure:-

100 ml of sample were placed in a conical flask, 1 ml of indicator was added and the mixture was titrated against mercuric nitrate until the green/blue to purple colour change was noted. The procedure was repeated for a blank.

$$Cl = \frac{(v_s - v_b)}{v} \times 1000 \text{ mg/l}$$

where

V = volume of sample

= volume used in titration of sample

= volume used in titration of blank

۷s

V<sub>b</sub>

Nitrates were determined directly without any sample treatment using a UV spectrometer at 203 nm. A linear calibration between 0 and 10 mg/l was recorded (Figure 28). The presence of dissolved organics has been reported as an interference for this determination, in this study, the levels encountered in stormwater samples were not found to cause any significant interference. If high dissolved organics were anticipated samples were digested with hydrogen peroxide prior to analysis.

# 4. DATA PROCESSING

A water quality monitoring programme of the type conducted in this research results in the generation of large quantities of data. The subsequent use and interpretation of this data is dependent upon the application of accurate and relevant data processing methods. The procedures utilised in this research fall into the following categories:-

a) Calculation of elementary statistics, e.g. mean, standard deviation, range, etc.

b) Investigation of correlations.

c) Calculation of storm flow.

d) Calculation of pollutant loadings

e) Mathematical modelling.

The results collected were stored on computer in one of four data file configurations dependent upon the type of data and the manner in which it was to be analysed. All computation was conducted in FORTRAN V using an interactive DEC 20 computer; graph plotting was performed using CALCOMP graph plotting software and the majority of statistics were generated using subroutines from the NAG (numerical algorithms group) library.

Mathematical modelling is considered separately in Chapter 5. The other data processing procedures may be conveniently considered in two sections:-

a) gully pot liquor quality data

b) storm runoff event data.

# 4.1. GULLY POT LIQUOR QUALITY DATA

The quality of gully pot liquor was monitored in three distinct manners:-

- a) Weekly/fortnightly analyses of four fixed pots. In this instance the gully pots were selected to represent widely varying situations and microcatchment characteristics.
- b) One day analysis of up to 25 randomly selected pots to assess catchment wide quality variation.
- c) Other miscellaneous analyses, e.g. sampling prior to storm runoff events, daily sampling during dry periods and other grab samples.

Data collected under a) and b) above were treated using the programme, Stats Package for Variation and Weekly Gully Analyses (STAT.FOR). The programme is presented in Appendix 1.1.it performed the following functions:-

- a) Calculation of basic statistics.
- b) Calculation of a correlation matrix for catchment wide variation analyses.
- c) A4 tabulation of data and statistics.

For the periodic weekly/fortnightly analyses of the four selected gully pots the following statistics were generated, for each pollution parameter, where the input data consists of N observations for each of M gully pots as an array  $(x_{ij})$ , i = 1 to N and i = 1 to M.

a) Annual maximum values for each gully pot.

b) Annual minimum values for each gully pot.

c) Annual range. Range = Max. - Min.

d) Interquartile range (IQR) Q1 =  $\frac{N}{4}$ , Q3 =  $\frac{3N}{4}$ , 1QR = Q3 - Q1 e) Annual mean values for each gully pot  $\overline{x}_{j} = \frac{1}{N}$   $\sum_{i=1}^{N} x_{ij}$ f) Weekly mean values  $\overline{x}_{i} = 1$   $\frac{M}{r}$  x

Weekly mean values  $\overline{x}_{i} = \frac{1}{N}$   $\sum_{j=1}^{M}$   $x_{ji}$ 

g)

Annual standard deviation for each gully pot studied and for weekly mean values

$$S_{j} = \sqrt{\frac{1}{(N-1)}} \qquad \begin{array}{c} N \\ \Sigma \\ i = 1 \end{array} \qquad (x_{ij} - \overline{x}_{j})^{2} \end{array}$$

h) The coefficient of skewness

$$s_{3_{j}} = \frac{1}{(N-1)} s_{j}^{3} \qquad i = 1 \qquad (x_{ij} - \overline{x}_{j})^{3}$$

i)

The coefficient of kurtosis

$$S_{j}^{4} = \begin{bmatrix} \frac{1}{(N-1)S_{j}^{4}} & \sum_{i=1}^{N} (x_{ij} - \overline{x}_{j})^{4} \\ \vdots = 1 \end{bmatrix} - 3$$

The application of the mean and standard deviation as reliable descriptive statistics of a sampled population is dependent upon the results being normally distributed. This may not be the case for limited numbers of water quality measurements and the assumption is made that the sample has been drawn from a whole population that is normally distributed.

Plotting the frequency distributions for each set of results (Chapter 6) shows that the majority of data sets are not normally distributed and the calculated standard deviations are commonly very high. This is due either to the presence of extreme high values that introduce a positive skew to the distribution or a preponderance of values approaching zero. A solution often employed to normalise distributions of this nature is to take logs of the individual results. However, this technique was also unsuccessful in creating a normal distribution because of the very wide spread of results, e.g. chloride, or the number of values close to zero where the limit of detection of the analytical method was the limiting factor. For these reasons the standard deviation cannot always be considered as a satisfactory measure of the dispersion of distribution. The interquartile range is presented as an alternative and was often 5 to 10 times less than the total range due to the elimination of extreme values. The coefficients of skewness and kurtosis were also presented to give some impression of the shape of the distribution. Skewness is a measure of the degree of asymmetry of the distribution. The coefficient has a value of zero for a normal distribution and takes a positive value when the distribution has a

longer 'tail' to the right (positive skewness) or a negative value when the distribution is skewed to the left. Kurtosis is a measure of the degree of peakedness of the distribution and is zero for a normal distribution. Values greater than zero result from a peaked (leptokurtic) distribution and values less than zero for a flat (platykurtic) distribution.

One-day catchment-wide sampling was performed not only to examine the dispersion of values for individual pots on the same day but also to test the significance of small sets of samples (3, 4 or 5) in representing the mean value of the population. Therefore, in addition to the statistics presented above, 95% confidence limits for the estimate of the population mean based on the sampled distribution were calculated using 'Students' t- distribution:-

Confidence limits:  $x_j + t_{0.95} \frac{s_j}{N-1}$ N - 1

The confidence limits were also expressed as a percentage of the mean.

Like the standard deviations, the confidence limits were often found to be very high (greater than 100% of the mean in some instances) this can again be ascribed to the presence of extreme values and serves to indicate the general unreliability of estimating the population mean from small sets of samples.

Investigation of inter-parameter correlations was not performed routinely for the periodic weekly analyses. Possible relationships were assessed individually. However, correlation analyses for the variation studies were performed routinely, the results being output as a matrix of correlation coefficients. The Pearson product - moment correlation coefficient was calculated:-

$$R_{jk} = S_{jk}$$

where

$$S_{jk} = \sum_{i=1}^{N} (x_{ij} - \overline{x}_{j}) (x_{ik} - \overline{x}_{k})$$

$$S_{jj} = \sum_{i=1}^{N} (x_{ij} - \overline{x}_{j})^{2}$$

Other miscellaneous gully pot quality data was not subjected to the statistical treatment outlined above but, where appropriate, weekly mean values were calculated.

# 4.2. STORM RUNOFF DATA

Storm runoff data was processed using the Urban Runoff Pollutant Loading programme (STQUAL.FOR). The programme is listed and its operation and parameters explained in Appendix 1.2. The programme performed the following functions:-

a) calculation of flow rate from lithium concentrations.

b) calculation of cumulative pollutant loads, total pollutant loads and mean storm concentrations.

c) tabulation of observed concentrations and cumulative loads in A4 format.

d) graphical plots of observed concentrations and percentage cumulative loads against time from start of sampling.

#### 4.2.1.Calculation of Flow Rate

Flow rates for the 4th sample and onwards were computed from the mass balance equation for dilution gauging previously derived in Chapter 2. Assuming an insignificant background concentration, then:-

$$Q = \frac{C_{i}}{C_{2}} q \times 10^{-3} 1/s$$

 $C_1 = \text{concn of lithium in dosing solution (mg/l)}$ 

 $C_2 = \text{concentration of lithium in the sewer (mg/l)}$ 

q = rate of dosage of lithium (ml/s)

No flow rate could be calculated for the first sample collected since the sample was collected at the same moment as dosing was initiated. However, since this first sample often contained high concentrations of solids and could contribute significantly to total pollutant load for the storm event an assumed value of 8 1/s was used. This value represented the average flow rate at which the float switch started operation. Since the doser required  $5\frac{1}{2}$  minutes to reach an equilibrium dose rate, then, flow rates for the second and third samples had to be calculated using the doser calibration curve (Figure 35) and an estimated low flow time of travel between the dosing and sampling locations of 3.2 minutes. Given that the mid-point time of the sampling interval corresponded to the effective average dosing rate over the whole sampling interval then the dosing rates applicable to samples 2 and 3 were 0.225 ml/s and 0.883 ml/s respectively (see Figure 35).

#### 4.2.2. Artificial Recession Curves

Towards the end of a storm event when flow rates dropped below 8 1/s the float switch acted as a circuit breaker preventing further sampling. On some occasions sampling was recorded as stopping at flow rates much higher than 8 1/s and as a result a significant portion of the storm recession was missed. On a small catchment such as Clifton Grove where runoff response is rapid and flow rates are low in comparison to larger urban areas the loss of this portion of the storm could introduce significant errors into the determination of cumulative discharge and cumulative pollutant loads. To overcome this problem a typical recession curve was artificially generated in the programme STQUAL.FOR to describe the decay in the flow rate that was missed by the field measurements. The equation was calibrated empirically to give a smooth decay consistent with available observations:=

$$Q_{t+1} = Q_t e^{-0.33T}$$

where T = sampling interval in minutes.

The equation operated until  $Q_{t+1}$  became less than 0.5 1/s when the storm was assumed to have ended.





Fig. 36 EXAMPLE OF A CALCULATED FLOW RECESSION CURVE FOR THE STORM OF THE 24TH MAY, 1978



Fig. 37 EXAMPLE OF CALCULATED RECESSION CURVES FOR SUSPENDED SOLIDS (SS), CHEMICAL OXYGEN DEMAND (COD) AND DISSOLVED SOLIDS (DS) FOR THE STORM OF THE 24TH MAY, 1978



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It was also necessary to apply a similar recession curve to suspended solids values since these were observed to decay at the end of a storm in a similar fashion. Their concentration was recessed down to a minimum of 3.0 mg/l, a value that was typical of the minimum concentrations observed in runoff.

$$SS_{t+1} = 3.0 + (SS_t - 3.0) e^{-0.33 T}$$

Other solids related pollution paramters, such as COD and BOD, were also recessed down to minimum concentrations of 5.0 and 0.3 mg/l respectively. However, soluble materials that were not generally observed to decay towards the end of a storm, but approached a constant value, were allowed to remain at their final observed value, e.g. nitrate, ammonium, dissolved solids:-

$$DS_{t+1} = DS_{t}$$

Figurs 36 and 37 illustrate the use and significance of the artificial recession curves for Storm 10, a small event of only 3 samples duration.

# 4.2.3. Calculation of Pollutant Loads

Pollutant loadings are expressed in three ways:-

a) Instantaneous storm loads (g/s) at time t.

$$IL_t = C_t Q_t \times 10^{-3}$$

where

C<sub>t</sub> = concentration in mg/l at time t Q<sub>t</sub> = flow rate in l/s at time t

b)

Pollutant load (kg) over time interval T, (minutes)

$$L_t = C_t Q_t T \times 60 \times 10^{-0}$$

where

 ${\rm C}^{\phantom{\dagger}}_t$  and  ${\rm Q}^{\phantom{\dagger}}_t$  represent average values over the time interval T.

c) Cumulative pollutant loads (kg) at time t

$$CUML_{t} = \sum_{t=1}^{t} C_{t} Q_{t} T \times 60 \times 10^{-6}$$

For the purposes of this research sampling was not discrete but continuous over a period of approximately 2 minutes with a pause of approximately 1 minute between samples for reverse pumping. The times against which concentrations and flow rates were recorded were the mid point time of the sampling interval and it was desirable that cumulative loads should be reported for these times. The rationale behind the calculation of cumulative loads for these times is illustrated in Figure 38. The situation was further complicated, since initial and subsequent sampling and reverse pumping times were different. For the cumulative load at time t = 1 only the first half of the initial sampling time is included in calculations. For the calculations of subsequent cumulative loadings (t = N) the pollutant load corresponding to the second half of the previous sampling period is calculated, the pollutant load corresponding to the first half of the present sampling period is calculated and average flow and concentration values are used to compute the pollutant load for the intermediate period of reverse pumping, the three factors are then summed and added to the cumulative load for t = N - 1 to give the cumulative load at time t = N. This approach minimised errors created by using discrete as opposed to continuous integration. Mathematically the calculation of cumulative loads can be expressed as follows:-

If:-	S1	=	initial sampling time
	S2	=	subsequent sampling time
	т1	=	initial sampling interval
	T2	=	subsequent sampling interval

Then:-

$$CUML_{1} = \frac{S1}{2} Q_{1} C_{1}$$

$$CUML_{2} = CUML_{1} + (\frac{S1}{2} Q_{1} C_{1}) + (\frac{S2}{2} Q_{2} C_{2})$$

$$+ (\frac{T1 - S1}{4}) (Q_{1} + Q_{2}) (C_{1} + C_{2})$$





$$CUML_N = CUML_{N-1} + \frac{52}{2} (Q_n C_n + Q_{n-1} C_{n-1})$$

+ 
$$(\frac{T2 - S2}{4}) (Q_{n-1} + Q_n) (C_{n-1} + C_n)$$

# 4.2.4. Multiple Regression Analysis

The combined data for all storm events monitored was used to investigate the relationship between pollutant loads and mean storm concentrations with meteorological parameters such as storm duration and average intensity using multiple linear regression analysis. The technique assesses the relationship between a dependent variable (Y) and a set of independent variables  $(x_1, x_2 --- x_n)$  and was of particular use in identifying the meteorological parameters of most significance in determining the generation of pollutant loadings. It was also used to formulate simple predictive models for determining pollutant loads when only basic meteorological data is available. The data was treated using a computer programme, MULREG FOR (Appendix 1.3) which made use of two subroutines for multiple regression analysis from the NAG Library, GO2BGF and GO2CGF. The subroutines fit a curve to the data points of the form:-

 $Y = a + b_1 X 1 + b_2 X 2 + b_3 X 3 \dots + b_r X N$ 

where

 $X1 \rightarrow XN =$  variables, i.e. meteorological parameters  $X1_1 \rightarrow X1_1 =$  observations on each variable (i.e. No. of storms in analysis) Y = pollutant loading under consideration a = regression constant  $b_1 \rightarrow b_r =$  regression coefficients.

The regression coefficients are calculated in such a way that the sum of the squared residuals is minimised. The accuracy of the resultant equation is indicated by the coefficient of determination  $(R^2)$ , the proportion of variance that is explained by the variables included in the analysis.

# 5. <u>MATHEMATICAL MODELLING OF URBAN RUNOFF PROCESSES AND</u> POLLUTANT REMOVAL FROM GULLY POTS.

The development of mathematical simulation as an important aid for the management of urban stormwater has been discussed previously. A wide variety of models now exists, each comprising of differing functions and capabilities geared to the needs of various potential users. The models available range from empirical design techniques, such as the Rational method, to highly complex and versatile computer packages, of which the Stormwater Management Model, (SWMM), is an example.

Although significant progress has been made in recent years in the simulation of hydrological and hydraulic processes occurring in urban runoff, the prediction of urban runoff quality and the modelling of pollutant accumulation and removal processes is still in its infancy. The complexity of the physical and chemical processes involved in determining runoff quality defies accurate mathematical description. In the models that have been proposed the role of roadside gully pots as sediment traps or pollutant generators has not been considered. Their effect has either been ignored, with the assumption that the pollutant accumulation function generally encompasses their contribution (Sartor and Boyd, 1972) or considered to be negligible (Price and Mance, 1978).

One of the main objectives of this research was an assessment of the significance of gully pots in determining the quality of stormwater runoff. Therefore, it was considered necessary not only to quantify the nature and amount of pollutant in the gully pot, but also the mode of removal of these pollutants during a storm event and their subsequent contribution to runoff quality at the sewer outfall. Thus, the criteria for the development of an urban runoff mathematical model from this research are two-fold, as:-

- a) A preliminary research tool for further catchment studies at Clifton Grove;
- b) A vehicle for the prediction of pollutant contributions from gully pots in urban runoff.

To achieve these aims the following approach was adopted:-

a) A laboratory study of the removal of material from gully pots;

- b) Derivation of functions describing these processes;
- c) Selection of an appropriate rainfall-runoff model;
- d) Development of the computer program and application of the model to the Clifton Grove catchment;
- e) Integration of the gully pot pollutant contribution functions into the rainfall-runoff model.
- f) Verification and calibration of the model with field measurements.

#### 5.1. THE REMOVAL OF MATERIAL FROM GULLY POTS

Two alternative approaches may be adopted for assessing pollutant contributions to urban runoff from gully pots, that is, measurements may be conducted either in the field or the laboratory. Field measurements may actually monitor the real situation but meaningful interpretation of the data is hampered, in fact, by the large number of unknown variables, the complexity of the processes involved and logistical problems of monitoring and installation of instrumentation. Problems of data interpretation can be ascribed to the following phenomena:-

- a) Unknown and variable flow rates into the pot;
- b) Unknown and variable solids contribution to the pot from urban surfaces;
- c) Unknown mass of stored sediment;
- d) Unknown quality of stored sediment.

The introduction of instrumentation for the elimination of some of these unknowns such as flow measuring devices or solid filtration systems will disturb the hydraulic stirring action of flow entering the pot and also render subsequent studies valueless by altering the usual properties of the stored sediment. In a laboratory study many of these unknowns may be eliminated or made constant and, therefore, this approach was adopted.

# 5.1.1 Theory of the Removal of Pollutants from Gully Pots

The physical parameters governing the amount of material contributed from a gully pot are illustrated below:-



M1 = total mass of material available for release from the settled solids.

M2 = mass of material initially in suspension

K = rate of release of settled material

Q = flow through the gully pot

V. = volume of fluid in the gully pot

c = concentration of material in the outflow

R = theoretical retention time of fluid in the pot = V/Q

The mass of material available for release from the settled solids (M1) is not readily defined even in a laboratory situation. M1 will be dependent upon the particle size of the sediments which is dependent in turn upon the soil types and other materials available for input in the immediate catchment locality. The degree of biological degradation of the sediments will also influence M1, however, the flow rate (Q) is likely to be the parameter of most influence in determining the size of M1. In the laboratory where the material is of constant composition and sediment input is either negligible or constant and temporarily ignoring material already in suspension then the change of concentration of material in the gully pot fluid can be expressed as:-

$$dc = \frac{K}{V} dt - \frac{Q}{V} dt c \qquad \dots (1)$$

This equation is applicable from t = o to t = M1/Ki.e. when all available material has been released.

Since R = V/Q

$$\frac{\mathrm{d}c}{\mathrm{d}t} = \frac{\mathrm{K}}{\mathrm{V}} - \frac{\mathrm{c}}{\mathrm{R}}$$

Using the integration factor:-

$$e \int \frac{1}{R} dt = e^{t/R}$$

Then:-

$$e^{t/R} \frac{dc}{dt} + e^{t/R} \frac{a}{R} = e^{t/R} \frac{K}{V}$$
$$\frac{d}{dt} (e^{t/R}c) = e^{t/R} \frac{K}{K}$$

dt V  

$$e^{t/R}c = \frac{K}{V} \int_{0}^{t} e^{t/R} dt$$
  
 $e^{t/R} = \frac{K}{V} \left[-R e^{t/R}\right]^{t} + A$ 

 $e^{t/R}c = \frac{K}{Q} (e^{t/R} - 1) + A$ when t = 0, c = 0 ... A = 0

for  $o \leq t \leq M1/K$ 

$$c = \frac{K}{Q} (1 - e^{-t/R})$$
 ... (2)

0

When all the mass available for release is in suspension, i.e.  $M1/K\leqslant t\leqslant 00$ 

$$\frac{\mathrm{d}c}{\mathrm{d}t} = -\frac{Q}{V}c, = -\frac{c}{R}$$

$$\frac{dc}{c} = -\frac{1}{R} \int_{M1/K}^{t} dt$$

$$\left[\log_{e} c\right]_{c_{1}}^{c} = -\frac{1}{R} \left[t\right]_{ML/K}^{t}$$

$$\log_{e} \frac{c}{c_{1}} = -\frac{1}{R} \left[t-\frac{M1}{K}\right] + A$$
when t = M1/K, c = c\_{1},  $\therefore A = 0$ 

$$\therefore \log_{e} \frac{c}{c_{1}} = -\frac{1}{R} \left(t-M1/K\right)$$

$$\frac{c}{c_{1}} = e^{\left(-t/r + M1/KR\right)}$$
since  $c_{1} = \frac{K}{Q} \left(1 - e^{-M1/KR}\right)$ 

$$e^{\left(-t/R + M1/KR\right)}$$

$$\cdots (3)$$

The same theory applies for dissolved material with an initial concentration of  $c_0$  or for material initially in suspension, where  $c_0 = M2/V$ . In this case:-

$$\begin{bmatrix} \log_{e} & c \end{bmatrix}_{c}^{c_{0}} = -\frac{1}{R} \qquad \begin{bmatrix} t \end{bmatrix}_{0}^{t}$$
$$\log_{e} & \frac{c_{0}}{C} = -\frac{t}{R} + A$$

when t = 0,  $c = c_0$   $\therefore$  A = 0 $c = c_0 e^{-t/R}$  ... (4)

The concentration of suspended material in the outflow derived from settled material and material in suspension in the gully pot is then described by the summation of the relevant equations:-

for 
$$0 \leq t \leq M1/K$$
  
 $c = \frac{K}{Q} (1 - e^{-t/R}) + c_0 e^{-t/R} \dots (5)$
for M1/K 
$$\leq t \leq \infty$$
  
 $c = \frac{K}{Q} (1 - e^{-M1/KR}) e^{(-t/R + M1/KR)} + c_0 e^{-t/R}$ 
(6)

If the inflow to the gully pot has a concentration of material itself ( $c_2$ ) then Eq (1) can be modified to:-

$$dc = \frac{K}{V} dt - \frac{Qdtc}{V} + \frac{Qdtc_2}{V} \dots (7)$$

Preliminary laboratory results from a study using sediment derived from road surfaces at a city centre site (Wilson, 1978) indicated the general validity of the proposed model and a full laboratory study was initiated.

## 5.1.2 Experimental

The apparatus assembled for the investigation is illustrated in Figure 39. The gully pot was of the type employed in the Clifton Grove sewerage system and contained a volume of 92.8 litres of water. A pumped flow of water was available from a main supply tank that fed three rotameters. The rotameters were arranged in parallel covering a range of differing flow rates. The fluid output from this arrangement was stilled by passage through a constant head tank. A rectangular, steel bed channel was used to simulate kerbside channel flow and was arranged with a gentle slope typical of that encountered at the Clifton Grove estate. The water entered the gully pot through a standard gully grating.

Since the proposed model assumes a state of complete mixing in the gully pot it was necessary to evaluate in the first instance whether this assumption was valid. This was achieved simultaneously with the investigation of the removal of dissolved material by monitoring the changes in concentration of a solution of sodium chloride added to the pot. An initial concentration of 1.5 to 2.5 g/l sodium chloride was prepared in-situ and the removal of this solution for flow rates between 0.03 and 1.0 l/s was monitored using a conductivity probe at the outflow point of the pot. A maximum inlet flow rate of 1.0 l/s was adopted, which was consistent with expected maximum flow rates in the field. A direct correlation between salt concentration and conductivity allowed the concentration of salt in the outflow to be determined.

Fig.39 LABORATORY APPARATUS FOR INVESTIGATING THE REMOVAL OF MATERIAL FROM GULLY POTS FOR DIFFERENT INLET FLOW RATES flow constant head tank dwnd rotameters. valves ' flat bed channel gully pot sediment gully grating conductivity meter

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Sediment for the solid removal tests was obtained from municipal gully emptying tankers that serviced the Clifton area. Prior to use, the material was sterilised, dried and sieved free of large twigs, paper, cigarette ends, etc. For each test 2 kg of the dry sediment were added to the base of the pot in the absence of water. This resulted in a depth of material of approximately 15 mm. Water was carefully added to the pot until the first overflow occurred and was then left for 2 to 3 days to allow the solid to settle and saturate with water.

Tests were performed for flow rates ranging between 0.1 and 1.0 l/s. A sample of water was taken prior to each test to determine the initial suspended solids content. Samples at time intervals of 0, 0.25, 0.5, 0.75, 1.0, 1.5, 2.0, 3.0, 4.0, 5.0, 6.0, 7.0 and 8.0 minutes were taken during each run. Samples of the inflow were also collected and the resultant background solids concentration was subtracted from the observed concentrations.

## 5.1.3 Interpretation of Results

For high flow rates the removal of soluble material was observed to follow the proposed exponential decay curve (Figure 40), where, for a completely mixed state, the curve is described by Equation 4. For the completely mixed state a plot of  $\log_e (c_1/c_0)$  against -t/R should yield yield a straight line with a gradient of one (Figure 41): gradients greater than one are indicative of incomplete mixing. The results obtained (Table 19) show that down to approximately 0.12 l/s complete mixing normally prevails in a gully pot. However, below this flow rate, although normal exponential decay curves were recorded, gradient values were greater than one, indicating that effective retention times were much greater than would be expected if complete mixing occurred. Stirring of the pot contents after the completion of tests revealed unmixed pockets of salt solution were still present. These observations are consistent with the occurrence of 'short circuiting', that is, some zones of the gully pot were being completely mixed while others remained unstirred. The effective retention was used to calculate the volume of the gully pot in a completely mixed state, the results being illustrated graphically in Figure 42.

Fig. 40 TYPICAL CURVE ILLUSTRATING THE MODE OF REMOVAL OF SOLUBLE MATERIAL (SALT) FROM A GULLY POT UNDER STEADY FLOW CONDITIONS (1 LITRE PER SECOND)



Fig. 41 PLOT OF LOGE (C1/Co) AGAINST -t/R ILLUSTRATING COMPLETE MIXING FOR A FLOW RATE OF 0.5 1/s







% Volume Mixed	μ1.7	48.2	58.5	41.8	6.49	60.2	75.4	0.72	96.1	106.1	8 <b>.</b> 69	100.0	21				
Theoretical Retention Time	3093	2320	1856	1856	1547	1326	1031	928	773	371	186	93		·			
Retention Time (S)	1289	1116	1086	777	1006	798	778	901	743	t2tt	185	93					
Slope	- 2.45	- 2.12	- 1.75	- 2.44	- 1.57	- 1.70	- 1.35	- 1.05	- 1.06	- 0.89	- 1.02	- 1.02					
Flow Rate	0.03	0*04	0.05	0.05	0.06	0.07	0.09	0.10	0.12	0.25	0.50	1.00					
	Flow Rate Slope Retention Time (S) Retention Time % Volume Mixed	Flow RateSlopeRetention Time (S)Theoretical% Volume Mixed0.03- 2.451289309341.7	Flow RateSlopeRetention Time (S)Theoretical Retention Time% Volume Mixed0.03- 2.451289309341.70.04- 2.121116232048.2	Flow Rate         Slope         Retention Time (S)         Theoretical         % Volume Mixed           0.03         - 2.45         1289         3093         41.7           0.04         - 2.12         1116         2320         48.2           0.05         - 1.75         1086         1856         58.5	Flow Rate       Slope       Retention Time (S)       Theoretical       % Volume Mixed         0.03       - 2.45       1289       3093       41.7         0.04       - 2.12       1116       2320       48.2         0.05       - 1.75       1086       1856       58.5         0.05       - 2.44       77       1856       41.8	Flow Rate         Slope         Retention Time (S)         Theoretical         % Volume Mixed           0.03         - 2.45         1289         3093         41.7           0.04         - 2.12         1116         2320         48.2           0.05         - 1.75         1086         1856         58.5           0.05         - 2.44         777         1856         41.8           0.05         - 1.75         1086         1856         41.8           0.05         - 1.75         1086         1856         41.8           0.05         - 2.44         777         1856         41.8           0.05         - 2.44         777         1856         41.8	Flow Rate       Slope       Retention Time (S)       Theoretical       % Volume Mixed         0.03       - 2.45       1289       3093       41.7         0.04       - 2.12       1116       2320       48.2         0.04       - 2.12       1116       2320       48.2         0.05       - 1.75       1086       1856       58.5         0.05       - 2.44       777       1856       41.8         0.05       - 2.44       777       1856       41.8         0.05       - 1.57       1006       1547       64.9         0.07       - 1.57       708       1326       60.2	Flow Rate       Slope       Retention Time (S)       Theoretical       % Volume Mixed         0.03       - 2.45       1289       3093       41.7         0.04       - 2.12       1116       2320       48.2         0.04       - 2.12       1116       2320       48.2         0.05       - 1.75       1086       1856       58.5         0.05       - 2.44       777       1856       58.5         0.05       - 2.44       777       1856       64.9         0.05       - 1.70       798       1547       64.9         0.07       - 1.70       798       1326       60.2         0.09       - 1.35       778       1031       75.4	Flow Rate       Slope       Retention Time (S)       Theoretical       % Volume Mixed         0.03       - 2.45       Retention Time (S)       Retention Time       #1.7         0.03       - 2.45       1289       3093       #1.7         0.04       - 2.12       1116       2320       48.2         0.04       - 2.12       1116       2320       48.2         0.05       - 1.75       1086       1856       41.8         0.05       - 2.44       777       1856       41.8         0.05       - 2.44       777       1856       41.8         0.06       - 1.57       1006       1547       64.9         0.07       - 1.70       798       1031       75.4         0.09       - 1.35       778       1031       75.4         0.10       - 1.05       901       928       97.0	Flow Rate       Slope       Slope       Retention Time (S)       Theoretical       % Volume Mixed         0.03       -2.45       1289       3093       41.7       41.7         0.03       -2.45       1289       3093       41.7       48.2         0.04       -2.12       1116       2320       48.2       48.2         0.05       -1.75       1086       1856       58.5       41.8         0.05       -2.44       777       1856       58.5       41.8         0.05       -2.44       777       1856       64.9       41.8         0.05       -1.70       798       1547       64.9       50.2         0.07       -1.70       798       1031       75.4         0.09       -1.35       778       1031       75.4         0.10       -1.05       901       928       97.0         0.12       -1.06       743       773       96.1	Flow Rate         Slope         Retention Time (S)         Theoretical Retention Time         % Volume Mixed           0.03         - 2.45         1289         3093         41.7         41.7           0.03         - 2.45         1289         3093         41.7         48.2           0.04         - 2.12         1116         2320         48.2         48.2           0.05         - 1.75         1086         1856         58.5         41.8           0.05         - 2.44         777         1856         41.8         60.2           0.06         - 1.57         1006         1547         64.9         75.4           0.07         - 1.57         778         1326         60.2         75.4           0.09         - 1.56         778         1031         75.4         75.4           0.10         - 1.05         901         928         97.0         07.0           0.12         - 1.06         743         773         96.1         106.1           0.25         - 0.09         424         371         106.1         106.1	Flow Rate         Slope         Retention Time (S)         Theoretical Retention Time         % Volume Mixed           0.03         - 2.45         1289         3093         41.7           0.04         - 2.12         1116         2320         48.2           0.05         - 1.75         1086         58.5         41.6           0.05         - 1.75         1086         1856         58.5           0.05         - 2.44         777         1856         41.8           0.05         - 2.44         777         1856         60.2           0.06         - 1.57         1006         1547         64.9           0.07         - 1.70         798         1732         55.4           0.07         - 1.05         778         1031         75.4           0.10         - 1.05         778         96.1         97.0           0.12         - 1.06         743         773         96.1           0.25         - 0.02         - 1.02         743         96.1           0.50         - 0.28         424         371         106.1	Flow Rate         Slope         Retention Time (S)         Theoretical Retention Time         % Volume Mixed           0.03         - 2.45         1289         3093         41.7         %           0.03         - 2.45         1289         3093         41.7         %           0.04         - 2.12         1116         2320         48.2            0.05         - 1.75         1086         1856         58.5            0.05         - 2.44         777         1856         58.5            0.05         - 2.44         777         1856         58.5            0.06         - 1.70         778         1326         60.2            0.07         - 1.70         778         1326         60.2            0.09         - 1.57         1006         1326         60.2            0.10         - 1.05         778         901         75.4            0.112         - 1.05         743         77.3         96.1            0.25         - 0.89         424         371         106.1            0.50         - 1.02         93	Flow Rate         Slope         Retention Time (S)         Theoretical Retention Time         % Volume Mixed           0.03         - 2.45         1289         3093         41.7         %           0.03         - 2.45         1116         2320         48.2            0.04         - 2.12         1116         2320         48.2            0.05         - 1.75         1086         1856         41.8            0.05         - 2.44         777         1856         41.8            0.05         - 2.44         777         1856         41.8            0.06         - 1.57         1006         1547         64.9            0.07         - 1.70         778         1326         60.2            0.09         - 1.35         778         1031         75.4            0.10         - 1.05         743         773         96.1            0.12         - 1.05         743         773         96.1            0.50         - 1.02         185         910.0           99.8            0.50         - 1	Flow Rate         Slope         Retention Time (S)         Theoretical Retention Time         % Volume Mixed           0.03         - 2.45         1289         3093         41.7           0.04         - 2.12         1116         2320         48.2           0.04         - 2.12         1116         2320         48.2           0.05         - 1.75         1086         1856         58.5           0.06         - 1.57         1006         1547         64.9           0.07         - 1.70         778         1326         60.2           0.07         - 1.70         778         1326         60.2           0.00         - 1.70         778         1326         61.9           0.00         - 1.65         778         1031         75.4           0.10         - 1.65         743         773         96.1           0.12         - 1.05         743         773         96.1           0.50         - 1.02         185         96.1         90.6           0.50         - 1.02         93         96.1         90.6           1.000         - 1.02         93         93         90.0           0.50	Flow Rate         Slope         Retention Time (S)         Theoretical Retention Time         % Volume Mixed           0.03         - 2.45         1289         3093         41.7         %           0.04         - 2.12         1116         2320         48.2            0.05         - 1.75         1086         1856         58.5            0.05         - 1.75         1086         1856         58.5            0.05         - 1.75         1086         1856         58.5            0.05         - 1.75         1006         1856         41.9            0.06         - 1.77         1086         1856         58.5            0.07         - 1.57         1006         1326         60.2            0.07         - 1.70         778         928         97.0            0.10         - 1.05         911         773         96.1            0.12         - 1.06         122         371         106.1            0.260         - 1.02         185         96.1             0.212         - 1.06         123	Flow Rate         Slope         Retention Time (S)         Theoretical         % Volume Mixed           0.03         - 2.45         1289         3093         41.7           0.03         - 2.45         1289         3093         41.7           0.04         - 2.12         1116         2320         48.2           0.05         - 1.75         1086         1856         58.5           0.05         - 1.77         777         1856         41.8           0.05         - 1.77         777         1856         41.8           0.05         - 1.70         777         1956         41.8           0.06         - 1.70         778         1031         75.4           0.07         - 1.95         778         1031         77.3           0.09         - 1.05         743         773         96.1           0.12         - 1.06         14.24         371         106.1           0.50         - 1.02         185         96.1         99.8           1.000         - 1.02         93         100.0         99.8           1.000         - 1.02         93         100.0         99.3

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TABLE 19 Laboratory results for the removal of dissolved material from gully pots for different intothe flow rates

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TABLE 20 Laboratory results for the removal of particulate material from gully pots for different inlet flow rates.

M (Total)	706.8	1673.6	3604.8	2799.0	3999.6	2902.2	8112.0	4719.6	8442.0		
(mg) M	56.2	3114.5	4075.9	3970.3	3097.1	2434.6	5413.4	4656.2	4663.3		
K (mg/s)	3.7	103.8	135.9	264.7	206.5	162.3	180.4	310.4	310.9		
tp (s)	15	30	30	15	15	15	30	15	15		
Cps (mg/l)	9*0	32.5	41.2	41.1	31.8	24.8	51.4	46.7	46.4		
Cpo (mg/1)	27.4	49.4	69.0	57.0	52.2	4.94	66.6	85.4	90.6		
Co (mg/1)	27.2	18.0	31.6	17.2	22.4	27.6	19.6	44.8	52		
R (s)	928	t19tt	232	186	155	133	116	103	92.8		
Inlet Flow (L/S)	0.1	0.2	<del>1</del> 4 0	0.5	0.6	0.7	0.8	0.9	1.0		

Cpo = observed peak concentrations, Cps = peak concentration attributable to resuspension of material (defined by Eqn. 9) tp = time to peak concentration, K = rate of resuspension of material (defined by Eq. 8) M txk = the mass available for release M(Total) = total measured mass removed after 8 minutes Where R = retention time, Co = concentration of material in suspension at the start of the test, monitoring. 的复数的人们的,也是有一个人,这个人的人,也不是有一个人的人们,你们是这个人的人们的,你们就是有这个人,就是有这个人的。""你们是有一个人,你们也是一个人,你们就是我们就是有一个人

The results for the removal of solid material from the pot are presented in Table 20. The rate of release of available material was normally very rapid with peak concentrations being reached within 30 seconds, after which the expected decay curve was followed, (Figure 43). The rate of release for each test, K (mg/s), was calculated from the equation:-

$$K = \frac{C_{ps}Q}{(1 - e^{-t_{p}/R})} \dots (8)$$

where  $C_{ps} = C_{po} - C_{o} e^{-t_{p}/R}$  ... (9)

C<sub>ps</sub> = peak concentration attributable to the release of bottom sediments.

C<sub>po</sub> = observed peak concentration minus the background concentration.

C\_ = initial suspended solids concentration

Q = flow rate

 $t_p$  = time to peak concentration (15 or 30 seconds)

R = Retention time

The total mass of material available for release could then be calculated from:-  $M = t_D K$  (Table 20)

The major difficulty in the analysis of the results was the rapid depletion of material available for release. The sampling interval of 15 seconds was the fastest practical sampling rate but remained too insensitive for the accurate determination of tp, the time to peak concentration. The value obtained for  $t_p$  in turn influenced the accuracy of the determination of M and K. A linear regression of K against flow rate was employed to reduce errors through unreliable t values and gave a good correlation (R = 0.85, Figure 45). Similarly, a plot of M against flow rate (Figure 44) gave a reasonable correlation (R = 0.75) but visually the points showed a much wider scatter and with the limited data available linear regression may not be a valid method of calibration for M.

It



Fig. 45 RELATIONSHIP BETWEEN THE RATE OF RELEASE OF MATERIAL, K, AND THE INLET FLOW RATE.



is possible, for the low flow rates investigated, that there is a maximum mass of solids available for release, i.e. only easily resuspended fine material and surface dusts are removed from the pot. Above 0.2 1/s the results could be interpreted in this manner with a mean mass available of approximately 4,000 mg, however, another plausible relationship is indicated by the dotted line in Figure 44. The rate of release, K, would not logically be expected to reach a plateau level but to be more rapid the faster the flow rate, and the regression equation was considered to be valid.

## 5.1.4 Conclusions on the Removal of Material from Gully Pots

For the removal of dissolved solids the laboratory tests probably give a reliable simulation of conditions that may occur in the field. The same may not be true for the sediment removal tests where a number of limitations inhibit the applicability of the results to field situations. Firstly, the material itself, although originally obtained from gully pots will have had its properties altered through the processes of sterilising, drying and sieving. Secondly bacterial degradation of solids creating fine solids and bacterial scums, a process that occurs continuously in the field, could not be taken into consideration. Thirdly, the depth of sediment used for all the tests was low and constant. Bearing in mind the limitations listed above the conclusion must still be that gully pots are effective in trapping coarse sediments that are not subsequently released by stirring of the pot under higher flow rates. A maximum mass released of only 0.2% of the original material input was recorded. In the field, material that is released from a gully pot probably represents wind blown dusts that settle on the surface of the liquor and fine sediments input and not removed on the recession limb of the previous storm event. The laboratory experiments are likely to represent the least polluting conditions that may occur in the field. Bacterial scums and greater depths of sediments could result in significantly larger sediment loads. Even with the low depth of solid used in the tests concentrations of up to 90 mg/l were recorded in the outflow, indicating that the resuspension of solids in gully pots may be of significance in the first flush of intense storms exhibiting high inlet flow rates.

The results indicate that the removal of dissolved material from gully pots may present a greater threat to storm water quality than the

removal of sediment. Greater loads are available for release and since gully liquors may contribute up to 20% of the total storm runoff volume their discharge could result in high dissolved pollutant concentrations. Again the adverse effect is likely to be of greater significance in the first flush.

## 5.1.5 Application of Results to Mathematical Modelling

Since the laboratory results were of limited scope, the calibration of the theoretically derived equations from this data and the application of these equations in modelling gully pollutant contributions to urban runoff may only be considered as tentative.

Considering first the removal of dissolved solids, one of two situations may be proposed. Firstly, above a flow rate of 0.12 1/s complete mixing occurs and, secondly, below this flow rate the percentage mixed is dependent upon flow rate and is defined by linear regression. The model approximation and likely relationship are shown in Figure 42. Now for the purpose of modelling, calculations have to be performed over a finite time element of normally one minute for the solution of equations. In the proposed model the volume of total outflow derived from original gully liquors (the fluid in the gully pot prior to commencement of the storm) was computed. Hence, the concentration of dissolved material derived from gully pots may be calculated as follows:-

Equation 4 becomes:-

 $V_{t} = V_{t-1} e^{-Q_{t}P/V_{0}} \dots (10)$   $V_{t} = Volume of original gully liquor remaining in$  $the pot at time t (litres)
<math display="block">Q_{t} = flow at time t (litres/min)$   $V_{o} = volume of gully pot, = 92.8 litres$  $at t = 1, V_{t}-1 = V_{o}$ 

= percentage mixed

for  $0 \leq Q \leq 0.12$ 

Ρ

 $P = 0.1107 Q_{t} + 0.1986$ 

for 0.12 < Q < 00

P = 1.0

The flow attributable to original gully liquors  $(Q_r)$  is then:-

$$Q_{gt} = V_{t-1} - V_t$$
 ... (11)

and the concentration of dissolved solids attributable to original gully liquors is:-

$$c_t = \frac{Q_g t}{Q_t} \times C_0 \qquad \dots (12)$$

where  $C_0$  = the initial concentration of dissolved solids. When another source of the pollutant occurs then the total concentration in runoff is:-

where  $F(C_p)$  is either a constant pollutant background concentration, e.g. rainfall, or a function describing other processes at work in generating and transporting the pollutant.

It has been mentioned that pollutant contributions from gully pots are likely to be of greater significance during the first flush portion of a storm event. In the literature the term 'first-flush' has generally been loosely used to describe the initial pollutant rich phase of runoff. Having mathematically described the contribution of gully pot liquors to runoff allows us, for the purposes of this research, to make a quantitative definition of first-flush based upon their discharge. Thus, the firstflush is here defined as "that portion of a storm event during which 90% of the total discharge of original gully liquors has occurred". The definition is so phrased for two reasons; firstly, the gully liquors that are present at the start of storm will never be totally removed, therefore, only those that are actually discharged may be considered and, secondly, a value of 90% is necessary since some gully liquor discharge will occur throughout the storm, the use of a 100% definition would result in the first-flush always composing the whole storm. Mathematically the first-flush is defined as being complete when:-

$$\begin{array}{c} t \\ \Sigma & Q & t \\ o & g \\ \hline TQG \end{array} = 0.1$$

where TQG = total volume of gully liquors discharged.

Using this definition for short low intensity storms the first-flush will comprise nearly the whole storm but for longer storm events it will consist of only the initial phases of runoff. The definition is therefore of more use than a fixed time definition (e.g. first 15 minutes) when the polluting potential of gully liquors is being considered.

The equations outlined above are the basis of a subroutine (GULLYS) for computing the contribution made by original gully liquors to flow at the storm sewer outfall. The subroutine was incorporated into the general urban runoff model and is further explained later in the chapter.

In computing the sediment contribution from gully pots both material in suspension prior to the storm event and material released from the bottom sediments have to be considered. Equation (12) can be used to describe the concentration of material from original suspended sediment ( $C_s$ ). The mass available for release is defined according to the model approximation illustrated in Figures 44 and 45, and thus Equations (5) and (6) can be rewritten as:-

for  $0 < (C_t - C_{st}) Q_t < M_t$  $C_t = \frac{K_q}{Q_t} (1 - e^{-Q_t/V_o}) + C_{st} \dots (13)$ 

for 
$$M_t$$
 < (C<sub>t</sub> - C<sub>st</sub>) Q<sub>t</sub> <  $\infty$ 

(that is when all the available mass has been released, then the remaining concentration  $C_t$  can be decreased exponentially)

$$C_t = C_{t-1} e^{-Q_t/V_0}$$
 ... (14)

where

$$M_{t} = M_{o} - \sum_{t=0}^{t} (C_{t} - C_{st}) Q_{t}$$
$$M_{o} = 0 \qquad \text{for} \qquad 0 < Q_{t} < 0.15$$

 $M_{o}$  = 3928 mg for 0.15 <  $Q_{t}$  <  $\infty$ 

 $K_{a}(regression) = (25.88 + 278.02 Q_{t}) \times 60$ 

Q<sub>+</sub> = flow in litres/minute at time t

V = volume of gully pot

K<sub>c</sub> = rate of release in mg/minute for flow q

 $M_{o}$  = total mass available for release

 $C_{+}$  = total concentration in suspension at time t

C<sub>st</sub> = concentration in suspension attributable to original suspended matter.

The performance of this model cannot be verified in the field because of the impossibility of accounting for the solids input to the pot. However, the results of modelling for a constant flow rate and using the continuous expressions, i.e. not discrete minute interval calculations, are shown for one set of laboratory results in Figure 46. The observed results have been compared to the calculated curve for the model calibrated from the experimental data for that test alone and the general calibrated model, where K and M are defined by regression analysis and a mean initial suspended solids concentration was assumed.

#### 5.2. MODELLING URBAN RUNOFF

The rainfall-runoff process in a sewered urban catchment is comprised of several linked complex physical phenomena, but may be conveniently divided into two principal phases. The overland or above ground phase (i.e. the conversion of a rainfall hyetograph to an inlet hydrograph to the sewer system), consisting primarily of hydrological processes. The pipe routing or below ground phase (i.e. the formation of the outlet hydrographs by routing the inlet hydrographs through the sewer system) consisting of mainly hydraulic processes. Both aspects have been the concern of a major research effort in the U.K. at the Institute of Hydrology and the Hydraulics Research Station for the development of new and efficient design techniques for the provision

Fig. 46 OBSERVED AND PREDICTED CURVES FOR THE REMOVAL OF SOLID MATERIAL FOR AN INLET FLOW RATE OF 0.8 1/s



of storm drainage in urban areas. The results of this work have been published in several papers and reports (Institute of Hydrology (1978), Price and Kidd (1978), Kidd (1978), Helliwell et al (1976), Kidd (1976) and have formed the core of the mathematical model employed in this research.

#### 5.2.1 The Overland Flow Model

The hydrological processes involved in the generation of the above ground runoff hydrograph from a measured rainfall input may be considered in three parts:-

- a) the determination of the percentage runoff, or the evaluation of the losses in runoff volume which may occur for any given rainfall volume.
- b) the distribution of these losses in time through the storm.
- c) the distribution of the runoff volume in time over the urban surface.

In the proposed model the Clifton Grove catchment has been divided into a number of sub-catchments that are defined to extend between the manhole points in the storm drainage system. Since any number of gully pots may exist in one sub-catchment, the above ground phase effectively comprises not only the processes of overland flow but also the routing of the flows generated through the gully pots and along the connecting pipework to the relevant manhole junction.

The assumption was made that runoff would only occur off impervious surfaces. Roofed and road surfaces were considered separately and assumed to totally contribute to runoff. However, footpaths and driveways were examined for each sub-catchment and were excluded from area calculations when their contribution to runoff was likely to be negligible e.g. footpaths in grassed areas, driveways sloping away from the road. Figure 47 illustrates the general methodology employed in defining a typical sub-catchment paved area. The assumption that runoff will only occur from impervious areas with no contribution occurring from pervious areas is an acknowledged simplification. In high rainfall intensity storms runoff may occur directly off pervious surfaces,

however, some impervious areas may drain into pervious land and the errors are assumed to cancel one another out. For the U.K., the Road Research Laboratory data (Watkins, 1962) showed percentage runoff from impervious surfaces to be consistently less than 100%. Kidd (1976) felt that pervious area contribution, whilst affecting the volume of runoff had little effect on the mechanics of the conversion from rainfall hyetograph to inlet hydrograph. However, it is clear that however the areas are defined as contributing, or not contributing, to runoff, they must have a significant effect upon the total predicted discharge.

Having acknowledged some of the assumptions necessary for the practical implementation of the above ground model the mechanics of the model is as follows. The calculation of the inlet hydrograph from the measured rainfall was achieved through the use of the three submodels:-

a) Percentage runoff submodel

- b) Depression storage submodel
- c) Surface routing submodel.

5.2.1.1 Percentage Runoff Submodel

The variation of percentage runoff with several catchment characteristics has been investigated by Stoneham & Kidd (1977) using data from 368 storms and 14 catchments. Their research led to the proposal of a regression equation for general application in the estimation of percentage runoff:-

PRO = 0.92 PIMP + 53 SOIL + 0.65 UCWI - 33.6

where PRO = percentage runoff PIMP = percentage impervious area SOIL = a soil index UCWI = urban catchment wetness index

The equation evaluates in an empirical fashion all likely losses that may occur over a catchment from processes such as surface infiltration, depression storage, infiltration and exfiltration from pipes, reduced catch by roofs and evaporation. A statistical treatment of the data was adopted due to the random and unquantifiable nature of

losses making the catchment averaged approach more appropriate than a deterministic solution of all the contributing processes.

For the purposes of this study the measured total storm discharge was used to compute the percentage runoff. This procedure has been termed as forcing the predicted discharge. Although unsatisfactory for use in 'free' simulation or design procedures, the technique was consistent with the aims of this research, i.e. examination of pollutant loadings from gully pots. 'Forcing' the percentage runoff in this manner increased the reliability of the pollutant runoff prediction and is thought to be valid.

Having ascertained the value of the percentage runoff it remains to subtract the resultant loss from the rainfall hyetograph to create a net effective rainfall hyetograph. A number of options are available, of which there are 3 main categories:-

- a) The constant proportion loss model (CPL). Losses are distributed as a fixed proportion of the rainfall intensity (Figure 48).
- b) The Phi-index model. Losses take place at a constant rate.
- c) The variable proportional loss model (VPL). This employs a Horton type equation to allow for a higher proportion of loss at the beginning of a storm than at the end.

The various loss models have been documented and evaluated in an International Workshop on Rainfall-Runoff Processes over Urban Surfaces (Institute of Hydrology, 1978) using data from the Netherlands, Sweden and the U.K. for 20 catchments. They concluded that the CPL and VPL models performed significantly better than the Phi-index mcdel. The CPL model has been commonly employed, is simpler than the VPL model and was adopted for this research.

5.2.1.2 The Depression Storage Submodel

Depression storage is comprised of the rainfall that falls on impervious surfaces but never enters the sewer system. It can be considered as a combination of water held on the surface by surface

Fig. 48 EXAMPLE OF THE APPLICATION OF THE CONSTANT PROPORTIONAL LOSS MODEL TO A RAINFALL HYETOGRAPH







tension forces, as in initial wetting, and water contained in surface hollows as puddles. These two factors may be lumped together and treated as an even depth of water over the subcatchment. Runoff is assumed to occur only after the depression storage has been satisfied and as such the depression storage assigned in millimetres is subtracted from the initial portion of the hyetograph (Figure 49). This assumption is clearly not an altogether realistic appraisal of what can occur in practice since runoff may occur before storage has been satisfied in more remote areas of the catchment. However, the complexity of describing such a situation which could vary greatly from one subcatchment to another, justifies its use.

The analysis of data from the International Workshop led to the development of a relationship between depression storage and catchment slope (Figure 50) and a generally applicable regression equation for the evaluation of depression storage:-

DEPSTO = 0.77 SLOPE  $^{-0.49}$  (R = 0.85)

The equation was employed in this study for the calculation of paved area storage values. No such equation exists for assigning a depression storage value to roofed areas but logically it is likely to be very small. Kidd (1976) had assumed a value of 0.1 mm, to cover surface wetting of roofs and this was adopted.

Depression storage was deducted from the net hyetograph after the application of the constant proportional loss model. This approach was considered to be more consistent with reality than the procedure of first deducting storage before applying the loss model, since infiltration losses will occur contemporaneously with the filling of depression storage.

### 5.2.1.3 Surface Routing Submodel

Surface routing is the process by which the effective hyetograph after the subtraction of losses is attenuated over the urban surface to form the inlet hydrograph at a manhole point. A wide variety of models have been used for the description of the surface routing process, these divide into two fundamental philosophies, the deterministic approach and the conceptual approach.

Fig. 50 RELATIONSHIP BETWEEN CATCHMENT SLOPE AND DEPRESSION STORAGE (KIDD, 1978)



Fig. 51 FORMATION OF A HYDROGRAPH FROM A ONE MINUTE RAINFALL INTENSITY USING THE LINEAR RESERVOIR MODEL.







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The deterministic approach to the problem attempts to mathematically simulate all of the specific processes involved, whilst making the minimum of simplifying assumptions to obtain a working model. The main objections to the use of deterministic modelling are the large data requirements for the accurate description of a subcatchment surface, which is generally not feasible (particularly in the consideration of design methods) and the corresponding lengthy computing times needed.

The conceptual approach may be considered as employing simplified descriptions of the actual physical processes involved. Conceptual models attempt to describe a system in terms of a reduced number of parameters that can be calibrated from easily determinable system characteristics. The resultant equations have been loosely termed as 'lumped parameter' models. The general method adopted for the simulation of the overland flow is known as storage routing, where storage is considered as the lumped parameter.

A number of such conceptual models were examined by the International Workshop (Institute of Hydrology, 1978) and their performance compared over 188 rainfall-runoff events on 16 urban catchments. The models examined were:-

- a) Linear reservoir
- b) Non-linear reservoir
- c) Non-linear reservoir with time-lag
- d) Nash cascade
- e) Muskingum
- f) Time of Entry

Regression equations were developed for the variable parameters (after optimisation) against catchment characteristics (slope and overland flow length). All the relationships showed a marked similarity and explained between 40% and 55% of the variance in the optimised parameter. An analysis of variance for all of the models, calibrated from the derived regression equations showed that in terms of overall fit the linear reservoir, nonlinear reservoir and Muskingum models were superior to the others. In terms of peak estimation there was little to chose between any of the models. The Workshop concluded that the choice of surface routing model

was less critical than the manner in which it was utilised, however a non-linear model could be expected to perform marginally better than a linear one.

Of all the conceptual models the single linear reservoir is one of the simplest in terms of both application and computing, whilst retaining a high degree of accuracy. It has been used by a number of researchers for the simulation of surface runoff, Viessman (1966), Watt and Kidd (1975) and Newmann & Marr (1976).

The basic equations that describe the storage routing relationship are:-

 $\frac{dS}{dt} = i - q$ ... (15) = ∝q<sup>h</sup> ... (16) S S = storage outflow q = i inflow = t = time

Differentiating equation (16) and substituting in (15)

 $n \approx q^{n-1} \frac{dq}{dt} = i - q \dots (17)$ 

The linear reservoir is a specific solution of the above equation when n = 1, leaving  $\backsim$  (the storage coefficient) as the only parameter to be determined. The storage coefficient has a fixed value for any one rainfall event on a given catchment. In respect of the non-linearity of the process, the International Workshop defined  $\backsim$  as a function of the rainfall intensity making the model pseudo linear.

$$\alpha = \alpha_{2} I^{-0.4} \dots (18)$$

where I is defined as the average rainfall intensity for the most intense 10 minutes of the storm. Regression analysis for optimized  $\sim_2$  values led to the proposal of the following equation for determining  $\sim_1$  from the subcatchment slope and overland flow length.

$$\approx_2 = 1.43 \text{ slope}^{-0.4} \text{ length}^{0.22}$$
 ... (19)

These equations are applicable only to paved areas. For roofed areas no generally applicable regression equations have been developed for defining routing constants. The technique commonly employed is the optimization of the parameters for a given event to produce the 'best fit' hydrograph. For a catchment, such as Clifton Grove, containing some 40 subcatchments optimization for each constant would prove to be a lengthy procedure. Alternative approaches to the problem are the assignation of a single averaged routing constant for all the subcatchments and to use this in optimization procedures, or the assignation of empirical routing constants for each subcatchment. The latter course was selected to facilitate the rapid application of the model. Values assigned for the roof routing constant, were 1.0, 1.5, 2.0 and 2.5 minutes, the values chosen being dependent upon the length of flow to the manhole point of interest. Future research may improve the models performance by rationalisation of the roof routing constants.

To determine the runoff hydrograph Equation 17 is applied over specified time intervals (usually one minute) and assumes that the runoff response for the rainfall intensity over this time increment is a function only of that particular rainfall intensity and not adjacent intensities. Thus a one-minute hydrograph is generated from the one minute rainfall intensity (Figure 51). The complete runoff hydrograph is obtained by addition of the individual one-minute hydrographs (Figure 52). The runoff decay curve is obtained from Equation 17 by considering the case when t > one minute, i is then equal to zero and for the linear reservoir n = 1.

 $\int_{1}^{t} \frac{dt}{\alpha} = \int_{q_0}^{q} \frac{dq}{q} \qquad \dots (20)$ 

After integration and simplification this yields:-

$$q = q_0 e^{-(t-1)/\alpha}$$
 ... (21)

 $\boldsymbol{q}_{0}$  is obtained by equating the area under the input intensity to the area under the response curve.

$$q_0 = \frac{1}{(\alpha + 0.5)}$$
 ... (22)

where i = rainfall intensity in mm/min. and  $q_0$  = runoff per unit area in litres/M<sup>2</sup>

Thus for a storm event the hydrograph is defined as follows:-

q(0)	=	0
q(1)	=	i(1) / ( $\alpha$ + 0.5)
q(2)	=	$i(2) / (\alpha + 0.5) + i(1) / (\alpha + 0.5) e^{-1/\alpha}$
q(3)	=	$i(3) / (\alpha + 0.5) + i(2) / (\alpha + 0.5)e^{-1/\alpha} + i(1) / (\alpha + 0.5) = 0$

$$q(n) = i(n) / (\alpha + 0.5) + i(n-1) / (\alpha + 0.5)e^{-1/\alpha} ... + i(1) / (\alpha + 0.5)e^{-1/\alpha}$$

It can be seen that the decay curve continues for an infinite time and in practice a cutoff point is necessary when the response may be effectively ignored. This has been selected as being the point when 99% of the total area has been included in calculations. Since the area under the curve from time T to infinity is equal to  $\propto q_0 e^{-T/\alpha}$  and the total area is equal to  $\propto q_0$ , then the cutoff time T can be determined from:-

$$\frac{\alpha q_o e^{-T/\alpha}}{\alpha q_o} = 0.01$$

thus T = 4.6 ∝

From this expression the cutoff time for the response curve for each minute increment may be determined.

The complete hydrograph per unit area derived as above is converted to the inlet hydrograph by multiplying by the subcatchment area.

### 5.2.2 The Pipe Routing Model

The pipe routing, or below ground, phase of urban runoff consists of combining the inlet hydrographs to each pipe at each time interval and routing the summed inlet hydrograph through the pipe to the next downstream junction. The procedure is then repeated for the following pipe in the network and so on through the system to the outfall. The modelling of this process is much more amenable to a deterministic approach than the above ground phase due to the simpler geometry involved. The differential equations that describe gradually varied, unsteady flow in uniform channels are derived from consideration of the conservation of energy and the conservation of mass of an element of water moving in a downstream direction. The equations have been presented in many different forms and are known as the St. Venant equations, they may be written as:-

a) The dynamic equation:-

$$s_{o} - s_{f} = \frac{\delta y}{\delta x} + \frac{v}{g} \frac{\delta v}{\delta x} + \frac{1}{g} \frac{\delta v}{\delta t} \qquad \dots (23)$$

b) The continuity equation:-

 $0 = y_{m} \frac{\delta v}{\delta x} + v \frac{\delta y}{\delta x} + \frac{\delta y}{\delta t}$ 

where y = water depth

ym = mean water depth
x = distance in a downstream direction
t = time

s<sub>o</sub> = channel slope

s<sub>f</sub> = friction slope

The equations involve two simplifying assumptions:-

- a) that the flow is so gradually varied that the vertical acceleration of the water particles may be neglected,
- and
- b) that the resistance coefficient is the same for the given depth and mean velocity regardless of whether the flow is uniform or non-uniform.

Many methods of solving the St. Venant equations have been proposed involving varying degrees of approximation or assumption. The interrelationship of many of these solutions has been reviewed by Weinmann and Laurenson (1979). Numerical methods of solution for both equations have been termed 'complete dynamic models' and involve the conversion of Equations (23) and (24) into algebraic expressions that can be solved for y and v at finite increments of x and t. Although solutions of this type give the most complete description of flow in channels accuracy cannot be guaranteed since they remain based on the underlying assumptions in the

formulation of the St. Venant equations. While all flood routing models use the equation of continuity in a similar manner further groups of models exist that can be distinguished according to the number of terms retained in the dynamic equation (Equation (25)).



Models that neglect the acceleration term have been called 'approximate dynamic models' whilst models that assume that the flow is at normal depth and thereby reduce the dynamic equation to  $s_f = s_0$ are known as 'kinematic wave models'. Such approximations of the full St. Venant equations are of interest since, not only are they simpler, but also solutions may be computed more easily than for the full equations. Several schemes that have been utilised in the solution of the full and approximated equations are documented in the Flood Studies Report (NERC, 1975). Bettess and Price (1976) reviewed a number of the schemes for the solution of the St. Venant equations with particular reference to their accuracy and precision in routing flow along a pipe. Of the 'approximate models' tested they concluded that the Muskingum-Cunge method, a non-linear kinematic wave model, was comparable in accuracy with the approximate dynamic model but was computationally considerably faster. They recommended that the method be utilised for the design of storm-sewer systems or the simulation of flows in such systems where the speed of computation is an important factor. However, a scheme for the solution of the full equations was still considered to give the highest accuracy.

Subsequently the Muskingum-Cunge method was employed by Price and Kidd (1978) in a new urban storm sewer design and flow simulation package aimed at improving upon and superseding the conventional TRRL method (Watkins, 1962). The Muskingum-Cunge method, as developed by Price and

Kidd, was utilised in this research.

5.2.2.1 Derivation of the Muskingum-Cunge Method.

The storage of water within a pipe length is defined by the difference between the average value of the inflow  $(Q_{inp})$  and the outflow  $(Q_{out})$  during the time interval  $\Delta t$ , known as the routing period, i.e.

 $\frac{\Delta S}{\Delta t} = Q_{inp} - Q_{out} \qquad \dots (26)$ 

If  $Q_{inp}^t$  and  $Q_{inp}^{t+1}$  are the rates of inflow at the start and end of the routing period and  $Q_{out}^t$  and  $Q_{out}^{t+1}$  are the respective rates of outflow then the change in storage is given by:-

$$S_2 - S_1 = \left[ \underbrace{\left( \underbrace{Q_{inp}^t + Q_{inp}^{t+1}}_{2} \right) - \underbrace{\left( \underbrace{Q_{out}^t + Q_{out}^{t+1}}_{2} \right)}_{2} \right] \Delta t \dots (27)$$

The storage at any given time may also be expressed in terms of the relative significance given to outflow and inflow by use of a weighting parameter,  $\epsilon$  :-

 $S = \beta \left( \epsilon Q_{inp} + (1 - \epsilon) Q_{out} \right) \qquad \dots (28)$ 

where  $\beta$  is a further model parameter.

Substituting this expression for  ${f S}$  in Equation (27) and rearranging:-

$$\begin{aligned} Q_{out}^{t+1} & (\beta - \beta \varepsilon + \frac{1}{2} \Delta t) = Q_{inp}^{t} & (\beta \varepsilon + \frac{1}{2} \Delta t) - Q_{inp}^{t+1} & (\beta \varepsilon - \frac{1}{2} \Delta t) \\ &+ Q_{out}^{t} & (\beta - \beta \varepsilon - \frac{1}{2} \Delta t) \end{aligned}$$

$$C_{1} = C_{1}Q_{inp}^{t} + C_{2}Q_{inp}^{t+1} + C_{3}Q_{out}^{t} \qquad \dots (29)$$
where  $C_{1} = \frac{\left(\beta\epsilon + 1/2\Delta t\right)}{\left(\beta - \beta\epsilon + 1/2\Delta t\right)}$ 

$$C_{2} = -\frac{\left(\beta\epsilon - 1/2\Delta t\right)}{\left(\beta - \beta\epsilon + 1/2\Delta t\right)}$$

$$C_{3} = \frac{\left(\frac{\beta - \beta\epsilon - 1/2\Delta t}{\beta - \beta\epsilon + 1/2\Delta t}\right)}{\left(\beta - \beta\epsilon + 1/2\Delta t\right)}$$

Equation (29) is known as the Muskingum equation and is the finite difference scheme whereby given an input hydrograph at the top end of a pipe and initial conditions down the pipe it is possible to calculate the discharge hydrograph at the bottom end of the pipe.

It now remains to define the model parameters  $\beta$  and  $\epsilon$ . Although the Muskingum method outlined as above was originally believed to be purely empirical, Cunge (1969), studying the finite difference scheme used in the computation of the outflow hydrograph, found that it approximated the 'diffusion analogy' equation. This enabled the parameters  $\beta$  and  $\epsilon$  to be related to the geometry and hydraulic characteristics of the pipe reach under consideration. Cunge (1969) derived the following equations for the two parameters which have subsequently been utilised for the routing of flow in pipes (Price and Kidd, 1978):-

 $\beta = \frac{L}{w} \qquad \dots \quad (30)$ 

where L = length of the pipe w = wave speed

Hence  $\beta$  can be interpreted as the time of travel of the flood peak down the pipe

and  $\varepsilon = \frac{1}{2} \left( 1 - \frac{\overline{Q}}{\overline{B} s_0 L \overline{W}} \right)$  ... (31)

where s = pipe slope

 $\overline{Q}$  = mean discharge

 $\overline{B}$  = mean surface breadth

w = mean wave speed

A value for the mean wave speed  $(\overline{w})$  can be found from the equation

$$\overline{W} = \frac{1}{Q_{fb}} \int_{0}^{Q_{fb}} \frac{dQ}{dA} dQ \qquad \dots (32)$$

$$\overline{W} = \frac{1}{Q_{fb}} \int_{0}^{d} \frac{1}{B} \left(\frac{dQ}{dy}\right)^{2} dy \qquad \dots (33)$$

with Q defined by the normal depth relationship of the Colebrook-White equation:-

 $Q = A (32g r S_0) \log_{10} (14.8r/k_s) ... (34)$ 

where g = acceleration due to gravity

d = pipe diameter

k = roughness coefficient

Q<sub>fb</sub> = full bore discharge

For typical values of  $k_s$  and for a water depth of 0.3d, then  $\overline{w} = dQ/dA$ , this value for  $\overline{w}$  is used in Equation (30). However, to give an appropriate mean vlaue for  $\varepsilon$ ,  $\overline{w}$ ,  $\overline{Q}$  and  $\overline{B}$  in Equation (31) are evaluated for a water depth of 0.5d.

The coefficients,  $C_1$ ,  $C_2$ ,  $C_3$ , have now been defined purely in terms of the pipe geometry and other fixed coefficients, therefore, the method has become known as the 'fixed parameter' Muskingum-Cunge method. In reality the wave speed and the amount of storage will vary with the discharge and it is possible to devise a scheme whereby the parameters  $\beta$  and  $\varepsilon$  are dependent upon the discharge. Such an improvement has been examined (Price and Mance, 1978), however, despite the marginal improvement in accuracy the method was judged to be unjustified because of the larger computing costs.

# 5.3. APPLICATION OF THE MODELS. THE COMPUTER PROGRAMME

The sub-models and pollutant removal expressions introduced in Sections 5.1 and 5.2 form the basis of the Trent Urban Runoff Simulation (T.U.R.S.). The computer programme is written in FORTRAN V language and was executed on a DEC 20/PDP 11 computer at Trent Polytechnic, Nottingham. The programme is currently limited to a sewer system containing up to 50 manhole points and rainfall events up to 200 minutes. These values are convenient for the Clifton Grove catchment and for the data available, but could be easily changed by re-diminishing the matrix storage allocations for application to larger catchments or events. In this format the programme required 9 pages of computer storage and took 10 to 15 seconds of CPU time to execute for the storm events encountered. The programme itself is

presented in Appendix 1.4. In its present form T.U.R.S. is strictly a simulation technique, its principal functions can be summarised as the computation of input hydrographs, output hydrographs and original gully liquor contribution to these hydrographs at any manhole point in the sewer system. In addition the dissolved pollutant concentrations derived from the discharge of original gully liquors can be computed. Any pollutant may be considered whose concentration in gully liquors prior to the storm event was known or was predictable. The programme comprises the following divisions and subroutines as illustrated in the flow chart. (Figure 53).

- a) MAIN PROGRAMME: performs data input and output operations, programme management and the calculation of relevant statistics;
- b) HYETO SUBROUTINE:- formulates a hyetograph from raw rain-gauge data, calculates the average intensity for the most intense
   10 minutes and applies the constant proportional loss model;
- c) DEPSTO SUBROUTINE:- calculates the depression storage values for each paved sub-catchment. Derives the hyetographs applicable to each paved sub-catchment and to roofed areas.
- d) LINRES SUBROUTINE:- calculates the storage routing coefficient for each paved sub-catchment. Computes the inlet hydrograph for paved and roofed sub-areas from each sub-catchment. Sums the two inlet hydrographs to form a total inlet hydrograph for each sub-catchment.
- e) MUSK SUBROUTINE (supplied by courtesy of the Institute of Hydrology and the Hydraulics Research Station):- calculates the Muskingum pipe routing coefficients from pipe geometry.
- f) ROUTE SUBROUTINE: routes the input hydrographs through the pipe network to form the output hydrographs at each manhole point.
- g) GULLYS SUBROUTINE:- computes the contribution to flow attributable to the input of original gully liquors at each manhole point. Calls ROUTE to form the output hydrographs attributable to original gully liquors at each manhole point. Calculates the concentration of pollutants at the outfall derived from original gully liquors.

Fig. 53 SIMPLIFIED FLOW CHART FOR THE TRENT URBAN RUNOFF SIMULATION.







#### 5.3.1.Data Preparation and Input

The data input requirements of the programme are four fold :-

- a) Fixed catchment data, i.e. pipe system data and sub-catchment characteristics.
- b) Rainfall event data
- c) Observed event data, i.e. measured flows, pollutant concentrations.
- d) Programme control parameters.

The catchment data was derived from a combination of field measurements and the use of a site map supplied by Nottingham City Council. Initially the total catchment was divided into a number of sub-catchments defined by the number of manhole points in the pipe system (Figure 54). The paved and roofed areas contributing to each sub-catchment were determined by planimeter off the site map. The average overland flow length and the number of gully pots draining each sub-catchment were also determined from the map. The mean sub-catchment slope expressed as a percentage was determined from field measurements. The pipe system was numbered according to the branch nomenclature employed in the TRRL method (Figure 54). Correct numbering is of some importance since the addition and routing of hydrographs through the system is dependent upon it. The sub-catchments were numbered according to the pipe length to which they contributed. The pipe dimensions, length, slope and radius were determined from field measurements. The roof area storage routing coefficients which were assigned empirically (see 5.2.1.3.) were also input with this data. Tables 21 and 22 are a summary of the sub-catchment and sewer system data for the Clifton Grove catchment.

Storm event data may be input in one of two possible formats, either as a rainfall hyetograph or as raw rain-gauge data available from the translation of rain-gauge logger tapes, i.e. time/number of tilts per minute. In the latter case the dynamic calibration for the rain-gauge employed, the intensity applicable to each tilt, must also be input. It is necessary to make some assumptions in formulating the hyetograph from rain-gauge data in order that a situation of least error applies to all storms, these are:-
Fig.54 STORM SEWER PIPE NUMBERING FOR THE CLIFTON GROVE ESTATE.



INDED 21 DUD-Catcimento dava	TABLE	21	Sub-catchment	data
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Pipe No	Pipe No Slope (%)		Slope (%) Paved Area (m <sup>2</sup> )		Roof Area (m <sup>2</sup> )	Length (m)		
$\begin{array}{c} 1.00\\ 1.01\\ 1.02\\ 2.00\\ 1.03\\ 1.04\\ 1.05\\ 3.00\\ 3.01\\ 3.02\\ 4.00\\ 3.03\\ 3.04\\ 3.05\\ 1.06\\ 1.07\\ 1.08\\ 5.00\\ 5.01\\ 5.02\\ 5.03\\ 5.04\\ 5.05\\ 6.00\\ 6.01\\ 6.02\\ 6.03\\ 6.04\\ 6.05\\ 6.06\\ 5.06\\ 1.09\\ 7.00\\ 1.10\\ 8.00\\ 8.01\\ 8.02\\ 1.11\\ 9.00\\ 9.01\\ 1.12\\ 1.13\end{array}$	$\begin{array}{c} 2.734\\ 1.813\\ 2.164\\ 3.813\\ 2.910\\ 2.995\\ 4.300\\ 1.500\\ -\\ .254\\ 2.987\\ -\\ .254\\ 2.987\\ -\\ .3.366\\ 2.000\\ .500\\ 5.102\\ 4.320\\ 4.000\\ 3.699\\ 3.903\\ .711\\ 2.000\\ 2.000\\ 4.065\\ 4.640\\ 4.120\\ 3.336\\ -\\ .315\\ 4.672\\ .440\\ 3.800\\ 4.400\\ 1.226\\ 1.200\\ 2.000\\ 2.482\\ 2.838\\ -\\ -\\ -\\ -\\ -\\ -\\ -\\ -\\ -\\ -\\ -\\ -\\ -\\$	960.00 620.00 520.00 320.00 410.00 760.00 755.00 1200.00 0.00 570.00 0.00 740.00 220.00 420.00 1597.00 688.00 324.00 940.00 350.00 260.00 210.00 100.00 264.00 270.00 280.00 1310.00 560.00 560.00 560.00 540.00 540.00 100.00 560.00 560.00 540.00 560.00 500.00 500.00 500.00 500.00 560.00 500.00 0.00	616.25 430.75 631.25 0.00 419.75 385.00 0.00 1504.25 0.00 747.50 0.00 521.25 0.00 916.00 0.00 1136.25 0.00 0.00 922.25 385.00 1255.75 0.00 0.00 922.25 385.00 1255.75 0.00 603.50 552.25 735.00 317.50 860.75 612.00 1221.75 154.00 338.50 0.00 0.00 0.00 0.00 0.00 0.00 1065.27 793.00 0.00 0.00 0.00 0.00 0.00 0.00 0.00 0.00 0.00 0.00 0.00 0.00 1065.27 793.00 0.0	$\begin{array}{c} 70.0\\ 90.0\\ 65.0\\ 90.0\\ 60.0\\ 70.0\\ 100.0\\ 90.0\\ 0.0\\ 0.0\\ 0.0\\ 90.0\\ 70.0\\ 0.0\\ 90.0\\ 70.0\\ 90.0\\ 85.0\\ 80.0\\ 95.0\\ 80.0\\ 95.0\\ 70.0\\ 70.0\\ 70.0\\ 70.0\\ 70.0\\ 95.0\\ 80.0\\ 95.0\\ 85.0\\ 85.0\\ 85.0\\ 85.0\\ 85.0\\ 85.0\\ 85.0\\ 85.0\\ 90.0\\ 65.0\\ 80.0\\ 90.0\\ 80.0\\ 90.0\\ 80.0\\ 90.0\\ 80.0\\ 90.0\\ 80.0\\ 90.0\\ 80.0\\ 90.0\\ 80.0\\ 90.0\\ 80.0\\ 90.0\\ 80.0\\ 90.0\\ 80.0\\ 90.0\\ 80.0\\ 90.0\\ 80.0\\ 80.0\\ 90.0\\ 80$				

# TABLE 22 Sewer System Data

1. NOT

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Pipe No	Length (m)	Slope (%)	Diameter (m)	Gully Pots
1.00 1.01 1.02 2.00 1.03 1.04 1.05 3.00 3.01 3.02 4.00 3.03 3.04 3.05 1.06 1.07 1.08 5.00 5.01 5.02 5.03 5.04 5.05 6.00 6.01 6.02 6.03 6.04 6.05 6.06 5.06 1.09 7.00 1.10 8.00 8.01 8.02 1.11 9.00 9.01 1.12 1.13	79.6 47.0 49.9 10.7 35.0 86.5 70.7 13.1 35.6 95.0 19.1 76.0 34.0 32.4 75.5 15.5 58.9 96.0 71.0 48.4 87.0 30.0 25.3 27.0 8.2 54.0 47.5 29.7 36.0 78.6 20.0 61.5 57.5 66.0 43.0 59.4 26.7 44.2 58.0 45.2 35.5 92.0	3.636 1.504 3.175 4.484 3.226 3.759 4.739 2.092 4.717 .440 .429 3.968 2.667 1.866 2.364 2.558 .639 5.051 5.051 4.854 4.115 1.186 1.799 2.646 7.752 4.405 4.854 2.342 3.650 4.525 2.041 5.882 1.972 2.786 4.016 .485 .100 4.386 2.288 4.405 5.102 .484	.2286 .3810 .2286 .3810 .2286 .3810 .2286 .3810 .2286 .3810 .2286 .3810 .2286 .3810 .2286 .3810 .2286 .3810 .2286 .3810 .2286 .3810 .2286 .3810 .2286 .3810 .2286 .3810 .2286 .3810 .2286 .286 .2286 .3810 .2286 .3810 .2286 .3810 .2286 .3810 .2286 .3810 .3810 .3810	6342264400340031375262231222503444251122200

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Fig. 55 EXAMPLE OF THE FORMATION OF A HYETOGRAPH FROM RAINGAUGE DATA



- a) the tipping bucket is half full at the beginning and end of a storm.
- b) the storm commenced at a time prior to the first tilt that is equal to the time gap (in minutes) between the first and second tilt.
- c) the storm ended at a time after the final tilt that is equal to the time gap between the final and penultimate tilts.

An example of the transformation of raingauge data to form a storm hyetograph is shown in Figure 55.

The programme operation is determined by the input of five programme control parameters (M1 - M5). The parameters determine the type of data input, the passage followed through the simulation subroutines and the nature of the output. The following alternatives are available:-

- M1 = 1 Raw raingauge data to be input
- M1 = 0 Storm hyetograph to be input
- M2 = 1 Contributions from original gully liquors wanted
- M3 = 1 Observed flow data to be input
- M4 = 1 Observed gully pollutant concentrations to be input
- M5 = 1 Graphical output

In addition it is necessary to input the desired output device number (IWRI) either a terminal or line printer and the constant proportional loss (CPL) coefficient. The latter is normally set to 1.0 for an initial run with a data set. For subsequent runs the predicted and observed discharges can be equated to determine the CPL coefficient and, therefore, 'force' the predicted discharge. Provision is also made for the input of observed flow data and observed gully pollutant concentrations, when these are available, for the calculation of relevant statistics and graph plotting.

#### 5.3.2 Output

Output may be obtained in two forms, A4 tabulated data and graph plots. Specifically these are as follows:-

a) Tabulation of calculated data:-

Time (Hours) Rainfall Intensity (mm/hour) Flow at outfall (l/s)

M2 = 1 Gully liquor contribution to flow at outfall (1/s)

M4 = 1 Gully pollutant concentration at outfall (mg/l)

It is a simple task to adjust the programme in order that the parameters may be output for other points in the system.

b) Tabulation of observed data:-

M3 = 1	Observed	times
--------	----------	-------

M3 = 1 Observed flows

- M4 = 1 Observed gully pollutant concentrations
- c) Graphical output (M5 = 1). Up to three plots are available in any execution:-
  - 1. Predicted runoff, Observed runoff, Hyetograph
  - 1a. Predicted runoff, hyetograph
  - 2. Predicted runoff, predicted gully liquor contribution, hyetograph
  - 3. Predicted runoff, predicted pollutant concentration, observed pollutant concentration
- d) Tabulation of storm statistics. A variety of statistics are available for output again, dependent upon the programme control parameters:-
  - 1) Storm duration (hours)
  - 2) Total rainfall (mm)
  - 3) Total rainfall volume on impervious area (litres)
  - 4) Depression storage loss (litres)
  - 5) Other losses, e.g. infiltration (litres)
  - 6) Average intensity (mm/hour)

7)	Average intensity for the most intense ten
	minutes (mm/hour)
8)	Total predicted discharge (litres)
M3 = 1 ( 9)	Total observed discharge (litres)
{ 10 )	Percentage runoff from impervious areas
( 11)	Total original gully liquor discharge (litres)
(12)	Percentage gully liquor contribution to total
(	runoff (litres)
(13)	First flush time (hours)
$M2 = 1 \begin{pmatrix} 14 \end{pmatrix}$	First flush volume (litres)
(15)	Percentage gully liquor contribution to the first
(	flush
(16)	Percentage original gully liquor discharged to
(	the total original gully liquor volume
(17)	Total predicted pollutant mass (g)
M4 = 1 ( 18)	Total observed pollutant mass (g)

#### 5.3.3 Application of the Model. Results and Conclusions.

The accurate calibration of the model was restricted by two factors, firstly the limitations of the model itself and secondly the small size and the limitations of the available data set. The limitations of the model have been discussed previously and could be minimised with calibration from a good set of data. The principal problems of calibration and therefore verification of the model's performance arise from restrictions imposed by the accuracy and reliability of the observed data. These in turn may be primarily ascribed to the use of lithium dilution for the determination of flow. Firstly, sampling and hence flow determination commenced only when discharge exceeded about 8 1/s. Many of the storms monitored were of low rainfall intensity and for the small Clifton Grove catchment resulted in long periods when discharge at the outfall was less than 8 l/s. As a consequence of this, for long periods of some storms no samples were collected and, therefore, no flow data was available, e.g. Storm 21 (Figure 56). Therefore, the total observed discharge could not be determined accurately and 'forcing' the predicted hydrograph by equating the observed and predicted total discharges to determine the CPL coefficient was impractical. The alternative approach that was adopted for the majority of storms lay in forcing the peak heights of the predicted hydrograph to be of the same magnitude as the observed hydrograph.



Figure 57



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Figure 59



1.19.1

Although it has been assumed above that the first sample was collected at a flow rate of about 8 1/s this may not have always been the case. Prevention of triggering, either by obstructing objects such as twigs washed down the sewer, or just unreliable action of the float switch was observed to result in portions of some large storms being missed, e.g. Storms 12 (Figure 57) and 17. Again this rendered the discharge based calculation of the CPL coefficient impractical.

The reliable use of smaller storm events for evaluating the model was restricted by two other factors in addition to those outlined above. Firstly sampling was conducted over a minimum time period of 2 minutes, therefore, observed flow rates represent an average value over the sampling time period. This results in a lack of resolution in the observed hydrograph. A second restriction for small events, where only 3 or 4 samples were collected, was the length of time required for dosing to reach equilibrium. Due to these two restrictions observed flow rates for small events of 10 to 15 minutes duration may only be considered as approximate.

Another problem with flow determination was the occurrence of spuriously low lithium concentrations for the occasional sample in an otherwise reliable set of data. This is probably attributable to fluctuations in the lithium dose rate and may be due to temporary partial blocking of the dosing orifice by small dust particles. When this occurred an estimated flow value was obtained by interpolation from adjacent flow values and consideration of the shape of the predicted hydrograph.

The timing of raingauge tilts and the commencement of sampling was achieved through the use of two separate clocks. Although these were both set to standard time when the instrumentation was prepared in readiness for a storm event, timing may not have been exactly synchronous when batteries started to run down. For some storms one clock or the other did not function and when this occurred observed and predicted peak flow rates were arranged to occur at the same time, e.g. Storms 7 (Figure 58) and 24.

When observed flow values for storm events where rainfall had been recorded were either absent or poor (Storms 6, 9, 10 and 11) then the predicted hydrograph was used as an estimation of flow for the purpose of calculating pollutant loadings, the first flush and gully liquor contributions. In this situation the CPL coefficient was estimated from

knowing the number of samples taken and assuming that sampling commenced at 8 l/s. The coefficient being determined by the reduction of the predicted hydrograph necessary to be consistent with the number of samples collected.

As a result of the limitations of the data outlined above no monitored storm was accurate in all facets to enable a reliable calibration of the model to be conducted. How the model was applied and the restrictions applicable to each storm are listed in Table 23. The lithium dilution method of flow determination is concluded as being unreliable and not of sufficent accuracy for model calibration on the Clifton Grove catchment. Accurate calibration will be possible when some alternative continuous flow measurement system is installed. Since it was impractical to perform a statistical error analysis on the performance of the model, conclusions can only be of a general nature and are presented after due consideration of the limitations applicable to each storm event.

Firstly, the overall shape of the predicted hydrographs would appear to be in broad agreement with that observed. This is more evident for the longer storms of greater intensity, e.g. Storms 7 and 24 (Figures 58 and 59) where a larger number of observed flow values were available. When the timing of rainfall and runoff was considered to be synchronous, then the time to peak flow for predicted and observed hydrographs was in reasonable agreement, e.g. Storms 9 (Figure 60), 10, 11 and 21 (Figure 56). However, for other events there was some tendency for the predicted peak to occur prior to the observed peak, as in Storms 19 and 20 (Figure 61. It is possible that the effect can be ascribed to the non-optimization of the roof storage routing constants.  $\checkmark$  values assigned are possibly too low and result in a rapid runoff response from roofs. The early contribution from roofed areas to runoff at the outfall would logically be expected but is possibly not as rapid as the model currently predicts. For low intensity storms the initial periods of runoff, up to 30 minutes, can be attributed solely to roof runoff, e.g. Storm 15. For Storm 11 (Figure 62), which consisted of a total rainfall of only 0.234 mm, the predicted hydrograph was comprised only of roof runoff. Few conclusions can be made on the effectiveness of the model in determining peak flow rate since this parameter was most frequently used to determine the CPL coefficient and optimise the predicted hydrograph. However, for situations where no adjustment of the CPL coefficient was necessary, i.e. CPL = 1.0,

	Storm	Quality Data	Flow Data	Rainfall Data	Model used to Predict Flow	Flow Data Questionable	Timing Inconsistent or Missing	Model Application	. Peak height 'forced	Peak times 'forced'	Gully Liquor Discharge Predicted	Gully Liquor Poll- utant loads predict	Gully Liquor Poll: utant loads estim.	
	1	*	/	/	/	/	/	/	/			/		
	2	*	/	/	/	/		/	/	/	/	/		
	3	*	*	/	/	/	/	/	/	/	/	/	/	
	4	*	*	/		/		/	/	/	/	/		
	. 5	*	*	/		/		/	/	/	/	/	/	
	6	*	/	*	*	/	*	*	*	¥	*	*	/	
	7	*	*	*	/	/	*	*	*	*	*	¥		
	8	*	*	/	/	/		/	/	/	/	/	/	
	9	*	/	*	*	/	/	*	*	/	*	*		
	10	*	/	*	*	/	/	*	*	/	*	*		
	11	*	/	*	*	/	1	*	/	/	/	/	/	
	12	*	*	*	/	*	*	*	/	*	*	*	$\square$	
	13	*	*	/	/	/	1	1/	/	/	/	/	*	
	14	/	/	*	/	/	1	4	/	/	/	/		
	15	*	*	*	/	/	*	i đ	$\square$	*	*	*		
	16	*	*	*	/	*	*	÷	/	*	*		/	
	17	*	*	×	/	/	*	*	*	*	*	/		
_	18	*	*	/	/	/	1	/	/	/	1/	1/	*	
-	19	*	*	*	/	/	/	*	*	/	*	*	/	
	20	¥	*	*	/	/	/	*	/	1/	*	*	/	
	21	*	*	*	/	K	/	*	*	/	*	/	/	
	22	*	*	/	V	/	/	1/	1/	/	/	/	/	
	23	*	*	*	/	/	1	*	/	/	*	/	/	
	24	*	*	*	/	/	*	*	1	1	*	*	/	1

TABLE 23 Data availability, application and limitations.



Figure 61







Figure 65





then for a multipeaked storm a fair agreement was obtained, e.g. Storm 24 (Figure 59). Adjustment of the CPL coefficient to a value greater than one was not found to be necessary, only Storm 15 (Figure 63) showed predicted peak runoff to be significantly lower than the observed peak runoff for a CPL coefficient of 1.0.

The verification of the gully liquor contribution model requires that each gully pot on the catchment is dosed with a similar concentration of some tracer and the concentration of that tracer is subsequently determined in storm runoff at the outfall. It is also important that the tracer used should not be available from any other source in the catchment. When the unreliability of sampling and storm flow determination are considered along with the practical difficulties associated with a task of this size the above approach can be discounted. However, gully pot liquors were considered for a naturally occurring parameter that could be used in a similar fashion. Calcium and ammonium were both observed to have high values over dry periods when gully pot liquors are likely to be the prime source of the parameter. However, for the period of study mean values of any parameters from five gully pot samples were never sufficiently high in respect to other possible sources to be used in this manner. Some limited results for calcium in Storm 7 were obtained and are presented in Figure 66, they indicate a reasonable agreement.

A full interpretation of the use of the model for predicting gully liquor pollutant concentrations at the outfall and defining the first flush is covered in Chapter 6. Figures 64 and 65 illustrate the gully liquor contributions to a runoff hydrograph for Storms 7 and 24.

In the light of the limitations of the currently available data set and bearing in mind the objectives of this research, then the following general conclusions may be drawn. Firstly, although verification can only be considered as tentative until reliable data is available the model would appear to give a good estimation of the storm runoff hydrograph and, as such, provides a useful tool for further catchment research. Future research aimed at optimizing the roof storage routing coefficients should improve the model's performance. The gully liquor contribution model, although not verified for the whole catchment, is based on simple mathematics and the laboratory results indicate that the model should provide a reliable estimation of original gully liquor contributions at the outfall. The removal of these liquors can be used

as a quantitative assessment of the duration of the first flush and for the purpose of this research it has been used to estimate the mass of pollutants derived from original gully liquors and, therefore, their significance in determining the overall quality of storm water runoff.

# 6.1. THE QUALITY OF WATER STORED IN ROADSIDE GULLY POTS.

The quality of water stored in roadside gully pots was examined in three ways:-

- a) Weekly and subsequently fortnightly sampling of four gully pots with differing catchment characteristics, (see Section 1.3.3), for a period of one year, to assess the magnitude of temporal and seasonal fluctuations of pollutant concentrations and the impact
   , of sources of pollution on stored water quality.
- b) Sampling on one day of up to 25 gully pots to assess the variability of water quality between pots, and to test the significance of small sets of samples in representing the mean quality of water stored in the drainage system.
- c) Other miscellaneous samples collected prior to rainfall events and over dry periods.

The mean and maximum results for all samples are shown in Table 24. The results are broadly comparable to those obtained for a similar catchment at Stevenage, U.K. (Water Research Centre, 1977, Table 15). Figure 67 illustrates the frequency distributions of the parameters where sufficient sample numbers were available. The majority of the distributions are leptokurtic and exhibit a strong positive skew. Whilst most pollutant concentrations were at low and reasonably steady values, occasional extreme values, which were attributed to the occurrence of a polluting event, resulted in this skewness. It is important to note that due to the skewness of the distributions, the arithmetic mean, as indicated on the diagrams, is no longer a strict representation of the central tendency of the distribution and is invariably greater than the modal value. Similarly, the standard deviation alone is not an adequate measurement of the variance of the distributions due to the occurrence of extreme values. This is indicated by the interquartile ranges (see results Appendix 2) which can be as low as one-fifth of the total range.

# 6.1.1 Temporal and Seasonal Variations

All of the results for the weekly/fortnightly analysis of the four gully pots are tabulated in Appendix 2.1. The annual mean values have been summarized in Table 25. Temporal variation is illustrated graphically for

TABLE 24 Mean quality of water stored in roadside gully pots: summary of all results (concentration in mg/l)

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23

Number of Samples	Mean	Max.
243	31.2	455
261	335	17,475
176	63	935
195	7.6	135
63	6.0	0 - 11.2
266	1.0	8.9
263	2.0	29.8
220	26,8	9800
201	8.0	6.3 - 12.9
246	31	272
221	125	5560
221	3.8	28
31	0.058	1.36
11	0.032	0.10
74	0.65	3.5
29	0.10	0.56
28	0.012	0.32
7	0.05	0.09
26	0.058	0.07
19	0.21	0.74
19	0.07	0.42
19	1.08	7.00
19	0.05	0.48
	Number of Samples 243 261 176 195 63 266 263 220 201 246 221 221 31 11 74 29 28 7 26 19 19 19 19 19	Number of Samples         Mean           243         31.2           261         335           176         63           195         7.6           63         6.0           266         1.0           263         2.0           220         268           201         8.0           221         125           221         3.8           31         0.058           11         0.032           74         0.65           29         0.10           28         0.012           7         0.058           19         0.21           19         0.07           19         1.08           19         0.05

Constituent	GP 1	GP 6	GP 7	GP 8
Suspended solids	15	27	29	15
Dissolved solids	197	343	231	818
COD	24.5	56.2	36.5	30.3
BOD	7.1	7.1	6.5	6.5
Ammonium	1.2	0.8	1.1	0.8
Nitrate	4.2	12.4	6.6	7.9
Dissolved oxygen	3.6	8.5	6.2	6.0
рН	7.5	10.5	7.5	7.9
Calcium	23	49	21	30
Sodium	28	9.7	34	232
Potassium	2.0	4.8	1.8	3.0
Chloride	44.7	26.4	71.9	426
			•	
		×		

# TABLE 25 Annual mean concentrations of parameters for individual gully pots (in mg/l)



Fig. 67 - Frequency distributions for the concentration of pollutants in gully pot liquors.

CONCENTRATION ( mg/l) 192 Fig. 67 (b) continued



some of the parameters in Figures 68 to 78, either as weekly mean values or values for individual gully pots.

A degree of temporal variation is exhibited in all of the results, but, changes in concentration may not be simply ascribed to seasonal fluctuations but are a result of numerous interactions, some of which are seasonally dependent.

Of all the parameters investigated ammonium concentrations behaved in the most predictable fashion (Figure 68). It retained consistently low values during the winter months and high values were closely associated with summer dry periods. Weeks 1, 2, 36, 42 and 46 all show peaks in ammonium concentrations and are all coincident with periods of dry weather in excess of 5 days. Figure 69 illustrates ammonium concentrations for gully pots 1, 6 and 7, the similarity in profile shown by these 3 pots indicated that the generation of ammonium is reasonably consistent across the catchment and has little dependency upon subcatchment differences. Mechanisms for the formation of ammonium over dry periods are discussed further in Section 6.1.3.

Nitrate concentrations were less consistent in their behaviour with peak values occurring in both winter and summer months, (Figure 70). Rainfall is an important source of nitrate and would be expected to result in complimentary increases for all gully pots indicating that pollution within the subcatchment was of more importance in defining the concentration. Possible sources of nitrate that could account for such differences include the application of garden fertilisers that was blown onto impervious surfaces and washed off during rainfall, or runoff of car wash liquors. There is some evidence in favour of the latter when the results for the summers of 1976 and 1977 are compared. During the summer of 1976 a long period of drought resulted in a public ban on car washing and Figure 70 shows that only low nitrate values were recorded. In the summer of 1977 nitrate values were high and variable, they showed no obvious relationship with either rainfall or dry periods, but may be explained by the regular input from some polluting activity within the gully pot subcatchment, i.e. car washing. Since sampling for the duration of the whole survey was always conducted on a Monday and car washing is a common activity on Sunday in urban areas, then it is hardly surprising that the concentrations recorded will show some dependency upon car wash liquor composition.

High nitrate concentrations resulting from the runoff of car wash liquors may be due to the following:-

Fig. 68 - Temporal variation of weekly mean ammonium concentrations (for key to week numbers see Appendix 2)



Fig. 69 - Temporal variation of ammonium concentration in gully pots 1, 6 and 7.





Fig. 70 - Temporal variation of nitrate concentration in gully pots 1, 6 and 8.

Fig. 71 - Temporal variation of weekly mean concentration of BOD and COD





Fig. 72 - Maximum and minimum temperatures, antecedent dry period and total weekly rainfall for the duration of the study.



Fig. 73 - Temporal variation of BOD concentration in gully pots 1, 6 and 7.





- a) Nitrate present in the washing liquid used (see 6.1.3.1).
- b) Higher nitrate levels in the local tap water. Typical concentrations in Clifton tap water of 8.0 to 9.6 mg/l (Severn-Trent Water Authority) are higher than the average nitrate concentrations recorded for gully pots 1, 7 and 8.
- c) Ammonium present in the washing liquid used which, in the aerated liquor, undergoes nitrification to form nitrate. Analysis of a commercial washing up liquid and a commercial car wash preparation has shown them to contain 6.5 and 0.6 mg/g of ammonium. The use of such liquids can result in ammonium concentrations in the region of 0.5 to 10.0 mg/l in the car wash liquor. Laboratory experiments indicate that, under aerobic conditions, such concentrations can undergo nitrification with the production of nitrite and nitrate over a period of 24 to 28 hours. The actual impact that this may have upon concentrations in gully pots is unpredictable being dependent upon the amount of washing liquid used and the volume of runoff reaching the pot.

Other peak nitrate concentrations were recorded in the winter during periods of road salting, the input of nitrate is thought to be attributable to the impure rock salt employed. Nitrate rarely maintains a constant concentration in a body of water, but, under anaerobic conditions may undergo bacterial reduction to ammonium; this is one plausible explanation for increases in ammonium over dry periods (see Section 6.1.3). Other evidence in favour of this conversion mechanism was recorded for a sample collected in week 18 from gully pot 8. Although the low temperatures prevalent at this time of year will normally inhibit such processes, exceptionally high concentrations of nitrate associated with street salting resulted in a small peak in ammonium concentration that was not mirrored in the other gully pots.

The oxygen demand (COD and BOD) of the gully pot liquors showed a similar but slightly more variable pattern to that observed for ammonium (Figure 72). Peak BOD concentrations were related to dry periods and were reflected in all gully pots, e.g. week 42, Figure 73. Such increases may be explained by the anaerobic bacterial degradation of trapped, settled sediments and other matter, with the release of soluble organic compounds resulting in rises in both the BOD and COD of the supernatant water. In the winter months BOD values remained consistently low reflecting both the rarity of dry periods and the lack of bacterial degradation due to the low

temperatures. COD also exhibited peak concentrations coincident with summer dry periods (Figure 74), but other peaks were also recorded which were probably attributable to the input of oxidizable, but not readily biodegradable compounds, from polluting activities within the subcatchment. Neither BOD nor COD showd any relationship with suspended solids concentration, further indicating that the bulk of the oxygen demand of the supernatant liquors was derived from soluble organic compounds created by degradation of bottom sediments or other organic compounds, such as oil, input from activity in the subcatchment. Maximum values of BOD and COD were both associated with the presence of large quantities of engine oil, presumably from the draining of car sumps - an activity, which though illegal, is still widely practised. Autumn leaf fall which has been reported as giving rise to a high oxygen demand through the accumulation and putrefaction of leaves in gully pots (Tucker, 1975), had no significant effect upon the quality of stored water in the Clifton Grove system. This can be explained by the fact that the catchment, only recently developed, has few deciduous trees of any size.

The concentration of dissolved solids in gully pot liquors mirror, in a general manner, the polluting events and seasonal changes already mentioned for specific parameters. Again the two events to have the greatest impact on dissolved solids concentrations were winter road salting and summer dry periods. Obviously, road salting was the cause of rapid and large increases in dissolved solids, as the applied salt melted ice and runoff occurred into gully pots. Values recorded in different gully pots during this period were highly variable, reflecting both the uneven application of the salt, but in particular, localized heavy application of salt by residents on drives and footpaths, which resulted in concentrations up to 17,000 mg/l. Road salting, not only resulted in increases in dissolved solids (Figure 75) chloride and sodium (Figure 76) but also caused increases in nitrate, potassium and calcium (Figure 77). These parallel increases were attributed to the impure rock salt employed. In the long dry period of the 1976 summer, coincident increases in dissolved solids concentration, up to 2000 mg/l, were recorded, but in the wetter summer of 1977 the increases, although remaining coincident with dry periods, were smaller and a maximum of 284 mg/l was recorded after 14 days dry weather. The increases that occur over dry periods are due to the following:-

- Anaerobic degradation of organic matter to soluble compounds
   by bacterial action.
- b) Input of soluble material from human activity, e.g. car washing.



Fig. 75 - Temporal variation of dissolved solids concentration in gully pots 1, 6 and 7.

Fig. 76 - Temporal variation of weekly mean chloride and sodium concentrations.



Fig. 77 - Temporal variation of weekly mean calcium concentration.



Fig. 78 - Temporal variation in pH for gully pots 1, 6 and 7.  $\cdot$ 



c) Input from road sweepings and wind blown material, e.g. grasscuttings and organic debris, which provide fresh material for a).

d) Evaporation which increases the effective concentration.

The pH of gully pot liquors, with the exception of gully pot 6, exhibited a degree of variability, but showed little temporal variation. In gully pot 6 exceptionally high pH levels were recorded (Figure 78), which were believed to be due to the leaching of lime since higher levels of calcium were also recorded for this pot. During dry periods the pH fell quite rapidly from 11.0 to a more typical pH value of 7 to 8. This drop is explained by the buffering capacity of natural water; at this pH an imbalance in the carbonate/bicarbonate equilibria can result in neutralization and precipitation reactions taking place (see Section 6.1.3). A similar, but less marked behaviour was observed for gully pot 8.

Suspended solids and dissolved oxygen concentrations, as expected, showed no temporal variation. However, dissolved oxygen is an important parameter in defining the nature of reactions that will occur during other polluting events (see Section 6.1.3), for instance, anaerobic degradation and putrescence of solid matter occurs when dissolved oxygen falls to zero. The results indicate that this happens after a period of 1 to 7 days dry weather. Both suspended solids and dissolved oxygen concentrations are dependent upon the time elapsed since the preceding storm event. For dissolved oxygen the concentration will also be dependent upon the catchment area of the gully pot, since the greater the volume of runoff through the pot the greater will be the degree of aeration of the stored Water. Plotting dissolved oxygen against the total volume of runoff for the week prior to analysis (estimated from the total rainfall and the impervious catchment area) indicates that for the Clifton Grove catchment, to keep the stored water fully aerated a runoff volume of approximately 6 cubic metres per week is necessary. (Figure 79). In order to keep the dissolved oxygen at a reasonable concentration and assuming a typical weekly summer rainfall of 6 mm, a catchment area of some 500  $\ensuremath{\mathrm{m}}^2$  would be needed. The majority of gully pot catchment areas at Clifton Grove are smaller than this, for example, gully pot 1 has an area of only 140 m<sup>2</sup> and can be seen from the graph to be rarely fully aerated. It would seem that one means of reducing the pollution threat due to the putrescence of organic matter is to increase the spacing between gully gratings and thereby increase their catchment area. This would have the additional advantage of reducing the number of

gully pots and thereby reducing the cost of installation, the time required for cleansing a given area and the volume of poor quality stored water, however, each gully pot would be subject to greater loads of solid material.

#### 6.1.2 Influence of Sources of Pollution on the Quality of Stored Water.

# a) Impervious Surfaces

The impervious surfaces were of a consistent type across the catchment: Roadways and pavements were of bitumen or tarmacadam and driveways were typically constructed of concrete strips separated by gravel. Erosion of these surfaces would typically produce large inorganic mineral particulates that are unlikely to undergo chemical change or decomposition, and so influence the quality of the stored water, but which would remain trapped in the bottom sediments until removed during council gully cleansing. An exception to this is the possibility of leaching of complex organic compounds from bitumen or tar coatings which may occur at low concentrations. Other particulate matter that gathers upon impervious surfaces, such as aerial dustfall, wind blown and rainfall eroded soils and vegetation, is likely to be of more significance. Fine particulates from dustfall are rapidly and easily removed by overland runoff and will normally be washed through gully pots into the sewer system. However, fine particulates reaching a gully pot on the recession limb of a storm, or during short low intensity storms where little runoff occurs, can influence gully pot chemistry. The slow settlement of such particles means that they will exert some oxygen demand on the stored waters over a period of a few days after the storm and probably comprise the bulk of the recorded suspended solids. Dustfall is likely to be of constant composition across the catchment and will not explain variation in quality between gully pots. Wind blown and eroded soils and vegetable matter consist of a variety of particle sizes and composition and are consequently of the most importance in influencing the chemistry of gully pots. The vegetable and organic components of this material will undergo decomposition over dry periods exerting an oxygen demand and subsequently, when anaerobic conditions prevail, releasing soluble organic compounds to the supernatant liquors. The degree of influence of soils on the quality of stored water will vary according to the presence or absence of soil and vegetation in a gully pot subcatchment, the amount of exposed soil and its

availability for removal. With the exception of treatment of soils by residents and the introduction of builders' sand during land developments, local soils should be of reasonably constant composition across a catchment, the size of Clifton Grove. Between catchments substantial variation in the nature of solids trapped in gully pots was recorded. Table 26 shows the results of an analysis of material retrieved from municipal gully emptying vehicles for the Clifton residential area and for Basford, a light industrial/old residential area of Nottingham.

### b) Vehicles

The presence of oil was commonly observed in gully pots as a blue/ green sheen covering the water surface. In some cases large quantities of oil were observed, which were attributed to the now illegal practice of emptying car sump oil into the drains. The presence of such large quantities of oil had a significant impact on the BOD and COD of stored waters. Tyre rubber and small rust fragments were observed in gully pot sediments and supernatant liquors had high concentrations of iron. Lead, of which the primary source in the environment is vehicle exhaust emission, was not found to be present in as high concentrations in the stored water as has been recorded in urban runoff in some studies. Levels were typically in the range of 0.025 to 0.200 mg/l compared to concentrations of 0.5 to 5.0 mg/l recorded by Hedley (1976) in motorway runoff. This is probably a reflection of the low traffic density on the catchment. However, the levels were higher than those expected for a typical surface water, where lead concentrations of less than 0.05 mg/l would be expected. High zinc concentrations of 0.2 to 1.0 mg/l, presumably derived from tyre rubber and vehicle corrosion, were also recorded.

# c) <u>Road Salting</u>

Road salting was the most discrete polluting event monitored. Since the Clifton Grove estate contained no trunk roads, it was of low priority for municipal salting and, in most instances of snowfall, only the circular feeder road, Fabis Drive, was gritted and salted by the Council. Therefore, the high dissolved solids concentrations recorded over such periods and the large variations in concentrations between gully pots indicates the importance of localised salting of driveways and footpaths by residents. Salt application rates by residents were in excess of those of the Council and in excess of the quantities needed to effect melting. The maximum dissolved

solids concentration was recorded for gully pot 8 located at the bottom of a cul-de-sac, which was not salted by the council. Salting by residents was not restricted to periods of snowfall alone. In the storm of the 6th December, 1976, high salt levels were recorded following a period of frosty weather, no council salt application had taken place and the high chloride levels could only be ascribed to salting by residents. The use of impure rock salt also results in increases in other substances, including nitrate, potassium, calcium, (see Section 6.1.1), nickel, lead zinc and chromium (Hedley, 1976). Additionally, the high concentrations of alkali metals in solution can upset water sediment equilibria and effect the release of further heavy metals, through ion exchange processes or complexation with chloride.

#### d) Human and animal activity

The nature and degree of human activity within a gully pot subcatchment is believed to be one of the most important factors determining the variation in composition of gully pot liquors. The variable and unpredictable nature of human activity leads to differences in stored water composition across a catchment. The occurrence of exceptional concentrations of parameters that are not mirrored in measurements from other gully pots may often be ascribed to human activity in the subcatchment. Some of the activities that lead to the pollution of pot liquors are:-

- a) Car maintenance and servicing results in oil, petrol, rust, paint, hydraulic fluid, antifreezes, etc., accumulating on roads and driveways. Such debris is commonly washed off or swept into kerbside channels after the completion of a job.
- b) Car washing is a widely practiced activity on residential estates and was considered to account for the frothiness of many of the samples collected from gully pots. Car wash liquids may have been an additional source of nitrate; samples exhibiting frothiness were commonly associated with high nitrate concentrations.
- c) Grass cutting. In the summer months grass cuttings were found in abundance in gully pots and at the roadside, providing further organic material for subsequent degradation over dry periods.
Fig. 79 - Variation of dissolved oxygen (DO) concentration with estimated runoff through gully pots 1, 6, 7 and 8 for week prior to analysis.



Fig. 80 - Relationship of annual mean BOD and COD concentrations with impervious subcatchment area.



Fig. 81 - Relationship of annual mean nitrate concentration with impervious subcatchment area.



Fig. 82 - Relationship of annual mean dissolved oxygen concentration with impervious subcatchment area.



# TABLE 26 Composition of gully pot sediments for two urban areas

	Clifton	Basford
BOD (MG/G) COD (MG/G) Volatiles (%) Zinc (MG/G)	43.8 34711.0 21.0 0.69	9.1 8462.0 7.6 0.22
Soluble Components:-		
Dissolvable Solids (MG/G) NO3 (MG/G) CL (MG/G) COD (MG/G)	4.0 0.165 0.037 0.75 3.2	4.5 0.185 0.035 1.25 2.7

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- d) Garden treatment. The application of manure, peat, fertilisers, pesticides and herbicides provide additional sources of pollutants which may accumulate in gully pots. Fertilisers can contribute nutrients and alkali metals to overland runoff, the use of lime may have been responsible for high pH values recorded in gully pots 8 and 6.
- e) Winter salting of driveways and footpaths to melt snow and ice.
- f) Litter dropping. Paper, cigarette ends and other garbage were observed at the roadside and in gully pots.
- g) Urination and defecation by dogs and birds. Dog faeces were observed on footpaths, gutters and wedged in gully grates. In addition to contributing degradable organic matter, they are a source of bacteria and even virus.

The influence of human activity on the quality of the water stored in gully pots might be expected to relate to the subcatchment area, i.e. the larger the subcatchment area, the greater the number of dwellings and the greater the likelihood of human influence. Figure 80, a plot of annual mean values of COD and BOD for the four pots investigated on a weekly basis against the contributing impervious subcatchment area, indicates (for the limited data available) that COD correlates well with subcatchment area, i.e. the degree of human influence, whilst BOD shows no dependency upon subcatchment area. This indicates that whilst the polluting input from human activity contributes a significant oxygen demand the materials are not readily biodegradable. This concurs with the analysis of oxygen demand of gully pot sediments from two areas of Nottingham (Table 26) where the BOD of the sediment is only 0.1 to 0.2% of the COD. A similar correlation exists for annual mean nitrate concentration against impervious subcatchment area (Figure 81). The use of garden fertilisers and car washing are believed to account for the difference in nitrate concentration between gully pot 1 containing no houses in its subcatchment and gully pot 6 draining 4 houses.

#### 6.1.3 The Chemistry of Gully Pot Liquors

The changes in composition and chemical reactions occurring in gully pot liquors can be conveniently considered in two parts:-

- a) during dry periods
- b) during rainfall

#### 6.1.3.1 Changes in Water Quality over a dry period

Changes in composition of the supernatant liquors over dry periods were evaluated in three ways:-

- a) Consideration of the antecedent dry period relevant to each weekly sampling date;
- b) Samples collected over a dry period in the field; and,
- c) A laboratory study using sediment collected from gully pots during municipal cleansing.

The results of the temporal variation study had indicated that gully pot liquors were at their most polluting state during long summer dry periods. A laboratory study was undertaken to investigate the chemical changes that migh occur over a dry period in a controlled environment where no unknown pollutant input could occur. A gully pot, of the same dimensions as those in the Clifton Grove sewer system, was filled to a depth of 30 mm with sediment obtained from municipal gully emptying vehicles. Distilled water was then added to the point of overflow and the contents were stirred to simulate the effect of a passing storm. Samples for analysis were drawn initially on a daily basis, but as changes in parameters became less marked, this interval was extended to every 2 or 3 days. The duration of the test was 36 days during which time the average temperature was  $25^{\circ}C$ .

Many chemical changes in natural waters can be related to the growth and decay of bacterial populations as they utilise material for energy and respiration. The type of bacteria present in a water, and consequently the nature of chemical changes, is largely determined by the dissolved oxygen level. When dissolved oxygen is present aerobic bacteria will predominate, that is, bacteria that utilise the oxygen for respiration. However, when dissolved oxygen is depleted, bacteria turn to other sources of oxygen, such as nitrate and sulphate, and anaerobic bacteria predominate. In water stored in gully pots both states occur. In the field, dissolved oxygen levels were monitored as falling to zero over a dry period of 1 to 7 days. The length of the period during which the stored waters remain aerobic is dependent upon the initial dissolved oxygen concentration and the degree of organic

pollution. The greater the concentration of organic material the greater will be the demand for oxygen by bacteria, utilising the material as a source of energy. The initial dissolved oxygen concentration is related to the antecedent weather conditions (e.g. size of the rainfall event) and the subcatchment area drained by the gully pot, i.e. the the greater the volume of runoff the greater the reaeration of the pot liquors (Figure 82). In the laboratory study, the dissolved oxygen decay over a dry period was monitored continuously using a probe and chart recorder (Figure 83). From a fully aerated condition the dissolved oxygen concentration fell rapidly to about 4 mg/l, when demand began to ease off. A point of inflexion was reached after which the dissolved oxygen demand accelerated resulting in a sharp cutoff, when oxygen became depleted, after a period of only 7 hours. This rapid utilisation of the dissolved oxygen was a result of the high oxygen demand of the sediment employed in the test. The different phases of dissolved oxygen consumption are believed to represent different phases in the growth of the two principal bacterial types. The initial, rapid consumption of oxygen corresponds to its utilisation by aerobic bacteria. The rate of consumption gradually declines as conditions becomes less hospitable for their development and the final accelerating stage of consumption corresponds to the gradual increase in population of bacteria capable of surviving under conditions of little or no dissolved oxygen.

Nitrogen containing species, that is, nitrate, nitrite, ammonia and organically bound nitrogen, can undergo a range of reactions that are bacteriologically and biologically dependent. In a water-sediment system a wide variety of transformations are possible, the situation is complex and, to date, poorly understood. The major transformations may be listed as follows:-

 a) Nitrification. This is the two stage oxidation of ammonia to nitrate as a result of the activity of autotrophic bacteria Nitrosomas and Nitrobacter:-

 $NH_4^+ \xrightarrow{NO_2} NO_2^- \xrightarrow{NO_3^-} NO_3^-$ 

Nitrification is believed to be one of the most important nitrogen transformations in both polluted, and unpolluted, river water (Curti et al, 1975). It is of particular significance in polluted waters

Fig. 83 - Laboratory simulation of the reduction in dissolved oxygen concentration in a gully pot liquor over a dry period.



Fig. 84 - Major transformations of nitrogen in a water sediment system (after Van-Kessel (1977)). The thickness of the lines gives an impression of the relative importance of the processes.







Fig. 86 - Relationship of weekly mean nitrate concentration in gully pot liquors with the antecedent dry period.



where ammonia discharge in sewage effluents can constitute, by its bacterial oxidation, a significant proportion of the biochemical oxygen demand of the water.

- b) Denitrification. This is the reduction of nitrate or nitrite to gaseous nitrogen, the process is impaired in waters that contain dissolved oxygen. However, bottom sediments are typically highly reduced and rich in organic matter making conditions for denitrification favourable.
- c) Nitrogen fixation. This is the uptake of gaseous nitrogen by algae and bacteria.
- d) Ammonification. The production of ammonia through the deamination of cell organic nitrogen, an important process in anaerobic sediments.
- e) Immobilisation. This, along with denitrification, is another process whereby aerobic heterotrophic bacteria can utilise nitrate under anaerobic conditions. Nitrate is thus reduced to ammonia.

These processe are illustrated diagrammatically in Figure 84 (van Kessel, 1977). In gully pot liquors the most significant change in nitrogen speciation over dry periods was the increase in ammonium concentrations. The temporal variation study had shown that high ammonium concentrations were invariably coincident with periods of dry weather and a linear correlation between the two could be obtained (Figure 85). Two mechanisms are available for the production of ammonium under anaerobic conditions: firstly, the breakdown of organic matter in sediments yielding ammonia as a by-product and, secondly, the reduction of nitrate by bacteria utilising it as a source of oxygen. There was some evidence for the occurrence of both reactions. Firstly, weekly mean nitrate values bore a loose inverse relationship with the antecedent dry period, (Figure 86), that is, high nitrate values were not coincident with dry periods. Secondly, for gully pot 8, week 18, high nitrate concentrations from road salting had resulted in the production of ammonium. Both observations indicate the occurrence of nitrate reduction. However, in the laboratory, whilst significant rises in ammonium concentration were recorded nitrate concentrations remained reasonably constant (Figure 87), indicating that ammonification was the prime source

of ammonium production in gully pot liquors over dry periods. In the laboratory test it is interesting to note that an initial decrease in ammonium concentrations occurred even though anaerobic conditions existed, additionally, gradual increases were observed in nitrate values contemporaneous with ammonium increases up to 24 days into the test. These observations may be tentatively explained by the following stages:

- a) Initial nitrification, before the commencement of anaerobic digestion, results in a decrease in ammonium concentration.
- b) Ammonium concentrations start to climb as breakdown of organic nitrogen commences.
- c) Nitrification of ammonium continues in surface layers where oxygen diffusion can take place resulting in contemporaneous increases in both nitrate and ammonium.
- d) Nitrate concentrations eventually start to fall as denitrification and immobilisation reactions assume greatest importance.

Over a dry period the reactions occurring in gully pot liquors may be compared to those that take place in anaerobic digestors for the treatment of sewage sludges. Anaerobic digestion is employed as a process for the reduction of the quantity of sewage sludge through conversion into gases or liquids:-

Organic material  $\rightarrow$  CO<sub>2</sub> + CH<sub>4</sub> + NH<sub>4</sub><sup>+</sup> + H<sub>2</sub>S

Digestion can be considered as occurring in two separate stages:-

- a) An acid fermentation by acid forming bacteria (pH4 → 6.5) comprising breakdown of organic matter to give simple organic compounds and organic acids resulting in a fall in pH and the sludge becoming slimey and foul smelling.
- b) An alkaline fermentation by methane forming bacteria (pH7  $\rightarrow$  7.8) comprising breakdown of the simpler organic compounds into methane and carbon dioxide accompanied by a rise in pH.

At ordinary temperatures complete digestion is a slow process and

may take between 3 to 6 months.

In a sludge digestion plant process a) takes hours to develop, whereas process b) may take many weeks. However, the sludge obtained in b) contains the bacteria and enzymes necessary to break down the simple organic compounds almost as quickly as they are formed, giving rise to a continuous breakdown of organic materials, via simple organic compounds, to carbon dioxide and methane. However, when digestion has become unstable or "gone sour" the production of hydrogen sulphide and other odourous compounds may occur. Additionally, the process is susceptible to inhibition by toxic substances, such as heavy metals, in the sludge.

The laboratory study confirmed the development of anaerobic digestion in gully pot liquors. After stirring of the pot contents (simulated storm) there was an initial period of settlement of suspended solids with closely correlating falls in COD and BOD (Figure 88). On day 4 the development of a thick surface bacterial scum was noted and, by day 8, the release of methane from the bottom sediments was confirmed by odour and ignition. It was about this time that the production of ammonium commenced. By day 23 the surface film had started to break up, bubbles of methane had become less frequent and suspended solids, BOD and COD had all attained reasonably constant levels. Over the duration of the test, a gradual rise in pH was recorded from 7.8 up to a peak value of 8.4 for day 35. There was a steady rise in dissolved solids (Figure 87) indicating that the breakdown of solid material to soluble compounds had occurred. However, no significant rises in soluble COD were recorded suggesting that process b) had been attained and the breakdown of organic material in the sediment to simpler soluble organic compounds was rapidly succeeded by complete breakdown to gases and simple inorganic compounds.

The development of anaerobic digestion was also noted in the field. The occurrence of surface bacterial scums over dry periods was common and the release of methane and hydrogen sulphide was recorded. Increases in dissolved solids, ammonium and oxygen demand were all noted and the liquors often turned a yellow colour, presumably through the release of soluble organic compounds. These last observations contradict the findings of the laboratory study, where no rises in oxygen demand were recorded and indicate that, in the field, the second stage of digestion was rarely achieved and anaerobic breakdown of solid organic matter results in a degradation of the quality of the stored water, through the

Fig. 87 - Laboratory simulation of the variation in gully liquor chemistry over a dry period - changes in dissolved solids (DS), nitrate  $(NO_2)$  and ammonium  $(NH_{\rm H})$  concentrations.



Fig. 88 - Laboratory simulation of the variation in gully liquor chemistry over a dry period - changes in suspended solids (SS), COD and BOD concentrations.



Fig. 89 - Laboratory simulation of the influence of car wash liquor runoff on COD, BOD and nitrate concentrations in an anaerobic gully pot liquor.



Fig. 90 - Relationship between annual mean values of pH and calcium.



Fig. 91 - Influence of lime on the pH of distilled water, gully pot liquor and tap water.







release of soluble organic compounds. The bacterial scum and the growth of suspended biomass (recorded in gully pot 8 over a dry period) may all contribute to the increases in oxygen demand recorded.

Thus the occurrence and development of anaerobic digestion in gully pots appears to be less predictable than during its controlled use for the treatment of sewage sludge. In gully pots digestion is unlikely to be efficient and does not result in the complete breakdown of organics to carbon dioxide and methane. Digestions may go 'sour', as indicated by the release of hydrogen sulphide in some gully pots. Conversely, in some circumstances digestion may not develop or may be inhibited by the input or presence of toxic substances.

Breakdown of solid organic material in gully pots may also have the undesirable effect of releasing loosely bound or chelated toxic heavy metals into solution; evidence for this effect has been recorded by Mance and Harman (1978). In the field additional pollutant input may occur, over a dry period, through human activity in the gully pot subcatchment. This can be a source of fresh material for degradation and also result in further increases in the oxygen demand of the liquors. The effect of car wash runoff on the quality of gully pot liquors was investigated in the laboratory through the addition of a dilute solution of a commercially available washing liquid on day 38 of the dry period investigation (Figure 89). Immediate increases in concentration were recorded for BOD, COD and nitrate, but, over the succeeding five days a steady decrease in all parameters was recorded. The steady reduction in oxygen demand may be explained by the fact that the second stage of digestion had been attained resulting in rapid breakdown of organic material.

Other changes in the concentration of species in gully pot liquors can result from natural buffering and precipitation reactions. Evidence of such reactions was recorded for gully pot 6 and to a lesser degree for gully pot 8. During wet weather gully pot 6 exhibited consistently high pH values, this was ascribed, either to the runoff of lime applied to local soils, or to the leaching of unfixed lime in cement and indeed a correlation between annual mean values of calcium and pH was indicated (Figure 90). Laboratory experiments showed that between 25 and 75 mg/l of lime were necessary to produce a pH in the region of 11 (Figure 91), corresponding to the runoff of between 2g and 7g of lime into the gully pot. The average difference in calcium concentration between gully pots 1 and 7 and gully pot 6 of about 30 mg/l is equivalent to 50 mg/l of lime and corresponds to the runoff of about 5g of lime into the gully pot and

may account totally for the difference in pH observed between these pots. Over dry periods the pH of gully pot 6 was observed to fall gradully to an equilibrium value of between 7.4 and 7.8 (Figure 92). This pH drop is attributed to a number of possible reactions; added lime may react with dissolved carbon dioxide to form calcium bicarbonate:-

$$Ca (OH)_2 + CO_2 (aq) \longrightarrow CaHCO_3 + H_2O$$

Additionally at high pH values the formation of carbonate anions from bicarbonate is preferred resulting in the release of hydrogen ions and a decrease in pH:-

$$HCO_3^{-} + H_2O \iff CO_3^{2-} + H_3O^{+}$$

In the presence of excess calcium it is also possible that the precipitation of calcium carbonate would occur. This would result in a drop in calcium in solution over dry periods for which there is little evidence, therefore, the reaction is likely to be of only minor significance:-

 $Ca^{2+} + CO_3^{2-} \longrightarrow CaCO_3 \downarrow$ 

At high pH similar precipitation reactions may occur for other species with the precipitation of either carbonate or hydroxide compounds:-

> $Cd^{2+} + 2H_2O \xrightarrow{pH 9.3} Cd (OH)_2 + 2H^+$  $Fe^{3+} + 3H_2O \longrightarrow Fe (OH)_3 + 3H^+$

In conclusion it is evident that changes in the chemistry of gully pot liquors over dry periods are complex and highly variable. In terms of mathematical modelling, this variability makes the prediction of concentration changes by simple conceptual methods, such as linear regression, unreliable.

6.1.3.2 Changes in Water Quality during Rainfall.

Changes in the concentration of parameters during a rainfall event are largely determined by three factors: the composition of the liquors prior to the event, the composition of the incoming runoff and the characteristics of the storm event.

The initial phases of overland runoff readily transports in suspension dustfall and other fine material that has accumulated on the



impervious surfaces over dry periods. The removal of these fine solids is more or less independent of flow rate, but, the subsequent erosion and transport of heavy particulates is determined by both the rainfall intensity and the flow rate. The first runoff entering the gully pot causes a rapid rise in suspended solids concentration, (Figure 91), due to the solids carried in the runoff, the resuspension of fine material settled in the pot and the mixing up of surface dusts and bacterial scums. Solid material in suspension is now flushed out of the pot in a similar manner to the removal of soluble components and constitutes the poor quality first flush of storm runoff. However, heavy particulates entering the pot will generally remain trapped in spite of the stirring action of runoff. Towards the end of a storm fine material carried in suspension may also be trapped in the gully pot as the rate of runoff recedes.

Soluble material is transported by runoff rapidly and more easily than solid material. The rate of flushing out of dissolved solids from gully pots is dependent upon the degree of mixing of the incoming runoff with the original gully liquors, which is determined by the rate of inflow (see Section 5.1.3). After a storm the concentration of dissolved components in the liquor will depend upon the volume of original gully liquors removed and the concentration of the components in the incoming runoff, which is, in turn, dependent upon the concentration in the rainfall and the amount of material dissolved or adsorbed during overland runoff.

The stirring action of incoming runoff also has the effect of increasing the dissolved oxygen concentration. The increase is strongly dependent upon the rate and volume of runoff.

To summarise, after a storm a completely new set of conditions prevail in a gully pot. The stored water is well aerated, different concentrations of dissolved parameters exist and fresh, undegraded solid material has been input.

## 6.1.4 <u>Variation in the Quality of Water Stored in Gully Pots across</u> the Catchment.

Variation in the quality of stored water across the whole catchment was evaluated by collecting, in one day, up to 25 samples from randomly selected gully pots. Five such sets of samples were collected at different times of the year, namely, 3rd September, 1977,

Table 27 Coefficients of Variation (  $\sqrt{\pi}.100\%$  ) for the catchment - wide variation analyses

July	#8%	33.5	66.3	159	25.6	88	116	1	18.3	160	77			
May	174%	93	124	188	44.5	105	206	14.7	31	314	56			
March	89%	717	61	108	25.7	80.3	185	3.66	71.4	180	92			
Feb.	91%	134	171	221	56.3	165	129	2.65	72	155	51			
Sept.	177%	37.7	171	89.6	71.3	75	36.4	2.54	14.8	149	300			
Constituent	Suspended solids	Dissolved solids	COD	BOD	Nitrate	Ammonium	Chloride	Hq	Calcium	Sodium	Potassium			

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TABLE 28 95% confidence limits (expressed as a percentage of the mean value) for the catchment - wide variation analyses

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TABLE 29 95% Confidence limits for estimating the mean quality of water stored in gully pots during the summer months with five random samples.

Confidence limits -  $\overline{X} \pm 2.776 \overline{\sigma} / \sqrt{5}$ 

where  $\overline{\sigma}$  = average standard deviation calculated from variation analyses <u>not</u> affected by winter road salting. and a second a second a second second and a second second second a second second second second second second s

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Constituent	Confidence Limits ( X ± )
Suspended solids	62 mg/l
Dissolved solids	66 mg/l
COD	76 mg/l
BOD	17 mg/l
Nitrate	4.6 mg/1
Ammonium	2.0 mg/l
Chloride	19.2 mg/l
pH	0.54
Calcium	8.3 mg/l
Sodium	11.1 mg/l
Potassium	4.2 mg/l

23rd February 1978, 20th March 1978, 15th May 1978, and the 17th July 1978. In all cases a wide variation in the concentration of parameters was exhibited, as shown by Table 27, the coefficients of variation (relative standard deviation) and Table 28, the 95% confidence limits (expressed as a percentage of the mean) for an estimate of the population mean based on the samples collected. The results indicate that the highest variations occur in the winter (February samples), presumably through the influence of road salting, and the lowest variations occur in the summer (July samples). As for the weekly analyses, the interquartile ranges were substantially less than would be expected from a normal distribution, indicating that occasional extreme values were the major cause of the high variability in quality.

In spite of the high variability in parameter concentrations it has been necessary, for the purpose of practicality, to assume that five randomly selected samples will provide some estimation of the mean quality of water stored in gully pots throughout the catchment. Taking an average standard deviation, calculated for each parameter from all of the tests, then the percentage range within which the population mean would be expected to fall is shown in Table 29. High deviations for chloride and other parameters similarly affected by road salting have been ignored since such variance is not applicable to the period when stormwater sampling was undertaken.

### 6.2. THE QUALITY OF STORMWATER RUNOFF

The quality of stormwater runoff from the Clifton Grove catchment was monitored for 24 storm events over a period of 2 years. The majority of events monitored occurred in May, June and July, 1978. Two events were monitored in June and July, 1977 and three in November and December, 1976. The results for each storm event were processed using a computer programme, STQUAL-FOR (see Section 4.2.). Tabulated pollutant concentrations, loads and graph plots formed the output of this programme and are presented in Appendix 2.3. Where reliable rainfall, runoff and quality measurements were available for an event, the mathematical model described in Chapter 5 was utilised to predict the mode of discharge of gully pot liquors and, thereby, to calculate the storm load attributable to their discharge. Graph plots of rainfall intensity, predicted and measured storm flows and predicted gully liquor discharge for storms analysed by this programme are presented in Appendix 2.4.

#### 6.2.1 Pollutant Loads and Concentrations in Runoff

The mean and maximum concentrations for all storm events are summarised in Table 30, and for individual events in Tables 31 and 32. The figures are broadly comparable with results obtained from other developed residential catchments reported in Tables 1 and 2, for example, Zurich (Roberts et al, 1977) and Stevenage (Mance and Harman, 1978). However, residential catchments, where building operations were still in progress, gave higher values of suspended solids and related pollutants, e.g. Oxhey (Wilkinson, 1956) and Rise Park (Tucker, 1975). Mixed, industrial, commercial and motorway catchments also yielded much higher levels of pollutants, e.g. M6 Motorway (Hedley, 1976), Oslo (Lindholm, 1978).

The distribution of the data is illustrated in Figure 94. Like the gully liquor quality data a positive skew is evident making the modal value commonly less than the mean value, e.g. suspended solids modal value is 0 to 50, whereas the mean value is 95; dissolved solids modal is 50 to 100, whereas the mean is 119. Runoff was not monitored during periods of road salting, therefore, extreme concentrations of dissolved solids and other parameters which were recorded in gully pot liquors have not influenced the distributions. Since the majority of storms monitored occurred in the summer months, the distribution may be considered as most representative of summer runoff only.

Comparing the average quality of the storm runoff with what might be expected from typical combined sewer overflow or secondary sewage effluent leads to the same conclusion as other studies. Namely, suspended solids concentrations are normally higher than the 30 mg/l limit recommended by the Royal Commission on Sewage Disposal for sewage effluent, whilst, BOD concentrations rarely exceed the recommended 20 mg/l. With the exception of nitrate, the levels of other parameters are generally lower than those in sewage effluent and combined sewer overflow.

# 6.2.1.1 Annual Pollutant Loads Discharged

The pollutant loads expected to be discharged from the catchment over a period of one year were calculated by three methods:-

 a) Pollutant loads for all storm events with known rainfall were summed, divided by the total rainfall and the impervious catchment area to give the load per unit area per mm of rainfall. The result was multiplied by a typical annual rainfall of 600 mm

Constituent	Mean	Maximum	No. of Samples.
Suspended solids	94.5	882	257
Volatile S. Solids	26.8	60	22
Dissolved solids	119	492	183
BOD	8.7	44.5	125
COD	63.4	600 +	147
Nitrate - N	1.8	7.3	225
Ammoniacal - N	1.3	4.1	151
Chloride	58.3	179	41
рН	7.54	8.22	21
Calcium	14.5	48.6	166
Sodium	90.7	211	52
Potassium	2.2	3.46	58
Lead	0.29	0.44	4
Zinc	0.25	0.33	4
Cadmium	0.013	0.03	4
Copper	0.05	0.10	4
Chromium	0.009	0.01	4
Iron	4.3	5	3
Nickel	0.02	0.03	4
Manganese	0.05	0.08	4

# TABLE 30 Mean quality of stormwater runoff: summary of all results (concentrations in mg/l)

TABLE 31 Mean Pollutant Concentrations for Individual Storm Events (mg/l)

1 NH4	0.63 0.76 0.756 0.77 0.74 0.77 0.74 0.74 0.74 0.74 0.74
NO <sub>3</sub>	- 200 m - 4 - 20 - 200 m - 200
COD	
BOD	
DS	77 77 375 375 375 110 110 110 110 71 71 71
SS	775 775 775 775 775 775 775 775 775 775
Q (L/S)	4 8 4 9 7 6 7 8 9 8 9 8 9 8 9 8 9 8 9 8 9 8 9 8 9 8
Storm	すららて 路 8 8 8 8 9 0 1 1 2 1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2

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TABLE 32 Maximum Pollutant Concentrations for Individual Storm Events (mg/1)

th NH	1	1.60	1.25	1.40	0.33	0.15	0.07	1.3	2.1	5.3	ı	1.80	0.70	0.34	0.36	2.0	0.44	0.56	ı	1	ı	ı			
NO3	1	22.0	11.7	12.3	5.1	6.9	5.4	12.5	32.4	1	T	32.1	10.0	7.5	6.7	11.8	5.8	10.9	6.1	12.5	6.6	ۍ ۲			
COD	1	ł	6†	157	18	1	1	121	214	439	109	133	107	53	102	67	29	36	504	147	717 717	254			
BOD	1	13.0	6.7	12.3	6.7	1	ı	20.2	36.6	44.5	. 1	15.5	9.7	8.4	8.6	14.2	11.0	11.8	I	1	1	1			
SQ	1	I	265	152	136	136	126	320	416	368	300	224	120	88	232	170	108	120	164	200	116	140		 ******	
SS	226	332	69	292	6†	30	52	202	205	413	130	178	017	60	201	64	68	25	882	279	42	323			
(S/7) Ő	14	138	24	65	21	48	t1		23	ω	52	85	31	29	36	51	13	. 99	18	20	11	106			
Storm	11	י ע	5 00	2	Sa Sa	8b	8c	σ	10	11	12	13	15	16	17	18	19	20	21	22	23	24			

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Fig. 94 - Frequency distributions for the concentrations of parameters in storm water runoff.





TABLE 33 Pollutant Loads for Individual Storm Events (kg)

NH 4	0.264 0.021 0.021 0.010 0.012 0.012 0.015 0.012 0.015 0.012 0.015 0.012 0.012 0.012 0.012 0.012 0.012 0.012
NO <sup>3</sup>	3.07 3.07 0.158 0.158 0.052 0.052 0.158 0.056 0.052 0.158 0.057 0.056 0.052 0.057 0.057 0.052 0.158 0.052 0.155 0.052 0.155 0.15
COD	7.126 7.12
BOD	0.112 0.421 0.421 0.128 0.154 0.154 0.154 0.154 0.154 0.154 0.154 0.154 0.154 0.154 0.154
SQ	3. 1 4. 6 3. 1 4. 6 3. 1 5. 0 3. 1 5. 0 3. 1 5. 0 5. 0 5. 0 5. 0 5. 0 5. 0 5. 0 5. 0
SS	82.56 10.03 10
Discharge (m <sup>3</sup> )	285 285 285 295 295 295 295 295 295 295 295 295 29
Storm	μωνμως αφωσος τνανματασος αφοσος τος τος τος τος τος τος τος τος τος τ

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TABLE 34 Estimated annual pollutant loads discharged from the Clifton Grove catchment by three methods of calculation (see 6.2.1.1.) (kg/ha/annum)

			<u> </u>						
	U	Discharge based calculation using all storms.	306	417	27.7	208	1.97	7.03	
1	В	As 'A' but excluding storms 16,17,21	295	371	11.9	237	1.61	5.92	
	A	Calculated for all events with known rainfall.	229	276	11.9	181	0.93	3.98	
			Suspended solids	Dissolved solids	BOD	COD	Ammoniacal - N	Nitrate - N	

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to give the estimated annual load:-

Annual Load (kg/ha/annum) =  $\Sigma(Storm Load) \times Annual Rain \Sigma(Rainfall) \times Imp. Area$ 

(13 storms were used in the calculation).

- b) The same procedure was followed as above but eliminating storms 16, 17 and 21 where the entire storm may not have been sampled.
  (10 storms were used in the calculation).
- c) Storm loads for all events with reliable discharge measurements were summed and divided by the total discharge and impervious catchment area to give the load per unit area per cubic metre of runoff. Using a runoff coefficient of 0.645 (an average based on the events monitored, Figure 95) the annual load for 600 mm of rainfall was calculated:-

Annual Load (kg/ha/annum) =  $\Sigma$  Storm Load) x Annual Rain x Runoff Coeff  $\Sigma$  Discharge

(20 storms were used in the calculation).

The results for the three methods are presented in Table 34. Variation between the three techniques serves to indicate how the method of calculation and the number of storms utilised may account for the wide variation in annual loads reported in the literature and further enforces the comments made in Section 1.2.3. The annual loads for the Clifton Grove catchment are broadly comparable to the values reported for other residential catchments (where a similar method of calculation was used, Table 3). The distribution of annual pollutant loads is shown in Figure 96. Difference in the annual rainfall used in the calculation may also account for some of the variation in the annual loads, for example, in the Zurich study (Roberts et al, 1977) low average concentrations were recorded, but a high annual rainfall resulted in the predicted loads being higher than average.

Since the Clifton Grove catchment is a new relatively clean and well kept residential estate, then the magnitude of the annual loads discharged serve to indicate the importance of stormwater runoff in influencing river water quality. When it is considered that a number of similar catchments and other catchments with potentially greater pollutant loads such as motorways, industrial and commercial areas may all discharge untreated runoff into one river, then the magnitude of pollution that

Fig. 96 - Distribution of annual pollutant loads for residential catchments.



may arise from this source is evident.

The relative importance of dustfall as a source of suspended solids in stormwater runoff can be estimated from the measured average dustfall deposition rate of  $46.5 \text{ mg/m}^2/\text{day}$ . Over one year this accounts for 170 kg or 55% of the 306 kg/ha of suspended solids load.

6.2.1.2 Relationships between Storm Loads, Discharge and Concentrations

It is to be expected that the greater the volume of runoff the greater will be the mass of pollutants discharged and, indeed, total storm loads show a good correlation for all parameters with total storm discharge (Figure 97). Because of the dominating influence of discharge in determining the pollutant load, relationships also exist between different pollutants, for example, suspended solids load with BOD and COD loads. Such relationships can be used with some success to predict the pollutant loads discharged on catchments where sufficient data is available to develop the regression equation, e.g. Bedient et al, (1978). These methods are advantageous in that only easily obtained data (e.g. total rainfall and the runoff coefficient) are needed for their application, however, they are limited since their use is only valid for the catchment for which the equation was developed.

To examine whether relationships exist between pollutants it is more meaningful to evaluate correlations between mean storm concentrations and thereby eliminate the dominating effect of discharge (mean storm concentration = total storm load/total storm discharge). Table 35 shows correlations between storm loads and Table 36 shows correlations between mean storm concentrations for all available data. It can be seen that, whereas strong correlations exist between many parameters when storm loads are considered, correlations between mean storm concentrations are fewer and more tenuous. For this data set the suspended solids concentration shows a good correlation with only one parameter, that is COD. This finding indicates that, although good correlations exist between suspended solids load and other pollutant loads, the technique of using suspended solids concentration as a means of predicting the concentrations of other pollutants in runoff though the application of 'pollutant potency factors' as practised in SWMM (Metcalf and Eddy, 1971) and STORM (Roesner et al, 1974), may not be valid. The use of these factors (simple ratio between pollutants) relies upon the strong correlation that exists between discharge and pollutant loads,

Fig. 97 - Correlations between pollutant loads and cumulative discharge for individual storm events.



Figure 97 (continued)



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Table 35 Correlation matrix for storm loads (all available data)

NO3	1	1	I	1	ı	0.99	
COD	I	1	1	1	0.47	0.84	
BOD	ſ	I	I	0.83	0.96	0.93	
DS	1	ı	0.97	0.54	0.96	0.94	
SS	I	0.65	0.94	0.93	0.80	0.77	
Discharge	0.85	0.90	0.97	0.76	0.93	0.96	
	SS	DS	BOD	COD	NO <sup>3</sup>	ή. HN	

Table 36 Correlation matrix for mean storm concentrations.

				•		**************************************	
e NO <sub>3</sub>	8	1	ı	1	I	0.79	
COD		I	ı	ı	0.18	0.87	
BOD	80	ĩ	ł	0.60	0.88	0.71	
DS	1	I	0.88	0.21	0.80	0.03	
SS	I	0.30	0.34	0.84	0.07	0.41	
ď	-0.32	-0.49	-0.54	-0.38	-0.23	-0.48	
	SS	DS	BOD	COD	NO3	<sup>th</sup> HN	

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the extension of this relationship to the prediction of concentrations and the pollutant runoff profile (pollutograph) will not be reliable, since discharge is no longer the determining parameter.

The strong correlations that exist, for both loads and concentrations, between COD and suspended solids indicate that the COD is largely determined by oxidizable organic material carried in suspension. However, with the exception of very high suspended solids concentrations, the results suggest that suspended solids exert less influence on BOD. Good correlations were also indicated between BOD, ammonium and dissolved solids, however, in this instance a direct relationship between the parameters is unlikely but, for example, common accumulation and removal processes or a common dependency upon some other factor is plausible.

## 6.2.1.3 Relationship of Pollutant Loads and Concentrations with Meteorological Parameters.

Since pollutant loads exhibit a strong correlation with storm discharge it is an attractive proposition that pollutant loads could be predicted with greater accuracy through the use of easily determined meteorological parameters such as rainfall volume and intensity. Multiple regression analysis is an ideal tool for obtaining such a relationship and also for investigating the dependency of storm loads upon a number of different factors. The analysis was undertaken using 13 storms, which satisfied the data requirements, and which were judged to be reliable. Pollutant loads for suspended solids, dissolved solids, BOD, COD, nitrate and ammonium were related to the following meteorological parameters (the data matrix used is shown in Table 37).

- a) Storm duration (DUR) in minutes.
- b) Runoff Volume (ROVOL) in cubic metres
- c) Average rainfall intensity (AVIN) in mm/hr.
- Average rainfall intensity for the most intense ten minutes (AV 10) - mm/hr.
- e) Peak rainfall intensity (PI) mm/hr.
- f) Rainfall volume (PPT) mm
- g) Volume of the previous event (SPE) in mm.
- h) Length of the antecedent dry period (ADP) in days.

Table 37 Meteorological parameters for storms utilised in the multiple regression analysis.

1 1															Contract Contractor day	 	-
SPE	1.900	0.200	3.300	0.200	1.900	3.100	4.500	2.800	0.200	0.100	1.200	2.200	0.200				
PPT	1.755	3.393	1.170	0.968	1.863	0.789	1.800	7.407	5.750	1.755	0.234	8.569	2.421				
AWI	46.200	44.300	46.300	48.700	45.100	46.900	47.700	48.000	49.700	57.000	56.000	45.700	45.100				
ADP	. 000 • 0	0.000	0.000	0.000	0.000	0 000	1.000	0.000	0.000	7.000	6.000	00000	0.000				
Id	7.020	21.060	7.020	3.379	3.997	6.808	5.657	19.131	7.020	14.040	3.510	13.756	3.947				
AV10	3.978	15.093	3.510	2.877	2.824	2.616	3.325	11.278	4.914	5.967	2.340	8.619	2.566				
AVIN	0.975	5.090	1.950	1.056	0.998	1.690	1.963	4.115	2.270	2.700	2.340	4.432	0.825				
ROVOL	60.369	116.713	40.246	33.299	64.081	27.125	61.901	254.778	197.780	60.369	8.049	294.743	83.274		4-2		
DUR	107.000	39.000	35.000	54.000	111.000	27.000	54.000	107.000	151.000	38.000	6.000	115.000	175.000				
Storm No.	6.	7.		. 15.	16.	19.	20.	24.	12.	10.		17.	21.				

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Antecedent wetness index (AWI) defined as:-

$$AWI = 50 + ADP - (R1 + \frac{R2}{2} + \frac{R3}{3} + \frac{R4}{4} + \frac{R5}{5})$$

where R1 = rainfall on the previous day R2 = rainfall two days prior to the event.

Not all of the parameters were used in any one analysis, parameters that reflected similar facets of a storm event were usually not paired up, e.g. runoff volume was not paired with rainfall volume. The results of some of the analyses for different combinations of parameters are presented in Appendix 2.5.

The strong relationship between pollutant load and runoff volume has been discussed previously and not surprisingly the total rainfall volume was found to be the most significant parameter for all pollutants. The measures of dry period, ADP and AWI, showed little correlation with any of the pollutants, nor was the size of the previous event of any significance. Measures of storm intensity, AVIN, AV10 and PI, were of considerable importance for suspended solids and COD, but were of less singificance for dissolved solids, where the storm duration assumed more importance. These differences may be explained by the fact that the removal of suspended solids is governed by the rainfall intensity needed to dislodge and transport particles trapped in small crevices on the road surface, whereas dissolved solids will be transported regardless of rainfall intensity.

All of the analyses gave high coefficients of determination indicating that the technique should provide a good means of predicting pollutant loads for any catchment where sufficient data is available to formulate an equation. Again, the applicability of such an equation to other catchments is likely to be limited. The results of the analysis confirm the findings of Mance and Harman (1978) that the characteristics of the current rainfall event are of the most importance in determining the magnitude of the pollutant load, the antecedent dry period and the size of the previous rainfall event are of little importance. However, the small size of the data set and the limited number of storms, with antecedent dry periods of one day or more, combine to make the results only tentative.

i)

Since storm loads have been shown to be strongly dependent upon the total rainfall volume, it is hardly surprising that there is much conflicting evidence in the literature concerning the influence of the antecedent dry period on storm runoff quality (see Section 1.2.3.2). In a regression analysis of ADP against storm load any influence the ADP may exert on runoff quality is masked by the characteristics of the current event. It is of more use to investigate the relative significance of other meteorological parameters by examining their relationship with mean storm concentrations and thereby eliminate the influence of the rainfall volume.

Multiple regression analysis of mean storm concentrations with meteorological parameters confirmed that the total rainfall volume (PPT) and the duration of the rainfall event (DUR) became of little significance for all of the pollutants examined. The analysis for suspended solids showed the limitation of the small size of the data set. When Storm 11 was excluded from the analysis, then measurements of rainfall intensity (PI, AVIN, AV10) were of the most significance. However, when it was included in the analysis, then the antecedent dry period became the most significant parameter. This anomaly can be explained by the nature of Storm 11, the event combined a strong first flush effect, and hence high concentrations for all pollutants, with a short rainfall event of low intensity (only one sample was collected) after a period of 7 days dry weather. For dissolved solids concentrations the only meteorological parameters of any significance, both including and excluding Storm 11, were the dry period measures. None of the parameters showed any significance for COD, BOD, nitrate and ammonium except when Storm 11 was included in the analysis when the dry period measures became of the most importance.

The multiple regression analysis for all of the pollutants was characterised by weak correlations, low coefficients of determination and high standard errors (see Appendix 2.5). However, the results point to the antecedent dry period, although being of no importance in determining the total storm load, being of some significance in influencing the concentration in runoff. This is probably particularly true for the early, or first flush, phases of runoff.

# 6.2.2. Variations in Quality during Runoff. The Mode of Removal of Pollutants.

Observations for individual storm events on quality variations

during runoff and the mode of removal of pollutants are presented below, along with possible limitations of the data. All data and graphs are presented in Appendix 2.3.

6.2.2.1 Observations for each Storm

Storm	1	-	6th	November	1976
Storm	2	-	6th	December	1976
Storm	3	-	17th	December	1976

No flow or rainfall data was available for these three storms which were used for preliminary testing of the sampling equipment.

Storms 1 and 2 showed suspended solids rising to a peak concentration, whereas, for storm 3 suspended solids remained low and constant. With the exception of storm 2, the soluble components show a steady decline in concentration throughout the storm. Storm 2 was influenced by road salting by residents following some frosty weather. Consequently, both dissolved solids and chloride rise to peak values, which precede the suspended solids peak, indicating the relative ease of mobilisation of soluble material on road surfaces compared to solid material. The initial low values of these components is explained by earlier runoff from roofed areas not influenced by road salting.

Storm 4 - 9th June 1977 Storm 5 - 27th June 1977

Although discharge was monitored for these two storms, no rainfall intensity data was available. The storms were used for testing the operation of the lithium chloride doser.

Storm 4 shows a clear relationship between flow rate and suspended solids concentration. The two main flow peaks are followed by peaks in suspended solids concentration lagging behind by 12 and 4 minutes. In spite of its duration, this storm shows no indication of solids exhaustion, which may be explained by the stage of development of the catchment: building was still in progress in two parts of the catchment and open ungrassed areas and piles of building sand provided a supply of solid material for easy removal by runoff. The storm provides an interesting comparison with later storms when the catchment was fully developed and confirms that far greater solids loads can be expected from catchments still under development.

Storm 5 is a very long and complex event for which the flow data







Fig. 101 - Storm 5 Percentage cumulative discharge (CQ) and suspended solids (CSS).



Fig. 102 - Storm 7 Discharge (1/s), suspended solids and COD concentration (mg/l)

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Fig. 103 - Storm 7 - Nitrate, ammonium and calcium concentration (mg/1)



Fig. 104 - Storm 9 Discharge (1/s), suspended solids and COD concentration (mg/l)



may be available. The storm was preceded by 11 days of dry weather and, as expected, exhibited a strong first flush of suspended solids. There was also some indication of solids exhaustion as peaks in suspended solids concentration became smaller towards the end of the storm.

Storm 6 - 2nd May 1978

The lithium doser failed to operate for this storm and flow values applicable to the six samples collected were predicted using the rainfall runoff model.

Storm 7 - 4th May 1978

The storm was characterised by a single peaked hydrograph. The suspended solids and COD curves showed a similar profile, with peak concentrations being coincident with peak runoff. High initial values of both suspended solids and COD when flow rates were low indicated an initial, or first flush, phase of flow independent removal of solid material. BOD and the soluble components - nitrate, ammonium and calcium showed similar removal patterns, a steady decline from initial high values. Although all the parameters exhibited a first flush effect in terms of concentration, the cumulative load plots do not indicate a first flush in terms of pollutant load.

Storm 8 - 5th May 1978

No rainfall intensity data was available for this long duration, but low volume, storm. Outfall flow rate was low, usually less than 30 1/s, and consequently the sampler was activated on three separate occasions (8A, 8B, 8C). The heavy storm on the 4th May, which effectively washed the catchment clean and the low flow rates, resulted in low concentrations for the majority of pollutants.

Storm 9 - 15th May 1978

This was a short duration, low intensity event for which only 4 samples were collected. Suspended solids and COD concentrations both exhibited high concentrations for the first sample, followed by a decline in concentration, which was not related to flow rate.

Storm 10 - 24th May 1978

This storm was of a similar nature to Storm 9, being of short duration and low intensity, during which only 3 samples were collected. However, it was preceded by 7 days dry weather. In spite of the dry period the suspended solids concentrations were similar in both storms, which

suggests that high initial values of solids in the runoff may be attributed, not to the wash-off of fine solids settled on impervious areas, but to the flow independent removal of sediments settled in the sewer from the previous storm event. The concentrations of COD, BOD, dissolved solids, and ammonium were all much higher than in storm 9 and indicate that some degradation of solids had taken place, either in the sewer or in gully pots.

Storm 11 - 31st May 1978

A very short storm of only 0.23 mm and 6 minutes duration (only one sample was collected), after a period of 6 days dry weather. The storm is of some interest, since the mathematical model predicted that the rainfall was insufficient to satisfy depression storage on the roads and runoff occurred only from roofs. In spite of probably being composed purely of roof runoff, the one sample collected was of very poor quality with a suspended solids concentration of 413 mg/l, COD of 439 mg/l and ammonium of 5.3 mg/l. Again the result is a strong indication that the degradation of sediments, settled in the pipe network after the previous storm and their subsequent resuspension by the early phases of runoff of the next storm, is of significance in determining the poor quality of the first flush. The high ammonium concentration suggest that the sediments, possibly trapped in shallow, sludge-rich pools within the pipe system, had undergone anaerobic degradation. It seems unlikely that roof deterioration could account for such high concentrations of both solids and ammonium.

Storm 12 - 1st June 1978

A long and complex storm of variable rainfall intensity which resulted in multiple flow peaks. The mathematical model indicated that sampling commenced some 30 minutes after the time expected.

Like discharge, the suspended solids concentration showed multiple peaks, but there was little relationship between the two. Some suspended solids peaks were coincident with peaks in discharge, but some apparently preceded them, whilst others lagged behind. COD concentrations exhibited a similar behaviour. Dissolved solids showed high initial values which may be due to the flushing out of gully pot liquors that were not removed during Storm 11 and had, therefore, some 7 days to degrade.

Storm 13 - 15th June 1978

No rainfall intensity data was available for this long storm which comprised four major peaks in discharge. The storm followed a period of 7 days dry weather.





Fig. 106 - Storm 10 Nitrate, ammonium and calcium concentration (mg/l)











Moderately high ammonium concentrations in the first flush phase of runoff suggest that some degradation of gully pot liquors and sewer sediments had taken place. A first flush, flow independent, removal of suspended solids, BOD and COD was also evident. The storm was of sufficient duration to exhibit exhaustion of solids from the catchment. Both suspended solids and COD show high initial values, which fall rapidly, with no significant rises in their concentration associated with peaks in discharge.

Storm 14 - No samples collected Storm 15 - 29th June 1978

Another short duration, low intensity storm with discharge not exceeding 31 1/s. Suspended solids and COD concentrations are also low and again show no relationship to discharge. For small storms such as this, the first flush concentration effect appears to dominate the shape of the pollutant runoff curve with the removal of solids showing little, if any, relationship to the shape of the outfall hydrograph. Soluble components again exhibited a steady decline from high initial concentrations.

Storm 16 - 1st July 1978

A short duration, low intensity storm of similar nature to storm 15 and exhibiting similar effects.

Storm 17 - 3rd July 1978

A long duration and complex event which, according to the mathematical model, was only partially sampled. All parameters again show peak concentrations for the first samples, with a general decline in concentration throughout the period of monitoring. The suspended solids curve showed only small humps that were coincident with peaks in discharge.

Storm 18 - 19th July 1978

No rainfall intensity data was available for this storm, which was another of short duration and low intensity. The storm followed a period of 10 days dry weather and, although high ammonium concentrations were recorded, the concentrations of COD and suspended solids were both low. Nevertheless, all parameters exhibited a first flush effect, with suspended solids and COD both showing the typical high initial concentrations followed by a steady decline, unrelated to changes in discharge. Storm 19 - 23rd July 1978

A very short storm for which only 2 samples were collected. Low concentrations of all parameters were recorded.

Storm 20 - 26th July 1978

A storm of medium duration but low intensity with flow rates not exceeding 40 l/s. All parameters exhibited the typical removal patterns outlined above for this type of storm; a general decline in concentration throughout the storm, with little relationship to discharge.

Storm 21 - 27th July 1978

A long duration but low intensity storm with flow rate rarely exceeding the value necessary for sampling to commence, resulted in only 4 samples being collected approximately 2 hours after runoff had commenced. In view of these circumstances and the fact that there had been no dry weather for some weeks, it is surprising that exceptionally high suspended solids and COD concentrations were recorded for the first two samples. Dissolved solids concentrations were low and constant and, therefore, consistent with the nature of the event.

Storm	22	-	30 th	July	1978
Storm	23	-	30th	July	1978
Storm	24	-	30th	July	1978

These storms form part of a complex series of rainfall events on the 30th July which were sampled in three sections. Four samples were collected for Storm 22 and two samples for Storm 23. Storm 24 was the main runoff event and was composed of several peaks in discharge.

The first of the sampling periods, Storm 22, showed the highest dissolved solids concentrations, which probably represented the first flush of soluble material from the catchment. The storm also exhibited a first flush of solid material.

The second sampling period, Storm 23, was judged to form the initial stages of the main sampling period Storm 24. The entire storm is a good example of both a strong first flush of solids and, subsequently, of solids exhaustion. The initial phases of runoff (Storm 22) were of low flow rate and were not efficient in the removal of either solids on the impervious surfaces or of solids settled in the sewer. Sediment may have been partially transported by this event but not discharged from the catchment resulting in greater quantities of solids in the sewer than

Fig. 109 -Storm 24 Discharge (1/s), suspended solids and COD concentration (mg/l)



Fig. 110 - Storm 24 Percentage cumulative discharge (CQ) and suspended solids (CSS)



existed prior to Storm 24. The rapidly rising flow at the start of Storm 24 caused an equally rapid flushing out of this material and resulted in a peak in suspended solids concentration of 323 mg/l prior to any peak in discharge. The solids concentration then fell sharply, with the occurrence of only small rises in concentration related to peaks in the discharge hydrograph, indicating that the catchment had been flushed clean of the majority of solids available for transport. COD concentrations bore a close relationship to suspended solids. Dissolved solids and nitrate concentrations, although variable maintained the same concentration at the end of the storm, as at the beginning, indicating that the majority of soluble material had been removed in the first phase of runoff (Storm 22).

# 6.2.2.2 Conclusions on the Mode of Removal of Pollutants by Stormwater Runoff.

The removal of pollutants in stormwater runoff is most conveniently discussed in two sections: the removal of soluble species and the removal of solid and solid related components.

Soluble material on urban surfaces and in the atmosphere are rapidly and easily taken into solution by flowing or falling water, the uptake per unit volume being independent of the intensity of the rainfall or the rate of flow of the overland runoff. Consequently, the shape of the pollutant removal curve of soluble species is characterised by high initial concentrations, followed by a steady decline in concentration, until a constant value is approached at the end of a storm (e.g. Storms 7 and 10). However, the rate of decline in concentration is dependent upon the volume of runoff; for an intense storm with high flow rates, the rate of decline will be more rapid than for a milder storm. The initial concentrations in runoff may be dependent upon a number of factors, but the results indicate that the antecedent dry period is probably of the most importance. The longer the antecedent dry period the greater will be the accumulation of soluble pollutants on urban surfaces and the greater will be the degradation of sediment trapped in gully pots and in sewers resulting in the release of soluble compounds, e.g. Storm 10 exhibits high dissolved species concentrations after 7 days dry weather. Additionally, the nature of the previous storm event must be taken into consideration, if the event was of low rainfall volume then efficient cleansing of the catchment or flushing out of the gully pots may not have occurred, e.g. Storm 12 had an effective antecedent dry period of zero,

however, the previous event, Storm 11, was of low volume and no road runoff occurred, resulting in high dissolved solids concentrations being recorded for Storm 12. The initial concentrations of components, whose prime source is washout from the atmosphere, will be influenced by local industrial activity, seasonal residential activity, e.g. burning of coal and the prevailing wind direction. In the winter the concentrations of particular soluble components will be largely determined by road salting.

Of the species monitored, the behaviour outlined above was typical of dissolved solids, nitrate and ammonium. Peaks in the pollutant runoff curve can occur for soluble components, but are considered to be due to the lag time between roof and road runoff and the configuration of the sewer system. For example, roofs are less susceptible to polluting activities than roads and runoff from roofs will generally have lower pollutant concentration. Since depression storage is minimal for roofed areas, the early phases of runoff will be dominated by roof runoff, which has lower dissolved pollutant concentrations. When the percentage of road runoff at the sampling point increases, this may result in a peak in concentration, e.g. dissolved solids and chloride in Storm 2. However, if high concentrations of soluble components are deposited in the sewer, then this effect may be masked.

The mode of removal of solid material is considered as occurring in two fractions:-

- a) Fine sediments and particulates that are removed independent of flow rate and comprise the first flush.
- b) Heavy particulates whose removal is dependent upon flow rate.

The fine dust fraction behaves similarly to soluble components and is believed to be of most importance in determining the quality of the first flush. As for soluble compounds, fine particulates are rapidly and easily transported by flowing water and their removal appears to be independent of the rate of runoff. This results in high concentrations of suspended solids occurring in the early phases of runoff, which were unrelated to peaks in discharge and were found to be characteristic of many of the events monitored, e.g. Storms, 7, 9, 10, 15, 16, 18, 22 and 24. Therefore the first flush of solid material may be ascribed to the flow independent removal of sediments from the following sources:-

- a) Fine solids accumulated on roof and road surfaces from atmospheric dustfall.
- b) Fine sediments deposited in sewers on the recession limb of the previous storm.
- c) Fine sediments deposited in gully pots on the recession limb of the previous storm.
- d) Dusts, sediments and other material (bacterial scums) in suspension and input to gully pots over dry periods.
- e) Break-up of sediments in sewers and gully pots through anaerobic degradation over dry periods.

For storms of short duration with low flow rates, the first flush removal of sediments, outlined above, dominates the shape of the suspended solids runoff curve, e.g. Storms 9, 10, 15, 16, 18, 20. The results suggest that sediments deposited in the sewer are of the most importance in defining the first flush of solid material for this type of event (see Storms 9, 10 and 11, Section 6.2.2.1).

The removal and transport of heavy particulates on impervious surfaces is largely determined by the rate of overland flow and the intensity of the rainfall event. Consequently, peaks in discharge result in peaks in solids concentration. It is usual for the solids peak to lag behind the runoff peak, due to the development of different vertical velocity and concentration profiles in the sewer. The position of mean solids concentration in a cross-section would typically be below the position of mean flow velocity, as a result, the mean velocity of solids transport is less than that of the fluid, causing a lag between flow and concentration peaks, e.g. Storm 4. The results show the time lag to be variable, which may be due to different particulate distributions being available for transport for different storms and/or variation in the removal of different particle sizes in individual storm events. Sources of material removed in this manner include soil eroded from pervious areas, such as gardens, during heavy rainfall, and material from the breakup of road surfaces, sands and grits. When all solid material on urban surfaces has been removed, or when rainfall is no longer sufficiently intense to erode material from pervious areas or transport heavy particulates, then solids exhaustion occurs. This is apparent, at the end of a storm, when peaks in runoff occur that are not matched by peaks in solids concentration, e.g. Storms 5, 13 and 24. The occurrence of solids exhaustion is believed to be dependent upon the state of development of the catchment. For example, Storm 4, in June 1977,

was a large event composed of two major runoff peaks and it showed no signs of solids exhaustion compared to similar storms in the summer of 1978. In June 1977, development of the catchment had not reached completion, exposed piles of builders sand and ungrassed areas provided a ready supply of erodable solids, whereas in 1978 development was complete. The result suggests that substantially higher pollutant loads can be expected from catchments under development.

The behaviour, outlined above, will also apply for other solids related parameters such as volatile suspended solids, oxygen demand and heavy metals. The ratio of COD and BOD to suspended solids for long events was commonly found to fall throughout the storm. This reflects the preferential occurrence of organic material in the fine solids fraction, heavier particulates being typically composed of the largely inorganic sand and grit fractions. COD removal was found to closely follow that of suspended solids, the ratio of COD to suspended solids tending to be greater for small storms than for large, due to the effect described However, BOD concentration, although influenced by high solids above. concentrations, did not imitate the pattern of solids removal, but tended to follow a hybrid curve between the solids and soluble behaviour. The removal of oil and organic compounds created during anaerobic degradation of sediment would typically be the same as the removal of soluble components and appears to be of equal influence as solids in determining the shape of the BOD removal curve.

## 6.3. THE CONTRIBUTION OF POLLUTANTS IN GULLY POTS TO STORMWATER RUNOFF

The contribution of stored water in gully pots prior to a runoff event to the total storm runoff volume was examined using the mathematical model described in Section 5.1.5. The model predicted the flow, throughout the hydrograph, attributable to the discharge of original gully liquors. A knowledge of the average concentration of pollutants in gully pots prior to the storm event enabled the pollutant load from gully pots to be calculated.

Since the discharge of gully pollutants was likely to be of more importance in the initial phases of a storm their contribution to the first flush was examined using a definition of the first flush based upon the discharge of gully liquors (Section 5.1.5). Pollutant contributions from gully pots were assessed for all storms where analyses of gully pot liquors prior to the storm event were available. The contribution for other storms, where no such information was available, was assessed using model values for the pollutants, judged from the annual survey.

## 6.3.1 Gully Liquor Contributions

As expected, the model indicated that, even for the heaviest storms, not all of the stored liquors were discharged. The maximum recorded discharge was 86% of the total original gully liquors for a storm of some 9 mm. Typically, between 50% and 70% were discharged for storms of 2 and 3 mm (Figure 111 Table 38). At the sewer outfall, gully pot liquors contributed up to 17% of the total runoff volume. Their contribution was of most significance for storms of 1 to 3 mm and became of less importance for high volume storms (Figure 112). For storms of less than 1 mm, roof runoff predominated and for storms greater than 5 mm gully liquors became depleted. A similar pattern of contribution was observed for the first flush with the maximum percentage contributions occurring for storms of 1 to 3 mm. However, for storms of 1 to 3 mm, the contributions were only marginally greater than to total storm runoff. For heavier storms a substantially greater contribution to the first flush occurred than to total runoff.

## 6.3.2 Pollutant Contributions

Tables 39 and 40 list the percentage contributions of pollutants stored in gully pots to the total storm load and to the first flush load. Again, pollutant contributions were of most significance for storms of 1 to 3 mm, percentage contributions decreasing for the larger storms, e.g. dissolved solids, Figure 113. Laboratory studies had indicated that the contribution of suspended solids through the resuspension of bottom sediments was only of significance for the more intense storms when inlet flows to gully pots exceeded 0.2 l/s. For the majority of storms this situation did not occur and the suspended solids contribution was derived only from material already in suspension in gully pots. Consequently, a low suspended solids contribution was recorded for most storms reflecting this fact and illustrating that gully pots are not an important source of solids in storm runoff. Of the other constituents investigated significant contributions were found for all of them in some events, contributions up to 49% for COD were recorded, 30% for BOD, 25% for nitrate and 40% for dissolved solids. The average percentage contribution for all constituents exceeded the average contribution of 11.3% for gully liquors to total storm runoff, showing that the liquor is relatively more polluted than some other components of storm runoff, ie. roof runoff. Ammonium showed the greatest contributions suggesting





Fig. 112 - The percentage contribution of gully liquors to total runoff volume and first flush runoff volume for varying rainfall volume.



Fig. 113 - The percentage contribution of dissolved solids in gully liquors to the total storm load and to the first flush load for warying rainfall volume.



Fig. 114 - The percentage contribution of nitrate in gully liquors to the total storm load and to the first flush load for varying rainfall volumes.



Table 38 Details of the first flush and gully liquor contributions to total runoff and the

		1					the second s	No. of Concession, name of				
	Mean	1	ħ6	73.6	5.64	37.5	53	54.3	11.3	14.0	69.4%	64.9%
	5tł	2.74	111	243.44	8.99	67.65	23	86.5%	3.4%	12.2%	27.8%	20.7%
	21	2.42	178	48.77	5.59	40.18	147	53.8%	11.5%	12.5%	52.4%	82.6%
	20	1.8	56	50.72	7.02	40.81	41	67.5%	13.8%	15.7%	80.5%	73.2%
	19	0.79	33	7.94	0.61	7.46	28	6.0%	7.7%	7.7%	94.0%	84.8%
10.02	17	8.57	116	131.72	8.82	68.79	69	84.8%	6.7%	11.6%	52.2%	59.5%
: emulo	16	1.86	105	52.90	6.79	42.21	85	65.3%	12.8%	14.6%	79.8%	81%
1 Jonbil	15	0.97	162	22.73	3.83	19.07	44	36.8%	16.8%	18.2%	83.9%	718
Gully.	12	5.87	153	190.53	8.94	71.63	63	86%	4.7%	11.3%	37.6%	41.2%
TEJOI) L	10	1.76	trtt	48.09	h.µ.	24.63	34	42.7%	15.7%	16.5%	51.2%	277%
ISNTI 1S	6	1.17	017	14.52	1.69	13.05	28	16.3%	11.6%	11.8%	89.9%	20%
JLI	7	3.39	011	42.53	6.53	32.53	22	62.8%	15.4%	18.2%	(58.5) 76.5%	55%
	9	1.8	86	28.94	4.42	22.15	54	42.5%	15.3%	18%	76.5%	62.8%
	Storm Number	Rainfall (mm)	Storm Duration (min)	Total rungff volume (m <sup>3</sup> )	Volume of gully :liquors dis. (m3)	Volume of first flush (m <sup>3</sup> )	First flush duration (min)	% Gully liquors discharged	% gully liquors contribution to total runoff	<pre>% gully liquors contribution to first flush</pre>	% first flush volume to total runoff.	Auration flush storm duration

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TABLE 39 Percentage of pollutant contributions from gully pots to the total

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<sup>††</sup> HN	20.2	12.3	31.0	29.8	ı	9.5	24.5	80.9	30	45.3	1	31.5			
R003	18.8	11.7	18.0	15.3	t	2.6	24.6	19.8	18	20.2	8.4	15.7			
BOD	30.4	12.0	12.2	6.4	1	2.7	19.5	12.6	10.4	19.3	1	13.9			
COD	42.2	6.5	12.1	5.1	19.4	3.9	9.7	47	27.6	49.2	2.9.	20.5			
DS	21.3	20.5	18.2	31.2	13.3	3.8	40.2	30.5	14.7	26.5	7.7	20.7			
SS	13	1.9	4.3	7.3	3.0	4.4	19.1	22.4	6.9	31.0	3.2	10.6			
	Q	7	6	10	12	13	15	18	19	20	24	Mean			

TABLE 40 Percentage of pollutant contributions from gully pots to the first flush load.

SS DS COD BOD NO3 MH4   6 11.7 19.6 38.1 27.6 17.4 19.1   7 2.3 28.5 7.6 15.5 16.1 16.1   9 3.9 16.9 10.9 11.1 16.8 28   10 7.3 31.2 6.4 5.1 15.3 29.8   12 3.5 16.7 21.9 - - -   13 11.8 8.9 7.0 4.8 19.7   14 8.9 7.0 4.9 5.5 27.6   15 18.5 41.5 9.7 19.4 25.5 27.6   19 6.3 13.8 25.0 9.4 17.0 24.7   20 31.0 27.6 46.9 19.0 20.2 43.7   24 5.4 11.1 22.9 20.2 43.7   24 5.4 11.0 20.2 14.0 7.0   24 5.4 11.0 20.2 14.0 7.0   24 5.4 19.0 20.2 14.0 17.6   24 5.4 10.0 20.2 43.7   24													 	 		
SS DS COD BOD MO3   6 11.7 19.6 38.1 27.6 17.4   7 2.3 28.5 7.6 16.1 16.1   9 3.9 16.9 10.9 11.1 16.8   10 7.3 31.2 6.4 5.1 15.3   12 3.5 16.7 21.9 - -   13 11.8 8.9 7.0 4.8   14 11.8 8.9 7.0 4.8   13 11.8 8.9 7.0 4.8   14 11.8 8.9 8.9 7.0 4.8   15 18.5 41.5 9.7 19.4 17.0   16 20.2 28.2 42.5 11.5 18.2   19 6.3 13.8 25.0 9.4 17.0   20 21.0 27.6 46.9 19.0 20.2   24 5.4 18.6 3.8 - 24.7   Mean 11.1 22.9 20.2 24.7 24.7	τt	19.1	16.1	28	29.8	ı	19.7	27.6	76.3	27.0	43.7	I	31.9			
SS DS COD BOD   6 11.7 19.6 38.1 27.6   7 2.3 28.5 7.6 15.5   9 3.9 16.9 10.9 11.1   10 7.3 31.2 6.4 5.1   12 3.5 16.7 21.9 -   13 11.8 8.9 6.4 5.1   13 11.8 8.9 6.4 5.1   15 18.5 14.15 21.9 -   13 11.8 8.9 8.9 7.0   14 5.1 13.8 25.0 9.4   19 6.3 13.8 25.0 9.4   20 31.0 27.6 46.9 19.0   21 5.4 18.6 3.8 -   19 6.3 13.8 25.0 9.4   20 31.0 27.6 46.9 19.0   24 5.4 18.6 3.8 -   Mean 11.1 22.9 20.2 14.0	NO3	17.4	16.1	16.8	15.3	ı	4.8	25.5	18.2	17.0	20.2	24.7	17.6			
SS DS COD   6 11.7 19.6 38.1   7 2.3 28.5 7.6   9 3.9 16.9 10.9   10 7.3 31.2 6.4   10 7.3 31.2 6.4   12 3.5 16.7 21.9   13 11.8 8.9 8.9   13 11.8 8.9 8.9   13 13.6 41.5 9.7   19 6.3 13.8 25.0   19 6.3 13.8 25.0   19 5.4 18.6 3.8   19 5.4 18.6 3.8   19 6.3 13.8 27.6   20 21.0 27.6 46.9   21 5.4 18.6 3.8   19 5.4 18.6 3.6   20 21.1 20.2 20.2	BOD	27.6	15.5	11.1	5.1	l	7.0	19.4	11.5	9.4	19.0	I	14.0		·	
SS DS   6 11.7 19.6   7 2.3 28.5   9 3.9 16.9   10 7.3 31.2   12 3.5 16.9   13 11.8 8.9   15 11.8 8.9   16 7.3 31.2   17 18 28.2   18 20.2 28.2   19 6.3 11.5   19 6.3 13.8   19 6.3 13.8   20 31.0 27.6   21 5.4 18.6   21 5.4 18.6   21 5.4 18.6	COD	38.1	7.6	10.9	6.4	21.9	8.9	9.7	42.5	. 25.0	46.9	3.8	20.2			
SS 6 11.7 6 11.7 2.3 9 3.9 10 7.3 12 13 13 13 11.8 13.5 13 13 11.8 15 18.5 18.5 18.5 18.5 18.5 18.5 18.5 11.1 Mean Mean 11.1	DS	19.6	28.5	16.9	31.2	16.7	8.9	41.5	28.2	13.8	27.6	18.6	22.9			
6 13 13 16 16 16 16 16 16 16 16 16 16 16 16 16	S S	11.7	2.3	3.9	7.3	3.5	11.8	18.5	20.2	6.3	31.0	5.4	11.1			
		Q	7	6	10	12	13	15	18	19	20	24	Mean			

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TABLE 41 Percentage of pollutant loads discharged in the first flush

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. then	95.2	68.6	100	91.3	1	43.4	80.0	95.4	100	93.3	F				
800	79	65.3	6.96	89.9	ł	48.4	86.6	97.6	96.7	89.7	30.5				
BOD	99.1	69.6	100	85.3	I	35.1	90.7	99 <b>.</b> 3	100	91.4	1				
, COD	2.66	76.9	6.99	71.3	79.6	39.7	89.6	7.99	99.2	4.99	. 68.6				
DS	97.6	64.9	96.7	89.8	71.6	0.04	87.2	97.3	95.9	86.4	37.3				
S	7.99	73.6	99.2	89.6	77.7	36.8	93.3	99.8	100	85	52.7				
	9	7	6	10	12	13	15	18	19	20	ъ.				

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that gully pots are the most important source of this parameter, particularly for storms occurring after periods of dry weather. A contribution of 80% was recorded for Storm 18 after 10 days dry weather. Dissolved solids and COD also exhibited high contributions after 10 days dry weather: 31% for dissolved solids and 47% for COD. Since gully pots only contribute small loads of suspended solids, by inference the high COD contributions indicate that the liquors are an important source of dissolved organic compounds.

For the first flush the percentage pollutant contributions were again dependent upon the size of the rainfall event. For low volume storms, the first flush composed the majority of runoff and the percentage contributions were similar to those for the whole storm. Maximum contributions were again recorded for storms of 1 to 3 mm and were between 10% and 30% greater than for the contribution to the whole storm. For high volume storms the percentage contribution was approximately double, but still less than for storms of 1 to 3 mm (Figure 113, for dissolved solids). Therefore, whilst pollutant contributions from gully pots are greater for the first flush than for the whole storm, other pollutant sources exhibit a first flush effect of similar magnitude, which tends to make the relative contribution of gully liquors to the first flush the same as to the whole storm.

### 6.3.3 Alleviation of Pollution by Treatment of the First Flush

It has been proposed that treatment of the more polluting first flush phase of runoff could substantially reduce the pollution threat from stormwater runoff. This hypothesis was investigated for the storms monitored using the definition of the first flush in Section 5.1.5. For the storms monitored, the first flush consisted of between 20 and 150 minutes, that is, between 20% and 85% of the total storm duration, and the volume of the first flush comprised between 28% and 94% of the total runoff volume (Table 41). For short storms, the first flush composed virtually the whole storm, therefore, these storms were of less interest in defining the percentage of pollutant load, that can be removed by treating the first flush. For larger events, the relative pollution capacity of the first flush was revealed; for example in Storm 12, the volume of the first flush consisted of only 38% of the total runoff volume, however, this accounted for 78% of the suspended solids load, 72% of the dissolved solids load and 80% of the COD load. Similarly, for Storm 24, the first flush comprised 28% of the runoff and consisted of 53% of the



Fig. 115 - Relationship between the volume of the first flush and the rainfall volume.

suspended solids load, 38% of the dissolved solids load and 69% of the COD. For Storm 10 the first 50% of runoff carried 90% of the suspended solids and dissolved solids, 71% of the COD, 85% of the BOD and 91% of the ammonium load. These figures indicate that an effective treatment of the first flush could result in a substantial reduction in pollutant loads discharged.

If pollution from the first flush is to be eliminated it is important to know what volume of runoff needs to be treated to achieve this aim. Figure 115, a plot of the volume of the first flush against total rainfall, shows that for storms greater than 6 mm the first flush volume approached a constant value. For the Clifton Grove catchment approximately 80 cubic metres of runoff would need to be treated to ensure an adequate treatment of the first flush for the heaviest events. This volume is equivalent to approximately 22 cubic metres for each impervious hectare drained and would provide complete treatment for a high percentage of rainfall events (of the storms modelled only Storms 12, 17 and 24 exceeded this volume), with adequate treatment of the more heavily polluted phases of the larger events. It is likely that similar treatment volumes per impervious hectare would be applicable to other residential catchments.

Another approach to partially alleviate the problem of pollution from the first flush might be the use of porous gully pots (Mance and Harman, 1978). Whilst this approach has the dual advantage of eliminating polluted gully liquors and also of providing an initial storage volume for road runoff, its application may be hindered by the need to modify current methods of emptying gully pots and a more rapid accumulation of sediment necessitating more frequent emptying. In the Clifton Grove drainage system, gully pots could provide a storage volume up to 10.0 cubic metres. For rainfall events of less than 1.5 mm this volume could provide virtually complete retention of road runoff, e.g. Storms 9, 15 and 19. For events of less than 3 mm this volume could provide sufficient storage for the first flush of road runoff, e.g. Storms 6, 7 and 10. However, for larger rainfall events this storage volume would not greatly reduce the volume of road runoff and the occurrence of high inlet flow rates could result in high solids concentrations from the discharge of fine sediments that may have accumulated over a number of rainfall events.

#### 7. SUMMARY OF CONCLUSIONS

## 7.1. LABORATORY STUDIES ON THE REMOVAL OF MATERIAL FROM GULLY POTS

a) The removal of dissolved material from gully pots is characterised by an exponential decay curve defined by the equation below. For inlet flow rates greater than 0.12 1/s the pot liquors are completely mixed, but, below this value 'short circuiting' occurs and the percentage of the pot liquors mixed is dependent upon the incoming flow rate:-

 $c = c_o e^{-tQP/V}$ 

when

0 > 0.12 P = 1.0 0 < 0.12P = 0.1107Q + 0.1986where c = concentration in outflow

- $c_{o}$  = initial concentration in the liquor t = time
- Q = inlet flow rate
- V = volume of the gully pot

Ρ = percentage mixed.

b) The removal of solid material occurs in two portions: material already in suspension, which obeys the same equations as dissolved material, and material resuspended from the bottom sediments by the hydraulic stirring action of incoming runoff. The mass of material released from the bottom sediments and the rate of uptake of this material is dependent upon the incoming flow rate:-

 $0 \leq t \leq M/K$ when

$$c = \frac{K}{\Omega} \quad (1 - e^{-tQ/V})$$

when

ι

 $M/K \leq t \leq \infty$ 

 $c = \frac{K}{\Omega}$  (1 -  $e^{-MQ/KV}$ )  $e^{(-tQ/V + MQ/KV)}$ 

c) The laboratory results show gully pots to be efficient in retaining trapped sediments, a maximum of only 0.2% of the bottom sediments

were removed by the highest flow rate examined of 1.0 1/s. The solid material that was removed was taken up rapidly and recorded concentrations of 90 mg/l suggest that it may be of importance in the first flush. However, the removal of dissolved pollutants from gully pots represents a greater threat to stormwater quality than the removal of sediment.

## 7.2. THE RAINFALL - RUNOFF MATHEMATICAL SIMULATION

- a) The rainfall-runoff model is composed of the following elements:
  - i) Hyetograph. Calculation of the rainfall hyetograph from raw raingauge data and application of a constant proportional loss model.
  - ii) Depression storage. Calculation of the depression storage for each sub-catchment (defined by manhole points in the pipe network) and formation of the net rainfall hyetographs applicable to each sub-catchment.
  - iii) Linear Reservoir. Calculation of the inlet hydrograph, for each sub-catchment, from the net rainfall hyetograph using a single linear reservoir.
  - iv) Muskingum-Cunge method. Calculation of the fixed parameter Muskingum-Cunge coefficients for the pipe network.
  - v) Pipe routing. Routing of the inlet hydrographs through the pipe network, using the Muskingum-Cunge coefficients, to form the hydrographs.

Within the limitations of the available data, the model has been shown to give a reliable simulation of the storm runoff hydrograph at the sewer outfall. Further refinement of the depression storage coefficients and the storage routing coefficients should improve the model's performance.

b) Integration of the equations, developed from the laboratory study, describing the removal of material from gully pots into the rainfallrunoff model enables the model to be used to predict the contribution of gully liquors to the outlet hydrographs and the concentration of dissolved pollutants in runoff derived from gully pots. Both facets of the model have given satisfactory results, but await further verification. Application of the model predicting the contribution

of solid material from gully pots to storm runoff is dependent upon further evidence that the laboratory results adequately describe the behaviour in the field.

c) Calculation of the discharge of gully liquors by the model can be used as a quantitative method of defining the duration of the first flush.

## 7.3. THE QUALITY OF WATER STORED IN GULLY POTS

- a) The quality of water stored in gully pots is highly variable and particularly polluting after periods of dry weather and winter road salting.
- b) Over dry weather periods dissolved oxygen is consumed and the subsequent anaerobic degradation of material trapped in gully pots results in a deterioration of the quality of the supernatant liquors, through the release of ammonium, hydrogen sulphide and soluble organics, which cause rises in the BOD and COD.
- c) Human activity in the gully pot sub-catchment is one of the major sources of pollution of the liquors and is the principal cause of variation in quality between gully pots. The activities affecting quality are car maintenance and servicing, car washing, grass cutting, garden treatment, salting of driveways and footpaths, litter dropping and urination and defecation by animals. COD concentrations in the pot liquors showed a dependency upon the degree of human activity in the sub-catchment (sub-catchment area), however, BOD showed no such dependency.
- d) Road salting is responsible for very high concentration of dissolved solids, salt and other species that are present as impurities in the salt. Localised heavy application by residents was the cause of extremely high concentrations.
- e) Dissolved oxygen concentrations in pot liquors were related to the sub-catchment area and the volume of runoff through the gully pot.
- f) Wide variation in the quality of stored water exists between gully pots across a catchment. The highest variation occurring during winter road salting.

### 7.4. THE QUALITY OF STORMWATER RUNOFF

- a) The concentration and annual loads of pollutants from the Clifton Grove catchment are comparable to other studies of suburban residential catchments. Average suspended solids concentration exceeded the Royal Commission standard of 30 mg/1, whereas, BOD concentration was substantially less than the 20 mg/l limit. The method of calculation of the annual pollutant loads was shown to be critical and may account for some of the variance reported in the literature.
- b) Pollutant loads are strongly related to the total storm discharge and are largely determined by the characteristics of the current rainfall event. The total rainfall volume and the rainfall intensity were of the most importance for all pollutants. The antecedent dry period was of little significance in influencing the pollutant load but was of importance in determining the pollutant concentrations in the early, or first flush, phases of runoff.
- c) Multiple regression equations can be developed for the prediction of pollutant loads from easily obtainable meteorological data. The reliable performance of such an equation is limited to the catchment from which the data was derived.
- d) The majority of events monitored exhibited a pronounced first flush effect, that is, the occurrence of high concentrations of pollutants in the initial phases of runoff that are unrelated to changes in discharge. For larger storms the first flush, although comprising only 30% to 50% of the total storm runoff, has been shown to carry the bulk of the pollutant load (between 60% and 90%).
- e) Soluble pollutants are easily transported by storm runoff and their removal is characterised by high initial concentrations followed by a steady decline until a constant value is approached at the end of a storm. Dilution by the less polluted roof runoff may result in peaks in their concentration when road runoff reaches the outfall.
- f) The removal of solid material occurs in two principal fractions, firstly the removal of fine particulates that compose the first flush, and secondly, the flow dependent removal of heavier particulates. The fine particulate fraction is composed of sediments deposited in sewers and gully pots on the recession limb of the previous storm event, dustfall accumulated on roof and road surfaces and sediments broken

up in gully pots and sewers through anaerobic degradation over dry periods. This fraction appears to be removed independent of flow rate and dominates the first flush portion of a storm. The results indicate, particularly for short storms, that solids deposited in sewers are the chief explanation of this phenomenon. The transport of heavy particulates on road surfaces is dependent upon the rate of runoff and results in peaks in solids concentration related to peaks in discharge.

g) Solids exhaustion is a phenomenon that occurs when all solid material available for transport for a given rainfall event has been removed. It is characterised by peaks in discharge that do not result in peaks in solids concentration and was observed for larger rainfall events when the catchment was completely developed. Solids exhaustion may not occur while a catchment is still under development, when piles of builders sand and ungrassed surfaces provide a supply of easily transportable material and much greater pollutant loads can be expected.

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h) The removal of solid material is also characteristic of the removal of COD and strong correlations exist between the two. However, the removal of BOD has characteristics of both solid and soluble pollutant removal. This is ascribed to the joint influence of oil and suspended solids on BOD concentration.

i) Provision of treatment for the first 22 cubic metres of runoff per impervious hectare would provide complete treatment for a high percentage of storms, ensure adequate treatment of the first flush for larger events, and substantially reduce the pollutant loads discharged.

## 7.5. CONTRIBUTION OF POLLUTANTS IN GULLY POTS TO STORMWATER RUNOFF

- a) Original gully liquors (water stored prior to a runoff event) are rarely, if ever, completely discharged. A maximum of 86% removal was recorded.
- b) Original gully liquors show the highest percentage contribution to total storm discharge (up to 17%) for storms of 1 mm to 3 mm. For heavier storms their contribution to the first flush is approximately twice their contribution to total runoff.
- c) Pollutant contributions from gully pots are also of most significance for storms of 1 mm to 3 mm. The contribution of suspended solids is not of importance compared to other sources, but significant contributions of other pollutants were recorded, up to 80% for ammonium, 49% for COD, 40% for dissolved solids, 30% for BOD and 25% for nitrate. Gully pots are the most important source of ammonium in storm runoff occurring after a dry period.
- d) In the first flush the percentage of pollutant contribution to runoff from gully pots is similar to their contribution to the whole storm. This indicates that other pollutant sources exhibit a first flush effect of similar magnitude to that from gully pots.

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#### QUALITY OF URBAN STORMWATER RUNOFF:

### AN APPRAISAL OF THE ROLE OF GULLY POT LIQUORS

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VOI IT

Submitted to the Council for National Academic Awards in partial fulfilment of the requirements for the degree of Doctor of Philosophy.

Sponsoring Establishment: Trent Polytechnic, Nottingham. Date of Submission: March, 1981. Bh.D Sic 81/Fie Bap

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VOLUME 2

## APPENDICES

Appendix	1	-	Computer	Programmes
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Appendix	3	-	Publicati	ions

### APPENDIX 1

## COMPUTER PROGRAMMES

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2	Urban Runoff Pollutant Loading Programme	6
3 .	Multiple Regression Programme	11
4.	Trent Urban Runoff Data Simulation	16

### 1. STATS PACKAGE FOR VARIATION AND WEEKLY GULLY ANALYSIS

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LIS	F OF SYME	BOLS	
a)	INPUT	(in	order of occurrence)
	NC	=	No. of rows
	NV	=	No. of columns
	NV2	=	No. of titled columns
	К	=	Output device number (5 or 3)
	M1	=	Program control parameter
	M2	=	11 11 11
			M1 = 0 - No correlation
			M1 = 1 - Correlation
			M2 = 0 - Weekly Gully Analysis
			M2 = 1 - Variation Analysis
	NAME(J)	=	Title of run (for labelling output)
	KVAR(J)	=	Column numbers
	TITLE(J)	=	Column names, e.g. GP1, GP6.
	X(I,J)	=	Data array
	Т	=	t-value for 95% confidence limits
	SEQR(J)	=	Interquartile Range
b)	OUTPUT	(exc	cluding above)
			minimum
	XMAX(I)	-	marinum
	RANGE(I)	_	
	YBAR(I)	-	mean
	STD(J)	-	standard deviation
	CSK(J)	_	coefficient of skewness
	CKII(1)	-	coefficient of kurtosis
	C95(J)	-	95% confidence limits
	PC05(J)	-	Percent 05% confidence limits
	WMEAN(T)	-	weekly mean
	R(I.J)	=	matrix of correlation coefficients

2

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C С С \*STATS PACKAGE FOR VARIATION AND WEEKLY GULLY ADALYSIS# Ć C C C C PROGRAMME NAME- STAT, FOR PROGRAMME USES TWO SUBROUTINES FROM THE HAG LIBRARY С GØ1AAF, GØ2PGF Ĉ THREE OPTIONS ARE AVAILABLE : ... С BASIC STATS FOR WEEKLY ANALYSIS A) M1=0 , 42=0 C C (INCOMPLETE ARRAYS MAY BE USED) B) "1=0 , M2=1 BASIC STATS FOR VARIATION ANALYSIS С (INCOMPLETE ARRAYS MAY BE USED) C CORRELATION AUALYSIS (OBLY COMPLETE ARRAYS) C) 1 = 1С DIMENSION X(100,100),SSP(100,100),STD(100),XBAR(100),XMAX(100) DIMENSION XMIN(100), RANGE(100), SEOR(100), KVAR(100) DIMENSION R(100,120), TITLE(100), NAME(50), X2(100), CSK(100) DIMERSION CKU(100),ESTD(100),C95(100),PC95(100),XAV(120) DIMERSION DEV(100), WMEAN(100) OPER(UMIT=1,FILE="G,DAT",ACCESS="SEQIN") READ(1,991) NC, LV, HV2, K, M1, M2 READ(1,992) (NAME(J); J=1,14) READ(1,993) (KVAR(J),TITLE(J),J=1,4V2) READ(1,994) ((X(1,J),J=1,"V),T=1,NC) READ(1,995) T, (SEOR(J), J=2, HV) 991 FORMAT(613)992 FORMAT(JOA5) 993 FORMAl(1 / (1 / (1 / (1 / (1 / 1)))) 994 FORMAT(5F10.4) 995 FORMAT(F8,4,7(F7,4)) IFAIL=0 IF(M1.EQ.1) GOT070 Ç C BASIC STATS С DO 10 J=2,NV N=0 DO 20 I=1,NC 1F(X(I,J).EQ.-1) GOTO20 N=N+1 X2(N) = X(I,J)20 CONTINUE CALL GOIAAF(N, X2, 0, WT, XMEAN, S2, S3, S4, XL, XG, WTSUM, IFAIL) IF(IFATL) 25,35,25 25 WRITE(5,996) IFAIL 996 FORMAT(I3) GOTOICO 35 CONTINUE XBAR(J)=XEEAN STD(J) = S2CSK(J) = S3CKU(J)=S4 XMIN(J)=XL XMAX(J)≈XG RANGE(J) = XG + XLESTD(J)=SGRT((N/(R+1))\*(S2\*\*2)) RN=N C95(J)=T\*ESTD(J)/SORT(RU)

.

C	10	PC95(J)=(C95(J)/XBAR(J))*100 CONTINUE
C C		WEEKLY MEAN CALCULATION
		Ni=0 DO 22 I=1,NC SUMW=0
		N=0 DO 21 J=2,HV IF(X(I,J),E0,-1) GOTO21
	21	N=N+1 SUMW=SUMW+X(1,J) CONTINUE
·		SUMX = SUMW + X(T, J) WMEAG(I) = SUMW/W IE(SUMW, EO, G) = MEAN(I) = -1
		JF(*PEAN(I),EQ1) GOTO22 $NI=NI+1$ $Y2(NI)=NWEAN(I)$
	22	CONTITUE CALL GUIAAF(N1,X2,0,WT,XMEAN,S2,S3,S4,XL,XG,WTSUM,IFAIL)
	75	$\frac{1}{2} \frac{1}{2} \frac{1}$
		CSK(AV+1)=S2 CSK(AV+1)=S3 CKU(FV+1)=S4
		XMIH(NV+1)=XL $XMAX(NV+1)=XG$ $RANGE(NV+1)=XG+XL$
C C	70	IF(M1,EQ,0) GOTO80 CORRELATION
C		CALL G02BGF(NC, NV, X, 50, NV2, KVAR, XAV, DEV, SSP, 50, R, S0, IFAIL)
	45	WRITE(5,996) IFALL GQT0120
С	55	CONTINUE GOT090
C.	0.0	OUTPUT
	80	WRITE(K,895)(NAME(U),U=1,10) IF(M2.EQ.0) GOTOB2 L1=2
		$\frac{1}{1} = 1 + T(((NV)/3) + 0.5)$ $DO = 84 L = 1, N10$ $WDITE(K = 89.5)(TITE(A))(TITE(A)) = 14 L25 (Y(T = A))(Y(T = A))$
	1	<pre>ARTIE(K,891)(ITTHE(I),(ITTHE(O),0=L1,L2),(X(I,1))(X(I,0)) ,J=L1,L2),I=1,NC)) WRITE(K,893)((XMIN(J),J=L1,L2),(XMAX(J),J=L1,L2),(RAHGE(J),</pre>
	1 1	J=L1,L2),(SEQR(J),J=L1,L2),(XBAR(J),J=L1,L2),(STD(J),J=L1,L2), (CSK(J),J=L1,L2),(CKU(J),J=L1,L2)) WRITF(K,894)((C95(J),J=L1,L2),(PC95(J),J=L1,L2))
		WRITE(K,897) L1=L1+5 L2=L2+5
		IF(L2.GT, NV) L2=NV

84	CONTINUE
	Geross
82	<pre>xRITE(K, 892)((TITLE(J), J=1, NV), ((X(I,J), J=1, NV), *MEA4(1),</pre>
1	] = 1 . !!(') )
	$I = (12 + EQ_{1} \times ) = V = 1 \times V = 1$
	WRITE(K, 893)((XMIN(J), J=2, NV), (XMAX(J), J=2, NV), (RANGE(J),
1	d=2.3V). (SEOR(J). $d=2.3V$ ). (XBAR(J). $d=2.3V$ ). (STP(J). $d=2.5V$ ).
Î	(CSK(J)   J=2   yy), (CKH(J)   J=2   yy))
ດຈົ	corefer
9 û	$x \in 1$ $x \in (K, SO(1) (N \land M \in (J)), J = 1, 10)$
4 8.7	$\mathbb{E}_{\mathcal{F}} = \mathbb{E}_{\mathcal{F}} = $
1	(1 + (-5V) + 1 + (-V))
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1	
891	FORMAT( (0) / 15X . 6(3X . A5. 2X) / / (16X . F3. 0. 6X . 5(F8. 2. 2X)))
802	$FOR (x_1 ( 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, $
803	$= \frac{1}{2} $
1	STER 2.2X)/(5X) $fRAMCET.SX.S(FR.2.2X)/(5X) ft.0 EANGET.IX.$
1	S(F, G, Z) > S(F, S) = S(F, G) = S(F, G) = S(F, S) = S
4	S(FR 2 2X)/(5X) (SKE $d$ FSKE $d$ FSKE $d$ FSKE 2 2X)/(5X) (KE2)(S) (S)
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### 2. URBAN RUNOFF POLLUTANT LOADING PROGRAMME

<u>LTS</u>	T OF SIME	SOLS	
a)	INPUT	(in	order of occurrence)
	К5	=	Output device number (5 or 3)
	NS		
	M1	=	program control parameters
	M2	=	11 11 11
			M1 = 0 → Recession curve
			M1 := 1 → No recesssion curve
			M2 = 0 - Graphical output
			M2 = 1 - No graphical output
	T1	=	Sampling interval (first sample)
	S1	=	Sampling duration (first sample)
	T2	=	Sampling interval (remaining samples)
	S2	=	Sampling duration (remaining samples)
	NAME(I)	Ħ	title of run for labelling output
	TIM	н	time sampling commenced
	N	=	No. of samples
	DOSEC	=	concentration of dosing solution
	DOSER	=	rate of dosing
	DATE	=	date of storm
	RLI(I)	=	lithium concentration
	SS(I)	=	suspended solids concentration
	DS(I)	Ξ	dissolved solids concentration
	BOD(I)	Ξ	BOD concentration
	COD(I)	=	COD concentration
	RNO3(I)	Ξ	nitrate concentration
	RNH4(I)	=	ammonium concentration
	CA(I)	=	calcium concentration
b)	<u>OUTPUT</u>	<b>(</b> e	xcluding above)
	DUR	=	sampling duration
	TIME(I)	=	time of each sample
	FLOW(I)	=	flow rate
	QM	=	mean flow rate

SSM	=	mean,	flow	weighted,	suspended	d solids	concentrat	ion
DSM	=	11	н	**	dissolved	d solids	concentrat	ion
CAM	=	11	11	11	calcium o	concentra	ation	
BODM	=	11	11	11	BOD conce	entration	ı	
CODM	=	41	Ħ	tt	COD conce	entration	ı	
RNO3M	1	f1	11	**	nitrate (	concentra	ation	
RNH4M	Ξ	n	**	11	ammonium	concent	ration	
CUMQ(I)	=	cumula	ative	discharge				
CUMSS(I)	=	58		suspended	solids lo	oad		
CUMDS(I)	=	11		dissolved	solids le	oad		
CUMCA(I)	=	11		calcium l	oad			
CUMBOD(I)	=	. 11		BOD load				
CUMCOD(I)	=	11		COD load				
CUMNO3(I)	=	11		nitrate l	oad		·	
CUMNH4(I)	=	"		ammonium	load			
TQ	1	total	disc	harge				
TSSL	=	11	susp	ended soli	ds load			
TDSL	=	11	diss	olved soli	ds load			
TCAL	=	"	calc	ium load				
TBODL	=	11	BOD	load				
TCODL	=	11	COD	load				
TNO3L	=	11	nitr	ate load				
TNH4L	8	11	ammo	nium load				

C C C C		会社会保险性保存合在法律保存条件资格条件资格条件资格条件资格标准保存资格存储存在资源并资金CREAN RUNOFF POLLANT LOADING PROGRAMMESS 资本在资格保存资格资格资格资格资格资格资格资格资格资格资格资格资格资格资格资格资格资格
C C C		PROGRAMME NAME 'STQUAL' LANGUAGE FORTRAN V
0000000		PROGRAMME CALCULATES FLOW VALUES FROM LITHIUM CONCENTRATIONS, CUMULATIVE POLLUTANT LOADS, TOTAL POLLUTANT LOADS AND MEAN FLOW WEIGHTED VALUES FOR OBSERVED PARAMETERS IN ANY ONE STORM, OUTPUT TABLES AND GRAPHS ARE IN A4 FORMAT,
	34	DIMENSION YAXIS(20,100),NAME(20),Y1(100) DIMENSION RLI(100),FLOW(100),SS(100),DS(100),BOD(100),COD(100) DIMENSION RNO3(100),RNH4(100),TIME(100),DATE(4),CUMSS(100) DIMENSION CUMQ(100),CUMBOD(100),CUMCOD(100),CUMNO3(100) DIMENSION CUMNH4(100),CUMDS(100),CA(100),CUMCA(100) DIMENSION Y2(100) OPEN(UNIT=1,FILE=*S.DAT*,ACCESS=*SEQIN*) READ(1,34) K5,NS,M1,M2 FORMAT(12,I3,I2,I2) READ(1,32) T1,S1,T2,S2
	32 20 23 33	FORMAT(4F5.2) READ(1,1)(NAME(I),I=1,18) FORMAT(9A4) READ(1,2)TIM,N,DOSEC,DOSER FORMAT(F6.2,I2,2F6.3) READ(1,33) DATE FORMAT(4A5) READ(1,3) (RLI(I),S5(I),DS(I),BOD(I),COD(I),RN03(I),RNH4(I),
C C	3	CALCULATION OF FLOW AND TIME ARRAYS
C	4	<pre>FLOW(1)=8 IF(RLI(2),EQ,0,0) FLOW(1)=0,0 FLOW(2)=(DOSEC/RLI(2))*DOSER*,2196 FLOW(3)=(DOSEC/RLI(3))*DOSER*,8615 DO4I=4,N FLOW(I)=(DOSEC/RLI(I))*DOSER TIME(1)=\$1/2 TIME(1)=\$1/2 TIME(2)=T1*(\$2/2) DOSI=3,N+10 TIME(I)=TIME(2)*((I=2)*T2) IF(FLOW(1),EQ,0,0) GOTO7</pre>
C C C		CUMULATIVE LOADS FOR THE FIRST TWO SAMPLES
		<pre>X=0 Y=0 Z=0 X=TIME(1)*FLOW(1)*0.06 CUMO(1)=X CUMSS(1)=X*SS(1)/1000 CUMDS(1)=X*DS(1)/1000 CUMBOD(1)=X*BOD(1)/1000 CUMCOD(1)=X*COD(1)/1000</pre>

CUMN03(1)=X\*RN03(1)/1000 CUMNH4(1)=X\*RNH4(1)/1000 CUMCA(1)=X\*CA(1)/1000 X=X/1000 Y=(S2/2)\*FLOW(2)\*0.06/1000 Z=(T1=S1)\*(FLOW(1)\*FLOW(2))/2\*0.06/1000 CUMQ(2)=CUMQ(1)+((X+Y+Z)\*1000) CUMSS(2)=CUMSS(1)+(Y\*SS(2))+(X\*SS(1))+(Z\*(SS(1)+SS(2))/2) CUMDS(2)=CUMDS(1)+(Y\*DS(2))+(X\*DS(1))+(Z\*(DS(1)+DS(2))/2) CUMBOD(2)=CUMBOD(1)+(Y\*BOD(2))+(X\*BOD(1))+(Z\*(BOD(1)+BOD(2))/2  $CUMCOD(2) \approx CUVCOD(1) + (Y \ll COD(2)) + (X \ll COD(1)) + (Z \ll (COD(1) + COD(2))/2)$ CUMNO3(2)=CUMNO3(1)+(Y&RNO3(2))+(X&RNO3(1)) +(Z\*(RN03(1)+RN03(2))/2) 1 CUMNH4(2)=CUMNH4(1)+(Y%RNH4(2))+(X%RNH4(1)) +(Z\*(RNH4(1)+RNH4(2))/2) 1 CUMCA(2)=CUMCA(1)+(Y\*CA(2))+(X\*CA(1))+(Z\*(CA(1)+CA(2))/2) ADDITION OF RECESSION CURVE IF(M1,EQ.1) GOTO50 DO 40 I=N, N+10 MaM41 FLOW(1+1)=FLOW(1)+EXP(=0.33+T2) IF(SS(I),EQ.0.0) GOTO41 SS(I+1)=((SS(I)+3.0)+EXP(+0.33+T2))+3.0 41 IF(BOD(I), EQ. 0, 0) GOTO42 BOD(I+1)=((BOD(I)=0,3)\*EXP(=0,33\*T2))+0,3 42 IF(COD(I),EQ.Ø.Ø) GOTO43 COD(I+1)=((COD(I)=5,0)+EXP(=0,33+T2))+5,0 DS(I+1) = DS(I)43 RNO3(I+1)=RNO3(I)  $RNH4(I+1) \approx RNH4(I)$ CA(I+1)=CA(I)IF(FLOW(I+1), LT.Ø.5) GOTO50 CONTINUE 40 50 NaNeM CUMULATIVE LOADS FOR REMAINING SAMPLES D061=3.N X wØ Y=Ø 2=0 X=(S2/2)\*FLOW(I)\*0.06/1000 Y=(S2/2)\*FLOW(I=1)\*0.06/1000 Z=(T2-S2)\*(FLOW(1)+FLOW(I-1))/2+0.06/1000 CUMQ(I) = CUMQ(I = 1) + ((X + Y + Z) + 1000)CUMSS(I) = CUMSS(I = 1) + (X \* SS(I)) + (Y \* SS(I = 1))+(Z\*(SS(1)+SS(1=1))/2) 1 CUMDS(I) = CUMDS(I = 1) + (X + DS(I)) + (Y + DS(I = 1))+(2\*(DS(I)+DS(1=1))/2) 5 CUMBOD(I) = CUMBOD(I=1) + (X \* BOD(I)) + (Y \* BOD(I=1))+(Z\*(BOD(I)+BOD(I=1))/2) 5  $CUMCOD(1) \approx CUMCOD(1 \approx 1) + (X \approx COD(1)) + (Y \approx COD(1 \approx 1))$ \*(Z\*(COD(1)+COD(1=1))/2) ٤ CUMNO3(I)=CUMNO3(I=1)+(X\*RNO3(I))+(Y\*RNO3(I=1)) \*(Z\*(RNO3(I)\*RNO3(I\*1))/2) 1 CUMNH4(I) = CUMNH4(I = 1) + (X = RNH4(I)) + (Y = RNH4(I = 1))+(2\*(RNH4(I)+RNH4(I\*1))/2)1 CUMCA(I) = CUMCA(I = 1) + (X = CA(I)) + (Y = CA(I = 1))6

C C C

C C C 1 +(Z\*(CA(I)+CA(I=1))/2)

CCC

C C C 17

CALCULATION OF MEAN VALUES AND TOTAL LOADS

Z=(S2/2)\*FLOW(N)\*0.06/1000  $TQ = CUMQ(N) + (2 \times 1000)$  $TSSL=CUMSS(N) + (Z \otimes SS(N))$ TDSL=CUMDS(N)+(Z\*DS(N))  $TBODL=CUMBOD(N)+(2 \otimes BOD(N))$ TCODL=CUMCOD(N) + (Z + COD(N))TNO3L=CUMNO3(N)+(Z\*RNO3(N)) TNH4L=CUMNH4(N)+(Z\*RNH4(N)) TCAL=CUMCA(N)+(Z+CA(N)) DURSTIME(N)+(S2/2) DURS=DUR\*60 QM=TO\*1000/DURS SSM=TSSL+1000/TO DSM=TDSL+1000/TQ BODM=TBODL\*1000/TQ CODM=TCODL+1000/TQ RNO3M=TNO3L+1000/TO RNH4M=TNH4L+1000/TQ CAM=TCAL=1000/TQ

#### OUTPUT

	CALL WRITE(DATE, L, K5)
	WRITE(K5,102)TIM
102	FORMAT(/23X, "TIME SAMPLING COMMENCED ", F5, 2)
	WRITE(K5,130)DUR
130	FORMAT(23X, "SAMPLING DURATION(MINS) ", F6, 2)
	WRITE(K5,131)DOSEC
131	FORMAT(23X, DOSER CONCENTRATION(G/L) , F5, 2)
	WRITE(K5,132)DOSER
132	FORMAT(23X, DOSE RATE(NL/MIN) °, F5, 2//)
	WRITE(KS,103)
103	FORMAT( * ,15X, MEASURED PARAMETER CONCENTRATIONS //16X,
Ć	$\beta$
	WRITE(K5.104)
104	FORMAT(23X, T(MINS) Q(L/S) SS(MG/L) DS(MG/L) CA(MG/L) //
•	LN=18
	DO9I#1.N
	IF(LN_LE_50)GOTO8
	CALL WRITE (DATE , L.KS)
	WRITE(K5,104)
	LNB
8	WRITE(K5.106)TIME(I).FLOW(I).SS(I).DS(I).CA(I)
106	FORMAT(19X.5F10.2)
,	LNSLNSI
Q	CONTINUE
r	IF(FLOW(1), EQ. 0, 0) GOTO152
	WRITE(KS.107)OH.SSM.DSM.CAM
107	FORMATIZE . 15% MEAN VALUES // P. 15% FLOW WEIGHTED.
6	AF10.2//1
152	TELLN, LT, AGICOTOIO
6 W M	CALL WRITECOATE L. KEY
	TNER
	eft fy max Pe

10	WRITE(K5,109)	
109	FORMAT(/* *23X,*T(MINS)	ROD(WG\P) COD(WG\P) ND3(WG\P)
2	NH&(M6/11/7) TNMP5112	
	114-1473 13/14/19:01.N	
	IF(LN_LE_S0)GOTO15	
	CALL WRITE(DATE, L, K5)	
	WRITE(K5,109)	
	$L_{N} \approx 1.0$	
15	WRITE(K5,110)TIME(I),BOD(	<pre>(I),COD(I),RNO3(I),RNH4(I)</pre>
110	FORMAT(19X,5F10,2)	
	LN=LN+1	
11	CONTINUE	
	$IF(FLOW(1), EG_0G_0)$ GOTO15	1098 DNRAV 21
	WELLEUND, JIL/BUUM, CUUM, KN ENDWATT ( / 0 0 0 EV 0 MEAN VAT	NUSMAKNHAM Lifefið é arv príðu urtæurrné
* * * M	$\Delta F (0, 0//)$	anno V. frovi chow untouting t
۰م	LNELNAA	
	IF(LN, LT, 45)GOT012	
	CALL WRITE (DATE, L, K5)	
	LN=7	
12	WRITE(K5,112)	
112	FORMAT( ° ,15X, CUMULATIV	VE PARAMETER LOADINGS %/16X,
Ç	का सभ क्या सा स्था स्था क्या क्या क्या का स्था का स्था का स्था था। संस्था सभ क्या सा स्था क्या क्या क्या क्या का स्था का स्था का स्था भाषा का स्था भाषा का स्था का स्था की स्था है	s es ta sa sa <sup>6</sup> / )
	WRITE(K5,113)	an un carran harran. Araranan
111	FORMAT(23X, T(MINS) Q(C	JU <sub>8</sub> M) SS(NG) US(NG) CA(NG)*7)
	TECLN, LE, 50)GOTOLA	,
	CALL WRITE(DATE.L.K5)	
	WRITE(K5,113)	
	LN=9	
14	WRITE(K5,114)TIME(I),CUMQ	<pre>&gt;(I),CUMSS(I),CUMDS(I),CUMCA(I)</pre>
114	FORMAT(19X,5F10,3)	
	LN∞LN+1	
13	CONTINUE	
4 4 52	WRITE(KS, 115)TQ, TSSL, TUSL FORMAT(//GR 457 / TOTAL LO	JATCAL
112	TNMALLY D', IDA, LULAD DU	ADTHOD. 14 2 3 34 10 6 31 1 1
	TECLN.LT.45)COTO16	
	CALL WRITE(DATE.L.K5)	
	LN#7	,
16	WRITE(K5,117)	
117	FORMAT(23X, "T(MINS) BOD	C(KG) COD(KG) NO3(KG) NH4(KG) <sup>6</sup> /)
	LN = LN + 3	
	D017I=1,N	
	IF(LN,LT,50)GOTO18	· ·
	CALL WRITE(DATE, L, K5)	
	WRITE(K5,117)	
1.8		ADITS CUNCADITS CUMBASITS CUMBASIS
118	FORMAT( $19X.5F19.31$	JON (*) & CONCONTINUOS (*) & CONNUS (*)
fr ap 14.	LN×LN41	
17	CONTINUE	
	WRITE(K5,119)TBODL, TCODL.	TNO3L, TNH4L
119	FORMAT(/ 00, 15X, "TOTAL LO	DADINGS ", F9, 3, 3F10, 3//)
151	D0191=1,N	
	YAXIS(1,I)=FLOW(I)	
	XAXIS(2,I) MSS(I)	
	IAALSUS, LJOCUULI]	

,如此是一个人的,我们有一个人的,我们们是一个人的,我们们们就是一个人,我们就能够帮助了。""我们们有一个人,我们有一个人,我们有不能来了。"

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YAXIS(4, I)=FLOW(I) YAXIS(5,I)=BOD(I) YAXIS(6, I)=DS(I)YAXIS(7,I)=RNO3(I) YAXIS(8,I)=RNH4(I) YAXIS(9,I)SCA(I) YAXIS(10,1)=(CUMQ(1)/TQ)\*100 YAXIS(11,1)=(CUMSS(I)/TSSL)\*100 IF(TSSL, EQ, 0, 0) YAXIS(11, 1)=0.0 YAXIS(12,1)=(CUMCOD(1)/TCODL)+100 IF(TCODL, EQ. 0, 0) YAXIS(12, I)=0.0 YAXIS(13,1)=(CUMQ(1)/TQ)\*100 YAXIS(14,T)=(CUMBOD(T)/TBODL)\*100 IF(TBODL, EQ.C. 0) YAXIS(14,I)=0.0 YAXIS(15,1)=(CUMDS(I)/TDSL)\*100IF(TDSL, EQ. 0.0) YAXIS(15, T)=0.0 YAXIS(16,I)=(CUMNO3(I)/TNO3L)+100 IF(TNO3L,EQ.0,0) YAXIS(16,I)=0.0 YAXIS(17,I)=(CUMNH4(I)/TNH4L)#100 IF(TNH4L,E0.0,P) YAXIS(17,I)=0.0 YAXIS(18,1)=(CUMCA(1)/TCAL)\*100 IF(TCAL, EQ. 0.0) YAXIS(18, I)=0.0 19 CONTINUE GRAPHICAL OUTPUT IF(M2,EQ.1) GOTO21 A=20,0 8=12.0 C=0.0 22 CONTINUE CALL PLOTS(0,0,18) CALL PLOT(5.0,5.0,=3) D02041=1,18 IF(YAXIS(I,1), EQ, 0, 0) GOTOG1 IF(C.NE.Ø.0)GOTO202 TIME(N+1) = 0.0TIME(N+2)=INT((DUR/20.0)+1.0) CALL AXIS(0,0,0,0,°TIME(MINS)°,...10,A,0,TIME(N+1),TIME(N+2)) 202 YMAX=0.0 DO 25 K=1.N Y2(K)=YAXIS(I,K) IF(Y2(K).GT.YMAX) YMAX=Y2(K) 25 CONTINUE IF(Y2(1),EQ.Ø.Ø) GOTO204 NAM#NAME(I) ¥2(N+1)≈0.0 Y2(N+2)=INT((YMAX/10,5)+1.0) IF(YMAX,LT,5,0) Y2(N+2)=0,5 1F(YMAX, LT. 2, 0) Y2(N+2)=0.2 IF(YMAX, LT. 1.3) Y2(N+2)=0.1 CALL AXIS(-C-(C\*0,2),0,0,NAM,8,B,90,Y2(N+1),Y2(N+2)) C2=C+1.0 C3=(C+0,5)=0.5 M5=INT(C2) CALL LINE(TIME, Y2, N, 1, ~1, M5) CALL SYMBOL((3\*A/4), Bod, 5-C3,0,28, M5,0,0,-1) CALL SYMBOL((3\*A/4)+0,3,B=0,5=C3,0,28,f===\*,0,0,4) CALL SYMBOL((3\*A/4)+1,3,8\*0,5\*C3,0,28,NAN,0,0,4) IF(C.NE.Ø.Ø)GOTO203

C C C

	CALL SYMBOL((2*A/10), B+1, 0, 42, STORM OF THE ", 0, 0, 13)
	CALL SYMBOL( $(2^{\alpha}A/10)+5$ , $1, 5+1, 0, 4\lambda, 0ATE/0, 0, 20)$
	CALL SIMBULLLEGATED, $beve brock block a block a block to bb = bb$
<b>*</b> * *	CAPP 21 0000000000000000000000000000000000
562	
	LF(C,GT,Z)G01060
01	14(1°60°2°0) COLORO
60	GUTU65
619	
	CALL FLUILOO, Sell, Sell
	CAUL FINISON SIN SIN SIN SIN SIN SIN SIN SIN SIN SI
	CALL DIAMERICA A A A -21
66	CAPR FRATERERERERERERERERERERERERERERERERERERE
20	
10	δ Δ C IV J of IZ kl Λ ká το ζλ
201	
600	(BRR, DLOP(AQ, Q, Q, Q))
	DOGOKEIN
90	YAXIS(I,K)::0
RØ	NAME(T)=0
21	CONTINUE
14 14	STOP
	END
	<b>我客学你会家你会家家家的家庭的</b>
	SUBROUTINE WRITE
	张武家李琼亦亦不不不不不不不
	SUBROUTINE WRITE(DATE,L,KS)
	DIMENSION DATE(4)
	WRITE(K5,100)
	WRITE(K5,101)(DATE(I),I=1,4)
	IF(L.EQ.Ø)GOTO20
	WRITE(K5,105)
100	FORMAT( 1 ///// ,35X, STORM RUNOFF DATA /35X,
C	E to be an
101	FORMAT(/'0*,15X, STORM OF THE 4A5/15X,
Ç	P (m) the second sec
105	FORMAT("+",51X,"(CONTINUED)"//)
20	RETURN
	END

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# 3. MULTIPLE REGRESSION PROGRAMME

### LIST OF SYMBOLS

a) <u>INPUT</u> (in order of occurrence)

NC	=	No. of rows
NV	8	No. of columns
NV2		dependent variable column number
К	=	output device number )5 or 3)
NAME(J)	=	title of run for labelling output
KVAR(J)	=	column numbers
TITLE(J)	=	column titles
X (I,J)	= `	data array

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b) <u>OUTPUT</u> (excluding above)

R(I,J)	н	matrix of correlation coefficients
COEFF(I,J)	=	matrix of multiple regression coefficients, standard error and t - values
CONST(I)	=	array of multiple regression constant, standard error and t - value.
RESULT(I)	=	array of multiple regression correlation coefficient, determination coefficient and corrected determination coefficient.

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c		
C	AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA	
C d	きょうちゃんをかきたまえんなななななななななななななななななななななななななななななななななななな	
C	ע היא דא דא איז איז איז איז איז איז איז איז איז אי	
č	PROGRAMME NAME MULREG, FOR	
C	PROCRAMME USES TWO SUBROUTIUES FROM THE	
č	HAG LIBRARY, GØ2CGF, GØ2BGF.	
C		
	DIMENSION X(50,200), SSP(50,50), STD(50), XHAR(50), KV	AR(50)
	DIMFUSION R(50,50), RESULT(20), COEFF(50,10), CONST(1	¢)
	DIMENSION RINV(50,50), C(50,50), WKZ(500,50), TITLE(5	2),NAME(52)
	OPEN(UNIT=1,FILE="M,DAT",ACCESS="SEQIN")	
	REAP(1,991) NC,NV,NV2,K	
991	FORMAT(413)	
	READ(1,993)(NAME(J),J=1,10)	
99(	FORMAT(10A5)	
	<pre>FEAD(1,993)(KVAR(J),TITLE(J),J≈1,NV2)</pre>	
99.	FORMAT(10(12,A5))	
	READ(1,992)((X(I,J),J=1,NV),I=1,NC)	
99;	FORMAT(6F10,4)	
	1FAIL=V	
	N1 = NV2 = 1	
,	NE=KVAR(NV2)	
	CALL GO2AGF(NC, NV, X, 50, NV2, KVAR, XBAR, STD, SSP, 50, R,	50; IFAIL)
	JF(1FATL)20, 42, 20	
20	WRITE(K, 995) IFAIL	
993	FORMAT(13)	
	GCTO60	
4	CONTINUE AND ADDADENA NUD NE VIND AGD FR D FR DACHLE ADDA	5) CA AANAA
	CALL GOZCGF(NC,NVZ,NI,XBAR,SSP,50,K,50,RESULT,COEF	r, SØ, CUEST
	1 ,RINV,50,C,52,WKZ,500,IFALL)	
5.	IF (IFALL) 80, 100, 80	
00	WKILELN, YYDJIFALL EODWARTYJ	
391		
1.01	60100M Comminum	
1 101	UNTOR WATAR (V 900) (NAME (1) 194 40)	
	ακτικίνα αθεριαματικί μου το αντάτει το του που του που του του του του του του του του του τ	۱ ·
	FDILLAR BOUND THE CHANNEL TELEVISION STATES (1) WE SHARE THE CTANK STARS THE STATES (1) A STAR STARS THE S	,
	WRITE(K 892)(TITLE(I) (CORFECT.1) $J=1$ .3) $J=1$ .21)	
	WRITE(K, 893) (CONST(T), $T=1, 3$ )	
	WRITF(K, 895)(RESULT(T), T=10, 13)	
891	FORMAT( 1 1 /// 15X. 18AS/15X. Common second and a second	9-7 TD
~ ~ ~ ~	1 Provide Antion COEFFICT	ENTS
	1 /15X, (@####################################	
89	FORMAT( 0 / 25X, A5//(15X, A5, 4X, F7, 4/))	
89	FORMAT('0',15X, 'PARAMETER', 5X, 'COEFF', 11X, 'STD ERR	* , 8X ,
	1 "T-VALUE"//,(16X,A5,8X,3(F9,4,6X)))	
89:	FORMAT(15X, 'CONSTANT', 6X, 3(F9, 4, 6X))	
89	FORMAT('0'/15X, 'STANDARD ERROR OF ESTIMATE =', F8, 4	
	1 //15X, MULTIPLE CORRELATION (R) = F8,4	
	2 //15X, DETERMINATION (R SQUARED) =",F8,4	
	3 //15X, CORRECTED R SQUARED = ", F8, 4//////)	
891	FORMAT( "0"/15X, "ANALYSIS OF REGRESSION "/15X, """""""	x3 m m F
	1 The set of an	
61	STOP	
	END	

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April March

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#### 4. TRENT URBAN RUNOFF SIMULATION

#### LIST OF SYMBOLS

a) INPUT (in order of occurrence) IWRI = Output device number (5 or 3) M1,M2,M3, program control parameters = M4,M5. No. of subcatchments NA = PIPE(I) pipe numbers Ξ PLENGT(I) ⇒ pipe length PSLOPE(I) pipe slope = RADIUS(I) pipe radius = GULLY(I) No. of gully draining each subcatchment Ξ SLOPE(I) subcatchment slope = PAVEA(I) = impervious paved area ROOFA(I) roof area = LENGTH(I) overland flow length Ξ CR(I) roof storage routing coefficients = Ν No. of raingauge data observations = TI(I) time of raingauge data = TILT(I) No. of raingauge tilts Ξ Κ dynamic calibration coefficient i.e. intensity/tilt = PERCEN constant proportional loss coefficient = date of storm DATE = NR No. of rainfall intensities Ξ AVIN average intensity for the most intense 10 minutes = RAIN(I) rainfall hyetograph = TIME(I) minute increments over hyetograph = NO = No. of samples QOBS(I) observed flow rates = TOBS(I) times corresponding to observed flows = MCA = mean pollutant concentration in gully pots FCA final pollutant concentration in storm runoff = CAOBS(I) observed pollutant concentrations Ξ

b) OUTPUT (excluding above)

HYETO:-CRAIN(I) = hyetograph adjusted for constant proportional loss b) <u>OUTPUT</u>, Contd...

DEPSTO:-		
SRAIN(I,J)	=	hyetograph for each paved subcatchment after depression storage
RRAIN(I)	=	hyetograph for roofed area
DEP(I)	=	depression storage for each subcatchment
DEPSTR	=	roof depression storage
LINRES:-		
NQ	=	No. of runoff data points
CP(I)	=	paved area storage routing coefficients
QR(I,J)	=	roofed area hydrographs (as intensities)
QP(I,J)	=	paved area hydrographs (as intensities)
QRL(I,J)	=	subcatchment input hydrographs (roofed areas)
QPL(I,J)	=	subcatchment input hydrographs (paved areas)
QINP(I,J)	11	subcatchment total input hydrographs
MUSK:-		
<b>C1,C2,</b> C3	=	Muskingum coefficients
ROUTE:-		
Q(I,J)	=	output hydrographs for each pipe
GULLYS:-		
GVOL	Ξ	volume of gully pot
VOL(I)	8	<pre>volume of original gully liquors in gully pot (VOL(I) = GVOL)</pre>
PM	=	fraction of gully liquor mixed
QG(I,J)	=	input gully liquor hydrographs for each subcatchment
QGO(I,J)	=	output gully liquor hydrographs
CA(I)	=	predicted pollutant concentrations
MAIN (Stor	rm i	Statistics):-
DUR	=	Storm duration
TR	=	total rainfall
TRV	=	total rainfall
DSL	=	depression storage volume
b) OUTPUT, Contd...

PIL	=	other losses (e.g. infiltration)
TQ2	=	total predicted discharge
AVIN	=	average rainfall intensity for the most intense 10 minutes
AVINT	=	average rainfall intensity
TOQ	H	total observed discharge
PR2	=	percentage runoff from impervious areas
TQG	=	total volume of gully liquors discharged
PGT	=	percentage gully liquor contribution to total runoff volume.
PG99	=	percentage gully liquor contribution to first flush volume
TIME 1	=	end time of first flush
TQ1	=	first flush volume
PGD	= .	percentage gully liquors discharged of the total gully liquor volume
TCA	=	total predicted pollutant load
TOCA	Ξ	total observed pollutant load

VERSION OVE NOVE!

NOVEMBER 1978 -

PROGRAMME NAME 'STMOD', LANGUAGE FORTRAN V

THE PROGRAMME CALCULATES THE STORM RUNOFF HYDROm GRAPHS, INPUT MYDROGRAPHS AND THE CONTRIBUTION TO THE RUNOFF MADE BY ORIGINAL GULLY LIQUORS AT ANY MANHOLE POINT IN AN URBAN STORMWATER SEWERAGE SYSTEM,

THE OPERATION OF THE PROGRAMME IS DETERMINED BY THE PROGRAMME CONTROL PARAMETERS (M1,M2,M3,H4) THEY DECIDE THE AMOUNT AND TYPE OF DATA INPUT, THE PASSAGE FOLLOWED THROUGH THE SIMULATION SUBROUTINES -AND THE NATURE OF THE OUTPUT, THE PROGRAMME MAY BE SUBDIVIDED INTO THE FOLLOWING PRINCIPAL SECTIONS :-

A) DATA INPUT

- B) STORM RUNOFF MODEL
  - 1. HYETO \* FORMS HYETOGRAPH FROM RAINGAUGE INPUT 2. DEPSTO \* SUBTRACTS DEPRESSION STORAGE
  - 3. LINRES . FORMS INPUT HYDROGPAPHS BY PASSING HYETOGRAPHS THROUGH A LINEAR RESERVOIR
  - 4. MUSK CALCULATES MUSKINGUM PIPE ROUTING COEFFICIENTS
  - 5. ROUTE ROUTES THE INPUT HYDROGRAPH'S THROUGH THE PIPE SYSTEM TO FORM THE OUTPUT HYDRO\* GRAPHS
  - 6. GULLYS CALCULATES THE CONTRIBUTION MADE BY ORIGINAL GULLY LIQUORS TO OUTPUT HYDROGRAPH
- C) CALCULATION OF RELEVANT STORM STATISTICS
- D) DATA OUTPUT FORMAT, A4 TABLES AND GRAPHS

DIMENSION QPL(50,200),QINP(50,200),QGO(50,200),QOUT1(50,200) DIMENSION DATE(4),QOBS(200),TOBS(200),CAOBS(200) COMMON/ONE/TI(100),TILT(100) COMMON/ONEA/RAIN(200),TIME(200) COMMON/ONEB/CRAIN(200) COMMON/TWO/SLOPE(50),SRAIN(50,200),RRAIN(200) COMMON/THREE/LENGTH(50),CR(50),PAVFA(50),ROOFA(50) COMMON/FOUR/ PLENGT(50),CR(50),PAVFA(50),ROOFA(50) COMMON/FOURA/ C1(50),C2(50),C3(50) COMMON/FOURA/ C1(50),C2(50),C3(50) COMMON/FIVE/PIPE(50) REAL K,LENGTH,MCA OPEN(UNIT=1,FILE=\*R,DAT\*,ACCESS=\*SEQIN\*)

#### DATA INPUT

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READ(1,4) IWRI 4 FORMAT(12) READ(1,10) M1, N2, N3, N4, M5, M6 FORMAT(612) 10 READ(1,30)NA 30 FORMAT(12) READ(1,35)(PIPE(I),PLENGT(T),PSLOPE(I),RADIUS(I),GULLY(I), SLOPE(I), PAVEA(I), ROOFA(I), LENGTH(I), CR(I), I=1, NA) ٤ 35 FORMAT(3F7, 3, F7, 5, F5, 1, F6, 3, 3F8, 3, F7, 3) IF(M1,EQ.0) GOTO40 READ(1,60) N,K,PERCEN,DATE,(TI(I),TILT(I),I=1,N) FORMAT(13, F8, 5, F5, 3, 4A5/(F6, 3, F3, 0)) 60 GOT070 PEAD(1,80) NR, AVIN, PERCEN, DATE, (RAIN(I), TIME(I), I=1, NR) 40 FORMAT(13, F7, 4, F6, 4, 4A5/(F7, 4, F6, 2)) 80 DO 45 K=1,NR RAIN(K)=RAIN(K)/60 45 CONTINUE 70 IF(M3,E0,0) GOT090 READ(1,100) NO, (GOBS(I), TOBS(I), I=1, NO) 100 FORMAT(13/(F7.2,F8.4)) IF(M4.EO.2) GOTO110 90 READ(1,120) MCA, FCA, (CAOBS(I), I=1, NO) FORMAT(2F8,3/(F7,3)) 120 CONTINUE 110 GOTO124 WRITE(IWRI,999)(PIPE(I),PLENGT(I),PSLOPE(I),RADIUS(I), GULLY(I), I=1,NA) 1 999 FORMAT("1"//15X, "SEWER SYSTEM DATA"/15X, " //15X, "PIPE NO", 5X, "LENGTH(M)", 3X, "SLOPE(%)", 4X, "RADIUS(M)", 1 3X, "GULLY POTS"//(17X,F5,2,7X,F5,1,7X,F5,3,7X,F6,4,6X,F4,1)) 2 WRITE(IWRI, 998)(PIPE(I), SLOPE(I), PAVEA(I), ROOFA(I), LENGTH(I) , I=1, NA) 1 998 FORMAT('1',//15X,'SUB CATCHMENT DATA'/15X,'\*\*\*\*\*\* //15X, 'PIPE NO', 5X, 'SLOPE(%)', 4X, 'PAVED AREA', 2X; 1 "ROOF AREA", 3X, "LENGTH(N)"//(17X,F5,2,7X,F5,3,7X,F7,2,5X, 2 3 F7.2.5X,F5.1)) STORM RUNOFF MODEL 124 CONTINUE CALL HYETO(K, M1, N, NR, AVIN, IWRI, PERCEN) CALL DEPSTO(NR, NA, IWRI) CALL LINRES(NA, NR, NG, AVIN, IWRI, QPL, QINP) CALL MUSK(NA, IWRI) CALL ROUTE(NG, NA, IWRI, GINP, GOUT1) IF(M2.EQ.0) GOTO140 CALL GULLYS(NQ,NA,IWRI,QPL,QGO,QOUT1,MCA,FCA,M4) 140 CONTINUE CALCULATION OF STORM STATISTICS DO 145 K=2.NQ L1 = L1 + 1IF (QOUTI (NA, K), GE, QOUTI (NA, K=1)) GOTO145 IF(QOUT1(NA,K),GE,0,5) GOTO145 NQ=L1 GOT0146 145 CONTINUE

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	146	CONTINUE
	0.20	DO 150 IFI.NO
		TRERAIN(I)+TR
		TQ=(QOUT1(NA,I)*60/PERCEN)*TQ
		TQ2 = (OOUT1(NA, T) + 6C) + TQ2
		IF(M2,EG, Ø) GOTO15C
		TQG = (QGO(NA, I) * 6C) * TQG
		TF(M4.EQ.0) GOTO150
		TCA = ((CA(I) * GOUT1(NA * I) * 60) / 1000) * TCA
	150	CONTINUE
	φα ( <sub>6</sub> 2 1.4	TPV=TR*34398,27
		AVINT=TR#60/NR
		ΑντηφΑντηφ60
		DSL #TRV ~TQ
		TE(M3, EQ, 0) GOTO170
		7=3.03
		IF(T,EQ.1) Zel.2
		$\mathbf{F}(\mathbf{F}_{2}, \mathbf{F}_{2}, \mathbf{Z}) = \mathbf{Z} = \mathbf{Z}$
		TOO=TOO+(OOBS(I)+60+Z)
		$TF(M4,EG,\mathcal{E})$ GOTO160
		TOCA=TOCA+((CAOBS(I)*QOBS(I)*60*Z)/1000)
	160	CONTINUE
	•	PR=(T00/T02)*100
		PERCEN=PR/100
		PIL=TO-TO2
		PR2=TQ2/TRV*100
	170	CONTINUE ,
		IF(M2,EQ,8) GOTO195
		DO 180 I=1,NQ
		TQG1=(QGO(NA,I)+60)+TQG1
		$TQ1 = (OOUT1(NA, I) \times 60) + TQ1$
		TIMF1=TIME(I)
		IF(TQG1/TQG,GE,Ø,9) GOTO190
	180	CONTINUE
	190	PG99=(TQG1/TG1)*100
		PGT=(TQG/TQ2)*100
	100	LOD=(IAC)(IIS+AS'R))*100
	195	LINEINT(TIME(NR)) TOLETNO(TIME(A))
		DID=1N#(DID/CG/TCDID"(IN#CDID/CG/TCG/TCG/TCG/T) DOKE((TIN#TIT)#DE)A((TIWE(UK)#TIN)#(ITWE(I)ATTI))#IRR
		DO 107 1-1 ND
	107	
C	1 2 1	
č		OUTPUT FORMAT
č		
Ŷ		WRITE(IWRI,200) DATE
	200	FORMAT( 1 //, 30X, STORM RUNOFF MODEL OUTPUT /
	1	30X, "
	2	, 4A5/15X, "
	3	//,15X, "PREDICTED DATA //15X, "==========="
		IF(M2,EQ.1) GOTO220
		WRITE(IWRI,230) (TIME(I),RAIN(I),QOUT1(NA,I),I=1,NQ)
	230	FORMAT(//,15X, TIME(HRS) ,9X, RAIN(MM/HR) ,6X
	1	<pre>/ FLOW(L/S) //(17X, F5, 2, 13X, F6, 3, 12X, F7, 3))</pre>
		GOTO 300
	220	IF(M4,EQ,1) GOTOZ40
		WRITE(LWR1, $250$ ) (TIME(I), RAIN(I), GOUT1(NA, I), GGO(NA, I),
	7	THE TORY

· · ·

	250	FORMAT(//,15X,°TIM	E(HRS)°, SX, °RAIN(MM/	HR) °,2X;
	2	8X, F7, 3, 7X, F7, 3))		1 0 0 0 p 7 p 7 p 7 p 7 p 7 p 7 p 7 p 7 p
	240	GOTOJEV Contteuf		
	6e '2 1)	WRITE(IWRI,260) (T	IME(I), RAIN(I), QOUT!	(NA, I), QGO(NA, I),
	260	CA(I),I=1,NQ) FORMAT(//15X_STIME	(HRS) . 3X . FRATN(HM/H)	R) <sup>e</sup> .1X <sup>e</sup> FLOW(L/S) <sup>e</sup> .
		3X, "G, FLOW(L/S)", 1	X, °CA(HG/L) °//(17X,F	5,2,7X,F6,3,6X,
	300	F7,3,5X,F7,3,4X,F7	,2))	
	308	IF(M3,EG,0) GOTO40	Ø	
	9 4 73 -	WRITE(INRI,310) DA	TE . TODU OF THE CANE OY	CONTINUED C
	320	15X Concessors	energienenen ander ander ander energienenenenen energienen bereitenenen einen	°//15X.
	2	OBSERVED DATA 115	X, <sup>e</sup> maximum maximum <sup>e</sup> )	
	900	WRITE(INRI, 320) (T	OBS(I), OOBS(I), CAOBS	(I),I=1,NO)
	320	*CA(MG/L)*//(16X,F	6.3.12X.F7.2.12X.F7.	2))
	400	CONTINUE		
	105	WRITE(IWRL, 405) DA	TE STORM OF THE (.4A5.2)	X . CONTINUED P
	1	115X, Presonante ante ante	CH FOT HID HID FOR CH FOR CH FOT HID	°//15X,
	2	STORM STATISTICS		ма <sup>#</sup> ) кууул кутым
	410	FORMAT(//15X, STOR	M DURATION (HRS) <sup>6</sup> ,1	4X. *= *. F5.2
	1	//15X, TOTAL RAINE	ALL (MM) ",15X, "= ",	F7.3
	2	//15X, TOTAL RAINF	ALL VOLUME (LITRES)	「,4X,「≒ 「,F9,1 】「,3X、「≒ 「,F9 1
	4	//15X, INFILTRATIO	N LOSS (LITRES)',9X,	fm f F8,1
	5	//15X, TOTAL PPEDI	CTED DISCHARGE (LITR	ES) = ",F9,1
	6 7	//15X, MAX INCENSI //15X, AVERAGE INT	ENSITY (MMZHR) . 19X.	$fm = f_* F f_* 4$
	,	IF((TQ-TQ2),GT,500	) GOTO416	
	A 6 6	WRITE(IWRI,415) PE	RCEN	
	416	IF(M3,EQ.Ø) GOTO42	5	·
	46.0	WRITE(IWRI,420) TO	Q, PR2	
	420	FORMAT(/)5X, TOTAL	RUNOFF 38X = .F6.	(DIIRES) = 'PEA <sup>E</sup> T
	425	IF(M2,EQ.0) GOTO44	5	
	420	WRITE(IWRI,430) TQ	G, PGT, PC /9, TIME1, TQ1	PGD TRESI ( AY (= (
	430	F7.1//15X, % GULL	Y RUNOFF/TOTAL RUNOF	F*,8X,*= *,F6,2//15X
	2	% GULLY/TOTAL FOR	THE FIRST FLUSH ,2X	, = , F6, 2//15X
	3	FIRST FLUSH TIME	$E (LITRES)^{*}_{*} BX^{*}_{*}$	F8.1//15X.
	5	S GULLY RUNOFF/TO	TAL GULLY VOLUME , 2X	, # , F6.2)
		IF(M4, EQ, Ø) GOTO44	5	
	440	FORMAT(/15X, TOTAL	PREDICTED CALCIUM (	GRAMS) 🗯 🐔
	1	F8.2//15X, TOTAL	OBSERVED CALCIUM (GR	AMS) = ',F8.2)
C	445	CONTINUE		
C		GRAPHICAL OUT	PUT	+
0		IF(M5,EQ.0) GOTOIR	00	
		A=20		
		L2=Ø		

	we was a star where the back where the transformed and the transformed
	TIDE(I)=(II======(I)=====(I)===============
6 9 /3	
0.2.45	TORS(1)=(TORS(1)=TKT(TORS(1)))+100
	IF(46,GT,C) 10ES(1)=TOBS(1)+(M6*60)
	TOBS(2) = TOBS(1) + 3
	DO 536 K=3, NO
530	TOBS(K) = TOBS(2) + ((K + 2) + 3, 03)
	CALL PLOTS(0,0,18)
M + 14	CALL PLOT(3, 0, 3, 0, m3)
550	DO 535 KHI, NG MCKA-MTNECKA
525	2 (K) = ( 1 ) = ( K) CONTTNUE
030	CALL SCALE(T(1), A, NQ, 1)
•	$T(NO+1) \approx T(NO+1) \approx 5.0$
	IF(T(NQ+1), DT, 0, 0) $T(NQ+1)=0, 0$
	$IF(T(1Q)_{0}GT_{0}(T(1Q+2)+20)*T(1)) T(1Q+2)=T(1Q+2)+1_{0}0$
	L2=L2+1
	CALL AXIS $(0, 0, 0, 0, 0, 0, 0, 1)$ (MINS) $(=10, A, 0, T(NQ+1), T(NQ+2))$
600	CONVINUE CONVINUE
500	CALL SCALF(Q(1), B.NQ.1)
	Q(NQ+2) = Q(NQ+2) + 1, 0
	CALL AXIS(0, 6, 6, 6, "FLOW(L/S)", 9, 8, 90, Q(NQ+1), Q(NQ+2))
	CALL LINE(T, G, NO, 1, 3, 1)
	CALL SYMBOL(6, 0, 14, 0, 0, 42, STORM OF THE , 0, 0, 13)
	CALL SYMBOL (11, 14, $0, 0, 42, DATE, 0, 0, 20$ )
	$\begin{array}{cccc} CVII & CVIIOI(C, 0, 13, 5, 0, 00, 14, 0, 01, 0, 0, 10, 0, 0, 10, 0, 0, 10, 0, 0, 10, 0, 0, 10, 0, 0, 10, 0, 0, 0, 0, 0, 0, 0, 0, 0, 0, 0, 0, 0$
	CALL SYMBOL(11.2.13.5.0.14. (EVERY 3RD POINT SHOWN) 0.0.23)
	IF(M3.EQ.0) GOTOGØØ
	IF(L2,E0,2) GOTO650
	IF(L2,E9,3) GOT0750
	DO 540 K=1,NO
	T(K) © TOBS(K)
S A (A	
240	CALL LINE(T.O.NO.1. $+1.2$ )
•	CALL SYMBOL(6.0.13.0.0.28.2.0.0.#1)
	CALL SYMBOL(6, 3, 13, 0, 0, 28, " OBSERVED FLOW", 0, 0, 17)
560	CONTINUE
688	CALL G2(RAIN, TIME, NQ, A)
	IF(M2,E0,0) GOT0699
	$\begin{array}{c} CAUL PLOT(0, \ell, 25, 0, *3) \\ COMOFER \end{array}$
660	CONTENTE
000	DO 660 KELNO
	Q2(K)=QGO(NA,K)
660	CONTINUE .
	DO 640 K=1,NQ
	Q(NQ+K+2)=Q(K)
C 4 0	Q(K) = Q(K)
040	CNTITUE CNTITUE CONTITUE
	$\begin{array}{c} \mathbf{C} \mathbf{A} \mathbf{L} \mathbf{L} = \mathbf{A} \mathbf{A} \mathbf{L} \mathbf{L} \mathbf{A} \mathbf{A} \mathbf{A} \mathbf{A} \mathbf{A} \mathbf{A} \mathbf{A} A$
	CALL SYMBOL(6.3.13.0.0.28. "mengully LTQUOR CONTRIBUTION".
1	0,0,29)
•	DO 670 K=1,NQ
	Q(K) = Q(NQ + K + 2)

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Q(NO+K+2)=0
  670
         CALL G2(RAIN, TIME, NQ, A)
         GOTOTES
         12=12+1
  699
         CONTINUE
  700
         IF(M4, EQ, 0) GOT0900
         CALL PLOT(0,0,25,0,-3)
         GOT0550
         CONTINUE
  750
         DO 755 K=1,NQ
         C(K)=CA(K)
  755
         CALL SCALE(C(1), B, NO.1)
         CALL AXIS(-1.5,0.8, "CALCIUM(MG/L)", 13, B, 90, C(NO+1), C(NQ+2))
         CALL LINE(T, C, NO, 1, -1,2)
         CALL SYMBOL(6,0,13,0,0,28,2,0,0,+1)
         CALL SYMHOL(6.3,13.0,0.28, "===PREDICTED CALCIUM(MG/L)"
         ,0,0,26)
      2
         DO 766 K=1,NQ
         T(K)=TOBS(K)
  760
         C(K) = CAOBS(K)
         CALL LINE(T, C, NQ, 1, -1, 3)
         CALL SYMBOL(6,0,12,5,0,28,3,0,0,*1)
         CALL SYMBOL(6, 3, 12, 5, 0, 28, " MANOBSERVED CALCIUM(MG/L)"
         .0.0.25)
      1
  900
         CALL PLOT (0.0,0,0,999)
 1000
         STOP
         END
0000000
               金雅乘水水水水水水水水水
               SUBROUTINE G2
               ***
               PROGRAMME PLOTS HYETOGRAPH
         SUBROUTINE G2(RAIN, TIME, NQ, A)
         DIMENSION RAIN(200), TIME(200)
         DIMENSION TIM2(200), RAIN2(200)
         CALL PLOT (0, \ell, 0, 0, 3)
         CALL PLOT(0.0,12.0,-3)
         CALL SCALE(TIME(1), A, NO, 1)
         TIME(NQ+1)=TIME(NQ+1)=5.0
         IF(TIME(NQ+1), LT, 0,0) TIME(NQ+1)=0.0
         IF(TIME(NQ),GT,(TIME(NQ+2)*20)*TIME(1)) TIME(NQ+2)=TIME(NQ+2)+1,0
CALL AXIS(0,0,0,0," ',1,A,0,TIME(NQ+1),TIME(NQ+2))
         CALL SCALE(RAIN(1), 3, 0, NO, +1)
         CALL AXIS(0,0, "3,0, "R(MM/HR) ", "9, 3,0,90, RAIN(NQ+1), RAIN(NQ+2))
         DO 5 K=1,NQ
         TIM2(K)=(TIME(K)+TIME(NQ+1))/TIME(NQ+2)
         RAIN2(K) = RAIN(K) / RAIN(NQ+2)
    5
         CONTINUE
         CALL PLOT(TIM2(1),0.0,3)
         DO 10 K=1.NQ
         CALL PLOT(TIM2(K), RAIN2(K), 2)
         CALL FLOT(TIM2(K+1), RAIN2(K),2)
   10
         CONTINUE
         CALL PLOT(A, \emptyset, \emptyset, 3)
         CALL PLOT(A, #12, 0, 2)
         CALL FLOT (0, 0, 012, 0, -3)
         RETURN
         END
```

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C CCCCCCCCCC 教育教育教育教育教育教育教育教育教育 SUBROUTINE HYETO 保持保持保持保持保持保持保持保持 PROGRAMME FORMULATES A HYETOGRAPH (RAIN(T)) FROM RAINGAUGE INPUT, APPLIES A CONSTANT PROPERTIONAL LOSS MODEL TO FORM A NEW HYETOGRAPH (CRAIN(I)) AND DETERMINES THE MEAN INTENSITY FOR THE MOST INTENSE TEN MINUTES (AVIN). C SUBROUTINE HYETO(K, M1, N, NR, AVIN, IWRI, PERCEN) COMMON/ONE/TI(100),TILT(100) COMMON/ONEA/RAIN(200), TIME(200) COMMON/ONEB/CRAIN(200) DIMENSION AVI(200) REAL K CCCC CALCULATION OF HYETOGRAPH FOR ONE MINUTE INCREMENTS, ASSUMING BUCKET TO BE HALF FULL AT THE BEGINNING AND END OF THE STORM Ç IF(M1, EQ. 0. 0) GOTO62 W=TI(2)IF(INT(TI(2)) -INT(TI(1)), EQ.0) GOTOS W=W=R.4 5 START=(2\*TI(1)=W)+0.01 S1=(START=INT(START))\*100 IF(S1,GE,60) START=START=0,4 W2=START IX=INT(START) IY=INT(T1(1)) IF(IYHIX,EO,0)GOTO10 W2=W2+0.4 10 R=((TI(1)=W2)\*100)+1.0 NR=INT(R+,5) 2=((TILT(1)=0.5)\*K)/NR D020J=1,NR RAIN(J)=Z .20 CONTINUE NR=NR+1 D0301=2,N NR1=0 W1=0 IY1=Ø IX1=0 W1=TI(I-1) IXI=INT(TI(I)) IY1=INT(TI(I=1)) IF(IX1-IY1,E0,0)GOTO50 W1=W1+0,4 R1 = ((TI(I) = W1) + 100)50 NR1=INT(R1+0.5)DO40J=NR,NR+NR1=1 RAIN(J)=TILT(I)\*K/NR1 40 CONTINUE NR=NR+NR1 30 CONTINUE DOGØJ=NR, NR+NR1=1 RAIN(J)=((TILT(N)=0.5)\*K)/NR1

	50	CONTINUE NR#NR+NR1#1
C C C		CALCULATION OF TIME ARRAY
<b>\$</b> -	62	TIME(1)=START IF (M1,EQ,1,0) GOTO63 START =TIME(1)
	63	$\begin{array}{l} IX=I0T(START) \\ COUTINUE \\ C=INT(START*10E) = (IX*100) \\ D015I=2,200 \\ TIME(I)=TIME(I=1) + 0,01 \end{array}$
		C=C+1 IF(C,LT,60)GOTO15 TIME(I)=TIME(I)+0,40 C=0
С	15	CONTINUE
C C		PROPORTIONAL LOSS MODEL
	75	DO7SJ=1,NR CRAIN(J)=RAIN(J)*PERCEN
C		DETERMINATION OF MEAN INTENSITY
ν.		D0701=1,NR=9 AVI(I)=0 D080J=1,I+9
	80	AVI(I)=AVI(I)+RAIN(J) CONTINUE
	70	AVI(J)=AVI(I)/10 CONTINUE
		AVIN=0 D0901=1,NR IF(AVI(I),LT.AVIN)GOT090
	<b>9</b> Ø	CONTINUE RETURN END
CCCC		★★非常有效和非常有效和非常常常常。 SUBROUTINE DEPSTO 和考察者和在在在中的中心中的中心。
CCCCC		PROGRAMME CALCULATES THE HYETOGRAPHS APPLICABLE TO EACH SUB AREA AFTER DEPRESSION STORAGE HAS BEEN SUBTRACTED, (SRAIN(J,I) FOR PAVED AREAS, RRAIN(I) FOR ROOFED AREAS)
C		SUBROUTINE DEPSTO(NR,NA,IWRI) DIMENSION DEF(50) COMMON/ONEB/CRAIN(200) COMMON/TWO/SLOPE(50),SRAIN(50,200),RRAIN(200)
C C		HYETOGRAPHS FOR PAVED AREAS
Ļ		DO 10 J=1,NA IF(SLOPE(J),EQ,=1,0) GOTO5 DEP(J)=0,77*(SLOPE(J)**=0,49)

	5 8	GOTO8 DEP(J)= $1,0$ DO 20 I=1,NR IF(DEP(J),EQ, $1,0$ ) GOTO15 IF(CRAIN(I),LE,DEP(J)) GOTO60 SRAIN(J,I)=CRAIN(I)=DEP(J)	
	60 15 20 10	DEP(J)=0.0 GOTO20 DEP(J)=DEP(J)=CRAIN(I) SRAIN(J,I)=0.0 CONTINUE CONTINUE	
C		HYETOGRAPH FOR ROOFED AREAS	
	40 30	DEPSTR=0,1 DO 30 I=1,NR IF(CRAIN(I),LE,DEPSTR) GOTO40 RRAIN(I)=CRAIN(I)=DEPSTR DEPSTR=0,0 GOTO30 DEPSTR=DEPSTR=CRAIN(I) RRAIN(I)=0,0 CONTINUE RETURN END	
00000000		**************************************	,K))
C C C C C C C		ARE ROUTED THROUGH A LINEAR RESERVOIR WHERE THE ROUTING CONSTANT(CP) IS DETERMINED FROM THE SUB AREA CHARACTERISTICS(LENGTH, SLOPE)	
		SUBROUTINE LINRES(NA,NR,NQ,AVIN,IWRI,OPL,OINP) DIMENSION CP(50),OR(50,200),QP(50,200),QRL(50,200) DIMENSION MAX(50),QPL(50,200),QINP(50,200) COMMON/TWO/SLOPE(50),SRAIN(50,200),RRAIN(200) COMMON/THREE/LENGTH(50),CR(50),PAYEA(50),ROOFA(50) COMMON/FIVE/ PIPE(50) REAL LENGTH	
C		CALCULATION OF PAVED AREA ROUTING CONSTANTS	
C	15	DO 10 J=1,NA IF(SLOPE(J).EQ.=1.0) GOTO15 C=1.43*(SLOPE(J)**=0.4)*(LENGTH(J)**=0.22) CP(J)=C*(AVIN**=0.4) GOTO18 CP(J)==1.0	
C C C		CALCULATION OF INPUT HYDROGRAPH FOR EACH SUB AREA (PAVED AND ROOF)	
C	18	DO 20 K=1,NR QP(J,K)=SRAIN(J,K)/(CP(J)+0.5)	

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		$QR(J,K) = RRAIN(K)/(CR(J) \neq 0.5)$
		$IF(CP(J), EQ, = 1, \emptyset) QP(J, K) \approx \emptyset, \emptyset$
		$\mathbf{TF}(\mathbf{CR}(\mathbf{J}), \mathbf{EQ}, \mathbf{v}) = 0$ $\mathbf{QR}(\mathbf{J}, \mathbf{K}) = 0$ $\mathbf{Q}$
	20	CONTINUE
	17	CONTINUE
	<i>(</i>	NTTDETNTICA ASCOLINISC SI
		$\mathbf{N}_{\mathbf{r}} = \mathbf{N}_{\mathbf{r}} + $
		AN(VERME) F. A. COMOEN NTTE-THTEER ONCUCHTER
		tt (MITt MITT) ROTORN
		no 30 vettukante
		X an to e to
		DO 40 L=1,NITP
		IF(K∞L,EQ,0) GOTO45
		IF((1-L), NE, C) GOTO64
		ZA=QP(J,K=L)*CP(J)
		GOTO63
	64	ZA=QP(J,K=L)+CP(J)+EXP((=L+1)/CP(J))
	63	ZB = QP(J, K = L) + CP(J) + EXP( = L/CP(J))
		X=X+ZA=ZB
	40	CONTINUE
	45	QPL(J,K) = (X + (QP(J,K)/2)) + PAVEA(J)
		IF(INT(PIPE(J)), EQ.6) QPL(J,K) = 0.0
	30	CONTINUE
	50	TECHTTR.LT.1.0) COTO65
	40	DO 60 KEL.NRANTTR
		X1=0
		2(\~^D); \~^) 1((1~D); \~^) 1((1~D); \~^)
	•	
	00	LIAHQK(U,KMU)*CK(U)*CXP((GU+1)/CK(U))
	67	A1859K(J,N=0)*CK(J)*EXP(0)/CK(J))
	• •	
	70	
	75	$QRD(J,K) = (X1 \land (QR(J,K)/2)) \land ROOFA(J)$
	60	CONTINUE
	65	IF(NITP.LT.NITR) GOTO80
		MAX(J)=NITP
		GOT090
	80	MAX(J)=NITR
	<b>9</b> Ø	DO $100 \text{ K}=1, \text{NR}+\text{MAX}(J)$
		QINP(J,K)=QPL(J,K)+QRL(J,K)
		GOTO100
	100	CONTINUE
	10	CONTINUE
		NMAX=Ø
		DO 110 J=1,NA
		IF(NMAX.LT.MAX(J)) NMAX=MAX(J)
	110	CONTINUE
		NQ=UR+NMAX
		RETURN
		END
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С		SUBROUTINE MUSK
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Č		PROGRAMME EVALUATES THE MUSKINGUM COEFFECCI. C2 C31
		The second second second contractions and second contractions and the second se

FOR NA PIPES OF GIVEN LENGTH (METRES), SLOPE (%) AND RADIUS (METRES) FOR A TIME INCREMENT OF ONE MINUTE SUBROUTINE MUSK(NA, IWRI) COMMON/FOUR/ PLENGT(50), PSLOPE(50), RADIUS(50) COMMON/FOURA/ C1(50), C2(50), C3(50) REAL KSD14, LN10D2 TINC=60 DATA G32,KSD14,XC,X1/313,92,0,000002027,1,159279,0,916515/ DATA X2. DRDY. LN10D2/0, 792673, 0, 468818, 1, 151295/ CALCULATION OF COEFFICIENTS IF(NA.GT.50) GOT0900 DO 200 1=1.NA PSLOPE())=PSLOPE(I)/100 FLAREA=X2\*RADIUS(I)\*\*2 WPm2, @wRADIUS(I) wXØ W=2.04RADIUS(I)\*X1 HR#FLAREA/4P FF==1.0/ALOG13(KSD14/HR) QP=FLAREA/FF\*SQRT(G32\*HR\*PSLOPE(I)) WS=0.5\*QP\*((w+W)/FLAREA+DRDY\*(1.0+FF/LN10D2))/W AM#QP/(PSLOPE(I)+(W+W)) GAM=0.0 VPL=WS\*TINC PVPL=1.0 IF(VPL.LT.PLENGT(I)) GOTO100 PVPL=PLENGT(I)/VPL GAM=PVPL-1.0 EPS=0.5\*(AM\*PVPL)/(WS\*PLENGT(I)) 100 CNU=2,0\*PLENGT(I)/(PVPL\*TINC\*WS) X3=1.0/(1.0+CNU\*(1.0\*EP3)) C1(I)=(1,0+CNU\*(EPS+GAM))\*X3  $C2(I)=(1, \partial * CNU*(EPS+GAM))*X3$ C3(I)=((1.0+EPS)\*CNU+1.0)\*X3 200 CONTINUE RETURN 900 WRITE(IWRI,600) FORMAT("1TOO MANY PIPES . EXECUTION HALTED") 600 END 教教教教学教教教教教教教教教教教 SUBROUTINE ROUTE \*\*\* PROGRAMME EVALUATES THE OUTPUT HYDROGRAPH FOR EACH PIPE (QOUT(J,K)) USING THE INPUT HYDROGRAPH (QINP(J,K)) AND THE MUSKINGUN PIPE ROUTING COEFFTS (C1, C2, C3)SUBROUTINE ROUTE(NQ, NA, IWRI, QINP, QOUT) DIMENSION NJ(20), GINP(50, 200), GOUT(50, 200) COMMON/FOURA/C1(50),C2(50),C3(50) COMMON/FIVE/PIPE(50) CALCULATION OF OUTPUT HYDROGRAPHS 1=0 DO 101 J=1.NA

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	6	CONTINUE
		IF(J_EQ.1) GCTO30
		IF(THT(PIPE(J)) ~IVT(PIPE(J~1)), GE. 1. 8) GOTO10
		IF(INT(PIPE(J)) »INT(PIPE(J=1)), LE, =1, C) GOTO20
		DO 102 K=1.VQ
		OTUP(J,K)=OINP(J,K)+OOUT(Je1,K)
	( (3 ")	
	1612	
		601030
	10	Y II T T T T T T T T T T T T T T T T T T
		$N \cup (I) = J = J$
		GOTO3C
	20	DO 103 K=1,NQ
		QINP(J,K)=QINP(J,K)+QOUT(J=1,K)+QOUT(NJ(I),K)
	163	CONTINUE
	4. • · · ·	T =: Y w 1
	20	DO 64 KE1 NO
	347	
		GOTO124
	40	OON1(1*K+1)=(C1(1)*J1N5(1*K))+(C5(1)*G1N5(1*K+1))
	1	+(C3(J)*QOUT(J,K))
		GOTO104
	104	CONTINUE
	101	CONTINUE
	* 17 *	DO 106 JE1.NA
	A	
	197	CONTINUE
	106	CONTINUE
		RETURN .
		END
C		
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č		SUBROUTINE GULLY
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à		BRACRANCE CONDUCTS THE CONTRIBUTION TO THE
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C		OUIPOI HIDROGRAPHS MADE BI ORIGINAL GULDI LIGOURS
C		(QGO(J,K))
C		
		SUBROUTINE GULLYS(NQ,NA,IWRI,QPL,QGO,QOUT,MCA,FCA,M4)
		DIMENSION QG(50,200),QGO(50,200),VOL(200),QPL(50,200)
		DIMENSION GOUT(S0.200)
		COMMON/SIX/GULLY(50), CA(200)
	λ.	
		DO 100 Umi, NA
		DO 200 KE1,NQ
		IF(GULLY(J),EQ.0) GOTO20
		QPL(J,K)=QPL(J,K)/GULLY(J)
		PMm1.0
		IF(K.EQ.1) GOTO10
		IF(QPL(J,K),EQ,0,0) GOTO40
		$\mathbf{T} \mathbf{F} (\mathbf{O} \mathbf{P} \mathbf{L} (\mathbf{J}, \mathbf{K}) \mathbf{G} \mathbf{T} \mathbf{Z} \mathbf{Z} \mathbf{I} \mathbf{I} \mathbf{G} \mathbf{G} \mathbf{T} \mathbf{O} \mathbf{Z} \mathbf{G}$
		DKmta (((7×0))()) () () () (0) (10)
	10	CHIMILE # 1 IV/ * XCULU/N//TEN/ 10010 NAT (V)#140/ * XCULU/N//TEN/ 10010
	4.10	YOULNJ#YOULN®1J#ENELL#GEULO,NJ#EMJ/GYOUJ
		QO((),K)≃(VOD(K#1)#VOD(K))#COPP1(0)
		GOTO200
	10	GOTO200 VQL(1)=GVOL
	10	GOTO200 VOL(1)=GVOL QG(J,1)≠0

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	GOTO200
20	$QG(J,K) = \emptyset$
200	CONTINUE
100	CONTINUE
G L'L	CALL ROUTE(NO,NA,IWRI,QG,OGO)
	IF(M4,EQ.0) GOTO50
	DO 150 K=1,NO
	IF (GOUT (NA.K), FQ.C.Ø) GOTO125
	CA1: (GGO(MA,K) ** CA)/QOUT(NA,K)
	CA2#(OOUT(NA.K)=QGO(NA.K))*FCA
	CA(K) = CA(+(CA2/QOUT(NA,K))
	GOTO150
(25	
140	
150	CONTINUE
50	CONTINUE
	RETURN
	END

## APPENDIX 2

2.00

## RESULTS

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		· .	
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### 1. GULLY LIQUOR QUALITY DATA

LIST OF SY	(MBO)	LS (in order of occurrence)
R	=	total rainfall (mm) for week prior to sampling
ADP	1	antecedent dry period days
ADP>2	=	days since rainfall greater than 0.2
T. MAX	=	maximum temperature on day prior to sampling
T. MIN	=	minimum temperature on day prior to sampling
GP	=	gully pot
C95 LMT	=	95% confidence limits
C95 (%)	=	95% confidence limits expressed as a percentage of the mean
SS	=	suspended solids
DS	=	dissolved solids
COD	=	chemical oxygen demand
BOD	=	biochemical oxygen demand
CL	=	chloride
N02	=	nitrate
NH4	=	ammonium
CA	=	calcium
NA	=	sodium
К	=	potassium
ZN	=	zinc
TOC	=	total organic carbon
PB	=	lead
CD	=	cadmium
MN	=	manganese
NI	=	nickel
CR	Ξ	chromium
FE.	=	iron
CU	=	copper

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(all concentrations are expressed in mg/l)

Number	Date
1	16/ 0/76
1	16/ 8//6
2	23/ 8/76
4	6/ 9/76
5	13/ 9/76
6	20/ 9/76
7	27/ 9/76
8	4/10/76
9	11/10/76
10	18/10/76
11	25/10/76
12	1/11/76
13	8/11/76
14	15/11/76
15	22/11/76
16	29/11/76
17	6/12/76
18	13/12/76
19	20/12/76
24	24/ 1/77
26	7/ 2/77
28	21/ 2/77
30	8/ 3/77
32	21/ 3/77
34	4/ 4/77
36	18/ 4/77
38	2/ 5/77
39	9/ 5/77
40	16/ 5/77
41	20/ 5/77
44	13/ 6/77
46	27/ 6/77

# WEATHER DATA FOR WEEKLY GULLY ANALYSIS

WEEK	R(MM)	ADP	ADP>2	Т <sub>в</sub> МАХ	T.MIN
1.	000	15,00	32,00	26,00	11,50
2,	0.00	22.00	39,00	27.00	14,00
4.	3,80	5,00	5,00	21.00	8,00
5.	18,40	0,50	0,50	13,50	9 aa
6.	13,40	3,00	4,00	22,00	9,50
7.	33,40	0,00	0,00	21,00	13,00
8.	15,30	0,50	0,50	15,00	9,00
9.	21,20	0.00	0,00	20,00	8.00
10,	35 8Ø	0,00	0,00	11.00	9,00
11.	12,60	0.00	0,00	12,50	7,50
12.	11,70	1.00	1.00	11,00	5,50
13.	11,80	1,50	1,50	12,00	1,00
14.	3,20	0,50	Ø <b>.</b> 5Ø	9,00	-1,00
15.	6,30	4,00	6,90	7,00	5,00
16.	9,90	0.00	1.00	8,50	4,00
17.	6,40	0,50	0,50	7,00	3,50
18.	6,10	0,50	1,50	4,50	1,00
19:	24,00	0.00	0,00	4,00	1,00
24.	5,30	0.00	0.00	9,50	4,50
26.	19,40	0.50	0,50	9,00	5,50
28.	33,7Ø	0,00	0.00	9,50	5,06
30.	12,00	0,50	5,00	12.50	5,00
32.	12,50	0,50	0,50	7,50	5,00
34.	14.40	0.00	0.00	10,00	1,50
36.	1.20	5,00	12.00	8,00	0,00
38,	16,20	0,00	3,00	12.00	6,00
39.	16,50	1,00	1.00	14.00	4.50
40.	15,60	0.00	1,00	12.00	5.00
42.	0.00	14.00	17.00	12,50	8,00
44.	46.40	0,00	0.00	15,50	10,50
40	0,00	11,00	12,00	19,50	10,00

1. 14 IV

SUSPENDED SOLIDS (MG/L); WEEKLY ANALYSES (1976-77)

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WEEK	GP 1	GP 6	GP 7	GP 8	MEAN
1 .	7,400	12,400	12.200	m1,000	10,667
2.0	11,800	21,200	8,600	-1. CMC	13.867
4 .	9.000	16,600	8,800	m1.0000	11.467
5.	5,400	13,200	1.800	•1. CAA	6.804
6,	6.600	13,640	7.200	· 1 . 08.0	9,133
7.	0,200	38,400	2.492	4.800	11.25
8.	13,600	16,400	10,800	18,800	14.900
9.	5,800	21,800	17.600	19.400	13.903
10.	26,600	88,500	28,629	15,400	39,775
11.	7,000	27,200	15,600	10,200	15.000
12.	22,600	44,100	30,500	7,800	26.250
13.	•1,000	3,000	5.600	0,400	3.000
14.	5,200	34,800	26,300	4.600	17.725
15.	1,600	16,600	34,600	4,000	14.200
16,	7,600	38,000	12.800	8.400	16.700
17.	39,200	35,000	62,600	16.500	38,200
18,	71,400	25,200	88,200	46,800	57,940
19,	13,200	5,500	20.300	7,500	11,625
24.	2.800	24,200	57.800	2,800	21,900
26.	17,600	14,400	213,600	36,800	73,600
28.	12.600	17.000	31,400	20.000	22,25>
30.	33,200	26.200	13,600	17,800	22.700
32.	13,600	32.000	39.800	9,400	23.700
34.	4.400	11.500	10.000	15,300	19,399
36.	15.000	8,400	8.000	45,400	19.200
38.	4,800	24.400	12.800	6.800	12.200
39.	-1.000	-1.000	25,000	-1.909	25.900
40.	22.400	25,200	14.800	4.000	16,600
41.	25,000	49.600	21.700	24.500	30,225
44.	10,400	36,800	21.600	14.800	20,900
46,	18.800	58.000	29,600	30,800	34.300
MINIMUM	0.200	3,000	1.800	0.400	3,000
MAXIMUM	71,400	88.500	213,600	46.800	70,600
RANGE	71.200	85,500	211.800	46,400	67,602
I,Q RANGE	17,000	23,200	22,600	15,400	3,000
MEAN	14,993	26,640	28,845	15.312	21,298
STD DEV	14,430	17,535	39,109	12,790	14,487
SKEWNESS	2.241	1,564	3,598	1,193	1.799
KURTOSIS	5,879	3,157	13.845	0.479	3.199

### DISSOLVED SOLIDS (MG/L); WEEKLY ANALYSES (1976-77)

WEEK	GP 1	GP 6	GP 7	GP 8	MEAN
1 -	464 00	1920 00	398 00	m 1 0.9	007 22
	100 00	1705 00	350 00	- 1 Co 17	227900 855 82
A 0 A	205 00	A15 00	218 00	- 1 00	000 p 83
12 B	60 00	338 00	72 20	-1 0 C	153 63
5,	03 00	303 00	125 00	- 1 (3.2)	172 67
17 N.	55 00	303 000	4409 000	50 00	1 1 2 8 5 7
<i>и</i> е су	95 00	136 00	4 12 12 10 13		1999 670 1999 670
0	60 <u>80</u>	120 00	103 00	60 00	16.6600
10	75 30	1150,00	63 00	69 60	80,010 20,010
1.6/ 0	70,00	115.00	00,00 75 00	10 03 K V	12,23
448	10,00	143.00	12,00	18,00	24.46
144	108,00	130,00	03,00	95,00	44 MM
13.	103,00	180,00	35,00	83.00	100.25
14.	128,00	248.00	143,00	83,00	152,50
15,	120,00	268,00	198,00	140.00	181,50
16,	120,00	165,00	138,00	95.00	129,50
17,	175,00	295,00	705.00	168,00	335,75
18.	1753,00	530,00	1552,00	17475,02	5327,59
19,	275,00	218.00	68,00	345,00	226,50
24.	158,00	328,00	1285,00	268,40	494,75
26,	65,00	210,00	83,00	98.00	114.00
28.	45,00	288.20	70,03	85,00	122.00
30.	160,00	223,00	100.00	158,00	160.25
32,	94,00	181.00	115.00	83.00	118,25
34.	143,00	243.00	115,00	123.00	156.30
36	170,00	233,00	200.00	135,00	184.50
38.	98,00	213.00	110,00	108.00	132.25
39	•1.00	-1.00	123.02	-1.02	123.00
40.	125.00	208.00	120,00	148.00	150.25
42.	284,00	182.00	222.00	232.00	230.00
44	66.00	184.00	64.00	84.00	99.50
46.	124.00	210,00	150,00	162.00	161,50
MINIMUM	45.00	115.00	35.00	50.00	78.25
MAXIMUM	1753.00	1920.00	1552.00	17475.00	5327.5
RANGE	1708.00	1805.00	1517.00	17425.00	5249.25
I.Q RANGE	150.00	119.00	127.00	57.00	0.00
MEAN	197.23	343.20	231.42	817.76	375.72
STD DEV	310.33	420.98	344.80	3470.86	940 62
SKEWNESS	4.35	3.19	2.86	4.60	4.86
KURTOSIS	18.82	8.68	7.22	19.10	22.61

## BIOCHEMICAL OXYGEN DEHAND (MG/L); WEFKLY ANALYSES

#### 

WEEK	GP 1	GP 6	GP 7	GP 8	FEAN
5.	3.70	4.00	4.10	-1.42	3.93
6.	4,80	4.10	7.00	-1.00	5.30
7.	8.50	7,90	12.30	10.30	9.75
10,	1.50	1,90	2.40	0,90	1.67
11.	4,00	4,90	4.50	4.60	4.53
14.	2.90	3.70	4.00	3.60	3,55
15,	1,40	1,80	1.60	1.62	1.62
16.	3,00	2.70	3.00	3.15	2.96
24.	-1.00	3.60	5.10	1.60	3.47
26.	3,50	2.20	5.10	3.20	3.50
28.	3.40	1.90	3.30	3.50	3.03
30.	2,80	8,90	3.60	3.70	4.75
32.	4.60	6.20	6.00	3.30	5.03
38.	9.60	11.00	7.70	5,78	8.50
39,	-1,00	-1,00	5,50	-1.00	5,50
40.	11.60	15.30	8.82	11.50	11.86
42.	46.00	35.00	31.00	40.00	38 0.3
44.	4,50	2,90	4.10	4.18	3.90
46,	4,10	10,40	4,00	2.42	5,23
MINIMUM	1.40	1.80	1.60	0.90	1.60
MAXIMUM	46.00	35.00	31.00	40.00	38.00
RANGE	44.60	33.20	29.40	39.10	36.42
I.Q RANGE	3,50	7.10	3.70	3.30	0.00
MEAN	7.05	7.13	6,48	6.45	6.63
STD DEV	10.41	7.92	6.44	9.40	8.03
SKEWNESS	3,24	2,53	3.04	2.99	3.28
KURTOSIS	9.27	6.14	8,78	7.85	9 99

## CHEMICAL OXYGEN DEMAND (MG/L); WEFKLY ANALYSES

WEEK	GP 1	GP 6	GP 7	GP 8	MEAN
2.4 .	10.30	49,90	~1.60	11.00	23.73
26.	16.50	23,80	50.00	24.52	28.70
28.	8.80	15.00	28.60	18.40	17.70
30.	13.00	163.00	20,00	17.00	53.24
32.	22.40	33.90	-1.00	23.70	26.67
34.	42.50	54.50	44.60	51.00	48 16
36.	44.70	118.40	68.40	49.10	70 1
38.	28.30	52.00	37.60	30.30	27 .44
39.	~1.00	×1.00	-1.00	-1.00	
40.	29.70	51.00	31.00	28.30	35 30
42.	60.20	75.40	39.80	85.63	65 26
44.	8.70	21.10	28.50	13 42	17 00
46 .	9,20	16.30	16,30	11,30	13.28
MINIMUM	8.70	15.00	16.30	11.00	13.29
MAXIMUM	60.20	163.00	68.40	85,60	70.15
RANGE	51.50	148.00	52.10	74.60	56.87
I.O RANGE	33.30	54.30	23.00	35.70	2.00
MEAN	24.53	56.19	36.48	30.30	36.40
STD DEV	17.00	44.52	15.33	21.86	18.92
SKEWNESS	0.80	1.28	0.67	1.40	11.54
KURTOSIS	-0.68	0.55	-0.33	1.00	-1.12

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DISSOLVED OXYGEN (MG/L); WEEKLY ANALYSES (1976-77)

WEEK	GP 1	GP 6	GP 7	GP 8	MEAN
10.	9,90	9,80	9,80	10.00	9,88
11,	2,90	10.70	9,90	7.33	7.7 1
12,	2,60	9,90	10.00	6.48	7.23
13.	0.80	6,60	2.20	4.42	3.50
14.	2,80	10,00	4.60	4.52	5.48
15,	0.50	4.10	0.60	0.85	1.51
16.	3.50	9.40	9.20	6.90	7.25
17.	2.30	8,90	9.30	6.00	6.63
18.	3.78	9,50	4.63	2.20	5.02
19,	-1,00	-1.00	-1.00	·1.00	-1.00
24.	3.40	10.20	5.80	8.00	6.85
26.	8,20	10.50	10.80	9.60	9.77
28.	7,90	10.60	11.20	9.90	9.90
30	1.40	9.70	2.30	6.30	4.93
32.	3,60	11.20	7.00	9.20	7.75
34.	4.10	10.00	7.10	8.50	7.43
36	0,50	1.20	0 40	1.60	0.92
46.	-1,00	1.70	0,90	0.00	0,57
MINIMUM	0.50	1.20	0.00	0.00	0.57
MAXIMUM	9,90	11.20	11.20	10.00	9,9,3
RANGE	9.40	10,00	11.20	10.00	9.33
I.Q RANGE	2.70	3,90	7.70	7.00	0.00
MEAN	3,63	8.47	6.16	5.98	6.92
STD DEV	2.77	3.14	3,93	3.25	2.97
SKEWNESS	1,00	-1.44	-0.29	-0.48	·2.52
KURTOSIS	-0,07	0,50	-1.47	-1.09	-21.83

### NITRATE (MG/L); WEEKLY ANALYSES (1976~77)

WEEK	GP 1	GP 6	GP 7	GP 8	MEAN
1.	3,20	4.70	2,20	··· 1 . 0P	3.37
2.	3,00	3,50	2,40	=1, 0°C	2.97
4 .	1,37	10,80	1.80	·1.0C	4.66
5 .	1.40	14.00	1,28	m1.00	5,56
6,	2,00	10.50	1,70	m1.00	4.73
7	1.20	6.40	0.85	1,20	2.41
8,	0,96	6,00	1.10	1.30	2.34
9.	1.99	4.80	1.10	1.72	2.38
10.	2.00	5,00	1.70	2.59	2.84
11.	1.30	7.10	2.60	2.60	3.40
12,	1.20	2,90	0.80	2.30	1.89
13.	3.00	5.00	1.70	2.70	3.10
14.	2.60	6.50	5.40	2.60	4.27
15.	1.50	6.20	2.30	2.80	3.20
16.	1.80	3.60	2.80	1.80	2.50
17.	1.90	3.70	8.30	2.09	3.97
18.	24.00	8.30	20.00	101.00	38.33
19.	7.50	8.60	3.20	10.00	7.33
24.	3.90	8.70	16.20	4.90	8.43
26	0.80	3.50	1.30	1.40	1.75
28.	1.80	4.20	2.60	2.60	2.84
30.	4.20	13.00	2.80	3.80	5.95
32	2.60	7.10	5.80	2.15	4.41
34.	2.90	6.00	3.70	3.00	3.94
36.	2.60	14.00	11.00	3.60	7 80
38.	3.30	32.00	10.50	5.02	12.70
39.	-1.00	-1.00	24.50	-1.00	24.5
40.	11.50	132.00	42.00	14.50	50.00
42	23.00	17.20	15.00	13.00	17.00
44.	6.40	9.00	7.40	7.00	7.45
46.	1.60	7,30	1,50	1,68	3.00
MINIMUM	0.80	2,90	0.80	1.20	1.75
MAXIMUM	24,00	132.09	42,03	191.02	50.00
RANGE	23.20	129.10	41,20	99,80	48.25
I.Q RANGE	2,10	5,00	7,15	3,20	8.80
MEAN	4.21	12.38	6.63	7 88	8.03
STD DEV	5,69	23,30	8 94	19,72	10.88
SKEWNESS	2,71	4.66	2.42	4.37	2.71
KURTOSIS	6.27	20.97	6.02	17.73	6.63

### AMMONIUM (MG/L); WEEKLY ANALYSES (1976-77)

#### ម្មល្លាស្ត្រស្នុសស្ត្រាដ្ឋាស្ត្រាយនេះ ស្ត្រាស្ត្រាល សេសស្តេលស្ត្រាស់ស្តេសស្តេសស្តេសស្តេសស្តែស្តែស្តែស្តែស្ត្រស

WEEK	GP 1	GP 6	GP 7	GP 8	MEAN
1	6,05	2.97	9,13	1 . 01 (X	6.05
2.	5.30	2.64	7.82	e1.00	5 25
4	0.75	1.10	1.54	1.0C	1 1 2
5.	0.10	0.31	0.13	m1.00	0 18
6.	0.34	1.00	0.21	<b>1</b> .60	3.52
7	0.35	0.39	0.35	0 42	0 38
8	0.65	0,93	0.77	0.57	4 73
9.	9.42	0.39	0.36	0.44	0.40
10.	0.29	0.32	0.53	0.30	9.36
11.	0.45	0.32	0.34	0.37	0 37
12.	1.14	0.32	0.66	0.48	0.65
13.	1.11	0.55	0.44	0.17	0.57
14.	1.17	0.51	0.70	0.50	0.72
15.	0.55	0.94	0.57	8.41	0.62
16.	0.24	0.20	0.18	0.28	10.23
17.	0.41	0.22	0.18	0.12	0.23
18.	0.42	0.31	0 28	7.04	2.01
19.	0.30	0.24	0.33	0.35	0.32
24	0.15	0.49	0.25	0.25	0.29
26.	0.14	0 21	0.20	0.13	0.17
28.	0,12	0.13	0.13	0 08	9.12
30.	0,19	0.39	0.25	0.24	0.27
32	0.25	0 48	0.57	0.07	0.34
34.	0.83	0.45	0.45	0.18	0.48
36,	1.43	1,17	1,37	0.14	1.93
38.	0.86	0.52	0.19	0.20	3.44
39.	-1.00	-1.00	0.19	-1.00	10.19
40.	1,90	0,62	0.35	0,31	6.83
42.	6.49	3,00	3,90	3,10	4.18
44.	0,64	0,57	0,41	0,54	0.54
46.	2.10	1.34	2.42	2.32	2.95
MINIMUM	0.10	Ø.13	0,13	0.07	2.12
MAXIMUM	6.40	3,00	9,13	7.04	6.05
RANGE	6,30	2,87	9,00	6,97	5,94
L.Q RANGE	0,89	0.65	0,51	0.30	0.00
MEAN	1,17	0.77	1,13	0,76	1.02
STD DEV	1,69	0.78	2,11	1.4B	1.47
SKEWNESS	2.27	1,97	2.90	3,36	2.41
KURTOSIS	3.80	2.76	7.30	10.84	4.66

### CHLORIDE (MG/L); WEEKLY ANALYSES (1976-77)

WEEK	GP 1	GP 6	. GP 7	GP 8	MEAN
1.	9,00	16,00	12,00	··· 1 . (1 ?)	12.33
2.	12,00	18,00	=1.00	=1.0e	15.00
4 .	12,00	15,00	16.00	· 1 . () ()	14.33
5.	2,50	18.00	2.00	m1.69	7.50
6,	3,00	14.00	7.00	. 1.06	8.00
7.	3.00	8.10	3.10	3.80	4.48
8.	1.80	2.40	1,911	2.30	2.10
9.	2.40	5.40	2.50	2.73	3.25
10.	2.60	3.00	2.40	2.00	2.50
11.	-1.00	m1.00	-1.00	. 1.00	-1.80
12.	7.70	7.50	5.80	8.30	7.32
13.	4.00	4.30	2.40	2.70	3.35
14.	31.00	29.60	28.50	7.60	24.18
15.	3.00	24.50	29.40	15.70	18.15
16.	6.50	14.70	20.80	6.00	12.02
17.	37.00	27.80	327.00	17.20	192.25
18,	875.00	160.00	700.00	9800.00	2883.75
19.	85.00	16.00	16.00	126.02	68.75
24.	23,00	81,80	610.00	36.00	187.70
26.	12,50	31.00	28.50	23.00	23.75
28.	13,00	21.50	27.03	20.00	20.38
30.	24.00	31,00	26.00	29.02	27.50
32.	14.50	27.00	25.00	14.00	22.13
34.	18.00	30.00	19.50	16.50	21.00
36, .	23,50	35,50	46.00	16.50	30.38
38.	11.50	31.00	21.00	12.5%	19.00
39,	-1,00	-1.00	29.23	-1.00	29.20
40.	18,00	33,00	22.00	18,00	22,75
42.	23,00	27,00	29,50	35.50	28.75
44.	4,10	10.50	6.50	6.50	6,92
46.	12.50	21,02	17,50	10,58	15,38
MINIMUM	1.80	2.40	1,90	2.00	2.10
MAXIMUM	875,00	160,00	700.00	9800,00	2883.75
RANGE	873,20	157,60	698,10	9798,00	2881,65
I.Q RANGE	18,90	20,59	22,30	17.03	(a _ (a )a
MEAN	44,66	26,36	71.88	426,35	121,13
STD DEV	160,53	2,9,86	172,21	1996.73	523,03
SKEWNESS	4,93	3,33	2,90	4,49	5.67
KURTOSIS	22,57	11,81	6,91	18,12	23.84

## PH; WEFKLY ANALYSES (1976-77)

WEEK	GP 1	GP 6	GP 7	GP 8	MEAN
1 .	8,23	7,92	8,13	m1.00	8.09
2.	8,03	7.84	7.95	«1. CC	7.94
4.	7.63	11.39	7.63	-1.40	8.88
5.	7.50	11.44	7.69	#1.CC	8,88
6.	7,99	11.34	7.59	-1.90	8.97
7	7.29	11.40	7.58	8.11	8,64
8	7.34	11.58	7.46	7.88	8.57
9	7.37	10.94	7.54	7.55	8 35
10.	7.48	10.58	7.58	9.22	8.72
11.	7.48	10.95	7.47	7.41	8.33
12.	7.53	11.00	7.54	7.55	8.40
13.	7.51	11.50	7.63	7.61	8.56
14.	7.70	10.88	7.42	7.64	R_41
15.	7.57	10.53	7,51	7.40	8 25
16.	6.54	10.50	7.42	7 70	8 12.1
17.	7.54	12.94	7.50	10.28	9 57
18.	7.52	11.48	7.50	7.64	8 42
19.	7.64	11.00	7.92	9.65	Q (45
24.	7.63	11.23	7.74	9.74	9.99
26.	7 00	11.26	7.31	8.63	2 S S S
28	7.32	11.45	7.36	7.45	8 30
30	7.89	9.74	7.70	7.96	8 23
32.	7.59	10,92	7.49	7.47	8 37
34.	7.09	10.95	6 96	7 20	8 27
36	7.68	8.03	7.63	7.64	7 75
38.	6.57	9.14	6.35	6.30	7 00
39	-1.00	-1.00	7.16	-1.00	7 16
40.	7.48	10.68	7.41	7 30	8 22
42	7.48	7.56	7.54	6 95	7 10
44	7.01	10.77	6 71	7 91	9 10
46.	7.69	7.67	7 81	7 66	·7 74
			1 1 1 2	1.6.00	1.11
MINIMUM	6,54	7,56	6,35	6.30	7.09
MAXIMUM	8,23	12.94	8.13	10.28	9.57
RANGE	1,69	5.38	1.78	3,98	2.48
I,Q RANGE	0,33	1,27	0 21	0.70	9.00
MEAN	7,48	10.49	7.49	7,92	8.33
STD DEV	0.37	1.37	0.34	0.92	0.55
SKEWNESS	-0.70	-1.62	-1.37	1.12	•0.33
KURTOSIS	0,93	0.03	2.90	0.71	0,13

### SODIUP ("G/L); WEEKLY ANALYSES (1976\*77)

WEEK	GP 1	GP 6	GP 7	GP 8	MEAN
1 .	3,30	19.30	5.70	∞ 1 e (1 C	9.43
2 .	3.80	13.50	4.50	au 1 . 6117	7.27
4 .	2.10	6.30	3,60	mj . 100	4. 20
5.	0.40	4.30	0.64	· 1.62	1.78
6.	0,55	3.78	1.26	m1.00	1.86
7.	0.30	1.20	8.44	6.82	0.69
8.	6.40	2.05	0.57	1.12	1.04
9.	0.78	1.57	1.38	1.52	1.31
10.	1.36	1.62	1.42	1.36	1.44
11.	1.45	1.78	1.57	1.67	1.62
12.	1.82	1.42	1.37	1.77	1.59
13.	2.10	1.83	1.36	1.70	1.75
14.	4.85	11,25	8.35	2 57	6.75
15.	3.28	6.46	10.52	3.54	5.95
16.	2.63	6.21	8.46	2.65	4.99
17.	22.30	18.70	212.00	7.00	65.00
18.	630.00	83.00	414.00	5560.00	1671.75
19.	94.80	17.20	20.70	140.02	68.18
24	31.30	38.80	267.00	22.30	89.85
26.	3.28	5.74	18.80	8.24	9.01
28.	2.12	6.14	6,56	3.98	4.70
30.	8.40	10.40	13.50	14.60	11.73
32	2.83	4.00	7.90	4.15	4.72
34.	3.60	2.73	5.64	3.36	3,83
36.	5 00	7.10	13.40	5.60	7.78
38.	2.82	3,14	2.87	2.77	2.90
39.	-1.00	-1.00	2.44	-1.00	2.44
40.	3.60	2.47	6.40	3.74	4.05
42.	3.08	3,38	7.76	5.14	4.84
44	0.80	1.72	2.48	1.05	1.51
46	1,12	2,43	2,57	2,12	2,06
MINIMUM	0.30	1.20	0.44	0.82	0.69
MAXIMUM	630.00	83.00	414.00	5560.00	1671.75
RANGE	629,70	81.80	413.56	5559,18	1671.06
I,Q RANGE	2,85	8.86	9.84	5.33	Ø Av
MEAN	28,14	9,65	34 04	232,11	54.74
STD DEV	115,05	15,96	91,92	1118.32	299.03
SKEWNESS	4,93	3,52	3,11	4.68	5,17
KURTOSIS	22,83	12,80	8,59	19.09	24.81

## CALCIUM (MG/L); WEEKLY ANALYSES (1976-77)

WEEK	GP 1	GP 6	GP 7	GP 8	MEAN
1. e	43,00	178,00	34.00	=1.00	85,03
2.	95,40	194.00	40.50	m 1 . C ?	109.97
4.	25.60	57.00	25.60	-1.08	36.07
5.	8.30	50.02	8,60	-1.00	22.3.1
6.	13.00	41.70	14.70	-1.00	23.13
7.	4.00	39.00	5.00	6.30	13.58
8	6.00	79.80	8.00	11.20	26.25
9	12.20	34.60	11.40	14.22	18.18
10.	8.90	26.20	8.60	10.00	13.42
11.	15.00	31.50	12.50	12.80	17.95
12.	19.50	28.50	9.50	16.50	18.5%
13.	19.60	39.20	13.30	16.52	22.15
14.	26.00	40.50	20.50	26.00	28.25
15.	22.30	37.00	29.00	23.22	27.88
16.	18,60	29.00	17.60	15.80	20.25
17.	20.20	42.80	20.80	27.60	27.85
18.	53.00	44.50	63.00	272.00	108.13
19	21.20	43.32	9.50	23.10	24.28
24	24,00	43.40	75.00	25.00	41.85
26,	12.30	43.78	7.70	13.30	19.25
28.	10.00	54.00	10.50	14.40	22.22
30.	27.50	34,30	20.20	22.79	26.17
32,	17,50	27.00	17.50	13.80	18.95
34.	19,60	43.40	13.30	15.00	22.83
36,	21.80	37.00	35.20	28.50	30.63
38	23,50	41.00	19.00	20.70	26.05
39.	-1.00	-1.00	17.50	-1.00	17.59
40.	28,00	42.00	21.60	42.50	33,53
42.	37.50	33,80	34.40	38,70	36,10
44.	12,80	27.00	13.00	13.90	16.67
46.	15,50	18,00	16,50	17,90	16,75
MINIMUM	4,00	18.00	5,00	6.30	13,42
MAXINUM	95,40	194,00	75,00	272.00	129,97
RANGE	91.40	176,00	70,00	265,70	96,54
I.Q RANGE	13.70	9,90	15,90	12.29	0.00
MEAN	22,73	49.37	21,08	29,63	31,34
STD DEV	17,32	38.88	15,74	51.21	24.41
SKEWNESS	2,65	2,98	1.93	4,40	2,43
KURTOSIS	8,23	7.78	3,55	17.91	4.69

## POTASSING (MGZL); WEEKLY AMALYSES (1976-77)

WEEK	GP 1	GP 6	GP 7	GP 8	MEAN
1. B	2,20	8,80	3,00	-1,00	4.67
2.	5.73	17.40	8,30	a1. 33	1(1,47
4 .	2,68	16,70	3.42	m1,000	7.59
5 .	0.57	10,20	1.03	m1,000	3,93
6.	1.00	1.80	1.32	* 1 . O C	1.37
7 :	0,53	3,10	0.45	1.75	1.46
8.	0.37	4.40	0.45	2.14	1.83
9,	1.00	2.14	2,73	2.42	1.57
10.	1.20	1.20	0.53	1.33	1.06
11.	. 1.17	1.80	1.07	1.25	1.32
12,	1,75	1.30	0.80	1,95	1.44
13.	1.40	2,38	0.70	1.80	1.57
14,	1,95	2.32	1.50	3.10	2.19
15,	1.60	3,36	2.37	3.20	2.63
16.	1,85	1.80	1.55	2.00	1.82
17.	1,90	2,38	1,55	3.65	2.37
18,	3,65	4.04	4,35	12,00	6.01
19,	2,00	6,60	0.79	2.43	2.93
24.	1,90	6.20	4.94	4.35	4.35
26.	0,95	3,10	0.55	1,90	1.63
28,	0.50	6,62	0.14	1,5%	2,19
30 .	2,23	4,20	1,05	4.60	3.02
32,	1,40	3,05	1,85	1.90	2.05
34,	1,70	2.37	0,75	1,85	1.67
36.	2,63	4.75	2,47	2,63	3,12
38.	2,42	3,20	1.25	2,38	2.31
39,	-1.00	-1,00	0,95	-1.00	2.95
40.	3,53	3,75	1.25	3,14	2,92
42.	6,72	6,50	3,69	6.22	5,75
44.	1,55	3,10	0,72	1,48	1.69
46,	2,27	5,00	2,43	3,2₹	3,23
MINIMUM	0.37	1.29	0.14	1,25	0.95
MAXIMUM	6,72	17,40	8.30	12,00	10,47
RANGE	6,35	16,20	8,16	10,75	9,52
I.Q RANGE	1,75	3,99	1,72	1,30	0,00
MEAN	2,01	4,79	1.80	2,96	2,94
STD DEV	1.40	3,98	1,70	2.21	2,13
SKEWNESS	1.80	2.01	2.11	2,99	1,95
KURTOSIS	3,38	3.54	4.85	8.88	3.68

### GULLY POT QUALITY VARIATION: 3RD SEPTEMBER 1977

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GP	88	DS	<b>C</b> 0D	вор	1,1)
2.	12.80	115.00	43,42	24.80	1.55
4.	23.60	185.00	17.90	22	5,33
7	10.4.1	125.0.	15.33	1 . 1 '	7 . 12 "
19.	20,03	127.50	15.3	3,52	7.83
95.	87.63	282.50	428,63	41.6"	7.92
32.	34.30	114.00	51.00	9.31	1.73
33.	16.80	118.00	35,70	17.8	7.85
35.	190.02	142.03	91.8.	26.50	8.15
39.	16.83	126,00	30,60	3.34	7.63
36.	4.80	124,03	7.72	12.30	7.75
35.	18,00	144, 12	250,23	40.4.3	. 7.73
47.	2.87	62.5.7	5.12	5.34	7.91
111.	4.33	85, C C	2,35	2.8.	7.89
1:5.	12.07	95,00	14 n 14	18:	1.0.1
97.	21.20	102.50	28,10	9.4.	7.92
68.	252,80	177.5~	150,50	23.	8.2 .
69.	m 1 . 13.7	132.57	318,90	1.03	7.76
70.	8.4.1	162.5.	7.74	6,33	7.84
72.	24.40	292.53	1.20	15.5 *	8,13
75.	6.82	173.20	30,62	18.82	7. 55
86.	12.20	142.50	2.6%	2	7.62
92.	3,40	145,60	0.20	12.00	7.44
42.	9,60	122.50	D. 20	0.13	7,92
MINTMON	2,82	62,50	0,00	0.00	7.55
MD-IXAN	252,80	292.52	428.68	41.67	P.36
RANGE	250,00	230.00	428.63	41.60	1.83
1.0 RANGE	16,53	49.00	59,22	18,27	3,17
HEAN	35,97	143.17	66,56	13.13	7.87
STU DEV	63,28	53,79	114.11	11.73	· . ? '
SKEWESS	2,57	1.51	2.26	0.99	. 73
KURTOSIS	5,28	2,12	3,06	G . 21	- 31
C95 LMT	23.16	19,26	40.85	4,23	A.37
C95 (%)	64,39	13.45	61,38	32.96	r', 91

 $C: \mathcal{F} \to \mathbb{T}^* \cup \mathbb{T}^*$  we set us to the form

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GP	CL	NO3	2月4	C۸	
2.	17.00	7, 40	3,13	2.2 . 8 .	2,27
4.	22.00	17.30	15,97	22.81	17.7 -
7.	13,50	5,92	1.1.7	26.6	1,67
1 7 .	12,50	17. 423	1.44	22.33	1.07
95.	36,00	m1.02	1,21	29.00	25. **
32.	16,53	20.50	1.97	23.25	25
33,	18.00	2,52	1.21	24,83	2.33
35.	24.50	5,32	0.73	31.01	3.27
39.	13,00	5 🖕 8 🖓	1,82	24,80	1.47
44.	9.50	7,72	1,15	23.62	1 . 47
45.	31.90	*1.02°	0.92	27.5.	4,65
47.	14,50	5.40	2,98	24.83	1.63
101.	13.50	3,60	1.0.4	27.57	1,72
1.25.	18,00	12.5/	0,95	26,4	1 2
97.	17.29	28,37	2.13	26.4.	2,12
5B *	11.30	2,40	4.62	35, 8 1	1.42
69 e	11,50	7 🔒 8 🔅	1.12	23.60	2,27
70.	19,20	3.3.	1.02	25,63	2.5
72.	19.50	12,22	2,31	35 ^	1.57
75.	15.50	5,30	1,94	27.2-	1.37
86.	22.00	4.80	1.010	17.63	2.52
92.	15.50	11.40	1,1-	21,62	2.03
42.	28,5×	10.50	3,52	26.77	1.82
MILLIUM	9,50	2,40	0.21	17.03	1.37
MAXINUM	36,40	28,00	4.62	35.87	25.83
RAFGE	26.50	25.60	4,41	18.24	21, 43
1.4 RANGE	8,50	7,20	4,47	4.40	2.25
MEAN	18.22	9,32	1.40	25.63	3. 55
STD DEV	6,63	6,62	1,05	3.79	5.22
SKEVIESS	1,12	1,28	1.78	8.47	32
KURPOSTS	0,60	8,97	2.27	0 <b>.</b> 81	7,55
C95 LMT	2,37	2.48	ð.38	1.36	2. 9
(95 (%)	13.24	26.63	26.97	5.23	54.12

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GP	К	ZH
2.	3.86	0.01
4	3 47	0.02
7	1.44	1. 40
19.	1.7.0	3.10
35	6.55	0.57
32.	2.27	12
33.	1.93	8.84
35.	2.78	0.28
30	1.8-3	0.17
44.	1.71	8,14
45.	28.20	0.00
47.	24.30	1.22
101.	5.00	0.21
j - "5 "	3,54	2.12
97.	3,15	6.05
ы8 <b>.</b>	2.88	4.20
69.	2,52	0.01
7 * .	2.27	Ø . 2.1
72.	6.00	8. 68
75.	1.65	0.04
86.	1,90	0.14
92.	3. 24	Ø, V1
42.	2,84	0.21
ититици	1.44	0.00
махтыам	28.20	0.57
RANGE	26.76	ð <b>,</b> 57
I.O RANGE	2,29	8,14
NEAD	5,36	9.89
STD DEV	6,91	0,13
SKENNESS	2,65	2,12
KURTOSIS	5,50	4,56
C95 LMT	2,47	0,25
(8)	48,87	52.73

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# GULLY POT QUALLTY VARIATION: 23RD FEBRUARY 1978

GÐ	SS	DS	COD	BOD	TOC
6.4	49,00	1065.00	53,90	6.82	0.00
9	85,50	2488.63	757.00	11,50	6.23
2.9 .	40,00	493,00	29,50	4.50	4.5%
25.	257,00	275.00	611.00	20.03	41.000
28.	35.80	808.00	57,00	35,00	0.0.
29.	43.5.1	726.00	41.00	3.51	6.5.
30.	29.52	618,00	32.00	4.00	2.23
32.	19,50	17.00	29,00	6 . 1.3	3,50
33.	88.50	645.03	76.00	3.50	13.00
34.	33,50	673.00	20.00	2.00	13 136
38.	51,03	756.00	45.02	1.50	2.40
56.	8,20	70.00	9,20	0.00	8 6.0,
58.	66.50	47.00	38.02	0.00	11.53
63.	155,00	31,88	112,00	6. 23	35.02
70,	196,50	522.60	935 PM	135.00	135.00
81.	35,50	172.00	29 814	0.00	Ø.5.
86.	63,00	75.00	48.00	17.517	14.85
89.	243.00	218.00	51.00	0.50	14
91.	61,00	1519.00	67.02	3.80	2
92.	20.50	4042.00	13.20	0.53	2.32
96.	155,00	35.00	121.00	6.00	18.00
104.	30.00	209,00	154.20	42.00	5.94
мінімим	8.30	17.00	9.20	0.00	8.82
MAXIMUM	257,00	4042.00	935.00	135.00	135.00
RANGE	249.00	4025.00	925.83	135.00	135.64
I,Q RANGE	72.50	695.00	85.20	7.50	14.7
MEAN	80.23	794,73	151.31	13.30	13.93
STO DEV	73,05	945,49	258,36	29.35	29.27
SKEWNESS	1,31	2,32	2,15	3,42	3.37
KURTOSIS	0,36	5,19	3,01	11,15	11,20
С95 ГЛТ	26.74	346,11	94.47	10.74	12.72
(95 (%)	33 33	49 11	62 43	00 70	76 07

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GP	103	NH4	CL	۳A	CA
6.	8,80	0.77	2603,03	1135,00	123.2.
9,	33,60	6.54	6270.20	3270,00	160.00
2.0%	16,60	0,31	1253,00	396 , 15	01.31
25.	15,60	1,90	680,00	220,00	38.43
28,	24,32	1.70	1700,00	535,00	183.1.3
29.	23,10	3,45	1733.00	635,00	112.62
30.	18,10	0.51	1320,00	562,20	93. 2.
32.	3.70	0.00	130,00	37.00	24. 113
33,	17.73	0.28	1650,00	730.00	\$7 . 4.1
34.	18,90	3,85	1532.20	736.02	80.00
38.	27.00	0.28	1650.00	785.00.	77.03
56.	5,62	0.76	80.03	27.00	22.00
58.	7,80	0.13	53,00	27.38	16.0
63.	9,21	0.07	52,00	22.00	6.5.
70.	40.51	3,44	1229,00	610,00	62,000
81.	12.33	8,09	463,23	277, 10	53,
86.	17.40	0.86	275.0.1	76.20	2.2 . 2.0
89.	36,33	0.28	84 12 30	5270.00	216.23
91.	36.83	2.40	3400.20	2320.03	133.00
92.	13.83	0.03	450.03	172.00	64.02
96.	9,30	0.00	140.32	46.00	24.2 .
104.	18,20	0,69	420.00	89.03	48.53
MINIMUM	3.70	3.00	52.00	22.30	6.53
MAXINUM	40.56	6.50	8400.00	5200.00	216,10
RANGE	36.80	6,50	8348 04	5178,22	2: 9.5
I.Q RANGE	15,32	0.76 -	1459.20	676.02	20.83
MEAN	18,79	0,85	1598,18	912.41	78,97
STD DEV	12,59	1.40	2049.67	1259.22	56,91
SKEWNESS	0.62	3,14	2.11	2,35	1.82
KURTOSIS	-ð,68	9,85	3,91	4.86	arst, 22
C95 LMT	3,88	0.51	758.31	460,96	24.84
C95 (%)	28.64	60.24	46.95	56.74	26.43

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GP	К	PH
· .	173 E 75	
0.	10,000 10,00	n , 1.5 7 0/
2.	10#48	1,80
2 VI	0 <b>0</b> 0 1	U . V B
20.	1.1.0	8.18
63.	14,00	8,28
29.	14,50	7.94
300	11,43	7,86
32.	2.50	/ . / 8
33.	9.39	8,07
34,	9.00	1.94
38.	13,60	7,86
56.	4,50	8.16
58.	4,20	7.87
63.	2.10	7,39
70.	6.00	7.68
81.	6,40	8.10
86.	5,20	7,56
89.	13,62	7,91
91.	17.60	7,96
92.	12.60	7,84
96.	3,30	7,86
104.	6,80	8.18
MINIMUM	2.10	7.39
MAXIMUM	17.60	8.28
RANGE	15.50	0.89
I.O RANGE	8.62	0.26
MEAV	8,99	7.93
STD DEV	4,55	0.21
SKEWNESS	9,16	-0.63
KURTOSIS	-1,23	0.16
С95 LMT	1.66	0,08
C95 (%)	18,52	0.99

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GP	SS	DS	вор	000	03
8.	3.10	194,00	5.00	4.1.24	6.1.
11.	3.4 *	164	11	25. 4 1	6.9
12.	1.6~	149,00	2 . 1900	32.00	<i>i</i> 5, 1
2.3 ,	1,40	178.00	2, 20	19.6 .	1 .
24.	3.10	122,03	2, ) )	33. 3 *	4
28.	1.30	252, 17	1.1	28,93	1 4 27
32,	1.73	68,133	4,00	22.7	4.3.
34.	3.00	148.00	3,03	43.31	5.4
37.	1 . 7 0	128,03	1,25	17,50	5,23
3 R .	3,53	1133 . 51	1.75	29,91	0 E
30 .	1,43	1360,00	1.25	36.1.2	9.2
44.	13,80	404,00	6.00	68, 1111	R 2 4
51.	1.67	228 * 1	3.5 1	35,1*	4 1
55.	5,40	1.724, 3.*	6, 31	128.94	6.2
6S.	4.4.	124.00	4 . 5 1	52.5	7.1
7.5.	1,91	112.1-1	2,25	25.65	5. 2. 1
87,	7.4.1	128.03	8,3	62.41	9.7
95.	3,00	124, 23	* 5	5.1.9.1	1 . a .
93.	1.28	120.00	4.54	22.2	6. 2
99 <b>.</b>	1, <b>6</b> 78	296.02	21.30	34 94	4.7.
125.	2,43	168,03	1.22	37.1.2	7.2.
мтатини	1.20	68.00	8.54	17.53	4.71
A /X1 K.1.X	13.80	1369,00	21, 2.3	128,01	4. 7 3
RANGE	12,63	1292.00	20.50	111.4	5.7 -
1.0 RANGE	1,90	161.20	4.5.	23,12	3.3 '
Y E A N	3,23	311,43	4.52	10.52	6.35
SID DEV	2,87	364,95	4.83	24.07	1.54
SKERNESS	2.56	1.94	2,17	2.20	3.30
KURTOSIS	6.45	2,16	4,31	5,49	-1,53
С95 ДЯТ	1,08	137.38	1.84	9,29	e.62
C95 (%)	33.45	44.11	40.63	22.92	9.72

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GP	*114	СЬ	ΡЦ	SA	Cis
R.	14.56	145,33	7,12	37,11	13, ••
11.	0.57	39.03	7.39	13,8 1	15. 23
12.	5,33	24,03	7.1"	5.9	11.
23.	0.30	28,03	7.32	9.6	ណ្ឌ ស
24.	10.20	14,111	7.27	6,23	
28.	0.35	125.01	7.46	04.51	15
32.	(1,39	23,00	7.12	11.12	ſ)
34.	1.01	14,00	7.22	17.21	12
27.	3.38	14.51	7.15	5.4 1	1
32.	0,27	945.01	7.89	68,8 -	50.00
39.	1,16	1350.00	7,45	400000	it " 5
14.	0.53	2-15,00	7.76	83. "	74.
51.	0,52	115,04	7.34	tī , , , , ,	0.*
55.	2.42	13,3, 33	7,54	314.00	22.
58.	1.37	23.33	7.19	6.3	· · · ·
7 🖓 🖌	, 17	19,00	7.12	7 . 3	0 <u>,</u> (,
P7.	17.46	25,00	7.36	1:1,7	15 5
25.	18,65	25.01	7.34	8.8.	\$ 3 . 5
98.	0,26	75,00	7.48	32.7	Q + +
99 <b>.</b>	0.24	142.83	7.74	50.30	22.5
105.	0.37	57,20	7.55	29.73	13. : .
<b>41519.1</b> 5	2.24	14.00	7.02	5.4-	÷
* AXINUM	2.42	1352,03	7.39	4.24.00	56. 14
RANGE	2.18	1336.21	3.87	394.63	43. 19
1.0 RANGE	12 . 30	121,20	0.44	41,91	11.3.
HEA I	0.61	225.31	7,37	59.17	16.10
STO DEV	0.49	416,39	8,27	126,73	:1.92
SKFAJESS	2,55	2.03	2,53	2,52	2.17
KURTOSIS	6,51	2.36	-1,99	4.75	3,93
C95 L 1T	9.18	156,74	0.12	4.4.18	4,40
(95 (8)	31 24	69 57	1 37	57 91	26 8 1

CO PI (HED)

(1)	К
8.	1 • 4 *
11.	1 • 4 *
12.	6 • 9 *)
23.	1 • 2 */
24	2,53
32	1,45
34	1,95
37	1,15
38, 39, 14, 51, 65	8,13 10,23 4,92 1,53
55.	2.95
70.	1.79
97.	2.97
95.	1.45
99. 135. ЭТКІМПМ	2,2% 3,35 1,85 Ø,90
RAXIMUM	19,23
RADGE	9,30
1.0 RANGE	1,33
MEAN	2,60
STD DEV	2,38
SAE NESS	2,19
КЧКТОБІБ	3,70
С95 БМТ	0,90
С95 (%)	34,53

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### GULLY POT QUALITY VARIATION: 15TH MAY 1978

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GP	SS	DS	BOD	COD	6 Q 3
2.	30,30	268,00	8,60	32.00	7.80
7.	20,00	352.00	3.43	48.00	15.00
9	42.46	148.00	9.03	62.00	6.6.4
39.	10.00	76.110	0.43	11.0.3	3 81
33.	22.00	204.00	0.93	38.00	10.23
34.	16.08	276,00	5.60	33.00	R.47
32.	12.20	212.13	25.64	315.00	19.00
29.	3.00	88.10	1.5.3	46.20	8.62
25.	20.00	164 00	7.30	25.00	10.7.
56.	11.00	176.80	0.00	9.00	6.40
45.	410 . 119	1116.60	23.73	24.00	9.30
63.	25,00	192.80	8.82	58.00	16.00
69	16.00	264.40	0.60	19.00	15.3
95.	49.05	224.00	8.40	64.00	14.4%
97.	6 . 15	336,00	0.40	55.00	21.6%
72.	5.02	96,00	0.90	84.05	7.1.)
101.	19.00	140.00	0.00	5.80	8.4 1
81.	1.60	48.00	0.00	0.20	5.8.
89.	14.00	268.38	2.43	42.00	13.4
92	249,119	160,00	72.60	148.20	7.20
MINIMUM	1.00	48,00	0.00	0.00	3.8/*
MAXIMUM	249.00	1116.00	72.60	315,00	21.80
RANGE	248.90	1068.00	72.60	315.40	18.00
1.0 RANGE	20.00	128,00	8,40	43.09	7.93
MEAN	30.50	239.40	8,90	55.50	10.76
STD DEV	53,05	222.56	16.71	68.88	4 48
SKEWNESS	3.64	3,13	. 2.93	2.81	6.72
KURTOSIS	12.16	9,76	8,23	7,79	-Y,50
C95 LAT	20.51	86.02	6,46	26,63	1.85
C95 (%)	67.25	35.93	72.55	47.98	17.26

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table with table with which with with strip

GP	1114	CL	РH	CA	s. A
2.	. 3°G9	73.00	7,26	28.00	27,00
7.	1.70	115.00	6,44	33.00	14 . 15
9.	1.60	54,000	6.70	16.00	11
39.	1.30	18,60	6,86	17.03	3.00
33.	1.7:3	49.14	6.91	30,00	4 . 35
34.	1.20	115,00	7.10	33,00	33.03
32.	6.70	90.00	6.70	24.00	11,00
29.	0,83	24.00	6.88	26.140	5.85
25.	1.20	38.60	6.82	22.103	4,00
56.	0.20	48,00	7,16	21 .00	22.5.1
45.	7.60	1150.00	6.81	34 . 190	, 600.00
63.	2.40	110.00	6,42	20.00	6
69.	1,3.	63.65	6.74	25.00	7.0
95.	3,40	67.00	6.73	23,00	6 . VC
97.	1.60	95.00	10.72	45.00	38,00
72.	0.22	23.12	7.38	19.30	4 51 1
161.	0.83	33,00	9,78	23.00	17.00
81.	0.18	11,50	7,56	13.03	4
89.	1.20	125,20	7,40	31,00	27.04
92.	0.11	79.00	7.18	17.20	6.83
NININIM	0.11	11.50	6.42	13.00	3.00
MAXIMUM	7.60	1150.00	10.72	45.00	690.30
RANGE	7.19	1138.50	4.30	32.00	597.00
1.Q RANGE	1.57	77.00	0.68	12.00	18.00
MEAN	1.91	119.02	7.28	24.95	42.45
STD DEV	2.00	245,19	1.27	7.74	131.67
SKEWNESS	1,80	3,89	2.31	8.72	3,98
KURTOSIS	2,34	13,46	4.20	0,18	13,92
C95 LMT	0.77	94,79	0.42	2,90	50,91
(95 (%)	40.37	79.64	5.70	11.99	119.92

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GP	ĸ	PB	ZU	CD	85 (S)
2.	3.00	0,00	0.49	61 . (11	r.74
7.	1,40	0,18	3,50	8.114	i, 39
9.	2. 3.3	0.00	1.30	6.12	0.63
39.	1.40	0.40	2.00	6.01	0.28
33.	1,90	10.45	3,40	S . 47	8.39
34.	2,30	0.05	2.40	6.02	.0.12
32.	4.03	M. DE	1.80	0.01	4.11
29.	3,40	0.03	0.80	12 101	0.05
25.	3,90	2,00	2,40	6.11	0,34
56.	8,90	PLUM	1.41	10.100	11 4
45.	4,50	0,01	0.80	0.01	11.42
63.	2.70	0.91	2.70	10.01	3,17
69.	2.30	13 <b>.</b> 1914	3,20	9.01	× * * 7
95.	3,10	-1.00	-1.27	· 1.03	m1 . 110
97.	4,00	0.00	1.20	0.01	5. 24
72.	3.40	10.64	1.80	£.6°	· • 1
171.	4.78	0.60	0.50	0.05	12. 8.
81.	1.70	63 . 1915	2,20	6.06	0.01
89.	7,90	0.09	0.19	R. Bu	2,31
92.	2.40	0.02	2.00	6.00	0.19
MINIMUM	1.40	0.00	0.09	0.00	0.00
MAXIMUM	8,90	0.45	3.50	8.07	0.74
RANGE	7.50	0.45	3.41	0.07	3.74
1.Q RANGE	1,70	0.02	1.70	0.01	0.35
MEAN	3,46	0.04	1,76	0.21	2.21
STD DEV	1,96	0,11	. 1.04	0.02	0.22
SKEWNESS	1.53	3,02	0.06	2,26	0,99
KURTOSIS	1,76	8,24	-1.07	4.78	• 2 . 1 1
C95 LMT	0,76	0.04	0.41	0.01	0.09
C95 (%)	21.87	96.88	23.37	53.21	41.44

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GP	ΝI	CR	FE	CII
2.	0,05	0,00	1.10	6.03
7 e	12.32	2.48	7,00	6.32
9.	0.33	0,00	1. 4.	P. E 4
39.	0.02	9.130	0,40	0.05
33.	C. 42	0.33	4.00	6.32
34.	0.03	0.00	0.60	0.03
32.	0.73	0.96	0.40	0.05
29.	0.20	0.20	9.30	6.04
25.	0.03	0.00	0,90	0.02
56.	0.01	0.00	3,20	0.03
45.	0.01	0.00	1.20	0.01
63.	0.03	0,00	0,90	2.03
69.	8.11	0.00	3.40	2.62
95.	-1.00	• 1 , Cris	·· 1 . C	•1 .C.C
97.	0.02	0.02	Ø <b>.5</b> :	14,03
72.	0.01	0.08	0.10	0.01
101.	0.02	0.01	0.20	0,06
81.	0.01	0,01	9,1e	8.92
89.	0.01	6.02	0.20	0.00
92.	0.02	0.00	Ø.70	0.03
MINIMUM	0.01	0.00	0.10	0.00
MAXIMUM	0.42	0,48	7.00	8.32
RANGE	0.41	0.48	6,92	6,32
I.O RANGE	0.02	0.01	8,98	0.03
MEAN	0.07	0,05	1.08	0.06
STD DEV	0.12	0.13	1.68	0.09
SKEWNESS	2,16	2.67	2,66	2.39
KURTOSIS	3.14	5.40	6,12	3.88
C95 LMT	0.05	0.05	9.67	0.04
C95 (%)	68.24	113.38	61.40	61.35

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## GULLY POT QUALITY VARIATION: 177H JULY 1978

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GP	នន	DS	COD	вор	1103
9.	20,00	176,03	78,90	2.60	7.33
10.	12,00	18:1. 00	26,30	0,00	5.6"
11.	63,00	136.00	31.4.3	0.0.5	7.7.4
94.	27, 34	196,93	59,21	3.40	5,3.
410 .	22,00	184.00	68,53	2.44	6.62
5,	24.33	132.00	87.30	5 . R.	ñ , 2 3
83.	13,36	128.00	39,44	1.91	6. 4.
R1.	24.20	104,00	25,45	0,91	4.1
78.	29,00	140.00	25.42	1,918	5,3%
76.	21,00	152,00	69,59	2.40	7 . 3
74.	16. 34	128,60	6,63	1,93	5,10
73.	25.89	152,00	16,2%	0.84	5.41
79.	21,00	96.03	51.60	6,93	5,6%
71.	23.24	272,03	115.50	8.67	2.01
99.	15.00	128,00	24.40	6.2 4	5
97.	26,83	168.30	164.30	2.83	9.1.
10.	18.10	108,40	38,50	5,23	÷ . 9
51.	53,00	152,00	53,27	5,82	7 . 1.
54,	29.30	3-10,13	99,57	26.2	15. 6.
25.	32.00	188.00	59,20	6,50	7,12
MININUM	12.00	96.00	6.69	0.00	4.13
махтици	63,30	300.20	164.34	26.24	1000
RANGE	51.00	204.00	157,70	26,2%	6.5
1.9 RANGE	11.00	76.00	53,52	4.60	2.7.
MEAN	25,65	159.00	57.47	3,57	6.97
STD DEV	12,40	53.22	38,12	5.71	1.77
SKEWNESS	1.82	1.21	1.12	3.22	2,39
KURTOSIS	2,78	6,99	8,99	10.07	- 2, 91
С95 LMT	4.79	20,58	14.74	2,21	8,69
C95 (%)	18.69	12.94	25.65	61.80	9.95

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GP	NH4	C1,	CA	NA	К
9.	8,10	12,50	17,60	3,50	2.30
100	3.63	17.50	18.19	9,81	2.1
11.	1.42	14.00	15,50	5.00	1,21
94.	8.10	16.00	10.60	3,00	3.5.
412	2,90	31,00	21.20	8.73	1.4.
5.	1.5%	8.70	13.80	1.62	
83,	1.60	113,80	16.50	3.40	1.5
81.	3,60	11,90	16.50	2.80	2.2.
78.	3.30	9,60	19.70	2.10	1.7
76.	0.40	14.60	21.7	3.645	3.1.
74.	0.62	11.50	18,10	3.94	3.3 -
73.	2,60	12.60	21,30	4.10	3.4
76.	1.50	13,00	15.53	3.43	2.3
71.	9.70	24 . 13	22.82	6.84	11.5
99,	1,80	13,59	16.57	4. 10	2.1 2
97.	1.40	15.60	21.24	0,6:	4.7.
16.	2.40	8,20	14.40	2.20	2.13
51.	6.42	46.00	17,70	11.70	2 8.
54,	11.50	118.00	23.60	57.00	4.00
25.	1,40	13,00	18,20	9,80	2.6 1
MININUM	0.40	8,20	10.60	1,60	A _ R.3
MAXIMUM	11,50	118.00	23.68	57.00	11.50
RANGE	11,19	109.8:	13.00	55,42	1. 7.3
I,Q RANGE	5,00	6.00	5.70	5.73	1,70
MEAU	3,69	21.44	18,03	7,48	2,93
STD DEV	3,25	24.46	3.30	12.02	2.24
SKEWMESS	1.14	3,30	-0.20	3.65	2.86
KURTOSIS	-2 e7	10.08	-0,51	12,2%	8,38
С95 БМТ	1,26	9,45	1,28	4.65	a.87
(95 (%)	34.03	14 94	7.08	62.15	29.56

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#### MISCELLAREOUS GULLY LIQUOR ANALYSES na the east we want the use of the set of the want we want the set of the set of the set of the set of the set want set may DATE 3RD MAY 1976 100 WO KO CH 100 CH 100 PP AV SC CS CB 100 PD 100 CS GP 7 GP 2 GP 3 GP 4 GP 5 NEAR 454.8 42.8 226,0 221.0 151.0 SS 218,4 2,35 3,59 \$.37 0,46 NH4 0.47 0.54 8,333 0.029 1.342 8,379 2.148 Z .... T 0,496 0,039 ₫,035 0.937 0.048 0,036 ZN-S 225,2 135,6 160,0 DS 120,0 185,0 135,0 N03 5,4 5.6 6,2 10.25 5.3 6.55 DATE 10TH MAY 1976 GP 5 GP 4 MEAN GP 1 GP 2 GP 3 29,2 82,4 21.6 49.7 71.6 43.6 SS 1,49 2,4 2,97 2,1 0,152 0,18 0,177 0,133 0,463 0,315 0,485 0,054 2.4 2.04 1.6 NH4 0,074 ZN-T 0,08 ZNOS en, 275.0 DS 385.0. 410.0 336.0 285.0 325.0 DATE 17TH MAY 1976 GP 2 GP 3 GP 4 GP 5 MEAN GP 1 6,0 29.6 14.8 18.4 12.4 96.4 SS 0.18 0.48 0,32 0.23 2.32 NH4 2.31 0.365 0.347 0.254 0.37 0.733 0.031 0.246 0.08 0.474 ZN-T 0.44 0,376 0.217 3,071 ZN-S 340,0 164.0 336.0 195.0 201.3 DS 185,2 1.9 1,85 2,73 2.82 2,95 2,27 NO3 DATE 12TH APRIL 1976 GP 1 GP 2 GP 3 MEAN 121.9 88.2 SS · 32.8 21.4 1.9 2.42 1.3 0.41 NH4 0.195 0.05 PB-T 0.11 0,085 0,035 PB-S 0,035 0.04 0,03 0.35 0.25 0,18 0,22 ZN-T ZN-S 0,064 0,18 0.122 -

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DATE 26TH APRIL 1976

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	GP 1	GP 2	GD 3	GP 4	GP 5	MEAL
55	20,0	4.8	38,8	23.6	20.4	21.5
HH4	0,03	0.76	0.36	0.79	3,78	1.14
7.1 - T	0.481	9,364	0,352	0.37	2,98	0,509
ZN-S	0.056	6.075	0,096	0.065	0,284	1 0,115
DS	525.0	310.0	500 .	485.0	695,0	503.0
N03	6.4	4.3	6.4	8,2	6.0	6.3

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GULLY	LIQUOR A	ALYSES	PRIOR TO	STORM R	UNOFF	
EVENT		ir 44 48 49 19 19 59	909 100 100 100 100 100 100 100	413 KCA 428 CBH MBT 99A 154 456		
99 89 69 69 99 vil	<del></del>					
DATE	20TH APRIL	1978				
	971: ent 87+ 93# 13# 53# 43# ent 98# 44	ar mit sin hut eye				
	GP 8	GP 7	GP 43	GP 44	GP 11	WEAH
NO3	10,6	4.8	4.3	5.4	6.0	6.2
NH4	1.12	11,13	0,23	3,21	0,23	4.18
COD	34.0	51.1	57,9	76.0	78.3	59,5
800	2.5	1.8	2.5	3.4	3.6	2.8
ι	23.0	1240	0 * 2	(*)	5.5	11.5
DATE	25TH APRI	1978				
Ar 21 10 40 40	90 811 90 271 89 28 90 56 56 56 56					
	GP 8	GP 7	GP 43	GP 44	GP 11	NEAN
NO3	10.6	9.3	8.9	7.1	6.1	8.4
NH4	0,33	0,33	0,18	0,25	0.24	0.23
COD	40.9	15.7	67,9	83.0	63,5	54.2
ns -	120.0	225.9	135,0	125,0	265,2	174.0
CA	17,0	16,5	25,6	14,5	12.8	17.3

DATE 23RD MAY 1978

	GP 7	GP 43	GP 89	GP 86	GP 84	MEAN
NO3	6.6	8.4	23.3	12.1	13.5	12.8
NH4	2,5	1.4	2.1	1.2	0.63	1.6
COD	4.9	0.0	34.0	23.9	44.1	21.4
BOD	7.7	1.2	8.3	4.3	1.5	4.6
DS	428.0	248.0	484 0	168.0	220.0	310.0
CA	72.8	24,9	29.1	13.8	47.1	37.1

	ED No en			
DA'CE 31	ST' № АХ 1978 он юг нэ тос нэ са на на на			
	GP 7 GP 43	GP 84	GP 86 GI	P 89 MEAN
NO3 NH4 COD BOD DS CA	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	13,8 2,1 123,3 0,4 252,0 35,0	11.3 3.3 104.7 2.1 188.0 29.0 2	1.3 1.8 8.8 4.4 9.0 116.8 5.3 4.8 2.0 2.0 2.0 2.0 2.0 32.4
DATE 12	TH JUNE 1978			
	GP 7 GP 43	GP 44	GP 52 G	P 53 JEAN
NO3 VH4 COD POD DS CA	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	7.1 1.7 22,6 0.9 84.0 9.4	7.0 1.8 37.7 12 5.2 1 104.0 22 16.0 2	$\begin{array}{cccccccccccccccccccccccccccccccccccc$
NETEORO	DLOGICAL DATA	FOR MISCELI	ANEOUS GP	ANALYSES THAN THE
12 A 76	K (PM)	ADP	ADP <sub>8</sub> Z	105 85
26,4,76		12	12	9,0 5,0
10,5,76	6.4	0,5	0.5	17.0 12.5
17.5.76	6,2 5 7,2	1 Ø.5	1 Ø.5	19,5 5.°° 23,0 17,°°
FOR GP	ANALYSES DURI	NG STORM RU	NOFF ANALY	SIS
29.4.78	16.2	3	Ø	12.7 7.5
23.5.78	9.5	1	1	15.0 8.3
31.5.78 12.6.78	3 1 <b>.</b> 2 3 4 <b>.</b> 8	6 4	6 4	14.6 7.6 12.0 8.5
FOR GP	QUALITY VARIA	TION ANALYS	ES	
23,9,77	0,0	13	13	14,6 10,5
23.2.78	5.6	Ø	0	10.4 4.9
15.5.78	3 <u>14</u> ,5	8 9	Ø	14.5 7.5
1 1 19 19 19 1	2 11 9	0	0	22.9 11.2

### HEAVY METAL ANALYSES - ZINC

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DATE	METHOD	SAMPLE	TOTAL	SOLUBLE
12.4.76	AAS	GP 1	2.183	0.064
12.4.76	\$1	GP 2	0.220	-
12.4.76	11	GP 3	a. 35A	0.180
12,4.76	<b>8</b> 8	GP 3	0.810	0.100
12.4.76	47	GP 3	1.74	3.56
5.4.76	19	GP 1	0.389	2.140
26.4.76	ti	GP 1	0.481	9.056
26.4.76	11	GP 2	0.364	3.075
26.4.76	8 <b>9</b>	GP 3	0.352	0,096
26.4.76	11	GP 4	0.370	8.065
26.4.75	11	GP 5	0.98.	a.284
3,5,76	11	GP 1	0.496	0.848
3.5.76	Ħ	GP 2	0.333	0, 235
3,5,76	11	GP 3	0.379	0,236
3.5.76	18	GP 4	0.354	0.829
3.5.76	12	GP 5	0.148	0,939
10.5.76	Ħ	GP 1	0.074	-
19.5.76	n	GP 2	0.386	
10,5,76	u	GP 3	0,152	0,063
10.5.76	ţ1	GP 4	0.180	0,015
10.5.76	11	GP 5	0.177	0.085
17.5.76	11	GP 1	0.440	0.217
17.5.76	F1	GP 2	9,474	0.071
17.5.76	11	GP 3	0.365	e. 033
17,5.76	11	GP 4	0.347	0,031
17.5.76	11	GP 5	0,254	0.046
13,7,76	ASV	GP 7	0.163	0.052
16.8.76	78	GP 6	0,283	1.205
16.8.76	If	GP 1	0,200	2.135
23,8,76	u.	GP 6		0.150
6,9,76	Ħ	GP 1	0.008	3.010
27.9.76	19	GP 6	0.022	0.112
мінімим			0,008	0.010
MAXIMUM			1.74	0,56
RANGE			1.732	0,550
I.Q.RANGE			0,301	9.115
MEAN			0.360	0.104
STD DEV.			0.328	8.110

## HEAVY METAL ANALYSES on LEAD

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DATE	MELHOD	SAMPLE	TOTAL	SOLUBLE
12,4.76	ANS	GP 1	1. 11	(A, 035
12.4.76	83	GP 2	0,095	1.940
12.4.76	<b>1</b> *	GP 3	0.050	0.030
12.1.76	11	GP 3	0.150	0,035
12.4.76	49	GP 3	1,360	9,103
5.4.76	H	GP 1	0.151	0.035
13.7.76	ASV	GP 7	0.362	8,912
16.8.76	67	GP 6	0.079	3.013
16,8,76	11	GP 1	**	0.012
23.8.70	14	GP 6	0.151	**
6,9,76	tt	GP 1	0.224	0,304
27,9,75	ц	GP 6	0,015	0,927
MINIMUM			0,015	8.004
MAXIMUM			1.36	0.100
RANGE			1,345	1, 196
1.0.RAJGE			2.120	0.022
MEAN			8.093	0,032
STD.DEV.			0.048	2. 225

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## HEAVY METAL ANALYSES - COPPER

DATE	METHOD	SAMPLE	TOTAL	SOLUBLE
26.4.76	AAS	GP 1	0,056	0,068
26.4.76	19	GP 2	0.056	0,259
26.4.76	41	GP 3	Ø,056	3.916
26.4.76	69	GP 4	0.058	3.488
26.4.76	#1	GP 5	0,073	8,285
13.7.76	ASV	GP 7	0.011	2.003
16.8.76	11	GP 6	0.080	0.018
16,8,76	11	GP 1	0,032	-
MININUM			0.011	8.003
MAXINUM			0,080	0,088
RANGE			0.069	3,285
MEAN			0,052	8,848

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### HEAVY METAL ANALYSES - CADMIUM

DATE	METHOD	SAMPLE	TOTAL	SOLUBLE
13.7,76	ASV	GP 7	0.010	0,996
16.8.76	11	GP 6	0,013	0,025
16.8.76	88 	GP 1	0.040	8.840
23.8.76	11	GP 6	0.042	3,225
6.9.76	11	GP 1	0.004	3.001
27,9,76	н	GP 6	0,006	0,003
MINIMUM			0.004	0.001
MAXIMUM			0,042	0,040
RANGE			0,038	0,039
MFAN			0.019	0.017

## DUSTFALL MEASUREMENTS

DATE	DURATION	MASS(MG)	MASS PER	WEEK
7-14/4/77	1	9.9	9.9	
7-21/4/77	2	26.0	13.0	
7-28/4/77	3	33,3	11.1	
18-25/4/7	7 1	8 8	8.8	
18-2/5/77	2	28.6	14.3	
9#23/5/78	2	27.1	13.55	
12=19/6/7	8 1	9.2	9.2	
26-3/7/78	1.	7.3	7.3	
6-20/7/78	2	15,8	7 🖁 8	
MEAN MASS	COLLECTED	PER WEEK =	10.55 MG	

DUSTFALL ACCUMULATION RATE = 46.5 MG/SQ.M/DAY DUSTFALL ACCUMULATION RATE = 169.6 KG/HA/ANNUM LOADING PER ANNUM FOR CLIFTON GROVE = 610 KG (IMPERVIOUS AREA ONLY) 

### 2. LABORATORY STUDIES

LABORATORY STUDY OF THE BEHAVIOUR OF POLLUTION PARAMETERS IN GULLY POT LIQUORS OVER A DRY PERIOD

THE SEDIMENT WAS OBTAINED FROM A COUNCIL GULLY EMPTYING VEHICLE, SOURCE CLIFTON ESTATE,

THE GULLY POT WAS FILLED WITH DEIONIZED WATER, STIRRED AND AERATED TO SIMULATE THE PASSAGE OF A STORM, THE FIRST SAMPLE WAS DRAWN AFTER ONE AND HALF HOURS, THE LIQUOR TURNING ANAEROBIC AFTER SEVEN HOURS,

GULLY POT VOLUME = 93 LITRES

SEDIMENT DEPTH = 30 MM

DAYS	SS	BOD	COD	COD(SOL)
Ø	1297.0	72,2	929.0	-
1	688.0	45.4	494.0	-
2	283.0	26,0	296.0	
3	245.0	22.5	253.9	44
4	223.0	25.0	204.0	73.0
5	191.0	15.5	184.0	75.6
7	128.0	18.5	158.6	53.7
9	106.0	15.0	127.9	61.4
11	110.0	11.0	148.3	98,6
14	62.0	7,5	103.8	40.3
16	13.0	2.0	59.3	16.9
21	18.0	7.0	61.9	19.0
23	16.0	2.0	51.9	39.5
28	3.0	5.0	43.5	42.5
31	13.0	0 8	30.3	30.3
35		1.5	26.7	26.7
38	12.0	4.3	30.0	30,0
ADDITION	OF 5.20	OF CAR WASH	LIQUID	
38	40.0	27.3	93.0	66.0
42	9.0	17.0	81.7	53.3
45	6.0	10.8	65.2	60.7
49			42.6	41.7
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## DRY PERIOD STUDY CONTINUED

DAYS	NO3	NH4	CL	DS	рн
Ø 1 2 3	12.0 11.0 10.7 14.0	2.1 1.7 1.7 1.75	80.0 95.0 110.0 125.0	240,0 265,0 295,0 295,0	7,84 7,86 7,85
7 7 9	12,5 13,5 12,8 15,0	1,45	96,0 115,0 110,0	265,0 345,0 300,0	7.74
11 14 16	14,5 13,5 17,0	1,45 1,78 1,95	155,0 120,0	330,0 380,0 330,0	7.83 8.09
21 23 28	20.0 22.0 22.0	2.4 2.8 3.9	125.0 110.0 125.0	390,0 405,0 400,0 410,0	7,94 7,90 8,36
31 35 38	19.0 12.0	4,5 4,0	175,0	455,0	8,41 8,06
ADDITION	OF 5.2G	OF CAR WA	SH FIGNI	D	
38 41 42 44	16.5 14.5 10.5 8.5	3.7 4.3 4.4	145,0 125,0 120,0	500,0 450,0	7.94 8.21 7.85
45 49	7.8 8.5	4.3 4.0	120.0	480.0 470.0	8,05 8,11
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## COMPOSITION OF GULLY POT SEDIMENT

TWO COMPOSITE SAMPLES WERE COLLECTED FROM COUNCIL GULLY EMPTYING VEHICLES :-A) CLIFTON (RESIDENTIAL)

B) BASFORD (RESIDENTIAL/ LIGHT INDUSTRIAL)

	CLIFTON	BASFORD
BOD (MG/G)	43,8	9.1
COD (MG/G)	34711.0	8462.0
VOLATILES (%)	21.0	7,6
ZINC (MG/G)	0,69	0,22

SOLUBLE COMPONENTS:-

4.3
0,165
0.037
0,75
3.2

4.5 Ø.185 Ø.Ø35 1.25 2.7

VISUAL OBSERVATIONS

MAINLY FINE SAND ,SILT AND GRIT. LEAVES, SMALL TWIGS, SMALL STONES, PAPER, CELLOPHANE WRAPPING PAPER (SWEETS), SILVER PAPER (CIGARETTE PACKETS), CIGARETTE ENDS,BOTTLE TOPS, CAN PULL KEYS.

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LABORATORY SIMULATION OF THE REMOVAL OF PARTICULATE

CUNULATIVE MASS OF SOLIDS REMOVED (MG) FOR VARIOUS FLOW RATES

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47~1 1	1359,6	2442ª	3417.6	4362,6	5664,6	6642.6	7518.	8628	8362.0	8442.6	8442.6	8442 .9
ຍາ ຍັງ	152,9	1647.9	2222,3	2343,5	2910,6	3348 ° C	3998°8	4298.4	4525°2	4622.4	4687.2	4719,6
6 <sup>4</sup> 0	770.4	1569,6	2340.0	3004.8	3998,4	4152,0	5836,8	6576,0	7123.2	7536.0	7891.2	8112.0
5. 5.	245.7	764.4	1077.3	1318,6	1671.6	1995.0	2440.2	2717.4	2877.8	2885.4	2893,8	2992,2
9 6	469,8	831.6	1164.6	1458,0	1986.4	2286.0	2725.0	3106.8	3373.2	3582.0	3895.2	3999.6
£.9	427.6	699.2	895.5	1047.0	1326.0	1551.0	1917.0	2187.0	2385.0	2511.0	2673.0	2799.0
0.4	327.6	741.6	1014.0	1233.6	1596.0	1857.6	2275.2	2635.2	2976.0	3211.2	3466.4	3604.8
0.2	58.8	291.6	428.4	519.2	678,8	802.4	989.6	1164.8	1346.0	1467.0	1570.4-	1673.6
0.1	41.1	72.6	194.7	129.0	184.8	239.4	343.8	465.0	533.4	599.4	655.2	706.8
TIME	0.25	0.5	0.75	0.	1.5	2.0	9.0 M	4	5.0	6.9	7.9	8.0

TOTAL MASS (G) 0.71

8.44

4° ₹

2.9

4.0

2,80

3.60

1.67

OBSERVED CONCENTRATIONS (MG/L) FOR VARIOUS FLOW RATES LABORATORY SIMULATION OF THE REMOVAL OF PARTICULATE MATERIAL FROM GULLY POTS 

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FLOW RATE (L/S)

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0,0

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LABORATORY SIMULATION OF THE REMOVAL OF DISSOLVED MATERIAL FROM GULLY POTS

FLOW RATE = LITRES/SEC

sLOPE = LOG(S/S1)/TIME(SECS)

R.T. = RETENTION TIME CALCULATED FROM SLOPE

%VOL, HIXED = THEORETICAL % COMPLETELY MIXED CALCULATED FROM RETENTION TIME RATIOS

FLOW	SLOPE	R, T, (SECS)	&VOL,MIXED
0,03	=2,45	1289,3	41,7
0.04	-2.12	1116.9	48.2
0,05	-1,75	1085,9	58,5
0.05	-2.44	777.0	41.8
0.06	-1.57	1005.7	64.9
0.07	a1.70	798.1	60.2
0.09	-1,35	777.9	75.4
0.10	*1.05	901.1	97.0
Ø.12	-1,96	743,2	96.1
0.25	ů,894	424.2	106.1
0.59	=1.02	185.3	99.8
1,00	=1,02	92,8	100.0

3. STORM RUNOFF QUALITY DATA

## STORM OF THE 6TH NOVEMBER 1976

TIME SAMPLING COMMENCED 0,00 SAMPLING DURATION(MINS) 65,42 DOSER CONCENTRATION(G/L) 0,00 DOSE RATE(ML/MIN) 0,00

MEASURED PARAMETER CONCENTRATIONS

T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
1,03	0,00	25,00	180,00	20,10
6.09	ØÖØØ	39,30	157,50	20.20
10.59	0,00	59,00	72.50	19.20
15,09	0,00	63,00	80,00	17.00
19.59	0,00	67,5Ø	90,00	16,50
24.09	0.00	76,50	82.50	16.50
28,59	0.00	80.00	75,00	16.20
33.09	0.00	74.40	60.00	16,00
37.59	0.00	52,50	65.00	15.60
42.09	0.00	41.00	62.50	14.80
46.59	0.00	34,20	57.50	14.40
51,09	0.00	34.50	60.00	13.30
55,59	0,00	32,30	70.00	12.80
60.09	0.00	28,50	65,00	12,80
64,59	0,00	31,00	65,00	12,20
T(MINS)	BOD(MG/L;	) CL(MG/L)	NO3(MG/L)	NH4(MG/L)
1,03	0.00	7.50	4.70	0,00
6,09	0,00	22,40	12,20	0,00
10,59	0.00	6,00	4.00	0,00
15,09	0.00	4,70	3,50	0.00
19,59	0,00	4,60	2,90	0,00
24,09	0,00	3,80	2,70	0,00
28,59	0.00	3,60	2,70	0,00
33,09	0,00	3,60	2,60	0,00
37,59	0,00	3,60	2,60	ØoØC
42,09	0,00	3,40	2,70	0,00
46,59	0.00	3,10	2,60	0,00
51,09	0.00	3,00	2,40	0,0A
55,59	0,00	3,00	2,30	0,00
60,09	0,00	2,90	2,20	0 <sub>8</sub> 00
64.59	0.00	3.40	2.20	0.00

STORM OF THE 6TH DECEMBER 1976

TIME SAMPLING COMMENCED 0,00 SAMPLING DURATION(MINS) 42,92 DOSER CONCENTRATION(G/L) 0,00 DOSE RATE(ML/MIN) 0,00

MEASURED PARAMETER CONCENTRATIONS

T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
1,03	0,00	96,20	396,00	37,00
6,09	0,00	196,20	492,00	34,30
10,59	0,00	271,00	452,00	28,50
15,09	0,00	284,00	352,00	24.40
19,59	0,00	270,00	304.00	22.50
24.09	0,00	248.00	332.00	20.20
28,59	0.00	196.00	352.00	19.20
33.09	0.00	164.00	224.00	17.80
37.59	0.00	159.00	204.00	17.80
42,09	0,00	156,00	200,00	18,10
T(MINS)	BOD(MG/L)	CL(MG/L	) NO3(MG/L)	NH4(MG/L)
1,03	0,00	117,00	5.50	Ø.15
6.09	0,00	176.00	6.80	0.15
10.59	0.00	179,00	6.20	0.15
15,09	0,00	142,00	5.40	0.12
19,59	0.00	133.00	5.00	0.16
24,09	0,00	97,00	4.50	0.10
28,59	0,00	103,00	4.10	0.15
33,09	0,00	84,00	3.70	0.11
37,59	0.00	82.00	3.90	0.10
42,09	0,00	75,00	4.10	0.19

## STORM OF THE 17TH DECEMBER 1976

TIME SAMPLING COMMENCED 0,00 SAMPLING DURATION(MINS) 78,92 DOSER CONCENTRATION(G/L) 0,00 DOSE RATE(ML/MIN) 0,00

MEASURED PARAMETER CONCENTRATIONS

T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
1.03	0,00	34,40	275,00	24.70
6.09	0.00	34.80	275.00	24,50
10.59	0.00	39,20	265.00	23,70
15,09	0.00	37.20	262.00	23.20
19.59	0,00	32,00	265.00	23,20
24.09	0.00	31,60	247.00	22.70
28,59	0.00	35.20	257.00	22.40
33.09	0.00	30.40	255,00	22.30
37,59	0.00	33,60	260.00	22,30
42.09	0 00	30,40	255,00	22.30
46,59	0,00	30,00	244,70	22,30
51.09	0,00	24.40	224.00	22,30
55,59	0,00	31.60	260,00	21.70
60.09	0.00	26,00	264,00	21,20
64,59	0.00	24,00	200,00	20.40
69,09	0,00	43,20	232.00	19,90
73,59	0.00	29,60	184.00	19,60
78,09	0,00	21,60	248,00	20,20
T(MINS)	BOD(MG/L)	CL(MG/L	.) NO3(MG/L)	NH4(MG/L
1,03	0,00	90,00	6,80	0.20
6, 09	0.00	90,00	6,50	0.27
10,59	0,00	90,00	6,40	0,28
15,09	0,00	90,00	6,42	0,26
19,59	0.00	90,00	6,30	Ø,25
24,09	0.00	90,00	6,40	0,25
28,59	0,00	90,00	6,60	0.22
33,09	0,00	90,00	6.40	0,25
37,59	0,00	90,00	6,60	0.22
42,09	0,00	78,00	6,00	0,14
46,59	6,00	78,00	6,50	0,25
51,09	0.00	78.00	6,00	P . 27

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# STORM OF THE 17TH DECEMBER 1976 (CONTINUED)

T(MINS)	BOD(MG/L)	CL(MG/L)	NO3(MG/L)	NH4(MG/L)
55,59	0,00	78,00	7,10	0.27
60,09	0,00	78,00	5,90	0,29
64,59	0,00	78,00	5,80	0,29
69,09	0,00	73,00	5.70	0,30
73,59	0,00	73,00	5,80	0,29
78.09	Ø,ØØ .	73,00	5,60	0,29

## STORM OF THE 9TH JUNE 1977

TIME SAMPLING COMMENCED Ø,00 SAMPLING DURATION(MINS) 195,92 DOSER CONCENTRATION(G/L) 48,00 DOSE RATE(ML/MIN) 1,04

## MEASURED PARAMETER CONCENTRATIONS

T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
1,03	8,00	5,20	0,00	0,00
6,09	9,97	15,20	0.00	0,00
10,59	7.23	16,40	0,00	0,00
15,09	8,05	19.00	0.00	0.00
19,59	8,91	22.00	0,00	0.40
24.09	10.74	34,00	0.00	0.00
28,59	14.47	42.00	0.00	0.00
33,09	30.25	58.00	0.00	0.00
37,59	35,66	66.00	0.00	0.99
42.09	31,20	76.00	0.00	0.00
46,59	38.40	98.00	0.00	0.00
51,09	50,94	107.00	0,00	0.00
55,59	31.20	116,00	0.00	0.00
60.09	25,60	128.00	0.00	0.00
64.59	21.70	140,00	6.00	0.00
69.09	20,38	144.00	0.00	0.00
73,59	21.70	136,00	0,00	0.00
78.09	22,69	105.00	0.00	0.00
82,59	20.38	70.00	0.00	0.00
87.09	17.83	60.00	0.00	0.00
91,59	16,64	54.00	0.00	0,00
96,09	16,64	45.00	0.00	0.00
100.59	16.64	39,00	0.00	0,00
105.09	17,83	37.00	0,00	0,00
109.59	20,38	34,00	0,00	0.00
114,09	20,38	36,00	0.00	0.02
118,59	21,73	42.00	0,00	0.00
123.09	23,77	52.00	0.00	8,09
127,59	38,40	74.00	0,00	0,00
132.09	58,73	123,00	0,00	0.03
136,59	71.31	193,00	0,00	0,00
141.09	71.31	226.00	0.00	0,00
145.59	58,73	214.00	0.014	0.03

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	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
	150.09	47.54	200.00	0.00	0,00
	154,59	33,28	190.00	0.00	0.00
	159,09	23,77	147.00	0.00	0.00
	163.59	17.52	122.00	0.00	0.00
	168,09	13,31	94.00	0.00	0.00
	172.59	10.76	70.00	0,00	0.00
	177.09	10.51	53,00	0.00	0.00
	181.59	10.62	39.00	0.00	0.00
	186.09	2.41	11.15	0.00	0.00
	198.59	0.54	4.85	0.00	0.00
	195,09	0.12	3,42	ଡ଼ୄ୕ୖଡ଼ଡ଼	0,00
MEAN	VALUES				
FLOW	WEIGHTED	24,28	114.38	0,00	0.00

## CUMULATIVE PARAMETER LOADINGS

T(MINS)	Q(CU,M)	SS(KG)	DS(KG)	CA(KG)
1.030	0.494	0.003	0.000	0,000
6,090	3,210	0,030	0.000	0,000
10,590	5,531	0.067	0.000	0.000
15.090	7.594	0.103	0.000	0,000
19,590	9,884	0,150	0.000	0.000
24.090	12,537	0.225	0,000	0.800
28,590	15,940	0.355	0.000	0.000
33.090	21.977	0,663	0.000	0,000
37.590	39,875	1,216	0.000	P. 000
42.090	39.901	1.856	0.000	0.000
46.590	19.297	2.677	0.000	0,000
51,099	61.358	3,916	0,000	A & B & A
55.590	72.447	5.148	0,000	0.000
60.090	84.115	6.082	0,012,00	0.000
64.590	86.501	6,937	0.000	0,000

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STORM OF THE 9TH JUNE 1977

	T(MINS)	Q(CU.M)	SS(KG)	DS(KG)	CA(KG)
	69,090	92.181	7.743	0,000	0.000
	73,590	97,862	8,538	0.000	0.999
	78.990	103.856	9.260	0.000	0.000
	82.590	109.670	9,771	0,000	0.000
	87,090	114.827	10.106	0,000	0.000
	91,590	119.480	10,372	0.000	0.000
	96,090	123,973	10.594	0.000	0.000
	100.590	128,466	10.783	0.000	0.000
	105.090	133.119	10,960	0.000	0.000
	109.590	138,277	11.143	0.000	0.000
	114,090	143.778	11,335	0.000	0.000
	118.590	149.459	11,557	0,000	0.000
	123,090	155,598	11.846	0.000	0.000
	127.590	163,991	12,383	0.000	0,000
	132.290	177.104	13,699	0.000	0.000
	136.590	194.660	16,495	0.000	0.000
	141.098	213.914	20.529	0.000	0.090
	145.590	231,470	24.395	0.000	0.000
	150.090	245,817	27.368	0.000	0.000
	154.593	256,728	29,500	0.000	0.000
	159,090	264.430	30.808	0.000	0.000
	163.590	270.004	31,561	0.000	0.000
	168,090	274.166	32.014	0.000	0.000
	172.590	277.415	32,282	0.000	0.000
	177.090	280.286	32,458	0.000	0.000
	181.590	283.139	32,589	0.000	0.000
	186,090	284.898	32,639	0.000	0.000
	190,590	285,296	32,643	0.000	0.000
	195.090	285,386	32,643	0.000	0.000
TOTAL	LOADINGS	285, 392	32,643	0,000	0.000

STORM OF THE 27TH JUNE 1977

TIME SAMPLING COMMENCED 0,00 SAMPLING DURATION(MINS) 222,92 DOSER CONCENTRATION(G/L) 42,40 DOSE RATE(ML/MIN) 1,04

## MEASURED PARAMETER CONCENTRATIONS

T(MINS)	Q(L/S)	SS(MG/L)	VSS(MG/L)	CA(MG/L)
1.03	8,00	225,00	44,00	0,00
6,09	40.35	245,00	46.00	0.00
10.59	108.54	295,00	53,20	0.00
15.09	73.49	332.00	60,00	0,00
19,59	47.93	320.00	53,00	0,00
24.09	29.40	230.00	33,60	0,00
28,59	18.00	187.00	30.00	0.00
33.09	10.76	172.00	28.00	0.00
37,59	7.41	162.00	26.40	0.00
42.09	5,65	137.00	24.00	0.00
46.59	8,09	115.00	23.00	0.00
51.09	9,91	113.00	22.80	0.00
55.59	100.22	160.00	25.20	0.00
60.09	41.21	175,00	26.00	0.00
64.59	21.51	214,00	28.00	0.00
69.09	15.47	238.00	41.20	0.00
73.59	13.57	178.00	34,00	0.00
78.09	137.80	148.00	24.00	0.00
82,59	88.19	176.00	29.00	0.00
87.09	97.99	187.00	44.00	0.00
91.59	61.24	151,00	37.00	0.00
96.09	44.10	138.00	29.00	0.00
100.59	28,09	125,00	33,00	9.90
105.09	20.51	117.00	39,50	0.00
109.59	13,78	87.00	34,00	0,00
114.09	8,91	74.50	25.50	0.00
118,59	6.73	70,00	20.00	0.00
123,09	5.24	68.00	15,00	0.00
127.59	6.50	57,00	19.00	0.00
132,09	22,05	51.00	5,00	0,00
136,59	29,40	39.00	5.00	0.00
141.09	23.84	37,00	5.00	0.00
145.59	18.76	33.00	5.00	0.00

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STORM OF THE 27TH JUNE 1977

MEAN FLOW 240

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T(MINS)	Q(L/S)	SS(MG/L)	VSS(MG/L)	CA(MG/L)
150.09	11,45	33,00	5,00	0.00
154,59	11.02	25.00	5,00	0.00
159.09	11.02	20,00	5,00	0.80
163,59	19.17	20,00	5,00	0.00
168,09	45,00	25.00	5,00	0.04
172,59	37.69	32,00	5.00	0.00
177,09	60.41	43,00	5,00	0.00
181.59	58,02	47.00	5.00	0.00
186,09	35,00	54,00	5,00	0.00
190,59	24,23	60,00	5,00	0,00
195,09	19,17	56,00	5,00	0.00
199.59	17,78	52,00	5,00	0.00
204.09	13,57	45.00	5,00	0.00
208,59	15,81	28,00	5.00	0.00
213.09	3,58	8,66	5,00	0.00
217.59	0.81	4.28	5,00	0.90
222,09	0,18	3,29	5,00	0.00
VALUES	31 43	1 4 7 0 4	00 75	6 60
«CTONTED	21992	141.34	20,410	81 • 61 61
T(MINS)	BOD (MG/	L) CL(MG/L	) NO3(MG/L)	NH4(%G/L)
1,03	9,80	9,90	22 00	1 60
6.09	11.50	0.00	13.00	0,90
10,59	13.00	0.00	8.40	0.70
15,09	10.80	0.00	6.70	0.59
19,59	9.00	0.80	6,50	0.55
24.09	8,50	0.00	6.40	0.51
28.59	7.90	0.00	7,20	0.57
33,09	6,30	0.00	7,60	0,49
37.59	6,80	0,00	7,60	0,50
42,09	7.80	0,00	10,20	0.40
46,59	8,60	0,00	9,70	0,18
51.09	9,30	0.00	10,40	0,53
55,59	7,60	0.00	5,50	0,53
66,09	6,30	0,00	5,60	0,56
64.59	6.00	0.00	7.20	0,80
69,69	6.50	0,00	9.10	1,00
13.59	7.20	0.00	10,20	1,30
78,09	7.60	0,00	7,70	13 <sub>#</sub> 7 8
82,59	0 6 /0	0 0 0	TT E7 /m	0 0 0
0 9 mm	n, 319	10 . 10 10	1.50	13 g 4 Z

## STORM OF THE 27TH JUNE 1977

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(CONTINUED)

	T(MINS)	BOD(MG/L)	CL(MG/L)	NO3(MG/L)	NH4(MG/L)
	91.59	9,80	0.00	6,70	0.53
	96.09	12.00	0,00	6.50	0.62
	100.59	9.80	0.00	6.70	0.59
	105.09	8.70	0.00	7.10	0.57
	109.59	8,50	0.00	7.20	0,59
	114.09	7.70	0.00	7.60	0.58
	118.59	7.50	0.00	8.60	0.56
	123.09	6.50	0.00	9.60	0.59
	127,59	6.30	0.00	8,80	0.62
	132.09	6.60	0.00	7.60	0.69
	136.59	6.50	0.00	7.00	0.60
	141.09	6.00	0.00	6.80	0.49
	145,59	6.60	0.00	8,20	0.54
	150.09	6.50	0.00	9.40	1.15
	154,59	6.00	0.00	9.40	4.52
	159,09	6.10	0.00	9.20	0.45
	163,59	5,50	0.00	7.80	0.47
	168.09	4.20	0.00	6.30	0.54
	172.59	5.20	0.00	5.70	0.50
	177.09	5.40	0.00	5.00	0.49
	181,59	4.80	0.00	4.80	0.47
	186,09	3,10	0.00	4.90	0.42
	190.59	3,30	0.00	5.60	0.45
	195.09	4.20	0.00	6,10	0.46
	199:59	2.80	0.00	6.30	0,45
	204,09	3,00	0.00	6.30	0.44
	208,59	3,20	0.00	6,30	0.45
	213.09	0,96	0.00	6,30	0.45
	217,59	0.45	0.00	6,30	0.45
	222,09	0.33	0,00	6,30	0,45
MEAN	VALUES				
FLOW	WEIGHTED	7.83	0,00	7,30	0.63
CUMU)	LATIVE PARAM	ETER LOADING	23 C ( 1 Fat 5		
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T(MINS)	Q(CU <sub>e</sub> M)	SS(KG)	VSS(KG)	CA(KG)
1.030	0.494	0.111	0.022	0,000 0 000
10.590	27,739	7.317	1.354	0.000

## STORM OF THE 27TH JUNE 1977

T(MINS)	Q(CU,M)	SS(KG)	VSS(KG)	CA(KG)
15,090	52.314	14,989	2.739	0.000
19.590	68.706	203 341	3.670	6 600
24 090	70 145	22,072	A 121	0 0 0 0 0
28.590	85.544	24 599	4 235	0,000
33 090	80: 425	25,200	A AA8	0 (3 Ø Ø
37.590	91.878	25 700	4 515	0 000
A2 090	92.641	25 074	4 550	0 000
46.590	95, 497	26 206	A 603	0 0 0 0 0 0 0 0 0 0
51.090	97,927	26.483	4.659	0 000
55.590	112.794	28.618	5.021	0.000
60.090	131,887	31.794	5 500	6 666
64.590	140.354	33.400	5 726	A GUE
69 090	145 347	24 547	5 967	0,0000
73,590	149.268	35,365	6.054	0 0000
78 090	169 700	38 603	6 616	0 300
82.590	200.211	42 511	7 448	0,000
87.090	225.346	48.376	8.340	3.000
91.590	246.843	51 742	9 9 9 9	6 6 6 6 6
96.090	261.664	53,802	0,680	0 000
100.590	270.808	55 089	0 000	5 260
105.090	277, 369	55,884	10.226	0 000
109.590	281,998	56.361	10.397	0.000
114,090	285,061	56.610	10 190	0 000
118.590	287.172	56,763	10.538	0.000
123.090	288.788	56.874	10.566	6.900
127.590	290.373	56.973	10,586	0,000
132.090	294,228	57.179	10.613	0.000
136.590	301.173	57.489	10.648	0,000
141.090	308.359	57.763	10.684	0.000
145.590	314.110	57,964	10,712	0.000
150,090	318.190	58.099	10.733	0.000
154.590	321.224	58.187	10.748	0,000
159.090	324.201	58,254	10.763	0.000
163.590	328.277	58.336	10.783	0,000
168,090	336.940	58.534	10.827	0.000
172.590	348.102	58,851	10.882	0.000
177,090	361,345	59.353	16.949	0.900
181,590	377.333	60.073	11.029	0.000
186,090	389.890	60.703	11.091	9.000
190,590	397.885	61.157	11.131	0.000
195, 394	493.744	61.497	11.161	0.840
199,590	408,733	61.767	11.186	0.200
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#### STORM RUNOFF DATA 611 62 63 63 63 63 61 61 63 63 63 63 ----

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DIUKM	OF THE 2	TH JUNE 1	911			
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	T(MINS)	Q(CU.M)	SS(KG)	VSS(KG)	CA(KG)	
				4		
	204,090	412,965	61.973	11,207	0.000	
	208,590	416.931	62.117	11,227	0.000	
	213,090	419.547	62,170	11,240	0.000	
	217,590	420.140	62.175	11.243	0.000	
	222,090	420.274	62.175	11.243	0.000	
LATÓT	LÖADINGS	420.284	62,175	11,243	0,000	
	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)	
	1,030	0,005	0,000	0.011	0.001	
	6.090	Ø.Ø82	0.000	0.129	0.009	
	10,590	0.331	0.000	0.336	0.025	
	15,090	0.625	0.000	0.523	0.041	
	19,590	0,789	0.000	0.632	0.050	
	24,090	0.880	0,000	0,699	0.056	
	28,590	0.930	0.080	0.742	0.059	
	33,090	0,956	0.000	0.771	0.061	
	37,590	0.972	0.000	0.790	0.063	
	42.090	0,985	0.000	0.805	0.863	
	46.590	1.000	0.000	0.824	0.064	
	51,090	1.022	0.000	0.848	0.065	
	55.590	1.144	0.000	0.955	0.073	
	60.090	1.279	0.000	1.061	0.084	
	64,590	1,331	0.000	1.115	0.089	
	69.090	1.362	0.000	1.155	0.094	
	73,590	1.389	0.000	1,193	0.098	
	78,090	1.541	0.000	1.368	0.118	
	82.590	1.786	0.000	1.600	0.144	
	87.090	2.020	0.000	1.791	0.161	
	91.590	2.234	0.000	1.947	0.172	
	96.090	2.388	0,000	2.041	0.180	
	100.590	2.495	0.000	2.105	0,186	
	105.090	2.556	0 000	2 152	2 100	
	109,590	2,506	0.000	2 193	01.102	
	114.090	2.671	0 000 0 000	5 202 A	67 £ 7 3 67 £ 6 7	
	118 503	2.627	0 000	* R 4 5 V 7 7 7 7 7	Q 100	
	123,090	2,640	6 6000	4 6 A A J 7 7 7 2 0	X 0 1 7 0	
	107 KOB	9 650	0 0000	6 n 6 3 0 7 7 5 7	00121	
	120 000	24030 5 203	V2 6 27 27 21 28 (3 (3 (3 )	3 402	Pe 1 7 6	
	126 600	2009 200 200	0 0 0 1 0 1 0 1 0 1 0 1 0 1 0 1 0 1 0 1	2 6 2 8 3	10,200 0,200	
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# STORM OF THE 27TH JUNE 1977

(CONTINUED)

	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	141.090	2.774	0,000	2,383	0.209
	145.590	2.810	0,000	2,426	2,211
	150.090	2.837	0.000	2,462	0.215
	154,590	2,856	0,000	2.491	0.217
	159.090	2.874	0.000	2,518	0.219
	163.590	2,897	0.000	2.553	0.221
	168.090	2,938	0,000	2,613	0.225
	172,590	2,991	0.000	2.680	0.231
	177.090	3,061	0.000	2.750	0.237
	181,590	3,143	0,000	2.829	0.245
	186.090	3.193	0.000	2,889	0.251
	190.590	3.219	0.000	2,931	0.254
	195.090	3.241	0.000	2.965	0.257
	199,590	3,258	0.000	2,996	0.259
	204.090	3.270	0.000	3.023	0.261
	208.590	3,283	0.000	3.048	0.263
	213.090	3.289	0.000	3.065	0.264
	217,590	3.289	0.000	3.068	0.264
	222,090	3.289	0.000	3,069	0,264
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TATAP	DANTLAD	30609	0,000	3.009	0.264
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## STORM OF THE 2ND MAY 1978

TIME SAMPLING COMMENCED 12.45 SAMPLING DURATION(MINS) 29,75 DOSER CONCENTRATION(G/L) 40,00 DOSE RATE(ML/MIN) 1.00

## MEASURED PARAMETER CONCENTRATIONS

	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(NG/L)
	1,19	8,00	36,00	215,00	25,60
	4.40	9,55	69,00	265,00	22,70
	7.44	11.88	49.00	160,00	18.70
	10,48	19,51	62,20	155,00	16.50
	13,52	21,86	56,00	142.00	15,00
	16,56	23,67	54.00	115,00	15,50
	19,60	8,68	21,70	115.00	15,50
	22,64	3,18	9,86	115.00	15,50
	25,68	1.17	5,51	115,00	15,50
	28,72	0,43	3,92	115,00	15,50
MEAN	VALUES				
FLOW	WEIGHTED	10,98	50,28	153,33	17.31

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	T(MINS)	BOD(MG/L)	COD(MG/L)	NO3(MG/L)	NH4(MG/L)
	1,19	6,20	11.70	11.70	0,65
	4.40	6,70	23,20	11,20	0.67
	7.44	6.70	43.10	16,40	0.76
	10,48	6,79	49,40	18,48	1.20
	13,52	6,50	54,20	10,00	1,25
	16,56	5,50	11,40	10,20	1.15
	19,60	2,21	7,35	10,20	1.15
	22,64	1,00	5,86	10,20	1,15
	25,68	0,56	5,32	10,20	1,15
	28,72	0,39	5.12	10,20	1,15
	des la contact				
MEAN	VALUES				
FLQW	WEIGHTED	5,70	30,92	10.41	1.06

STORM OF THE 2ND MAY 1978 EXAMPLE AND MAY 1978 (CONTINUED)

# CUMULATIVE PARAMETER LOADINGS

	T(MINS)	$Q(CU_{B}M)$	SS(KG)	DS(KG)	CA(KG)
	1,195	8.574	0,021	0.123	0.015
	4,400	2.253	0,109	0,527	0,055
	7,440	4,208	0.223	0,935	0.095
	10,480	7.071	0,385	1,384	0.145
	13,520	10,844	0,607	1,944	0,205
	16,560	14,996	0,835	2,476	Ø.268
	19.600	17,946	0,962	2,815	0.314
	22,640	19,028	0.981	2,939	0,330
	25,680	19,425	0,984	2,985	0.337
	28 <sub>e</sub> 72Ø	19.570	0,985	3,002	Ø.339
TOTAL	LOADINGS	19,596	ø <b>.</b> 985	3,005	Ø,339
	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
•	1,195	0.004	0.007	0,007	0,000
	4.400	0.014	0,036	0.026	0,001
	7.440	0.027	0.102	0.047	0.003
	10,480	0.047	0.236	0.077	0.206
	13,520	0.072	0.432	0.115	Ø, e10
	16°560	0,096	0,566	0.157	0.015
	19,600	Ø <sub>e</sub> 109	0,595	Ø . 187	0,019
	22.640	0.111	0.603	@e198	0,020
	25,680	0.112	0,605	0.202	0,020
	28,720	0.112	Ø . 606	0,204	0,021
TOTAL	LUADINGS	Ø,112	0,606	0.204	0.021

## STORM OF THE 4TH MAY 1978

TIME SAMPLING COMMENCED 8,05 SAMPLING DURATION(MINS) 47,99 DOSER CONCENTRATION(G/L) 38,00 DOSE RATE(ML/MIN) 1,03

## MEASURED PARAMETER CONCENTRATIONS

	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
	1,19	8,00	226,00	124,00	14,50
	4.40	7,13	166,00	116,00	15,50
	7,44	11.37	120,00	152,00	15,50
	10,48	31,93	150.00	126,00	12,30
	13,52	51,93	185.00	90.00	11.10
	16,56	64,92	292,00	56,00	9,70
	19,60	33,87	226,00	28,00	8,90
	22,64	24.34	140,00	84,00	8,90
	25,68	21,05	99,00	88,00	9,40
	28,72	22,91	122,00	56,00	9,20
	31.76	21,64	55,00	76,00	8,90
	34,80	13,20	47.00	72,00	9.40
	37,84	4,84	19:13	72,00	9,40
	40.88	1.78	8,92	72,00	9,40
	43,92	0,65	5.17	72,00	9.40
	46,96	0,24	3,80	72,00	9,40
MEAN	VALUES				
FLOW	WEIGHTED	20,23	172,20	78,88	10,39
	T(MINS)	BOD(MG/	L) COD(MG/L	) NO3(MG/L)	NH4(MG/L)
	1.19	12.30	122.00	12.00	1.85
	4. 40	8,40	81,40	12,30	1,40
	. 7.44	8,80	79,10	12,00	1.25
	10,48	9,90	110,70	11.00	1,10
	13.52	8,70	157,00	10.10	1,00
	16,56	6,90	137,90	6,70	0,91
	19,60	7,70	93,80	6,30	0.77
	22,64	7.10	84,20	7.00	0,75
	25,68	4.70	79,10	6,93	0,67
	28,72	6,50	62 ,70	7.40	0,69
	31.76	4.90	27.70	7 40	0.66

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STORM OF THE 4TH MAY 1978 (CONTINUED)

	T(MINS)	BOD(MG/L)	COD(MG/L)	NC3(MG/L)	NH4(MG/L)
	34.80	3,40	9,00	7,90	0.71
	37,84	1.44	6,47	7,90	0.71
	40,88	0,72	5,54	7,90	0,71
	43,92	0 36	5 6 2 0	7,90	Ø e 71
	40 e 20	ស្ដ្	5801	1020	0012
MEAN	VALUES				
FLQW	WEIGHTED	7,23	99 <sub>8</sub> 87	8.31	Ø . 88

## CUMULATIVE PARAMETER LOADINGS

	T(MINS)	Q(CU,M)	SS(KG)	DS(KG)	CA(KG)
	1,195	0.574	0.130	0.071	0.008
	4,400	2,032	0.420	0,247	0.030
	7.440	3,720	0,655	0.478	0,056
	10.480	7,669	1.207	1.010	0.109
	13,520	15,317	2,510	1,814	0,198
	16,560	25,974	5,094	2,578	P. 308
	19,600	34,983	7,491	2,983	0.393
	22,640	40.292	8,488	3,264	0.440
	25,680	44.432	8,987	3,620	0.478
	28,720	48,442	9.431	3,907	0,515
	31,760	52,505	9,793	4.174	0,552
	34,800	55,683	9,958	4.410	0,581
	37,840	57.328	10.019	4,529	0.596
	40.880	57,932	10,029	4,572	0.602
	43,920	58,153	10,030	4,588	0,604
	46,960	58,234	10,031	4,594	0,605
TOTAL	LOADINGS	58,249	10,031	4。595	0.605
	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	1,195	0,007	(1, (17)	0.007	0.001
	4,400	0.022	0,221	0,025	0.002
	7.440	0,037	C. 356	0,345	0.005
	10,480	0.075	0.751	0,090	0,009
	13,520	0.145	1,803	6.170	0.017
	16,560	0.227	3,367	0.258	Par 27

STORM OF THE 4TH MAY 1978

(CONTINUED)

	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	19,600	0,292	4.453	0,317	0,035
	22,640	0,332	4,928	0.352	0,039
	25,680	0.357	5,267	0.381	0.042
	28,720	0.379	5,550	0.410	0.045
	31.760	0.402	5,735	0.440	0.047
	34,800	0,416	5.798	0.464	0.052
	37,840	0.420	5.812	0.477	0.051
	40,880	0.421	5.816	0.482	0.051
	43,920	0.421	5.817	0.483	0.051
	46,960	0,421	5,817	0.484	0.051
TOTAL	LOADINGS	Ø . 421	5,817	0,484	0,051

# STORM OF THE 5TH MAY 1978

TIME SAMPLING COMMENCED 2.00 SAMPLING DURATION(MINS) 44.95 DOSER CONCENTRATION(G/L) 38.95 DOSE RATE(ML/MIN) 1.03

MEASURED PARAMETER CONCENTRATIONS

	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L).
	1,19	8,00	18,00	40,00	0,00
	4,40	2 a 36	21,00	40,00	0,03
	7.44	14,64	25,00	36,00	0,00
	10.48	18,15	32.00	54.00	0.00
	13,52	17.36	49.00	136.00	0.03
	16,56	15.07	45.00	100.00	0.00
	19.60	12.48	43.00 .	84.00	0.00
	22,64	20.47	36.00	76.00	0.00
•	25.68	17.74	30.00	60.00	0.00
	28.72	21.01	28.00	68.00	0.03
	31.76	11.09	26.00	96.00	0.00
	34.80	17.36	29.00	88.00	0.00
	37.84	16.63	32.00	60.00	0.00
	40.88	15.07	20.00	82.00	0.00
	43,92	17,36	12,00	88,00	0.09
MEAN	VALUES				
FLOW	WEIGHTED	14,98	30,92	76,71	0 <sub>8</sub> 0 0
	T(MINS)	BÒD(MG∕L)	COD(MG/L)	) NO3(MG/L)	NH4(MG/L)
	1.19	6,50	4.50	5.10	0.27
	4.40	6.60	10.10	5.10	0.27
	7.44	6.70	18,10	4.60	0.28
	10.48	6.70	16.10	4.40	0.25
	13.52	6.70	12.40	4.10	0.21
	16.56	5.60	10.20	4.20	0.21
	19.60	4,90	6.80	3.60	0.21
	22.64	5,20	8.40	3.80	0.22
	25,68	6,20	9.00	4.40	0.23
	28,72	6.10	8.20	3.70	0.24
	31.76	5,80	6,82	3,60	0.33
	34,80	5,60	5,00	3,50	0,21

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STORM OF THE 5TH MAY 1978

T(MINS) BOD(MG/L) COD(MG/L) NO3(MG/L) NH4(MG/L) 4,90 4,30 4,10 3,50 3,30 3,20 37,84 2,80 0,20 2,50 40,88 6,19 43,92 0,19 MEAN VALUES FLOW WEIGHTED 8,33 3,91 5,68 0,23

(CONTINUED)

## CUMULATIVE PARAMETER LOADINGS

	T(MINS)	Q(CU,M)	SS(KG)	DS(KG)	CA(KG)
	1,195	0,574	0.010	0,023	0,000
	4.400	1,597	0.030	0,064	0.000
	7.440	3.147	0.067	0.121	0.000
	10.480	6,137	0,153	0.258	0.000
	13,520	9,375	0.284	0.563	0.000
	16,560	12.332	0.423	0.915	0.000
	19,600	14.844	0.533	1.147	0.000
	22,640	17,849	0,650	1,386	0.000
	25,680	21.334	0.766	1.624	0.000
	28,720	24,869	0,868	1.851	0.000
	31,760	27.797	0.948	2,083	0.000
	34,800	30.391	1,020	2,320	0,000
	37.840	33,491	1.114	2,550	0.000
	40.980	36.382	1.190	2,754	0.200
	43,920	39,339	1,237	3,006	0.200
TOTAL	LOADINGS	40,412	1.250	3,100	0,000
	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	1,195	0.004	0.003	0.003	0.000
	4,400	0.010	0,009	0.008	0,000
	7.440	0.021	0.034	0.015	0.001
	10,480	8.041	0.085	0,029	0,002
	13,520	0.062	0.231	0.043	0.002
	16,560	0,082	0,164	0,055	6,803
	19,600	0.894	0,186	0,065	0.201
	22,640	(1,109	2,209	0.076	(1 , 13 (1 )
	25,680	3,129	6.240	0.090	0,005

STORM OF THE 5TH MAY 1978

(CONTINUED)

	T(MINS)	BOD(KG)	COD(KG)	NQ3(KG)	NH4(KG)
	28.720	0.151	0,270	0.104	0,006
	31.760	0.168	0,292	0.115	0.007
	34.800	0.183	0,307	0.124	0.297
	37,840	0.199	0,319	0.135	0,208
	40.880	0.213	0.327	0.145	0.008
	43,920	0,225	0.334	0.155	0,009
TOTAL	LOADINGS	0.229	0.337	0.158	0.009

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## STORM OF THE 5TH MAY 1978

TIME SAMPLING COMMENCED 3,00 SAMPLING DURATION(MINS) 60,15 DOSER CONCENTRATION(G/L) 38,95 DOSE RATE(ML/MIN) 1,03

## MEASURED PARAMETER CONCENTRATIONS

	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
	1.19	8,00	7,00	88,00	Ø . ØØ
	4,40	3,81	8,00	96,00	0,00
	7.44	12,74	17,00	88,00	Ø, ØØ
	10,48	15,84	29,00	84,00	0,00
	13,52	13,09	18,00	84,00	0,00
	16,56	15,36	10,00	88,00	0,00
	19,60	14,52	15,00	96,00	0.00
	22,64	16,99	8.00	98,00	0,90
	25.68	20.47	9.00	108,00	0,00
	28,72	47.53	8,40	115.00	0,00
	31,76	38,02	8,80	124.00	0.20
	34,80	16,23	9,00	136,00	0,00
	37,84	21,58	8,80	120.00	0.00
	40,88	21.58	19.00	116,00	0.00
	43,92	20,58	30.00	100.00	0,00
	46,96	16,99	12.00	96,00	0,00
	50.00	6,23	6.30	96.00	0,00
	53,04	2,28	4,21	96,00	0,00
	56,08	0,84	3.44	96,00	0.00
	59,12	0,31	3,16	96,00	0,00
MEAN	VALUES				
FLOW	WEIGHTED	15,79	12,88	106,62	Ø, NA
	T(MINS)	ROD(MG/L)	COD(MG/L)	NO3(MG/L;	NH4(MG/L
	1.19	0.00	0.00	3.00	0.15
	4.40	0,00	0.00	2.98	0.12
	7.44	0.00	0.00	2,90	0.10
	10,48	0.00	0.00	2.90	0.07
	13,52	0,00	0,00	2.83	69,89
	16.56	0.00	0.00	2.79	0.10
	19,60	0,00	0.00	2.70	0,08

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### STORM RUNOFF DATA 671 588 68 F21 621 621 628 629 63 659 639 639 631 631 631 631 63

STORM OF THE 5TH MAY 1978 

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·	T(MINS)	BOD(MG/L)	COD(MG/L)	NO3(MG/L)	NH4(MG/L)
	22,64	0,00	0,00	3,20	0,08
	25,68	0,00	0,00	3,00	0.07
	28.72	0.00	0.00	6.90	0.06
	31,76	0,00	0.00	6.30	0,06
	34.80	0.00	0.00	5.80	0.05
	37.84	0.00	0.00	5.30	0.94
	40,88	0.00	0.00	5.10	0.74
	43.92	0.00	0.00	5.20	0.03
	46,96	0.00	0.00	5.10	0.04
	50.00	0.00	0.00	5.10	0.04
	53.04	0.00	0.00	5.10	0.04
	56.08	0.00	0.00	5.10	0.04
	59,12	0,00	6.66	5,10	0,04
MEAN	VALUES				
FLOW	WEIGHTED	0.00	0.00	4.70	0,06

#### CUNULATIVE PARAMETER LOADINGS FE CT (2 E2 C) -----

	T(MINS)	Q(CU,M)	SS(KG)	DS(KG)	CA(KG)
	1,195	0.574	0.004	0,050	0.000
	4,400	1.730	0.013	Ø.156	0.000
	7,440	3,239	0,034	0,292	0.646
	10,480	5,846	0,095	0.516	0,000
	13,520	8,485	0,158	0,738	0.000
	16.560	11,079	0.194	0.961	0.000
	19,600	13,803	0,228	1.211	0.000
	22,640	16,677	0,260	1,490	9.000
	25,680	20,093	0,289	1.843	0,000
	28,729	26,295	0,343	2,541	0.000
	31,760	34,097	8,410	3,470	0,000
	34,800	39,045	0.454	4,105	0.9.90
	37,842	42,493	0.484	1.544	0,000
	40,880	46,436	0.539	5,009	C. MAD
	43,920	54.275	0,633	5.424	9.289
	46,960	53,701	0.707	5,761	0,000
	50,000	55,818	(* 729	5,964	0.200
	53,940	56,595	0,732	6,438	0.000
	56,080	56,880	1.734	6.066	0.000
	59,120	56,984	0.734	6.076	0,000
TOTAL	LOADINGS	57,003	0.734	6,078	0. VA0

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STORN OF THE 5TH MAY 1978

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(CONTINUED)

	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	1,195	0,000	0,000	0,002	0,000
	4,400	0,000	0,000	0,005	0,000
	7.440	0,000	0,000	0.010	0,000
	10,480	0.000	0,000	0.017	0.001
	13,520	0,000	0.000	0.025	0.001
	16,560	0.200	0.000	0,032	0.001
	19,600	P.000	0.000	0,039	0,001
	22.640	0,000	0,000	0,048	0,002
	25,680	0,000	0.000	0,058	0,002
	28,720	0.000	0,000	0.092	0.002
	31,760	0,000	0,000	0.144	0.003
	34,800	0,000	0,000	0,174	0,003
	37,840	0,000	0.000	0,193	0.003
	40,880	0,000	0,000	8.214	0.003
	43,920	0.000	0,000	0,233	9.803
	46,960	0.000	0,000	0.251	0.204
	50.000	0,000	0.000	8,262	0.004
	53,040	0,000	0.000	0.266	0.004
	56,080	3.000	0.000	0.267	0.004
	59°159	0,000	0.000	0.268	0.004
TOTAL	LOADINGS	0,000	ଷ, ମଷଷ	0.268	0.004

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## STORM OF THE 5TH MAY 1978

TIME SAMPLING COMMENCED 4.00 SAMPLING DURATION(MINS) 41.91 DOSER CONCENTRATION(G/L) 38.95 DOSE RATE(ML/MIN) 1.03

MEASURED PARAMETER CONCENTRATIONS

	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
	1,19	8,00	8,00	126,00	0,00
	4,40	7,31	9,00	110,00	0,00
	7.44	41e19	52,00	96,00	0.00
	10,48	31,94	46.00	92,00	0,00
	13,52	18,57	29,00	88,00	0,00
	16,56	28,12	35,00	84.00	0,00
	19,60	29,57	19,00	88.00	0,00
	22,64	24.34	24,00	96,00	0.00
	25,68	19,47	27,00	124,80	0.00
	28,72	16,23	11,00	96,00	0,00
	31,76	5,95	5,93	96.00	0.00
	34,80	2,18	4.08	96,00	0.00
	37,84	Ø . 8 <i>9</i>	3,39	96,00	0.00
	40,88	0.29	3,14	96.00	0,00
MEAN	VALUES				
FLOW	WEIGHTED	16,94	30,02	94.60	0,00
	T(MINS)	BOD(MG/L)	COD(MG/L)	NČ3(MG/L)	NH4(MG/L)
	1.19	0,00	0,00	4.80	0.04
	4,40	0,00	0,00	4.50	0,03
	7,44	0,00	0,00	5,20	0,02
	10,48	0.00	4,90	0.04	0.00
	13,52	0,00	4,50	0,05	0,00
	16,56	Ø. 2 C	3,80	¢ " @ 7	0,00
	19,63	0,00	Ø n 90	0.07	Ø,ØQ
	22,64	P . 09	4.50	0,86	Ø , Ø Ø
	25,68	0,00	0,50	0.05	0.00
	28,72	0.00	5,42	0,04	0,00
	31,76	(1 g (2 Ø)	5,15	0,04	0.00
	34,80	0,00	5,05	6. 8 6 4	0,00
	37,84	(* <sub>6</sub> (* 6	5,02	0,04	0,00

STORM OF THE 5TH MAY 1978 (CONTINUED) T(MINS) BOD(MG/L) COD(MG/L) NO3(MG/L) NH4(MG/L) 40,88 0.00 5.01 0,04 0,00 MEAN VALUES FLOW WEIGHTED 0,00 2.71 1,23 0.01 CUMULATIVE PARAMETER LOADINGS T(MINS) Q(CU,M) SS(KG) DS(KG) CA(KG) 1,195 0.574 0.005 0.072 0.009 0.017 0.247 4.400 2.049 0,000 7.440 6.472 0.197 9.688 0.000 10.480 13.141 0.526 1,316 0.000 0.705 1.733 13,520 17.747 0.000 16.560 22.005 3.843 2.097 0.000 0.000 19,600 27,266 0,985 2.550 1,090 22.640 32,183 3,001 0.000 25.680 36,180 1.191 0,000 3.400 28,720 1.254 39,436 3.726 0.000 31.760 41,459 1,273 3,920 0.000 42.201 1.277 34.800 3,991 0.000 1.278 37.840 42,473 4.018 0.000 42.572 40.880 1.278 4.027 0.000 TOTAL LOADINGS 42.590 1,279 4.029 1.000 T(MINS) BOD(KG) COD(KG) NO3(KG) NH4 (KG) 1,195 0,000 0,000 0.003 0.000 4,400 0.000 0,000 0.010 0.000 7.440 0,000 0.000 6.000 0.032 10,480 0.015 0.200 0.000 0.051 0.037 13,520 0.000 0.999 0.151 16,560 0.000 0.054 0,051 00,000 0,0MM 19,602 3.066 0.052 0.000 10 . C79 22,649 0.000 8.052 0.200 25,680 0,000 0.090 0.052 0.000 (\* , () (\* (? 0,099 6.452 (1,000A 28,720 31,760 Ø, ØØR a . C. O. U 0.110 0,252

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STORM OF THE 5TH MAY 1978 (CONTINUED) 医生生素医生素有有的学生的有有有有有有有有有有有有有的的现象。 T(MINS) BOD(KG) COD(KG) NO3(KG) NH4(KG) 37,840 40,880 Ø.115 Ø.115 Ø,052 0,052 0,000 0,000 0,000 0.000 0.115 TOTAL LOADINGS 0.000 0,052 0.000

### STORM RUNOFF DATA 10 th 69 01 th 10 10 th 60 60 60 10 10 40 to 10 10 60 10 10

### STORM OF THE 15TH MAY 1978

TIME SAMPLING COMMENCED 16,34 SAMPLING DURATION(MINS) 23,67 DOSER CONCENTRATION(G/L) 41,50 DOSE RATE(ML/MIN) 1,03

## MEASURED PARAMETER CONCENTRATIONS

	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)	
	1.19	8,00	202,00	320,00	26,80	
	4,40	11,12	165,00	272,00	24,70	
	7,44	9,40	147,00	224,00	23,28	
	10,48	11,19	125,00	242,00	19,60	
	13,52	4.10	47,74	240,00	19,60	
	16,56	1,51	19.41	240.00	19,60	
	19,60	0,55	9,02	240,00	19.60	
	22.64	0,20	5.21	240,00	19,60	
MEAN	VALUES					
FLOW	WEIGHTED	5,86	130,89	257,86	22,80	
	T(MINS)	BOD(MG/	L) COD(MG/I	5) NO3(MG/U)	NH4(MG/	[, ]
	4 40	00 00	131 00	12 50	1 00	
	4 40	12 0.0	110 00	12 00	1 27	
	7 A A	15 10	96 00	11 40	1 301	
	10.48	17.20	74.00	11.30	1 30	
	13.52	6.50	30.30	11.30	1.30	
	16.56	2.57	14.28	11.30	1.30	
	19,60	1.13	8,40	11.30	1.30	
	22,64	0.61	6,25	11.30	1,30	
MEAN	VALUES					
FLOW	WEIGHTED	14.19	89.74	11.69	1.24	

### CUMULATIVE PARAMETER LOADINGS

T(MINS)	Q(CU,M)	SS(KG)	DS(KG)	CA(KG)
1,195 4,400	0.574	Ø.116 0.448	0.184	0,215
7.443	4.268	0.741	1.187	8,107

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	T(MINS)	Q(CU,M)	SS(KG)	DS(KG)	CA(KG)
	10,480	6.146	0,995	1,624	0,147
	13,520	7,541	1,133	1,958	0,174
	16,560	8,053	1,152	2.081	0,184
,	19.600	8,240	1,155	2.126	0,188
	22,640	8,309	1,156	2,143	Ø.189
TOTAL	LOADINGS	8,321	1.156	2.146	0,190
	T(MINS)	BOD(KG)	COD(KG)	N03(KG)	NHĄ(KG)
	1,195	0.012	0,069	0,007	0,001
	4,400	0.040	'0,288	0,029	0,893
	7.440	0.066	0.490	0,051	0,005
	10,480	0.096	0.649	0,073	0.008
	13,520	0.115	0.731	0.088	0,009
	16,560	0,118	0.744	0.094	0.010
	19.600	0.118	0.746	0.096	0.010
	22,640	0.118	3.747	0.097	0.010
TOTAL	LOADINGS	0,118	0,747	0,097	0.010

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## STORM OF THE 24TH MAY 1978

TIME SAMPLING COMMENCED 9.41 SAMPLING DURATION(MINS) 20.63 DOSER CONCENTRATION(G/L) 41.50 DOSE RATE(ML/MIN) 1.03

## MEASURED PARAMETER CONCENTRATIONS

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	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
	1,19	8,00	189,00	416,00	48,60
	4,40	23,35	205,00	380,00	44,18
	7.44	20,03	155,00	360,00	41,40
	10,48	7.34	58,74	360,00	41.40
	13,52	2,69	23,44	368.80	41,40
	16,56	0,99	10,50	360,00	41,40
	19,60	0.36	5,75	360,00	41.40
MEAN	VALUES				
FLOW	WEIGHTED	9,18	155.47	374,94	43.37
	T(MINS)	BOD (MG/	L) COD(MG/I	L) NO3(MG/L)	NH4(MG/L)
	1.19	33,70	180.60	32,40	2,00
	4.40	36,60	213.60	32.10	2.04
	7.44	26,19	159,20	30,80	2,10
	10,48	9,76	61,55	30,80	2,10
	13,52	3,77	25.74	30,80	2,10
	16,56	1.57	12,60	30,80	2,10
	19,60	0.77	7,79	30,80	2,10
MEAN	VALUES				
FLOW	WEIGHTED	27,14	159.11	31.48	2.05

## CUMULATIVE PARAMETER LOADINGS

T(MINS)	Q(CU',M)	SS(KG)	DS(KG)	CA(KG)
1,195	0,574	0,108	0,239	0.028
4,400	3,512	0,694	1,393	0.162
7 . 440	7.468	1.412	2,658	0,332
10,480	9,964	1,716	3,757	0.435
13,520	10.880	1,759	4.086	0,473

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STORM	OF THE 2	4ТН МАХ 19 Ферманска с	78 P KK 03 67 95 1/6	(CONTINUE	(0
					•
	T(MINS)	$Q(CU_nM)$	SS(KG)	DS(KG)	CA(KG)
	16,560 19,600	11,215 11,338	1,765 1,766	4 a 207 4 a 252	Øn 487 Øn 492
TOTAL	LOADINGS	11,361	1.766	4,260	Ø # 493
,	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	1 , 195 4, 400 7, 440 10, 480 13, 520 16, 560 19, 600	0 0 0 1 9 0 1 2 4 0 2 4 9 0 3 0 0 0 3 0 7 0 3 0 8 0 3 0 8	0 • 10 4 0 • 697 1 • 440 1 • 754 1 • 799 1 • 806 1 • 807	Ø . 019 Ø . 113 Ø . 238 Ø . 315 Ø . 343 Ø . 353 Ø . 357	0 • 001 0 • 007 0 • 015 0 • 020 0 • 022 0 • 023 0 • 023
TOTAL	LOADINGS	Ø <b>, 3</b> Ø8	1,808	0.358	0.623

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STORM OF THE 31ST MAY 1978

TIME SAMPLING COMMENCED 17,13 SAMPLING DURATION(MINS) 11,51 DOSER CONCENTRATION(G/L) 41,50 DOSE RATE(ML/MIN) 1,03

MEASURED PARAMETER CONCENTRATIONS

	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(VG/L)	)
	1,19	8,00	413.00	368,00	25.00	
	4.40	2.94	151.00	368,00	25,00	
	7,44	1.08	56.00	368,00	25,80	
	10,48	0.40	22,44	368,00	25,00	
MEAN	VALUES					
FLOW	WEIGHTED	3,15	294,97	368,00	25,00	
	T(MINS)	BOD(MG/	L) COD(MG/	L) NO3(MG/	L) 184()	(G/L)
	1.19	44,50	439,00	0,09	5,30	
	4,40	16,30	161,00	0,00	5,30	
	7,44	6,00	59,00	0,00	5,36	
	10,48	2,39	24.80	6.00	5.30	
MEAN	VALUES					
FLOW	WEIGHTED	31,79	313,65	0,00	5,39	
CUMUI	ATIVE PARAM	ETER LOAD	INGS			
173 EX 45 197 1	19 CH EO HT 60 91 CH ET 67 ET (8 F)	1 153 169 463 \$5 815 912 844 963 5:8	43 43 43 ( <u>1</u> 3			
	T(MINS)	Q(CU.M)	SS(KG)	DS(KG)	CA(KG)	
	1,195	0,574	0,237	0.211	0.014	
	4,400	1.650	0,592	0.647	6.241	
	7,440	2,017	0,635	0.742	2,050	
	10,480	2,151	0.641	0.792	0,054	
TOTAI	, LCADINGS	2,176	0.642	6.801	0.054	

STORM * ex ca ca ca ca ca	OF THE 31	1 ST MAY 197 3 10 min 19 10 10 10 10 10 10 10	9 469 669 689 683 683	(CONTINUE)	))
	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	1,195 4,400 7,440 10,480	Ø • 926 Ø • 964 Ø • 968 Ø • 969	0,252 0,629 0,675 0,682	ୠୢୢ୶ଡ଼ଡ଼ଡ଼ ୠୢୢ୶ଡ଼ଡ଼ଡ଼ ୠୢୢୢ୶ଡ଼ଡ଼ଡ଼	0,003 2,009 0,011 0,011
TOTAL	LOADINGS	ø, ø69	0,682	0,000	0,012

## STORM OF THE 1ST JULE 1978

TIME SAMPLING COMMENCED 17.30 SAMPLING DURATION(MINS) 81.43 DOSER CONCENTRATION(G/L) 41.59 DOSE RATE(ML/MIN) 1.63

MEASURED PARAMETER CONCENTRATIONS

	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CACIGZES
	1,19	8,00	130,00	336,79	28.50
	4.40	17.30	53,02	168.00	18,03
	7.44	51.61	88,00	164.45	16,32
	10.48	30.38	68.00	152,20	15.09
	13.52	30,38	83,03	136,00	13.82
	16.56	48.13	79,00	142.03	12.83
	19.60	32.72	14,00	148.00	11.43
	22.64	25.78	42.03	152.00	10.03
	25.68	45,25	92.02	162.03	9.81
	28,72	42.54	114.93	156.00	12.52
	31.76	36.99	23.03	156.00	10,6%
	34.80	34.03	48.92	136,00	8.83
	37.84	38.67	125.00	108.00	9.00
	42.88	34.03	127.83	140.08	11,60
	43,92	19.34	22.00	168.03	10.01
	46,96	6.49	78.02	164.00	11.60
	50,00	27.44	26.00	164.00	10.00
	53,04	28.17	30.00	160.00	9,92
	56,08	27,09	52.33	158,00	10,00
	59,12	21.81	39,03	152,30	9.52
	62.16	32.72	63.60	144.00	8.80
	65,20	36,99	115.00	128.00	11.00
	68.24	13,56	44,07	128.00	11.00
	71,28	1,97	18,06	128,00	11,00
	74,32	1,82	8,52	128.00	11.00
	77.36	0.67	5,03	128,03	11.03
	80.40	0.25	3.74	128.03	11.00
MEAN	VALUES				
FLOW	WEIGHTED	25.71	68,17	149.97	11.67

### STORY RUNOFF DATA 153 68 49 151 40 44 lin ku 48 42 49 19 19 -10 -10 50 50 48 58

STORY OF THE 1ST JUNE 1978 tt 612 100 610

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T(MINS)	BOD(MG/L	) COD(MG/L)	NOBCMG/L	) NP4(MG/L)
1.19	0.00	109.00	0.00	0.03
4.40	0.00	63.00	0.00	0.00
7.44	0.00	60.00	0.00	0.60
10.48	0.00	60.00	0.00	ວ. ດວ
13.52	0.00	59.00	0.00	0.02
16.56	0.00	55.00	0.00	0.00
19.60	0.00	17.30	0.00	0.00
22.64	0.00	32.00	0.00	3.00
25.68	0.03	54.00	0.00	0.00
28.72	0.00	60.00	0.03	0.00
31.76	0.00	13.02	0.00	0.03
34.80	0.00	31.00	0.00	0.62
37.84	0.00	47.92	0.00	0.00
40.88	0.00	51.00	0.00	0.00
43.92	0.00	15.00	0.00	0.00
46,96	0.00	36.00	0.03	0.00
50.00	0.00	14.00	0.00	0.00
53.04	0.00	14.20	0.00	0.00
56.08	0.00	22.00	0.00	0.00
59.12	0.00	8.10	0.00	0.00
62.16	8.00	33.00	0.00	0.00
65.23	0.00	72.40	0.00	0.00
68.24	0.00	29.72	0.00	0.00
71.28	0.00	14.06	0.00	0.03
74.32	0.00	8.32	0.00	0.00
77.36	0.00	6.22	0.00	0.00
80.40	0.00	5,45	0.00	0.02
MEAN VALUES				
FLOW WEIGHTED	0.00	41,37	0.03	Ø " 0 0
CUMULATIVE PARAME	TER LOADI	INGS		
bin they are the call of the stat with the stat she and the stat the stat	a 188 493 684 494 683 693 594 69 9	4 MP (C) FR		
T(M1NS)	Q(CU.M)	SS(KG)	PS(KG)	CA(KG)
1.195	8,574	0.075	2.172	2.216
4.400	2,960	1.274	0.698	5.269
7.14.	9.245	. 751	1,737	3,175
13.480	16,723	1,35%	2,325	\$ , 293
13,520	22,265	1.769	3.724	1.373

STORM OF THE 1ST JUVE 1978

(CONTINUED)

	T(MINS)	0(CU.M)	SS(KG)	DS(KG)	CA(KG)
	16.560	28.696	2,289	4.620	0.458
	19.600	35.340	2.612	5.582	0,539
	22,640	40.675	2,751	6,381	0,596
	25,683	47.154	3,210	7,397	0.868
	28,723	55,160	4.033	8.662	0.749
	31,760	62.413	4.545	9.794	0,833
	34.800	68,890	4.773	1.0.741	0,896
	37,843	75,520	5,288	11,546	3,955
	49.880	82.150	6,054	12.364	1,223
	43,927	87.017	6,464	13,100	1.077
	46,960	89.364	6,559	13,492	1,101
	50,292	92,450	6,686	13,998	1.134
	03.840 66 303	100 660	0,828	14.619	1,184
	50.000 60.100	102,002	1,004	13,021	1,233
	69 163	111 006	1.239	17 044	1+6/0
	65 220	110 354	8 073	17 000	1 386
	68 243	100 064	8 491	10 100	1.300
	71.283	124.655	8,551	18 715	1 156
	74.320	125,275	8.560	18.795	1 460
	77.360	125,502	8,562	18.824	1.465
	80.400	125.586	8.562	18.835	1.466
<b>ምፅ</b> ፕ አቤ	LOADINGS	125,601	8 562	18 837	1 465
4 V 4 M 4 J	10402-00	1.609033	9 <b>8</b> 4 9 8 8	108023	4 4 12 (7 1 )
	T(MINS)	BOD(KG)	COD(KG)	N03(KG)	MH4 (KG)
	1,195	0,900	6,063	0.000	3,000
	4.400	0.000	8.256	0.000	6 . 2 19 (1)
	7,490	0.303	0.643	0.980	1.200
	10.480	0.000	1.088	7.796	3.292
	13,520	0.300	1,418	0000	6.000
	16,568	6,000	1,783	0,000	0.000
	19.600	0.000	2,431	0.300	7. 30.3
	22.642	000.0	2,159	it, iteld	0. 83
	25,680	6.265	2,451	3 4 14 3	6. 2 (2 (3
	28.721	Ant CC	2.907	13, 14PS	1.000
	31.760	0,500	3,179	a)0a	1. 699
	34,803	5 . OVA	3,329	No 211 22	20101
	37,843	12 . OF 3	3,581	Cathe S	
	49,888	0 · 9 9 9	3,025	"1" " I ( K3	-1 * -44 :)

STORY OF THE 18	T JUNE 19	78		
free \$29 for \$18 for 126 520 mm 64 6/0 for 129 521 520 500 400 for	- 1 phi bat est tat sat site bat tat per	- 	(CONTINUE	D)
T(MLNS)	ROD(KG)	COD(KG)	№03(KG)	NH4(KG)
43,920	0.200	4,082	0.003	0.000
46,967	0,200	4.134	0,000	3,002
50,000	0.000	4.197	6.909	0.000
53,040	4.289	4,268	0.20%	0,000
56.280	0.000	4.358	0.000	2. 4.214
59,120	0.000	4,427	0,000	0.000
62.160	0.000	4,538	0.200	9.0000
65.200	0.000	4.878	0.000	9.000
68.240	0.200	5,145	1. 122	2.420
71,280	0,000	5,186	0.000	3 . 3 6 11
74.323	0.000	5,193	3.023	3,22.1
77.360	0.203	5.195	0.000	6. 820
80,400	0.000	5.195	0.000	6.000
TOTAL LOADINGS	0.000	5,196	0.000	0.900

# STORM OF THE 15TH JUNE 1978

TIME SANPLING COMMENCED 10,12 SAMPLING DURATION(MINS) 130,07 DOSER CONCENTRATION(G/L) 41,50 DOSE RATE(ML/MIN) 1,03

MEASURED PARAMETER CONCENTRATIONS

T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(HG/L)
1,19	8,00	178,00	224,00	27,00
4,40	9,16	161,00	208,00	24.00
7.44	36,28	126,00	212,00	22,40
10,48	47.26	97.00	208,00	21,00
13,52	46,74	114,00	216,03	18,00
16,56	36,99	83,00	132,00	16,00
19,60	38,67	68,00 .	192,00	20,40
22.64	30,38	58,00	188,00	12,00
25,68	30,38	49,00	166,00	10,00
28,72	31,51	44,00	152,00	11,20
31,76	27.80	42.00	144,00	9,40
34,80	25,63	40.00	112,00	13.80
37.84	29.54	42.00	108,00	10.00
40,88	42,97	46.00	104.00	9.10
43,92	58,27	56,00	108,00	8.00
46,96	67,52	68,00	104,00	6.80
50.00	81,82	60.00	108.00	7.00
53,04	47,26	52,00	168,00	6,50
56,08	45,25	44.90	116,00	6,20
59,12	50,04	38,00	112,03	6,20
62 <sub>e</sub> 16	85,08	35,00	112,00	5,90
65 ° 3 Ø	77.34	62,00	104,00	6,50
68,24	52,52	45.00	112,00	5,90
71.28	55,97	47,00	156,00	5,90
74:32	42,54	25,00	116.00	7,40
77,36	41.70	68 ° QQ	184.00	7.70
80.40	35,45	24,00	120,00	6,50
83.44	34,03	22,00	92.00	5,90
86,48	31.51	22.00	50.00	7 , 70
89,52	28,94	17.00	20,00	6,20
92,56	29,54	18,00	18.00	6.50
95,60	55,97	50,00	13,02	5,93
98,64	50,64	52,00	20,00	6,20

### STORM RUNOFF DATA CO 60 10 10 50 50 60 50 50 FR 1.1 (11) 53 68 63

(MG/L)

STOR	OF THE 1	STH JUNE 1	978		
) (C2) (C2) (C2) (C2) (C2) (C2)	n m m # 69 69 69 69 13 13 13 14 61	53 fin 43 ca bit (p) D= 65 69 69	Ci 4a (3 Ka (4 Ki	(CONTINUED	)
	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/
	101.68	49,46	41,00	28,00	7,70
	104,72	40,51	54,00	26,00	5,30
	107,76	29,54	8,00	22.00	5,90
	110,80	23,24	5,00	14,00	5,90
	113,84	17,36	3.00	20.00	6,80
	116,88	18,10	6,00	27.00	7.00
	119,92	6,64	4.10	27.00	7.00
	122,96	2,43	3.40	27.00	7,00
	126.00	0.89	3,15	27.00	7.00
	129,04	0,33	3,05	27,00	7,00
MEAN	VALUES				
FLOW	WEIGHTED	37,41	51,37	109,74	8,97

52

T(MINS) BOD(MG/L) COD(MG/L) NO3(MG/L) NH4(MG/L) 1,19 15,50 133,00 32.10 1,20 4.40 11,20 120,00 22,80 1,80 7.44 10,10 92,00 22,00 1,60 9,00 1.30 10,48 69,80 23,80 13,52 9,00 75,00 21,60 1,40 16,56 9,20 50,90 13,20 1.70 19,60 8,80 50,00 1,50 13,00 22.64 8,40 50,00 12,60 1,40 25,68 7,00 1.30 36,00 12.10 28,72 6,20 28,00 11,60 1.10 31,76 6,50 11,50 28,00 1,20 34,80 7.10 2010 26,00 12,00 37,84 7,00 27,60 11,50 0,95 40.88 6,70 9,30 0,85 34,00 7,80 43,92 6,20 38,00 0,70 46,96 6,90 5,60 42,00 0,61 38,20 50,00 5,30 6,60 0,65 53,04 4,90 6.10 38,03 0.64 56,08 4.50 32,20 5,80 0,65 59,12 4,10 23,00 5,20 0,67 62,16 4,60 26,00 4,80 0,68 65,20 5,40 31 , 00 4.70 0.70 60,24 5.20 22,00 4,60 0,66 71,28 4,39 15,00 4,40 0,56 74,32 4,00 4,60 19,00 0.54 77,36 3,49 4,80 21,00 0,53 9. 70 80,40 3,60 20,00 0.51

STORM OF THE 15TH JUNE 1978 នាក់តាក់តាត់នាងនាងសតមមនុត្តជាតាងដោះមានដោយនាងនោះដោយនាងសតម្លាស់ (CONTINUED)

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	T(MINS)	BOD(MG/L)	COD(MG/L)	NO3(MG/L)	NH4(MG/L)
	83,44	5,80	16,00	4,60	0,49
	86.48	3,80	12,00	4,60	0,53
	89,52	2,60	11,00	4.40	0.57
	92,56	4,30	22.00	4.40	0.46
	95,60	5,60	27,00	4.50	0.31
	98,64	5.00	30,00	4,40	0,36
	101,69	4,70	34,00	4.30	0.37
	104,72	6,50	26,00	4.80	0,37
	107,76	5,40	20,00	4,60	0.36
	110.80	5,20	16,00	4.10	Ø.34
	113,84	4.80	12,00	4.60	0.30
	116.88	4,30	6,00	4.80	0.28
	119,92	1.77	5,37	4,80	0.28
	122,96	0.84	5,13	4,80	0,28
	126.00	0.50	5.05	4.80	0.28
	129.04	0.37	5,02	4,80	0.28
MEAN	VALUES				
FLOW	WEIGHTED	5,77	33,65	8,02	0.76

## CUMULATIVE PARAMETER LOADINGS

T(MINS)	Q(CU,M)	SS(KG)	DS(KG)	CA(KG)
1,195	0,574	0,102	0,128	0,915
4.400	2.218	0.381	0.484	0.057
7,440	6,362	0,946	1.357	0.152
10.480	13,981	1.786	2,956	0.317
13,520	22,555	2,690	4.774	0.484
16,560	30.191	3.452	6.128	0.615
19.600	37.092	3.972	7,249	0.741
22,640	43,389	4.371	8.446	0.845
25,680	48,931	4.668	9.427	0.906
28,720	54,576	4.930	10.324	0,966
31.760	59,985	5.163	11.126	1.021
34,800	64.858	5,363	11.751	1.078
37,840	69.889	5,569	12.304	1.137
40.880	76.502	5.862	13.204	1.200
43.920	85.734	6.337	13.984	1.278
46,960	97.226	7.052	15,199	1.363
50,000	110,825	7,920	16,644	1,457

## STORM RUNOFF DATA (a) 400 GS 401 GB 409 GS 403 GB 403 GB 403 GB 403 GB 403 GB

STORM OF THE 15TH JUNE 1978 encommente a commente a commente a commente a commente a commente commente a comme

	T(MINS)	Q(CU,M)	SS(KG)	DS(KG)	CA(KG)
	53,040	122,596	8,588	18,205	1,537
	56,080	131,033	8,993	19,406	1,591
	59,120	139,724	9,349	20,396	1.644
	62,160	152.047	9,795	21,776	1.719
	65,200	166,859	10,507	23,378	1.810
	68,240	178,702	11,154	24,651	1.884
	71 . 280	188.596	11,609	25,981	1,943
	74.320	197,580	11,942	27,220	2,002
	77,360	205,263	12.298	28,371	2,060
	80.400	212,299	12,630	29,452	2,119
	83,490	218,635	12,776	30,125	$2_{R}149$
	86,480	224,613	12,908	30,553	2,190
	63°550	239,125	13,015	30.748	2,228
	92,500	235,458	13,109	30,850	2.262
	00 CVU 20000	243,207	13,400	. 30,966	2,310
	101 COM 200030	2020200	13,870	31,120	2,309
	101000	2020103	14.320	31.344	2.432
	1021760	210,310 376 70A	14 1/07	31,30/	2:480
	110 800	201 517	14.760	31 1/21	
	113 840	2018-21/ 205 201	140334	34 074	5 5 5 5 5
	116.880	288.455	14 982	31 0/1	6 1 J / J 5 K06
	110,920	200,300	11 001	22 000	21220
	122.960	291.539	14.997	32 031	28911
	126.000	291.842	14,998	32.039	2 610
	129,040	291,953	14,999	32,942	2,620
mon t	10501000	304 073	4 4 000	20 040	
TOTAL	TOADINGS	2730913	14,999	32,042	2.620
	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	1,195	0.009	0,076	Ø,018	0,301
	4,400	0.031	0,284	0,064	0.003
	7.440	0.071	0.700	0.156	ØBCIM
	10,480	0.146	1,309	0.331	0,021
	13,520	0.224	1,930	0,525	0,033
	16,560	0.293	2,418	0.661	Ø . @ 4 1
	19,600	Ø , 355	2,766	0,751	0,255
	22,648	0.409	3,081	0.832	0,264
	25,680	0.452	3,319	0.900	0.072
	20,720	0 489	3,499	0.967	0.079

STORM OF THE 15TH JUNE 1978 

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T (	MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
31	. 768	0.524	3,651	1,030	0,085
34	. 800	Ø.,557	3,782	1,087	0,091
37	.840	0.592	3,917	1.146	0,096
42	6880	0.637	4.124	1.214	0.102
43	,920	0,697	4,458	1,292	0.109
46	.960	0,764	4.918	1.376	0.116
50	,000	0,838	5,462	1,468	P. 125
53	.040	Ø,899	5,911	1,543	0,132
56	080	0.938	6,208	1.594	0.138
59	a 120	0,976	6,446	1.641	0.144
62	a 160	1.030	6 e 751	1.702	0.152
65	,200	1.104	7.172	1.773	0,162
68	.240	1.166	7,493	1.828	0.170
. 71	,280	1.212	7,675	1.872	0,176
74	. 320	1.249	7.826	. 1,913	0.181
77	,360	1.277	7,980	1.949	0.185
. 82	1.400	1.302	8,124	1.982	0,189
83	.440	1.332	8.239	2.012	0,192
86	.480	1.361	8,323	2.039	0.195
89	,520	1.378	8.386	2,064	0.198
92	a 560	1.397	8.474	2,088	0.201
95	600	1,436	8,669	2,122	0,204
98	.640	1,488	8,946	2,166	0,207
101	0880	1.532	9,238	2,205	0,210
104	,720	1,578	9,486	2.243	0,213
107	,760	1.616	9,635	2.273	0.216
112	,800	1.642	9,723	2,294	0 8 2 1 7
113	.840	1,660	9,775	2,310	0.219
116	.880	$1_{e}675$	9,804	2,325	0,220
119	1,920	1,683	9,817	2,336	Ø.220
122	.960	1.684	9,822	2.340	0,220
126	0000	1.684	9,823	2,341	0.221
129	, 040	1 e 6 8 4	9,824	2,342	0,221
TOTAL LOA	DINGS	1.684	9,824	2.342	0.221

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STORM OF THE 29TH JUNE 1978

TIME SAMPLING COMMENCED Ø,19 SAMPLING DURATION(MINS) 32,79 DOSER CONCENTRATION(G/L) 40,50 DOSE RATE(ML/MIN) 1,03

# MEASURED PARAMETER CONCENTRATIONS

	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
	1,19	8,00	40,00	120.00	16,00
	4,40	10.36	33.00	102.00	13.20
	7.44	13.25	28,00	56,00	11.60
	10.48	21.29	27.00	96.00	12.20
	13.52	30.52	21.00	28.00	9.40
	16,56	24,42	23.00	28.90	9.40
	19.60	25.16	12.00	36.00	9.70
	22,64	9.23	6.30	36.00	9.70
	25.68	3.38	4.21	36.00	9.70
	28,72	1.24	3.44	36.00	9.70
	31,76	0,45	3,16	36,00	9,70
MEAN	VALUES				
FLOW	WEIGHTED	13,61	21.62	51.41	10.71

T(MINS) BOD(MG/L) COD(MG/L) NO3(MG/L) NH4(MG/L)

. . .

	1,19	9,70	98,30	10,00	0,70
	4.40	8,20	102,70	7.42	0,62
	7.44	6,50	107.00	5.40	0,58
	10,48	4,79	94,00	5.70	0.59
	13,52	6,20	86,00	4,60	0.54
	16,56	6.50	99,00	4,30	0,53
	1,9,60	4,70	79,00	3,80	0,53
	22,64	1,91	32,14	3,80	0.53
	25,68	6 * 8 0	14,95	3,80	0,53
	28,72	0,52	8,65	3,80	0,53
	31,76	Ø. 38	6.34	3,80	0,53
MEAN	VALUES				
FLOW	WEIGHTED	5,65	85,61	5,05	0,56

# STORM OF THE 29TH JUNE 1978

(CONTINUED)

# CUMULATIVE PARAMETER LOADINGS

	T(MINS)	Q(CU <sub>s</sub> M)	SS(KG)	DS(KG)	CA(KG)
	1,195	0.574	0,023	0,069	0,009
	4.400	2.327	0,087	0,263	0.035
	7,440	4.480	0,152	0,429	0,061
	10,480	7,629	Ø,238	0,678	0.099
	13,520	12,355	0,350	Ø,952	0,149
	16,560	17.366	0,460	1,073	0,196
	19,600	21.887	0,539	1,200	0,239
	22,640	25,023	0,570	1,313	0.270
	25,680	$26_{n}173$	0.577	1,355	0,281
	28,720	26,595	0,578	1,370	0.285
	31,760	26,749	Ø <b>.</b> 579	1.375	Ø,287
TOTAL	LÖADINGS	26.777	Ø,579	1.376	0,287
	T(MINS)	BOD(KG)	COD(KG)	N03(KG)	NH4(KG)
	1,195	0,006	0,056	0,006	0.000
	4.400	0.021	0.233	0,021	0.002
	7.440	0.037	0,459	0.035	0.003
	10,480	0.054	0.772	0,052	0.005
	13,520	0.080	1,195	0.076	0.007
	16,56Ø	0.112	1,656	0.098	0,010
	19,600	0.137	2,058	Ø.117	Ø . Ø 1 2
	22,640	0.149	2,255	0,129	0.014
	25,680	0.151	2,286	0,133	0.015
	28,720	0,151	2,291	0,135	0,015
	31,760	0,151	2.292	0.135	0,015
TOTAL	LOADINGS	0.151	2.292	0,135	0.015

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# STORM OF THE 1ST JULY 1978

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TIME SAMPLING COMMENCED 12.11 SAMPLING DURATION(MINS) 29,75 DOSER CONCENTRATION(G/L) 40,50 DOSE RATE(ML/MIN) 0.38

## MEASURED PARAMETER CONCENTRATIONS

	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
	1,19	8,00	39,00	56,00	15,70
	4,40	8,99	60,00	52,00	11,70
	7.44	9,44	27,00	64,00	9,40
	10,48	29,34	20,00	80,00	13.30
	13,52	20,74	25,00	88,00	12,20
	16,56	10.23	21,00	52,00	11.10
	19,60	3,75	9,60	52,00	11,10
	22,64	1,38	5,42	52,00	11,10
	25,68	0,50	3,89	52,00	11.10
	28,72	0.19	3,33	52,00	11.19
MEAN	VALUES				
FLOW	WEIGHTED	9.41	26,71	70,03	12,23
	T(MINS)	BOD(MG/L)	COD(MG/L)	) NO3(MG/L)	NH4(MG/L)
	1 10	3 010	24 60	7 50	0.34
	A AO	$\Omega \Delta \alpha$	18 00 18 00	7 90	12 D 3
	7 44	2 80	51 00	6 70	0 20
	10.48	2.40	28 00	5 70	(A 21
	13.52	9.20	53.00	6 00	G 75
	16.56	3.20	29.00	6 10	0,23
	19.60	1.36	13.80	5 10	a 23
	22.64	ê 69	8,23	5 10	0 23
	25.68	0.44	6.18	6 10	0 23
	28,72	0,35	5,43	5.10	0,23
MEAN	VALUES				
FLOW	WEIGHTED	2.59	37.77	6.08	0.24

## STORM OF THE 1ST JULY 1978

(CONTINUED)

# CUMULATIVE PARAMETER LOADINGS

	T(MINS)	Q(CU <sub>e</sub> M)	SS(KG)	DS(KG)	CA(KG)
	1,195	0,574	0.022	0,032	0,009
	4.400	2.202	0,103	0.120	0,031
	7.440	3,882	0,175	0.218	0.049
	10,480	7.419	0,254	0,482	0,092
	13,520	11,986	0,356	0.864	0,150
	16,560	14,810	0.422	1.073	0,183
	19,600	16.086	0.444	1,139	0.198
	22,640	16,553	0.448	1,164	0,203
	25,680	16,725	0,448	1.173	0,205
	28,720	16.788	0.449	1,176	0,205
TOTAL	LOADINGS	16,799	0.449	1.177	0,205
	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	1,195	0,002	0.020	0.004	0.000
	4.403	0.011	0,086	0.016	0,001
	7.440	0.020	0.169	0.028	0.001
	10.480	0.029	0,295	0.049	0,002
	13,520	0,036	Ø.473	0.076	0.003
	16,560	0.040	Ø.597	0.092	0.004
	19,600	0.043	0.627	0,099	0.004
	22,640	0.043	0,633	0.101	0.004
	25,680	0.043	0.634	0.102	0,004
	28,720	0.044	0.634	0,102	0,004
TOTAL	LOADINGS	0.044	0,634	0,102	ø,cea

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STORM OF THE 3RD JULY 1978

TIME SAMPLING COMMENCED 14.27 SAMPLING DURATION(MINS) 41.91 DOSER CONCELTRATION(G/L) 40.50 DOSE RATE(ML/MIN) 6.11

## MEASURED PARAMETER CONCENTRATIONS

	T(MINS)	Q(L/S)	SS(MG/L)	DS(AG/L)	CA(MG/L)
	1.19	8,00	201.00	232.03	17.60
	4,40	9,43	158,02	72.20	12,82
	7.44	28.45	123,00	98,50	13,51
	10.48	35.78	132.00	82.40	8,00
	13,52	22,59	92.00	102.40	8,30
	16.56	18.67	87,02	148.00	8,86
	19,60	13,85	72.00	156.00	11.40
	22,64	9,13	33.00	160.00	11.00
	25.68	21,46	57.00	168,82	8.54
	28.72	21.46	40.00	128.00	7.20
	31.76	7.87	16,57	128.00	7.20
	34.80	2.89	7,98	128.90	7.88
	37.84	1.26	4.82	128.00	7.00
	40.88	0.39	3,67	128.00	7.22
MEAN	VALUES				
FLOW	WETGHTED	14.55	91,81	123.74	9.77

BOD(MG/L) COD(MG/L) HC3(MG/L) T(MINS) 2114 ("G/L) 6.79 7.50 89,00 0.27 1,19 72.20 4.40 8,61 6,03 1.16 7.44 8.63 98,50 5,50 0.28 7.30 10.19 10.48 82.40 4.30 3,20 4,28 13.52 102,44 P. 36 4.53 1.23 16.56 74.10 4.60 0.23 19,60 4.30 49.80 4,10 C . 24 22.64 4,73 22.42 4.62 25,68 4,90 37.10 3.50 01,26 28.72 3,80 2.60 28,80 F . 17 31.76 1,14 13.73 3,80 13,17 34.83 0.61 8,20 3.84 0,17 (\* "41 37.84 6.17 3,83 0.17

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OF THE 3R	8.D. JULY 1978 1. CO XI 66 50 50 100 100 100 100 100 100 100 100 1	ang ang kiti kiti	(CONTINUED)	)
T(MINS)	BOD(MGZL)	CODCHGY	L) NO3(MG/L)	) 584(MG/L)
40.88	0.34	5,43	3.8%	0,17
ALUES EIGHIED	5,33	65,72	4.58	0.24
TTVE PARAS M# AB 49 49 49 48 40 49 49 49 49	KETTER DOADIN	G S Ra ne		
T(MINS)	0(CU.M)	SS(KG)	DS(KG) (	CA(KG)
1.195 4.403 7.44A 10.482 13.523 16.563 19.672 22.643 25.689 28.723 31.769 34.800 37.840 40.880	0, 574 2, 242 5, 697 11, 554 16, 877 20, 640 23, 645 25, 701 28, 492 32, 407 35, 283 36, 555 36, 579	0,115 0,415 v,879 1.628 2,241 2,578 2,816 2,932 3,667 3,256 3,342 3,355 3,358 3,358 3,358	0.133 0.386 0.696 1.223 1.706 2.172 2.621 2.952 3.413 3.992 4.335 4.460 4.523 4.523	0.010 0.035 0.081 0.143 0.146 0.218 0.218 0.248 0.271 0.298 0.328 0.328 0.347 0.354 0.357
T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NHA(KG)
1.195 4.400 7.440 10.480 13.520 16.560 19.688 22.64 25.688 28.720 31.764	0,604 0,018 0,047 6.047 6.094 0.123 6.138 0.151 0.151 0.174 0.174 0.198 0.194	0,051 N=185 N=496 1,122 1=506 1,841 2,328 2,103 2,103 2,107 2,326 2,399	0.004 0.014 0.034 0.062 0.105 0.102 0.115 0.120 0.135 0.135 0.149 0.149 0.149	4,000 0,000000
	DF THE 3F T(MINS) 40.88 ALUES EIGHIED TIVE PARAN T(MINS) 1.195 4.403 7.443 10.482 13.523 16.563 19.602 22.643 25.680 28.720 31.760 34.300 37.840 40.880 37.840 40.880 LOADINGS T(MINS) 1.195 4.400 7.440 37.840 40.880 LOADINGS T(MINS) 1.195 4.400 37.840 40.880 37.840 40.880 37.840 40.880 13.523 1.760 31.760 31.760 31.760 31.760 31.760 31.760 31.760 31.760	DF THE 3RD JULY 1978   T(MINS) BOD(MG/L)   40.88 0.34   ALUES 5.33   TIVE PARAMETER LOADIN   T(MINS) Q(CU.M)   1.195 0.574   4.403 2.242   7.443 5.697   10.482 11.554   13.520 16.877   16.560 20.640   19.662 23.605   22.640 25.701   25.680 28.492   28.720 32.407   31.766 35.283   34.800 36.555   LOADINGS 36.579   T(MINS) BOD(KG)   1.195 0.604   37.840 36.555   LOADINGS 36.579   T(MINS) BOD(KG)   1.195 0.604   37.840 36.579   T(MINS) BOD(KG)   1.352.0 12.38   13.52.0 12.38   13.52.0 12.38   19.602 0.151   22.64 0.151   22.64	DF THE 3RD JULY 1978   T(MINS) BOD(MG/L) COD(MG/L)   40.88 0.34 5.43   ALUES EIGHIED 5.33 65.72   TIVE PARAMETER LOADINGS   T(MINS) Q(CU.M) SS(KG)   1.195 0.574 0.115   4.402 2.242 0.415   7.443 5.697 0.879   16.560 20.640 2.578   19.662 23.695 2.816   22.642 2.5781 2.932   25.680 28.492 3.667   28.720 32.447 3.256   31.766 35.283 3.342   34.340 6.464 3.355   37.840 36.423 3.358   LOADINGS 36.579 3.358   LOADINGS 36.579 3.358   20.447 0.496 0.671   40.880 36.555 3.358   LOADINGS 36.579 3.358   2.448 6.647 0.496   19.662 6.6	DF THE 3RD JULY 1978   T(MINS) BOD(MG/L) COD(NG/L) NC3(MG/L)   40.88 0.34 5.43 3.80   ALUES EIGHIED 5.33 65.72 4.50   TIVE PARAMETER LOADINGS DS(KG) DS(KG) 0.5(KG) 0.5(KG)   1.195 0.574 0.115 0.133 4.403 2.242 0.415 0.386   7.440 2.697 V.879 0.696 1.433   4.403 2.242 0.415 0.386 7.443 5.697 V.879 0.696 0.133   1.352.1 15.54 1.628 1.223 1.352.1 1.678 2.172   1.9672 2.3695 2.816 2.621 2.952 2.5680 2.929 2.952   25.680 28.492 3.667 3.413 28.728 3.342 4.335   24.728 3.2407 3.256 3.992 3.358 4.523   LOADINGS 36.579 3.358 4.523   LOADINGS 36.579 3.358 4.526

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STORS OF THE 3RD JUL		D JULY 19"	1978 		TINPED)	
	T(MINS)	BOD(KG)	COD(KG)	N03(KG)	NH4(KG)	
	37.840 40.860	0,195 D.195	2,403 2,404	0,164 0,165	0,009 0,009	
TOTAL	LOADINGS	0,195	2,404	0.165	(, (0)	

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## STORM OF THE 19TH JULY 1978

TIME SAMPLING COMMENCED 19.58 SAMPLING DURATION(MINS) 29.75 DOSER CONCENTRATION(G/L) 43,60 DOSE RATE(ML/MIN) 0,99

MEASURED PARAMETER CONCENTRATIONS

	T(MINS)	0(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
	1,19	8,90	44,00	170,00	23.60
	4.40	12,84	64,00	165.00	20,20
	7.44	10.83	40.00	164.00	18,60
	10.48	12,57	25,00	145,00	16.00
	13,52	23,51	22,00	110,00	14.40
	16.56	21.63	24,00	122.02	15.40
	19,60	7.93	10.70	122.00	15.40
	22,64	2.91	5.82	122.98	15.40
	25,68	1.07	4.04	122.00	15,40
	28,72	0,39	3,38	122,00	15.49
MEAN	VALUES				
FLOW	WEIGHTED	10.34	29,87	135,98	16,84
	T(MINS)	BOD(MG/L)	COD(MG/L;	) NÖB(MG/L)	NH4(⊮G/L)
	1 10	14 20	66 70	11 80	1 30
	A 40	10 80	20 80	11 00	1 70
	7.44	7.30	39.90	10.70	2.00
	10.48	8.20	28.90	8.40	1.10
	13.52	6.50	28.90	8.20	0.95
	16.56	6.50	27.90	8,10	0.84
	19.60	2.57	13.40	8,10	2.84
	22.64	1.13	8,08	8.10	0.84
	25.68	0.61	6.13	8.10	0.84
	28,72	0,41	5,41	8,10	2.84
MEAN	VALUES				
FLOW	WEIGRTED	7.35	31.88	9.09	1.17

STORM OF THE 19TH JULY 1978 and the sea on the

#### CUMULATIVE PARAMETER LOADINGS

	T(MINS)	Q(CU.M)	SS(KG)	DS(KG)	CA(KG)
	1,195	0,574	0,025	0,098	0,014
	4,400	2,553	0,134	0.429	0.857
	7.440	4.711	0,248	0.784	0,098
	10.480	6.846	0.317	1,112	0,135
	13.520	10,136	0.393	1.528	0,185
	16,560	14,252	0.487	1,997	0,246
	19,600	16,948	0.540	2,326	0,288
	22,640	17,936	0.549	2,446	0,303
	25,680	18,299	0.551	2,491	0.308
	28,729	18,432	0,551	2,507	0.310
TOTAL	LOVDINGS	18,456	0,551	2.519	0.311
	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	1,195	0,008	0.038	0.007	0.001
	4.400	0.033	0.141	0.029	0.204
	7.440	0.052	0.227	0.053	0.008
	10.480	0,069	0.300	0,073	0.011
	13,520	0.092	0.395	0.100	0.014
	16,560	0.119	0.512	0.134	0.018
	19,600	0.133	0.573	0.156	0.020
	22,640	0.135	0,585	0.164	0,021
	25,680	0.136	Ø.588	0,167	0.021
	28,720	Ø.136	0,588	Ø s 168	0,022
				-	
TOTAL	LOADINGS	0.136	0,588	0,168	0,022

#### STORM OF THE 23RD JULY 1978

TIME SAMPLING COMMENCED 19,53 17,59 SAMPLING DURATION(MINS) DOSER CONCENTRATION(G/L) 43,60 DOSE RATE(ML/MIN) 0,95

#### MEASURED PARAMETER CONCENTRATIONS

10,489

13,520

16,560

TOTAL LOADINGS

	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
	1,19 4,40 7,44	8,00 15,16 5,56	36,00 68,00 25,84	108,00 107,00 107,00	0,00 0,00 0,00
	10,48 13,52 16,56	2,04 0,75 0,27	11,74 6,21 4,18	107.00 107.00 107.00	ଡ <b>୍</b> ଡମ ଡ୍ଡେମ ଡ୍ଡେମ
MEAN FLOW	VALUES WEIGHTED	5,41	45 <b>,</b> 58	107,26	0.00
	T(MINS)	BOD(NG/L)	COD (MG/I	.) NO3(MG/	L) NH4("G/L)
	1,19	11,00	25,40	5,80	0.44
	4 g 4 Ø	2 2 0 0	29,30	5.63 5.43	0,40
	10 10	3 8 0 Z	13,31	5,00	0 40
	13 53	a 77	6 20	5 00	0 AC
	16,56	0.47	5,44	5,03	0,40
MEAN	VALUES				
FLOW	WEIGHTED	8,11	23,02	5.21	10.41
CUMUI		ETER LOADIN	1 G S		
	T(MINS)	0(CU,M)	SS(KG)	DS(KG)	CA(KG)
	1,195	0,574	0,021	0,062	8 * 6 (3 1)
	4,403	2.765	0.1417	0,297	8°603
	7.440	4,655	0.242	0,500	10 p 1 C *

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STORM OF THE 2	3RD JULY 19	978	CONTINUE	1 1
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T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
1,195	0,006	0,015	0,003	0,000
4.400	0.029	0,075	0,015	0.001
7.440	0,044	0.121	Ø. C24	0,002
10,480	0,946	0.129	0,028	0,072
13,520	0.046	0.131	0,029	0.002
16,560	0.046	0,131	0.030	0,002

	100200	0 C D A D	A8791	<i>N 8 6 3 6</i>	510 X C V
TOTAL	LOADINGS	0.046	0,131	0.030	0,002

### STORM OF THE 26TH JULY 1978

TIME SAMPLING COMMENCED 5,13 SAMPLING DURATION(MINS) 44,95 DOSER CONCENTRATION(G/L) 43,60 DOSE RATE(ML/MIN) 0,95

# MEASURED PARAMETER CONCENTRATIONS

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	T(MINS)	Q(L/S)	SS(MG/U)	DS(MG/L)	CA(MGZL)	
	1,19	8,00	23,00	120,00	0,23	
	4.40	10,58	25,00	112,00	10,00	
	7.44	25,49	25.00	104.00	0.00	
	10.48	23,80	24.00	97.00	0.00	
	13,52	18,83	22.00	104.00	0,00	
	16,56	17,85	15,00	91.00	0.00	
	19,60	32,36	14.00	75.00	0.00	
	22.64	35,40	3.00	73.03	0.00	
	25,68	23.01	10.00	73.00	0.00	
	28,72	20,29	10.03	75,00	3,00	
	31,76	16,57	13,00	82.00	0.03	
	34,80	6,08	6,67	82.00	0.0.1	
	37,84	2,23	4,34	82.00	10,00	
	40,88	0,82	3,49	82.00	0,00	
	43,92	0,30	3,18	82.00	0.49	
N	VALUES			-		
넦	WEIGHTED	16 e 30	14,91	87,14	0.00	
	T(MINS)	BOD(MG/L)	COD(MGZL)	NO3(MG/L)	NH4(MG/L)	
	1,19	11.80	38.20	10,90	0.56	
	4,40	11.60	31,22	18.18	0.41	
	7,41	9,90	27.30	9,20	13,41	
	10,48	8,69	36.10	8.80	1 . 41	
	13.52	7.10	23.43	8,33	0,50	
	16,56	6,90	11,70	7.50	2,46	
	19,60	7,30	17,62	6,20	0.30	
	22,64	2.29	15,00	5.40	*,28	
	25.68	4,70	15,00	5,30	C , 28	
	28.72	4,93	6.8%	4,60	P . 24	
	31,76	4.30	8,88	4.78	1,22	
	34.89	1.77	6.10	1.70	8.22	

# STORM OF THE 26TH JULY 1978

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	T(MINS)	BOD(MG/L)	COD(MG/L)	N03(MG/L)	NH4(MG/L)
	37,84 40,88 43,92	0,84 0,50 0,37	5,40 5,15 5,05	4,70 4,70 4,70	0,22 0,22 0,22
MEAN FLOW	VALUES WEIGHTED	6,36	18,77	6,86	0,35

# CUMULATIVE PARAMETER LOADINGS

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	T(MINS)	Q(CU,M)	SS(KG)	DS(KG)	CA(KG)
	1,195	0,574	0.013	0,769	0.007
	4.400	2,347	0,056	8.274	0.000
	7.440	5,636	0.138	0.626	0.000
	10,480	10,132	0,248	1.078	0.000
	13,520	14,020	0,338	1.468	0.000
	16,560	17,365	0.400	1.794	0 000
	19,600	21,944	0.466	2.167	0.880
	22,640	28.124	0,518	2.624	0.000
	25,680	33,451	0.549	3.013	0.202
	28,720	37,393	0.589	3,305	0.000
	31,760	40,746	0,627	3,567	0.000
	34,800	42,811	0,649	3,736	0,000
	37,840	43,569	0,654	3,798	2,000
	40,880	43,846	0,655	3,821	0.032
	43,920	43,948	Ø,655	3,830	6.000
ΤΟΤΑΙ,	LOADINGS	43,967	0,655	3,831	0.000
	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	1,195	0,007	2.017	0006	0.000
	4,490	0,028	0.072	0.025	0.001
	7,440	1,062	:. 166	0,056	8.603
	19,480	0.104	0,348	2,297	8.824
	13,520	0.134	0,426	0,130	0,006
	16.560	C. 158	0,485	0,156	0,000
	19,630	0,191	0,555	0,187	0.209
	22,643	0.219	2,655	2,223	0.211
	25.680	0.237	2,735	4,252	13,013

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	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	28,720	0,256	0,779	0.271	0,014
	31.760	0.271	0,803	0.287	0.014
	34.800	0.278	0.819	0,296	0.015
	37.840	0.279	0.823	0,300	0.915
	40,880	0.280	0,825	0.301	0,015
	43,920	0,280	Ø,825	0,302	0,015
TOTAL	LOADINGS	0,280	0,825	0,302	0,015

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### STORM OF THE 27TH JULY 1978

TIME SAMPLING COMMENCED 11,48 SAMPLING DURATION(MINS) 23,67 DOSER CONCENTRATION(G/L) 43,60 DOSE RATE(ML/MIN) 1,03

#### MEASURED PARAMETER CONCENTRATIONS

	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
	1,19 4,40 7,44	8,00 10,78	882,00 750,00 250,00	164.00 152.00	0,00 0,00 0,00
	10,43 13,52	15,41 5,65	160,00 60,57	136,00 136,00	0,00 0,00 0,00
	16,56 19,60 22,64	2,07 0,76 0,28	24.11 10.74 5.84	136,00 136,00 136,00	0,00 0,00 0,00
MEAN Flow	VALUES WEIGHTED	7,75	372.3ð	138,05	0.00

#### T(MINS) BOD(MG/L) COD(MG/L) NO3(MG/L) NH4(MG/L)

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	1,19	0,00	504,00	5,40	0,00
	4.48	0,00	446,00	5,20	0.00
	7.44	0.00	161,03	6,10	0,00
	10,48	0,00	138.00	5,40	0,00
	13,52	0.00	53,77	5,40	0.00
	16,56	0.00	22,88	5,40	0.00
	19,60	0.00	11,56	5.40	0,00
	22,64	0,00	7,40	5,40	0,00
MEAN	VALUES				
FLOW	WEIGHTED	0.00	233,47	5,56	0.00

#### CUMULATIVE PARAMETER LOADINGS

T(MINS)	Q(CU,M)	SS(KG)	DS(KG)	CA(KG)
1.195	c.574	0.506	0.094	(* . 300
4.400 7.440	2,366	1.962	0,311	0,000

S). OR M 100 400 400 400 400 400 400 400	ЭКЧ ОГ ТНЕ 27ТН ЈИЦУ 1978 пасти власка со са со по со са на на на кој са на			(CONTINUED)	
	T(MINS)	Q(CU,M)	SS(KG)	DS(KG)	CA(KG)
	10.480	8,006	3,814	1.111	0,000
	13.520	9,927	4,256	1.373	0,000
	16.560	19,631	4.090	1,468	0.000
	19,600	10.890	4, 895	1,504	0.000
	22.640	10,984	4,096	1,516	0,000
TOTAL	LOADINGS	11.001	4.095	1,519	0,000
	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	1,195	a.000	0,289	0,003	0,000
	4.400	0.000	1,138	0,013	0,000
	7.440	0.000	1,868	0,028	0.000
	10,482	0.000	2.322	0,045	0,000
	13,520	2.000	2,532	0.055	0,000
	16,562	0.000	2,562	0.359	0.000
	19,600	0.000	2.567	0.061	0.000
	22.640	<b>୬</b> ,ମରଡ	2,568	0.961	ଡ, ମହର
ТОТАЦ	LOADINGS	0,000	2,568	0,061	ଡ <b>୍</b> ଟଡଡ

#### No 12 CA 12 Ca 12 Ca 12 CA 12 CA 13 CA 13 CA 13 CA 10 CA 10 CA 10 CA

# STORM OF THE 30TH JULY 1978

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TIME SAMPLING COMMENCED 7,00 SAMPLING DURATION(MINS) 17,59 DOSER CONCENTRATION(G/L) 43,60 DOSE RATE(ML/MIN) 1,03

#### MEASURED PARAMETER CONCENTRATIONS

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	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
	1,19	8,00	42,00	116,00	0.00
	4.40	11,41	37.00	108.00	0,00
	7.44	4,18	15,47	188,80	0,00
	10.48	1.53	7.57	108,00	0,00
	13.52	0.56	4.68	108,00	0.00
	16,56	0,21	3,61	108,00	Ø.Or
MEAN	VALUES				
FLOW	WEIGHTED	4,40	31,72	110,47	0,00
	T(MINS)	BOD(MG/L)	COD(MG/L	) NO3(MG/L)	NH4(MG/L)
	1.19	0.00	46.70	6.50	0.03
	4.40	0.00	44.00	6.67	0.09
	7.44	0.00	19.30	6.60	0.00
	10.48	0.00	10.24	6.60	0,00
	13.52	0.00	6,92	6.60	0.0.3
	16 56	aaa	5 71	6.60	0.02

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MEAN	VALUES				
FLOW	WEIGHTED	0.00	37.00	6,57	0,00

#### CUMULATIVE PARAMETER LOADINGS

	T(MINS)	Q(CU,H)	SS(KG)	PS(KG)	CA(KG)
	1,195	0.574	0.024	0.067	a.cee
	4. 117,7	2,423	0.097	1.273	2.000
	7.440	3,846	0,139	1,427	0,000
	10,480	4.367	1.146	0,483	6,864
	13.520	4,558	1. 147	0,504	0.010
	16.560	4,629	13,147	0.511	6,000
TOTAL	LOADINGS	4,641	0.147	0.513	() * () (**)

STORM OF THE 3	MTH JULY 19	) 7 8 9 69 13 69 69 69	(CONTINUE)	))
T(MINS)	BOD(KG)	COD(KG)	NQ3(KG)	EH4(KG)
1,195	Ø <u>,</u> 800	0,027	0,004	0,000
4,400	0.000	0.110	0.016	6.000
7,448	0.000	0,161	0,025	0.000
10.480	0.000	0.169	0,029	C. C03
13,520	0.000	0.171	0,030	0.000
16,560	0,000	0.172	0.030	<b>ព្</b> ំ ចំផង
TOTAL LOADINGS	0,000	0,172	0.030	0,000

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## STORM OF THE 30TH JULY 1978

TIME SAMPLING COMMENCED 6,00 SAMPLING DURATION(MINS) 23,67 DOSER CONCENTRATION(G/L) 43,60 DOSE RATE(ML/MIN) 1,03

### MEASURED PARAMETER CONCENTRATIONS

	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
	1.19	8,00	92,00	200.00	0,00
	4.40	12.27	279,00	188,00	0,00
	7.44	19,54	144.00	177,00	0.00
	16.48	17,66	75,00	174.00	0.00
	13,52	6,48	29.40	174.00	0.00
	16.56	2.38	12.68	174.00	0.00
	19.60	0,87	6,55	174.00	0,00
	22,64	0,32	4.30	174,00	0,00
MEAN	VALUES				
FLOW	WEIGHTED	8,61	125,93	180,53	0,00

7	(MINS) BO	D(MG/L)	COD(MG/L)	NO3(MG/L)	NH4(MG/L)
	1.19 0	.00 1	43.00	12,50	0,00
	4.40 0	.00 1	47.03	10.00	0,00
	7.44 0	.00 1	107.00	9,60	0.00
1.	0.48 0	.00	94.00	9,60	0.00
1	3,52 0	.00	37,64	9,60	0,00
. 1	6,56 Ø	. 80	16,97	9,60	0,00
1	9,60 0	.00	9,39	9,60	0,00
2	2.64 0	,00	6,61	9,60	Ø. 00
MEAN VALU	JES				
FLOW WEIG	HTED Ø	,00	192,83	10,02	0.80

#### CUMULATIVE PARAMETER LOADINGS

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T(MINS)	0(CU,S)	SS(KG)	DS(KG)	CA(KG)
1,195	0.574	0 <sub>e</sub> 115 3	0.115	0,000
4.400	2.501	0,425	$\emptyset_{R}$ 488	6 6 10 12
7,440	5.402	1. 111	1,015	N. 180

STORM OF	THE 30		)"/ (} 54 63 67 69 89	(CONTINUE	D)
T	(MINS)	Q(CU,M)	SS(KG)	DS(KG)	CA(KG)
1	0,480	8,796	1.386	1,610	0.000
1	3,520	10,997	1.517	1,993	0,020
	6,560	11.805	1,536	2.134	0,000
1	9,600	12,101	1.539	2,185	0,000
2	2,640	12,209	1 · 540	2,204	0.000
TOTAL LC	ADINGS	12,229	1.540	2,208	0,000
Ţ	(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	1,195	0.000	0,082	0,007	0,000
	4,400	0,000	0,362	0.029	0,000
	7.440	0.000	0,721	0,057	0,000
1	0.480	0.000	1,063	0.790	0,000
1	3,520	0,000	1,223	0.111	0.100
1	6,569	0.000	1,252	0,118	0.000
1	9,600	0.000	1,256	0,121	0,000
2	2.640	4 6 0 0 Q	1,257	0.122	9,000
TOTAL LO	ADINGS	0,000	1,257	0.123	0,000

## STORM OF THE BOTH JULY 1978

1. 2

> TIME SAMPLING COMMENCED 8.00 SAMPLING DURATION(MINS) 114.87 DOSER CONCENTRATION(G/L) 43.60 DOSE RATE(ML/MIN) 1.03

NEASURED PARAMETER CONCENTRATIONS

T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
1,19	8,00	30.00	72,00	a , aa
4.40	9,81	39,00	100.00	0,00
7.44	11,32	24.00	140.00	0.00
10,48	13,75	22,00	116.00	0.00
13,52	19.02	59,00	112,00	0.99
16,56	26,29	323,90	88.00	0,00
19,60	66,70	279,00	76.90	0.00
22,64	61.22	138,00	72.08	0.00
25,68	84,32	138,00	52,00	0.00
28,72	106,40	171.00	72,00	0.00
31.76	93,10	128.02	44,00	0.00
34.80	101.57	72.00	27.00	0,00
37,84	93,10	59,00	60.00	0.00
40,88	59,59	52,00	60.00	0.00
43,92	27.93	44.00	76.00	0.00
46,96	18,24	28,00	87.00	0,00
50,00	13,97	21.00	100.00	0.00
53.04	14,90	24,00	124.00	0,00
56,08	25,54	29,00	116,00	0.00
59,12	44,69	31,00	72,00	0.00
62,16	34,38	40.00	60.00	0.00
65,20	20,79	28.00	60,00	0.00
68,24	14,65	18,00	87.00	0.00
71,28	12,95	26,00	116,00	0.00
74,32	59,59	36,00	120,00	0.00
77,36	48,05	49,00	56,00	0.20
80.40	33,10	31,00	60,00	0.00
83,14	23.04	29.03	72.00	0.00
86,48	22.01	21.00	74.00	0.00
89.52	21,80	11,00	68,06	0.00
92,56	31,92	15,00	80,00	0.00
95.60	38,86	12,00	80,00	0.00
98.66	21.02	11 133	<u>ຊ ຊັ ຜ ຊ</u>	0 20

STOR 101 101 101 101 101 101 101	б ОЕ ТНЕ <u>3</u> 6 г до ех са ео са ео Ад во ра еи ее	TH JULY 1	978 	(COWTINUED	)
	T(MINS)	Q(L/S)	SS(MG/L)	DS(MG/L)	CA(MG/L)
	101.68	11.71	5,93	88,00	0.00
	104,72	4,29	4.08	88.00	0.00
	107,76	1,57	3,39	88.00	0.00
	110,80	0,58	3,14	88,00	0,00
	113,84	0,21	3,05	88,00	0,00
M 6" A M	VALIDO				
FLOW	WEIGHTED	34,68	84,38	70,50	0.00
	T(MINS)	BOD(MGZ	L) COD(MG/L		
				a constraint and a second and	
	1.19	0,00	42,00	5,20	0.00
	4,40	0.00	58,10	5,30	0.00
	7.44	0.00	57,00	5,30	0,00
	10,48	0.00	56,20	5,30	0,00
	13,52	0.20	84,60	5,00	0.00
	16.56	0,00	254,00	5,50	0.00
	19,60	0,00	170.00	4.90	0,00
	22,64	0.00	178,00	4.18	0.00
	25,68	0.00	129.00	3.80	0,00
	28,12	0,00	156,00	3,70	0.00
	31,70	00,0	85,00	3,30	0,00
	34,80	0.00	75,00	2,80	0,99
	3/ 6 2 2	0,00	57,20	3,50	0,00
	40,00	0.00	48,60	3.20	0.00
	43,94	0,00	42,00	3.20	0,00
	40,30	0 00	31,00	3.30	0.00
		0,00	21,20	3,62	0.00
	23808 66 00	0,00	30,00	3 e 4 1)	0,00
	30 6 9 0 6 0 6 9	0,00	36,30	3,10	0,00
	69 16	0,00	10,20	3,80	0.00
	65 20	12 0 0 11 13 0 13	32,20	3,02	0.00
	60 24	0 00	40 00	3,00	0,00
	71 78	a aa	10,00	12 g 3 V/ A (3 (3	0,00
	74 30	6 60	30 <b>3</b> 0	4.00	N 9 8 10
	77 36	3 20	30 50	3 0 3	A Qu
	80.40	0,00	21 63	2 2 3	8 B 3 3
	93.44	0.00	R. TO	1 7 A	e a a a
	86.48	9.03	15.89	4 00	Pava DA 0.2
	89.52	0.00	13.70	A 010	G 433
	92.56	0,00	17.70	4 20	6 6.1
	95,69	BeWH	30, 50	5,21	0.00

STORM OF THE 30TH JULY 1978

	MIC MET MICH	DODAHAAAN	ADDONALS	NANCHAN	511) A C 51 C 15
	rturnos	BOD (MG/LJ	COD(MG/4)	NOSCACIDI	NH4 (MG/L
	98.64	0.00	18.10	4.40	0.00
	101.68	0.00	9,80	4,40	0.00
	104.72	0,00	6.76	4,41	0.00
	107.76	0.00	5,65	4.40	0,00
	110,80	0.00	5,24	4.40	0.00
	113,84	0,00	5,09	4,40	0,00
MEAN	VALUES				
FLOW	WEIGHTED	0,00	75,13	3,88	0,00

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# CUMULATIVE PARAMETER LOADINGS

T(MINS)	Q(CU,M)	SS(KG)	DS(KG)	CA(KG)
1,195	0,574	0,017	0.041	0,000
4,400	2,277	0,076	0.188	0.000
7,440	4,205	0.136	0.421	0,000
10,480	6,492	0,189	0.712	0,000
13,520	9.480	0,316	1.052	0.000
16,560	13,612	1.164	1.460	0,000
19,600	22,093	3,662	2.141	0,000
22,640	33,759	6.118	3,005	0.000
25,682	47,032	7,950	3,813	0,000
28,720	64,427	10,660	4,905	0,000
31,760	82,622	13,398	5,972	6.000
34,800	100,376	15,159	6,598	0,000
37,840	118,130	16,325	7,362	0.000
40,889	132,056	17,105	8,197	0,000
43,920	140.037	17,496	8.724	0,000
46,960	144,248	17,652	9,064	6,000
50,000	147.185	17,725	9,337	6.000
53,040	149,818	17.784	9,606	0,000
56,380	153,565	17,884	12.015	0.000
59,120	159,910	18,077	10,591	0,000
62.160	167.121	18,330	11,071	2,000
65,200	172,152	18,506	11.373	0,000
68,240	175,384	18,583	11,605	8,600
71,200	177 . 901	18,637	11,259	0.000
74,320	184.517	18,857	12.646	0,000
77,364	194.334	19,270	13,532	13,000
80.400	201.735	19.574	13,960	0.080

STORM		NTH JULY 1	978 ≈ ** ** ** ** ** **	(CONTINUE	D)
	T(MINS)	Q(CU,M)	SS(KG)	DS(KG)	CA(KG)
	83,440	206,855	19,728	14.294	0,000
	86,480	210,964	19,831	14,594	0,000
	89,520	214,960	19,895	14.878	0.000
	92,560	219,859	19,960	15,244	0,000
	95,600	226,315	20,047	15.760	0,000
	98,640	232,770	20,121	16,301	0,200
	101.680	236,749	20,158	16,651	0,000
	104.720	238,208	20,166	16.780	0,000
	107,760	238,743	22,168	16,827	6,000
	110,800	238,939	22,168	16.844	0,000
	113,840	239,011	20,169	16,850	0,000
TOTAL	LOADINGS	239,024	20,169	16.851	0,000
	T(MINS)	BOD(KG)	COD(KG)	NO3(KG)	NH4(KG)
	1,195	0,000	0.024	0.003	0.000
	4,400	0,000	0,110	0.012	0,000
	7.440	0,000	0.221	0.022	0.000
	10,480	0.000	0,350	0.034	0,000
	13,520	0.000	0,565	0,050	0,000
	16,560	0.000	1,302	0,071	0,000
	19,600	0.000	2,995	0,115	0.000
	22,640	0,000	5,024	0,157	0,000
	25,680	0.000	7,027	0.220	3,990
	28,720	0.000	9,524	0.285	0,000
	31,760	00000	11,745	0.349	0.960
	34,800	0,000	13,163	0.403	0,000
	37,848	0,000	14,341	0.458	A.000
	40,880	0,000	15,087	0.505	A, 090
	43,920	0.000	15,455	0,531	0,000,0
	46,960	0,000	15,612	2.545	6.663
	50,000	0.000	15,698	0,555	8,848
	53,240	0.009	15,774	0.564	0.000
	50,000	0,000	15,890	9.577	11.040
	59,120 69 460	0.000	16,033	0,601	0.000
	02,100	0.000	16,208	0,629	A.000
	00,200	(1) (1) (1) (1) (1) (1) (1) (1) (1) (1)	10,300	v,640	0,000
	00024P	C G G G G	10,138	A, 501	12 8 1.
	11.200	C 000	10,012	0.6/1	0,8900
	14,320	0.000	10,112	0,69B	0,000

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STORM OF THE 30TH JULY 1978 (CONTINUED) T(MINS) BOD(KG) COD(KG) NO3(KG) NH4(KG) 77.360 0.000 17,112 0,738 0,200 80,400 0,000 17,308 0.765 0,000 0,000 83,440 17,383 0,785 0.000 86,480 0,000 17,427 0,803 0.000 89,520 0,000 17.485 0,819 0.000 92,560 0,000  $17_{e}562$ 0.839 0,000 95,600 0,000 17,720 0.870 0.000 98,640 8,000 17,880 0,901 0,200 0,000 101.689 17,941 0,919 0.000 0,000 104,720 17,953 0,925 0.000 107,760 17,957 0.000 0,927 0,000 17,958 0,928 110.800 0,000 0,000 113,840 0,000 17,958 0,929 0,000 TOTAL LOADINGS 0.000 17,958 0,929 0.000

#### 3.2. Storm Runoff Quality Data Graphs

<u>Note</u>:

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Flow rate (Q) is in litres per second

All concentrations are in mg/l

Cumulative pollutant loads (e.g., C COD, C BOD) are expressed as percentages.



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4. STORM RUNOFF MODEL OUTPUT

STORM OF THE 2ND MAY 1978 (6)

STORM STATISTICS

STORM DURATION (HRS)		1.47
TOTAL RAINFALL (MM)	ų	1,755
TOTAL RAINFALL VOLUME (LITRES)	N	60369.0
DEPRESSION STORAGE LOSS (LITRES)	a a	16514.1
INFILTRATION LOSS (LITRES)		14910.7
TOTAL PREDICTED DISCHARGE (LITRES)	3	28944.2
MAX INTENSITY FOR 10 MINS (MM/HR)	H	3,9780
AVERAGE INTENSITY (MM/HR)	8	0.9750
TOTAL OBSERVED DISCHARGE (LITRES)		18840.7
PERCENTAGE RUNOFF	: <b>1</b> 3	47,95
TOTAL GULLY DISCHARGE (LITRES)	พ	4420.4
& GULLY RUNOFF/TOTAL RUNOFF	R	15,27
S GULLY/TOTAL FOR THE FIRST FLUSH	Ħ	18.03
FIRST FLUSH TIME		13,05
FIRST FLUSH VOLUME (LITRES)	Ħ	22153,7
& GULLY RUNOFF/TOTAL GULLY VOLUME	8	42.53

STORM OF THE 4TH MAY 1978 (7)

STORM STATISTICS

STORM DURATION (HRS)	14	0.39
TOTAL RAINFALL (MM)		3,393
TOTAL RAINFALL VOLUME (LITRES)	ą	116713.3
DEPRESSION STORAGE LOSS (LITRES)	ų	24257.2
INFILTRATION LOSS (LITRES)		49926.3
TOTAL PREDICTED DISCHARGE (LITRES)		42529.8
MAX INTENSITY FOR 10 MINS (MM/HR)	. #	15.0930
AVERAGE INTENSITY (MM/HR)	-	5,0895
TOTAL OBSERVED DISCHARGE (LITRES)	#	57334.0
PERCENTAGE RUNOFF	:	36,44
TOTAL GULLY DISCHARGE (LITRES)	Ħ	6527.5
& GULLY RUNOFF/TOTAL RUNOFF	-	15,35
& GULLY/TOTAL FOR THE FIRST FLUSH	ij	18,17
FIRST FLUSH TIME		8.34
FIRST FLUSH VOLUME (LITRES)	開	32533,9
& GULLY RUNOFF/TOTAL GULLY VOLUME	ų	62,80

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STORM OF THE 15TH MAY 1978 (9)

STORM STATISTICS

STORM DURATION (HRS)	8	0,35
TOTAL RAINFALL (MM)	8	1.170
TOTAL RAINFALL VOLUME (LITRES)	8	40246.0
DEPRESSION STORAGE LOSS (LITRES)	a,	16442.0
INFILTRATION LOSS (LITRES)	.8	9283,5
TOTAL PREDICTED DISCHARGE (LITRES)		14520.4
MAX INTENSITY FOR 10 MINS (MM/HR)	. 8	3.5100
AVERAGE INTENSITY (MM/HR)	ų	1,9500
TOTAL OBSERVED DISCHARGE (LITRES)	· II	7610,5
PERCENTAGE RUNOFF	:8	36,08
TÔTAL GULLY DISCHARGE (LITRES)	Ħ	1689.3
% GULLY RUNOFF/TOTAL RUNOFF	:#	11.63
& GULLY/TOTAL FOR THE FIRST FLUSH	- FI	11.78
STROM BLIGH MTNE		16.53
FIRST FLUSH TIME	. 49	**************************************
FIRST FLUSH VOLUME (LITRES)	-	13045.0

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STORM OF THE 24TH MAY 1978 (10)

### STORM STATISTICS

STORM DURATION (HRS)	ę,	Ø <b>.3</b> 8
TOTAL RAINFALL (MM)	I	1.755
TOTAL RAINFALL VOLUME (LITRES)	8	60369.0
DEPRESSION STORAGE LOSS (LITRES)	Ħ	17232,6
INFILTRATION LOSS (LITRES)	=	14882.0
TOTAL PREDICTED DISCHARGE (LITRES)	Ŋ	28254:3
MAX INTENSITY FOR 10 MINS (MM/HR)	Ą	5,9670
AVERAGE INTENSITY (MM/HR)		2,7000
TOTAL OBSERVED DISCHARGE (LITRES)	1	10769.5
PERCENTAGE RUNOFF	8	46.80
TOTAL GULLY DISCHARGE (LITRES)	4	4441,9
& GULLY RUNOFF/TOTAL RUNOFF	:8	15,72
& GULLY/TOTAL FOR THE FIRST FLUSH	Ħ	16.53
FIRST FLUSH TIME	調	10.06
FIRST FLUSH VOLUME (LITRES)	2	24632,7
& GULLY RUNOFF/TOTAL GULLY VOLUME	9	42.74

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STORM OF THE BIST MAY 1978 (11)

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# STORM STATISTICS

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STORM DURATION (HRS)	n	0.05
TOTAL RAINFALL (MM)	8	0.234
TÒTAL RAINFALL VÒLUME (LITRES)	8	8049.2
DEPRESSION STORAGE LOSS (LITRES)	1	5746.8
INFILTRATION LOSS (LITRES)	H	0.0
TOTAL PREDICTED DISCHARGE (LITRES)		2302.4
MAX INTENSITY FOR 10 MINS (MM/HR)	10	1.4040
AVERAGE INTENSITY (MM/HR)	ii	2,3400
PERCEN	3	.61222
TOTAL OBSERVED DISCHARGE (LITRES)	14	1409.5
PERCENTAGE RUNOFF	E	28.60
TOTAL GULLY DISCHARGE (LITRES)	11	0.0
% GULLY RUNOFF/TOTAL RUNOFF	11	0.00
% GULLY/TOTAL FOR THE FIRST FLUSH	11	***
FIRST FLUSH TIME	n	17.09
FIRST FLUSH VOLUME (LITRES)	=	0,0
& GULLY RUNOFF/TOTAL GULLY VOLUME	1	0.00

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STORM OF THE IST JUNE 1978 (12)

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#### STORM STATISTICS

STORM DURATION (HRS)	Ŧ	2.42
TOTAL RAINFALL (MM)	H	5.867
TOTAL RAINFALL VOLUME (LITRES)		201804.8
DEPRESSION STORAGE LOSS (LITRES)	5	11278.5
INFILTRATION LOSS (LITRES)		0.0
TOTAL PREDICTED DISCHARGE (LITRES)		190526.3
MAX INTENSITY FOR 10 MINS (MM/HR)		4.9140
AVERAGE INTENSITY (MM/HR)		2.1595
PERCEN	=	.62508
TOTAL OBSERVED DISCHARGE (LITRES)	=	119093.4
PERCENTAGE RUNOFF	3	94.41
TOTAL GULLY DISCHARGE (LITRES)	R	8938.8
% GULLY RUNOFF/TOTAL RUNOFF	8	4.69
& GULLY/TOTAL FOR THE FIRST FLUSH		11.32
FIRST FLUSH TIME	=	18,42
FIRST FLUSH VOLUME (LITRES)		71632.1
% GULLY RUNOFF/TOTAL GULLY VOLUME		86,00

STORM OF THE 29TH JUNE 1978 (15)

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#### STORM STATISTICS

STORM DURATION (HRS)	5.R.	Ø.54
TOTAL RAINFALL (MM)	8	0,968
TOTAL RAINFALL VOLUME (LITRES)	11	33299.2
DEPRESSION STORAGE LOSS (LITRES)	8	10571,7
INFILTRATION LOSS (LITRES)	28	0.0
TOTAL PREDICTED DISCHARGE (LITRES)	11	22727.6
MAX INTENSITY FOR 10 MINS (MM/HR)	55	2.8773
AVERAGE INTENSITY (MM/HR)	57	1,0561
PERCEN	1	1,14427
TOTAL OBSERVED DISCHARGE (LITRES)	W	26006.4
PERCENTAGE RUNOFF	15	68,25
TOTAL GULLY DISCHARGE (LITRES)	2	3827.2
& GULLY RUNOFF/TOTAL RUNOFF	ä	16.84
% GULLY/TOTAL FOR THE FIRST FLUSH	ų	18.19
FIRST FLUSH TIME	=	0.50
FIRST FLUSH VOLUME (LITRES)	IF	19066.5
% GULLY RUNOFF/TOTAL GULLY VOLUME	*	36.82

#### STORM OF THE 1ST JULY 1978 (16)

STORM STATISTICS

STORM DURATION (HRS)	11	1.51
TOTAL RAINFALL (MM)	8	1.863
TOTAL RAINFALL VOLUME (LITRES)	i.	64081.7
DEPRESSION STORAGE LOSS (LITRES)	8	11178.6
INFILTRATION LOSS (LITRES)	8	Ø,Ø
TOTAL PREDICTED DISCHARGE (LITRES)	14	52903.1
MAX INTENSITY FOR 10 MINS (MM/HR)	31	2,8245
AVERAGE INTENSITY (MM/HR)		Ø <b>.</b> 9980
PERCEN	2	.30321
TOTAL OBSERVED DISCHARGE (LITRES)	H	16040.7
PERCENTAGE RUNOFF	=	82,56
TOTAL GULLY DISCHARGE (LITRES)	n	6788.6
% GULLY RUNOFF/TOTAL RUNOFF	-	12.83
& GULLY/TOTAL FOR THE FIRST FLUSH	11	14,56
FIRST FLUSH TIME	15	1.40
FIRST FLUSH VOLUME (LITRES)	#	42205,6
% GULLY RUNOFF/TOTAL GULLY VOLUME	5	65.31

## STORM OF THE 3RD JULY 1978 (17)

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STORM STATISTICS

STORM DURATION (HRS)		1.55
TOTAL RAINFALL (MM)		8,569
TOTAL RAINFALL VOLUME (LITRES)	8	294743°, 3
DEPRESSION STORAGE LOSS (LITRES)	8	23157.5
INFILTRATION LOSS (LITRES)		139866.7
TOTAL PREDICTED DISCHARGE (LITRES)	R	131719.1
MAX INTENSITY FOR 10 MINS (MM/HR)	8	8,6194
AVERAGE INTENSITY (MM/HR)		4.4320
TÒTAL ÒBSERVED DISCHARGE (LITRES)	8	33407.5
PERCENTAGE RUNOFF	: 🛃	44.69
TÒTAL GULLY DISCHARGE (LITRES)	#	8814.7
& GULLY RUNOFF/TOTAL RUNOFF		6 . 69
& GULLY/TOTAL FOR THE FIRST FLUSH		'i1.61
FIRST FLUSH TIME		1.18
FIRST FLUSH VOLUME (LITRES)	Ħ	68784.5
& GULLY RUNOFF/TOTAL GULLY VOLUME	=	84,81

STORM OF THE 23RD JULY 1978 (19)

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STORM STATISTICS

STORM DURATION (HRS)	8	1,07
TOTAL RAINFALL (MM)	8	0.789
TOTAL RAINFALL VOLUME (LITRES)	-	27125.9
DEPRESSION STORAGE LÖSS (LITRES)	19	14521.4
INFILTRATION LOSS (LITRES)	8	4663.7
TOTAL PREDICTED DISCHARGE (LITRES)	Cļ	7940.8
MAX INTENSITY FOR 10 MINS (MM/HR)	H	2,6162
AVERAGE INTENSITY (MM/HR)	ų	1.6898
TOTAL OBSERVED DISCHARGE (LITRES)	. 11	5053.8
PERCENTAGE RUNOFF	-	29.27
TOTAL GULLY DISCHARGE (LITRES)	7	609.9
% GULLY RUNOFF/TOTAL RUNOFF	. <b>2</b>	7.68
& GULLY/TOTAL FOR THE FIRST FLUSH	Ŗ	7.70
FIRST FLUSH TIME	8	20.05
FIRST FLUSH VOLUME (LITRES)	#	7459.0
& GULLY RUNOFF/TOTAL GULLY VOLUME	- 54	5.87

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STORM OF THE 26TH JULY 1978 (20)

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STORM STATISTICS

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STORM DURATION (HRS)	: 🔛	0,54
TÒTAL RAINFALL (MM)	A	1.800
TOTAL RAINFALL VOLUME (LITRES)		61900.8
DEPRESSION STORAGE LOSS (LITRES)	IJ	11177.4
INFILTRATION LOSS (LITRES)	. 2	0.0
TOTAL PREDICTED DISCHARGE (LITRES)	I.	50723.5
MAX INTENSITY FOR 10 MINS (MM/HR)	H	3,3252
AVERAGE INTENSITY (MM/HR)	:8	1.9631
PERÇEN		.85045
TOTAL OBSERVED DISCHARGE (LITRES)	ŧ	43137.9
PERCENTAGE RUNOFF	-	81.94
TOTAL GULLY DISCHARGE (LITRES)	8	7016.0
& GULLY RUNOFF/TOTAL RUNOFF	8	13.83
% GULLY/TOTAL FOR THE FIRST FLUSH	8	15.67
FIRST FLUSH TIME	:	5.28
FIRST FLUSH VOLUME (LITRES)	F	40806.7
& GULLY RUNOFF/TOTAL GULLY VOLUME	10	67.50

STORM OF THE 27TH JULY 1978 (21)

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STORM STATISTICS

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STORM DURATION (HR5)	*	2.55
TOTAL RAINFALL (MM)	#	2.421
TOTAL RAINFALL VOLUME (LITRES)	Ħ	83273,6
DEPRESSION STORAGE LOSS (LITRES)	Ħ	15541.5
INFILTRATION LOSS (LITRES)	:8	18965,0
TOTAL PREDICTED DISCHARGE (LITRES)	. #	48767.2
MAX INTENSITY FOR 10 MINS (MM/HR)	Ħ	2,5660
AVERAGE INTENSITY (MM/HR)	8	Ø.8253
TOTAL OBSERVED DISCHARGE (LITRES)		10279.5
PERCENTAGE RUNOFF	: (2	58,56
TOTAL GULLY DISCHARGE (LITRES)		5588 <sub>8</sub> Ø
& GULLY RUNOFF/TOTAL RUNOFF	- 19	11.46
& GULLY/TOTAL FOR THE FIRST FLUSH	R	12.52
FIRST FLUSH TIME	-	24.04
FIRST FLUSH VOLUME (LITRES)	.有	40181.3
& GULLY RUNOFF/TOTAL GULLY VOLUME	8	53,76

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STORM OF THE 30TH JULY 1978 (24)

STORM STATISTICS

STORM DURATION (HRS)		1.47
TOTAL RAINFALL (MM)	29	7.407
TÒTAL RAINFALL VOLUME (LITRES)	Sais	254778.2
DEPRESSION STORAGE LOSS (LITRES)	ʻ <b>H</b>	11339,4
INFILTRATION LOSS (LITRES)	1	0,0
TOTAL PREDICTED DISCHARGE (LITRES)	, F	243438.8
MAX INTENSITY FOR 10 MINS (MM/HR)	H	11,2780
AVERAGE INTENSITY (MM/HR)	Ŗ	4,1148
PERCEN	Ħ	.96904
TOTAL OBSERVED DISCHARGE (LITRES)		235901.9
PERCENTAGE RUNOFF	. E	95,55
TOTAL GULLY DISCHARGE (LITRES)	. #	8991,3
% GULLY RUNOFF/TOTAL RUNOFF	8	3,69
& GULLY/TOTAL FOR THE FIRST FLUSH	=	12.21
FIRST FLUSH TIME	- <del>[]</del>	6,26
FIRST FLUSH VOLUME (LITRES)	19	67645,2
& GULLY RUNOFF/TOTAL GULLY VOLUME		86,51







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5. MULTIPLE REGRESSION OUTPUT

LIST OF SYMBOLS					
DUR	=	storm duration (Mins.)			
ROVOL	=	runoff volume (M <sup>3</sup> )			
AVIN	=	average intensity (mm)			
AV10	=	average intensity for the most intense 10 minutes (mm)			
PI	=	peak intensity (mm)			
ADP	=	antecedent dry period (days)			
AWI	=	antecedent wetness index			
PPT	=	total rainfall (mm)			
SPE	=	total rainfall of previous event (mm)			

STORN PARANETERS

STORM NO	DUR	ROVOL	NIAN	olva	E di	ADP	INV	Ĕdd	(s) (s) (v)
ů,	107.000	60°369	9.975	3,978	7.020	0,666	46,200	1 = 755	000° -
*	39,060	116,713	5,090	15,093	21,660	0000	44°300		6,260
e e	35,000	40.246	1,950	3,510	7.620	9.000	46,300	1.279	3,300
ະ ເມ	54,000	33,299	1,056	2,877	3°379	0°005	48,709	C,958	6,200
16,	111,000	64.981	865° 8	2.824	166 * 8	0,690	45, 29, 8	1,863	1 - 908
° 67	27,000	27.125	1,690	2,616	6,808	6,000	46,950	987.89	3,100
20.	54,000	61,901	1,953	3,325	5,657	1,000	201°75		4,500
24.	107,900	254.778	4 4 4 5	11°278	19,131	0,092	48,000	704.7	2.800
• 27 • 7	151,000	197,780	2,270	4.914	7.026	000.0	49.700	5,756	9,200
е О т	38,000	68,369	2,700	5°967	14.040	2,405	57,988	5 5 7 7 8 7 8 7 8 7 8 7 8 7 8 7 8 7 8 7	s. 188

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2.800 961.6 1 . 200 2,200 806\*7 C. 200 3.106 0.200 0.200 4.500 0°200 3,360 006 -(1) (1) (1) 1.863 0.789 7.467 - 155 1,755 2,421 1.179 5.750 0.234 8,569 0,968 1.800 3,393 Edd 48,000 001.65 56,000 45.700 44,300 46,309 48.700 45,100 906 95 001.150 57,000 45,100 46.200 INV 000° 0 800.3 0.000 1.000 0,090 0.000 7.096 5995 9 0.000 000.0 0000 0 900° 8 330.0 ADP 146.0 1.020 3,379 320"1 7.920 5.657 14.040 3.518 13.756 21,069 6,808 19-131 100 8 3,325 5,967 2,349 3,978 5,093 3,510 8.619 2.616 4.916 2.565 7.8°77 2.824 11.278 OTAV 1,963 4.432 6,975 1,950 1,056 2,270 2.726 1,699 4.115 2.340 Ø**.** 825 5.090 \$ 998 AUIN 60.369 116,713 40.245 33°299 27.125 254.778 197.780 60,369 8.049 294.743 63,274 64°081 61.901 ROVOL 000.70 39,000 35,999 54.000 111.000 27,000 54,000 107.000 151,003 38.000 115,000 175,000 6,000 DUR 02 ه س 6. C) 4....] 20. 24 0 نگ جسم إسبو - 1 - 5 - 5 å -Ch. 16. 2 STORM

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STORM PARAMETERS

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STORM NO	FLOW		SS	03	BOD	cop	CON	NH4
• 9	16,98	8	50,280	158.330	5.700	30,928	10.419	39'8 1
	20.23	30	172.200	78.850	7.230	978.99	8.310	0,880
.6	5,86	20	138.890	257,850	14.199	89,740	11.690	1,240
1 S.	13.61	6	21.620	51.410	5.650	85,610	5°050	8,568
16.	9.41	61	26.719	060.01	2,590	37,770	5 . បទល	0,240
19.	5.41	01	45,580	107,260	8.1.0	23, 026	5.210	9.419
20.	16,30	08	14.910	87.140	6.360	18.776	. 6,862	0,358
24.	34,65	30	84,380	70.500	0.000	75,130	3.380	0,996
12.	25.71	0	68.170	019.970	0.000	41,370	0000	. 0,000
10.	. 9.16	0	155,470	374.940	27.140	159,110	31.480	2,052
11.	3.1 "	20	294.970	368,000	31.790	313,650	0,000	5,398
. 17.	14.55	20	018-16	123,749	5,330	65,720	4.500	9,249
21.	1. T	20	372.309	138,055	000.0	233.470	5.569	000 000
MEAN	13.6	692	118.253	156.624	8.776	98.012	7.618	6,948
STD DEV	р, д	5	109.219	103.300	19.022	88.325	7.925	964.1.

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STORN LOADS 

0,012 0000.0 0,009 0.014 0,000 0,023 0.015 0.002 0.615 0,000 0.016 0.921 \$00.0 0.051 γHN 0,221 0,929 000000 9.358 0,661 0.258 9.102 0.230 0,302 0.135 0000.0 0.165 0.204 6.484 169.0 N03 3.205 2,292 1.809 2,404 2,568 117.2 0.825 17.958 5.196 \$,682 5.817 1.147.0 0,634 0.606 0 ± 1 3 1 000 3.603 6.669.9 0,308 0.069 0,195 000.0 12.469 g.118 0.044 0.112 0.286 45,100 0.421 0.151 0.046 BOD 4,888 18.837 4.526 4,268 1,519 5,937 3,005 4,595 2.146 1.376 0.613 3.831 16.851 0,801 1.177 50 4,054 1.766 4.096 5.787 0.985 1.156 0.579 0.449 0.269 0.655 8.562 0.642 3,358 20.169 10.031 50 46,550 STORM NO DISCHARGE 58,249 8,312 16.799 5,712 43,967 11.361 2.176 26.777 36,579 965.61 239,024 125,500 100.11 66,547 STD DEV 15. 20. 24. 12. 6 17 \* 6. 01 \*\*\* 21. MEAN

MULTIPLE REGRESSION ANALYSIS FOR SUSPENDED SOLIDS

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# CORRELATION COEFFICIENTS

	88
DUR	0,3767
AVIN	0,7731
`AV10	0.7794
P1.	0,7591
ADP	-0,1743
AWI	∞0 <b>,</b> 0539
PPT	0,9368
SPE	-0.1107

## ANALYSIS OF REGRESSION

....

PARAMETER	COEFF	STD ERR	T-VALUE
DUR	₩Ø.1260	0,0068	-18,4621
AVIN	₩4 • 9852	0,2592	*19,2333
AV10	0,6970	0,0936	7,4450
ΡI	0,1706	0.0470	3,6331
ADP	0,8246	0,1518	5,4311
AWI	ů,7885	0.1022	.7.7145
PPT	5,2270	0,2026	25,8034
SPE	mØ,4754	0,0761	-6.2474
CONSTANT	43,4995	5,3717	8,0979

STANDARD	ERROR	OF ESTIMATE		0,1289
MULTIPLE	CORREI	ATION (R)		i,0000
DETERMIN	TION (	(R SQUARED)	14 10	1,0000
CORRECTED	R SQL	JARED		0,9996

MULTIPLE REGRESSION FOR SUSPENDED SOLIDS LOAD (KG)

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## CORRELATION COEFFICIENTS

	55
DUR	0,3767
AVIN	0,7731
AV 1 Ø	8,7794
AWJ	-0,0539
ppr	0.9368

### ANALYSIS OF REGRESSION

PARAMETER	COEFF	STD ERR		T ··· V A L U E;
DUR	<i>∞0,0</i> 979	0.0138		-7,0940
AVIN	-4,3487	0,8713		•4,9909
AV10	1.1830	0,2085		5,6747
AWI	······································	0.0593	+	-1,8891
ррт	1.2495	0.3122		13,6133
CONSTANT	8,8338	3,0730		2.8747

STANDARD	ERROR	OF ESTIMATE	879 434	0,5765
MULTIPLE	CORREL	ATION (R)		0,9983
DETERMINA	TION (	R SQUARED)	1	0,9966
CORRECTED	RSQU	ARED	11	0,9923

MULTIPLE REGRESSION FOR SUSPENDED SOLIDS LOAD (KG)

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## CORRELATION COEFFICIENTS

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	66
DUR	Ø,3767
PI	6,7591
ADP	-0,1743
PPT	0,9368

## ANALYSIS OF REGRESSION

PARAMETER	COEFF	STD ERR	TEVALUE
DUR	~P.0298	0.0187	<b>≈1</b> ,5952
РΙ	0,2742	0.1261	2,1745
ADP	••0,4362	0,2012	m2,1682
PPT	2,6250	0.4487	5,8498
CONSTANT	≈2,6382	1,4234	··· 1. 8535

STANDARD	ERROR	OF ES	TIMATE	8-14 822)	1,2031
MULTIPLE	CORREL	ATION	(R)	1013 1717	0,9906
DETERMIN	ATION (	R SQU	ARED)	8	0.9814
CORRECTE	R SQU	IARED			0.9664

# MULTIPLE REGRESSION FOR DISSOLVED SOLIDS LOAD (KG)

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## CORRELATION COEFFICIENTS

DS		
0:6771	DUR	
0,4449	ΛΥΙΝ	
Ø <sub>n</sub> 3975	A V 1 O	
؄3856	ΡI	
«0 <sub>8</sub> 0895	ADP	
Ø.1933	ANI	
0,9444	<b>PF</b> T	
-8.1747	SPE	

## ANALYSIS OF REGRESSION

PARAMETER	COEFY	STD ERR	T ~ VALUE
DUR .	0,1523	0,0505	3,0165
AVIN	6,9554	1,9181	3.6263
AV10	*1.1326	0,6927	<b>~1.6321</b>
だよ	5000 1000 1000 1000	1 1936	9 # 8047 m 2 1/319
AWI	2,3809	0,7563	3,1480
PPT	*1 4562	1,4990	∾0 <b>.</b> 9714
SPE	0,7449	0.5631	1,3229
CONSTANT	×126,6374	39,7509	*3,1858

STANDARD	ERROR	OF ESTIMATE	10	0,9538
MULTIPLE	CORREI	ATION (R)	85	Ø,9988
DETERMINA	TION (	(R SQUARED)	8	6,9977
CORRECTEL	RSQ	JARED	55	0,9789

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## MULTIPLE REGRESSION FOR DISSOLVED SOLIDS LOAD (KG)

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# CORRELATION COEFFICIENTS

	bs
DUR	0,6771
AVIN	0,4449
AV 1 Ø	Ø <sub>8</sub> 3975
AWI	Ø,1933
rqq	0,9444

## ANALYSIS OF REGRESSION

PARAMETER	COEFF	STD ERR	T=VALUE
DUR	0.0371	0,0446	Ø,8321
AVIN	2,9587	2,8137	1,8515
AVIØ	m1, 1274	0,6732	~1.5261
AWI	0,2091	0,1914	1,0927
PPT	2,4049	1,0080	2,3857
CONSTANT	#14,4123	9,9235	*1:4523

STANDARD	ERROR	OF ESTIMATE	814 213	1,8618
MULTIPLE	CORREL	ATION (R)	914 914	0,9820
DETERMINA	ATION (	R SQUARED)		Ø,9643
CORRECTER	R SQU	ARED	S.	0,9197

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## MULTIPLE REGRESSION FOR COD LOAD (KG)

## CORRELATION COEFFICIENTS

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	COD
DUR	0,3502
AVIN	0,6745
AV10	0,6833
AWI	0,0056
ppr	6.8952

## ANALYSIS OF REGRESSION

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PARAHETER	COEFF	STD ERR	T≈VALUE
DUR	···Ø·1492	0,0263	~5,6704
AVIN	+7,7486	1,6608	* 4. 6655
AV10	1,6576	0,3974	4.1714
AWT	∞Ø,0267	0.1130	······································
PPT	5,0104	0,5950	8,4296
CONSTANT	9,6818	5,8575	1,6529

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STANDARD	ERROR	OF ESTIMATE	22	1,0989
MULTIPLE	CORREL	ATION (R)	6.19 X(2	0,9908
DETERMINA	TION (	R SQUARED)	EA.	0,9817
CORRECTED	R SQU	ARED		0,9589

### MULTIFLE REGRESSION FOR BOD LOAD (KG)

#### 

# CORRELATION COEFFICIENTS

	BQD
DUR	∞0.3390
AVIN	0,8527
AV 1 Ø	0,8318
AWI	-0.2517
PPT	0,7912

## ANALYSIS OF REGRESSION

PARAMETER	COEFF	STD ERR	T⇔VALUE
DUR	······································	0,0160	~0,9843
AVIN	···0,6424	0.7271	mØ₊8835
AVIO	0,0516	0.0624	0.8267
AWJ	··· Ø . 0134	0,0787	∞0,1699
PPT	0,7950	0,6563	1,2113
CONSTANT	1.4221	4.6498	0,3058

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STANDARD	ERROR	OF ESTIMATE	te " ejny	0,0256
MULTIPLE	CORREI	ATION (R)	613 747	0,9971
DETERMINA	TION (	R SQUARED)	400 829	0,9942
CORRECTED	RSQL	JARED	8/8 4 at	0.9651

MULTIPLE REGRESSION FOR NITRATE LOAD (KG)

## CORRELATION COEFFICIENTS

.

	NO3
DUR	0,3624
AVIN	0.7711
A V 1 Ø	ø,7792
AWI	0.1406
TAA	0,9748

### ANALYSIS OF REGRESSION

PARAMETER	COEFF	STD ERR	T-VALUE
DUR	0,0054	0,0022	2,4370
AVIN	0,2721	0,1154	2,3581
AVIØ	···Ø, 0228	0.0211	-1,0790
AWI	0,0839	0,0231	3,6372
PPT	···Ø,0143	0,0481	-0,297Ø
CONSTANT	-4.4315	1.2357	*3.5861

STANDARD	ERROR	0F	ESTIMATE		0,0379
MULTIPLE	CORREI	LTA	ON (R)	8	0,9977
DETERMINA	TION (	(R 5	QUARED)		0,9953
CORRECTED	R SQL	JARE	:D		0.9837

## MULTIPLE REGRESSION FOR AMMONIUM LOAD (KG)

-.32

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CORRELATION COEFFICIENTS

	NHA
DUR	-0,1463
AVIN	0,8461
AVIØ	0,9432
TWA	-0,4703
ppT	0.8645

## ANALYSIS OF RECRESSION

PARAMETER	COEFF	STD ERR	T . VALUE
DUR	-0.0015	6,0039	mk . 3947
AVIN	m0,0760	0,1779	-0,1271
AVIØ	0.0111	0.0153	0.7264
AWI	mØ,0036	0,0192	♥Ø.1852
rey	0,0685	0.1606	0.4257
CONSTANT	Ø.2564	1,1375	0,2254

STANDARD	ERROR	0F	ESTIMATE	*** ***	0,0063
MULTIPLE	CORREI	AT I	ON (R)	17	Ø,9878
DETERMINA	TTON (	RS	QUARED)	714 5.22	0,9758
CORRECTED	RSQL	ARE	D	-	0.8549

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#### MULTIPLE REGRESSION FOR SS MEAN CONCENTRATION

### CORRELATION COEFFICIENTS

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	Ci ti
DUR	-0,3536
AVIN	0,6982
AV19	0,6461
рŢ	e,7316
ADP	0,4186
A₩1	0.2379
ррт	0,1592
SPE	≈0.3596

### ANALYSIS OF REGRESSION

12

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PARAMETER	COEFF	STD ERR	↑ •• VALUE
DUR	0.9027	4,1402	0.2186
ΑΥΙΝ	90,8515	157,2591	0,5777
AV10	-38,0297	56,7974	≈0,6696
PI	18,3360	28,4949	0,6435
ADP	·5.1181	92.1193	-0,0556
AWI	-2,9626	62.0093	-0.0478
PPT	≈29,3979	122,9039	-0.2392
SPE	#17.2739	46,1669	×0.3742
CONSTANT	101,4061	3259,1244	0,0311

STANDARD	ERROR	OF	ESTIMATE	**	78,2009
MULTIPLE	CORREL	ATI	0N (R)	2	0,8935
DETERMIN	TION (	RS	QUARED)	87	0.7984
CORRECTEI	R SQU	ARE	D		ů,8147

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MULTIPLE REGRESSION FOR SS MEAN CONCENTRATION

CORRELATION COEFFICIENTS (inc. S.11)

	នន
DUR	-8,5421
AVIN	8,4602
NV10	0,2185
ΡŢ	0,2282
ADP	0,6747
VAI	Ø.5699
PPT	<i>∞0</i> ,1551
SPF	

ANALYSIS OF REGRESSION

PARAMETER	COEFF	STD ERR	TAVALUE
DUR	≈Ø,238Ø	4,7053	······································
AVIN	89,7369	182,8414	0,4908
AV10	<b>*13,5075</b>	62.3266	······································
РÏ	-7,2304	24,0625	•0,3005
ADP	29.6452	102,5324	0,2891
AWI	+ +12,7599	71,5677	-0.1783
PPT	1,9946	140.1368	0.0142
SPE	-15,9849	53,6646	-0,2979
CONSTANT	652,3719	3757.4477	0.1736

STANDARD	ERROR	OF ESTIMATE		90,9236
MULTIPLE	CORREI	LATION (R)	Ba Ba	0.8798
DETERMIN	ATION	(R SQUARED)	22	0.7741
CORRECTED	D R SQI	JARED	W/S Roc	-0.1294

# MULTIPLE REGRESSION FOR SS MEAN CONCENTRATION

CORRELATION COEFFICIENTS

	SS
DUR	«Ø,3536
VAJN	0,6982
AV10	0,6461
AWI	0,2379
PPT	0,1592

## ANALYSIS OF REGRESSION

PARAMETER	COEFF	STD ERR	TAVALUE
DUR	Ø,6798	1,1571	0.5875
AV1N -	64,6574	73,0361	0,8853
AVIO	w2,2887	17,4745	m@ + 1310
AWI	4.7925	4.9678 .	0,9647
PPT	×25,8563	26,1662	mØ,9882
CONSTANT	≈266,9644	257,5879	-1.9364

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STANDARD	ERROR	OF ESTIMAT		48,3260
MULTIPLE	CORREL	ATION (R)	9(27) #192	0,8319
DETERMINA	TION (	R SQUARED)	n	0.6920
CORRECTED	R SQL	ARED	जन्म इन्द्र	8.3070

# HULTIPLE REGRESSION FOR DS MEAN CONCENTRATION

## CORRELATION COEFFICIENTS

	DS
DUR	ů.2542
AVIN	mØ,0232
AV10	wØ,1335
Рĭ	0,0943
ADP	0,7809
AWI	0,7055
ррт	~0,2133
SPE	-0.1763

## ANALYSIS OF REGRESSION

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PARAMETER	COEFF	STD ERR	T -VALUE
DUR	3,7663	7,5976	Ø.4957
AVIN	175.9614	288,5850	0.6097
AV1Ø	~78,5510	104,2285	mØ,7536
ΡĮ	40,6870	52.2909	0,7781
ADP	≈36.1919	169,0474	-8,2141
AWI	28.2162	113,7928	0.2480
PPT	-105.0068	225,5400	-0.4656
SPE	-1.3534	84,7189	-0.0160
CONSTANT	m1519.9772	5980.7943	······································

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STANDARD	ERROR	OF ESTIMA	TE =143,5058
MULTIPLE	CORREL	ATION (R)	≈ Ø,8846
DETERMINA	TION (	R SQUARED	) = 0.7825
CORRECTED	RSAL	JARED	= -0.9573

MULTIPLE REGRESSION FOR DS MEAN CONCENTRATION

# CORRELATION COEFFICIENTS (inc.S.11)

	DS
DUR	×0.4407
AVIN	∞ <i>0</i> ,0110
AV1.0	-0.2452
61	-0,0936
VDb	0.8563
A \v 1	0.8042
PPT	-0,3542
SPE	··· 0 . 2137

## ANALYSIS OF REGRESSION

PARAMETER	COEFF	STD ERR	T*VALUE
DUR	2,9639	5,8720	0.5047
AVIN	175.1775	228,1775	Ø.,7677
A V 1 Ø	-61.3001	77.7731	······································
ΡI	22.7017	30,0289	Ø. 756Ø
ADP	e11.7368	127,9556	·Ø,0917
AWI	21.3241	89.3132	0.2388
PPT	<b>#82,9228</b>	174.8842	-0.4742
SPE	~0.4466	66,9709	······································
CONSTANT	-1132.3865	4689,1190	-0.2415

STANDARD ERROR OF ESTIMATE ≈113,4684 MULTIPLE COPRELATION (R) = ∅,9046 DETERMINATION (R SQUARED) = ∅,8183 CORRECTED R SQUARED = ∅,0913 MULTIPLE REGRESSION FOR DS MEAN CONCENTRATION

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CORRELATION COEFFICIENTS

	u a
DUR	mØ,2542
AVIN	-0,0232
AV 1 Ø	ů,1335
AWI	0,7055
PPT	«C. 2133

ANALYSIS OF REGRESSION

PARAMETER	COEFF	STD ERR	T .VALUE
DUR	1,5834	2,2551	0.7021
AVIN	111,9079	142,3366	0,7862
AV10	*19,9381	34,0555	*0.5855
AWI	20,1233	9,6815	2,0785
PPT	-49.1260	50,9941	mØ.9634
CONSTANT	=951.441Ø	502,0011	=1.8953

STANDARD	ERROR	OF ESTIMATE	ŝ	94,1803
MULTIPLE	CORREL	ATION (R)	414 9418	Ø <b>.7</b> 908
DETERMINA	TION (	R SQUARED)	924 8/3	0,6253
CORRECTED	R SQL	ARED	10 Le 10 Le	0.1570

## MULTIPLE REGRESSION FOR COD MEAN CONCENTRATION

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## CORRELATION COEFFICIENTS

	COD
DUR	ů,3928
AVIN	Ø . 4446
AV10	0.4141
AWI	0,6015
PPT	0.0165

ANALYSIS OF REGRESSION

PARAMETER	COEFF	STD ERR	T ** VALUE
DUR	-0,3442	0.7754	-0,4440
AVIN	-24.8510	48,9399	Ţ,5078
AV10	13,5613	11.7094	1,1582
ΛWI	9,0671	3,3288	2,7238
PPT	-2.0677	17.5334	-0.1179
CONSTANT	<b>~358,3856</b>	172.6041	P2.0763

STANDARD	ERROR	OF ESTIMATE	11	32,3822
MULTIPLE	CORREL	ATION (R)		0,8724
DETERMIN	TION (	R SQUARED)		0,7611
CORRECTEI	R SQL	ARED	1	0.4626

MULTIPLE REGRESSION FOR BOD MEAN CONCENTRATION

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## CORRELATION COEFFICIENTS

	BOD
DUR	-0,6776
ΑΥΙΝ	0.2289
AV10	0,0405
AWI	0,0427
PPT	-0,2298

## ANALYSIS OF REGRESSION

PARAMETER	COEFF	STD ERR	T .VALUE
DUR	×2,5949	2,4672	<b>≈1,0518</b>
AVIN	-114,3796	112.2446	-1,0190
AV 1 Ø	9,5842	9,6362	0,9946
AWI	-13,4325	12,1462	-1,1059
PPT	101.6819	101,3188	1,0036
CONSTANT	796.2875	717.8065	1,1093

STANDARD	ERROR	OF ESTIMATE	873 174	3,9531
MULTIPLE	CORREI	ATION (R)	11	Ø,8917
DETERMINA	TION (	R SQUARED)	11	0,7951
CORRECTED	R SQU	ARED	11	₩Ø,2296

## MULTITPLE REGRESSION FOR MOS MEAN CONCENTRATION

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## CORRELATION COEFFICIENTS

### NO3

DUR	-9,1703
AVIN	-0,1115
AVIØ	∞C.0790
AWI	-0.4780
PPT	<i>∞Ø</i> ,3828

## ANALYSIS OF REGRESSION

PARAMETER	COEFF	STD ERR	T-VALUE
DUR	ר.0191	0,2410	-0.0793
AVIN	··1.0177	12,5996	∞Ø,0808
AV 1 Ø	0,1917	2.3044	0,0832
AWI	-0,9538	2.5197	rØ,3785
PPT	mØ.0701	5.2542	~0.0133
CONSTANT	54.3034	134.9487	0.4024

STANDARD	ERROR	OF ESTIMATE	14	4,1438
MULTIPLE	CORREL	JATION (R)	e.a	Ø,5899
DETERMIN	ATION (	(R SQUARED)	45%) 1-10	0,3480
CORRECTEI	R SQL	JARED	407 876	-1,2821

MULTIPLE REGRESSION FOR NH4 MEAN CONCENTRATION 

#### CORRELATION COEFFICIENTS

	NH4		
DUR	-0.1411		
AVIN	0,2543		
AV1Ø	0.3067		
A ₩ I	•0.2378		
PPT	0.1450		

•

#### ANALYSIS OF REGRESSION

PARAMETER	COEFF	STD ERR	T-VALUE
DUR	•Ø,3108	0.4004	······································
AVIN	-14,3215	18,2172	·Ø.7862
AV10	1,3400	1.5639	0.8568
AWT	-1.5887	1.9713	······································
PPT	12.3141	16.4439	0.7489
CONSTANT	94.3179	116.4993	0.8096

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STANDARD	ERROR	OF ESTIMATE	149 159	0,6416
MULTIPLE	CORREL	ATION (R)	61 <b>5</b> 614	0.7323
DETERMINA	TION (	(R SQUARED)	8	Ø.,5363
CORRECTED	R SQL	JARED	8474 6472	-1.7821

## APPENDIX 3

SECTION	TITLE	Page
1	"An Assessment of the Importance of Roadside Gully Pots in Determining the Quality of Stormwater Runoff".	2
	Urban Storm Drainage, Proceedings of the First International Conference, 586-602, Pentech Press, London, 1978.	
2	"Mathematical Simulation of Pollutant Contributions to Urban Runoff from Roadside Gully Pots".	19
	<u>Proceedings</u> of the Second International Conference on Urban Storm Drainage, University of Illinois, Urbana, U.S.A., 1981.	

AN ASSESSMENT OF THE IMPORTANCE OF ROADSIDE GULLY POTS IN DETERMINING THE QUALITY OF STORMWATER RUNOFF.

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#### ABSTRACT

Roadside gully pots on an urban estate have been monitored for a period of one year to obtain measurements of various pollution parameters. The results of an analysis of these parameters for the first year of study is presented and initial conclusions drawn on the relative importance of the different sources of pollution and of the effect of rainfall, dry periods, seasonal variations and human activity upon the quality of the stored water within the gully pots.

Examination of the stormwater runoff, (using an automatically triggered sampling machine) indicates that the quality of the 'first flush' may be substantially attributed to the discharge of gully pot liquors.

Three different forms of pollutant run off curve have been identified dependent upon the origin and nature of the material in question. Equating the levels of pollutants in gully pots to those found in the stormwater runoff has provided an indication of the principal origins of the pollutant.

The preliminary results of a laboratory simulation of the removal of solids from a gully pot are presented.

#### INTRODUCTION

Gully pots are placed along the roadside to act as inlet points for rainwater and stormwater runoff to the sewerage system. Their principal purpose is to remove solid material transported by the runoff which may subsequently cause sewer blockage. They also act as a water seal to prevent the release of bad odours from the sewer.

Although in recent years the nature and polluting character of stormwater runoff has been the attention of several studies
(1,3,4,5,6,7 and 9) the role of gully pots in removing sediments and concentrating polluting matter is not known to have formed the core of any major study. As early as 1900 Folwell (2) noted the accumulation with subsequent putrefaction of organic matter and drew attention to the inefficiency of cleansing cycle. The U.S. Dept of the Interior (9) produced a general report on the pollution of storm runoff which included a minor section on gully pots. The report indicated BOD values from 35 to 225mg/l in the supernatant liquid and concluded that "catch basins (gully pots) may be one of the most important single sources of pollution from stormwater flows". Other works including Sartor et al. (5) and Tucker (6) have come to similar conclusions. Tucker noted the anaerobic breakdown of organic material and reported wide variations in stored water quality between gully pots at the same and different locations, with values of BOD up to 350mg/1 and the COD up to 965mg/l.

### Pollution of stormwater

The pollution of stornwater results from the contamination of rainwater through contact with various substances from the time of origin in the atmosphere until the moment of its discharge into a receiving body of water. This quality variation may be brought about by such a wide variety of different events that the complex changes which result are difficult to isolate. However, in broad outline pollution of stornwater runoff may be attributed to the following principal sources:

- Open land contributes dust and soil particles which accumulate in roadside gutters and it acts as a source of nutrients and micro-organisms. In heavy storms runoff can occur directly off open land resulting in high solid loads.
- 2 Vegetation and animal activity leads to the deposit of grasscuttings, leaves, faecal matter, etc.
- 3 Roof and road surfaces provide collecting areas for pollutants and are themselves sources of inorganic solids, cement, sand, eroded road material and salt.
- 4 Motor Traffic acts as a source of oil, exhaust gases, petrol, particles from car bodies and tyres.
- 5 Factories are a sources of dust and gaseous emissions in the atmosphere that subsequently settle or are washed down in rainfall.
- 6 Human activity within a catchment leads to the accumulation of litter, detergents from car washing and garden fertilizers.

The relative importance of these sources of pollution at any given location will be governed by the type of land use ie urban, industrial or commercial, and by individual catchment characteristics such as building density, slope, soil type etc.

For any given storm the degree of pollution is determined by the following factors:-

1 Intensity and duration of rainfall;

- 2 Length of the preceding dry period, which controls the build-up of pollutants in gutters and the quality of stored water;
- 3 Seasonal variations that occur in the rainfall pattern, temperature(which affects the degradation of organic matter), leaf fall, use of salt as a de-icer; and
- 4 Effectiveness of Council cleansing procedures.

The present study is being undertaken to gain an insight into the relative importance of the above listed factors in influencing the quality of stored water in roadside gully pots. The significance of this water in determining the degree of pollution of stormwater runoff is also being considered.

#### CATCHMENT CHARACTERISTICS, SAMPLING AND ANALYSIS

The catchment selected for study is at the Clifton Grove estate on the south side of Nottingham, England. Clifton Grove is a small urban estate covering an area of some 10.6 hectares, of which approximately 5.5 hectares are impervious, and which consists totally of middle income private housing. Built between 1973 and 1976 the development is spaciously arranged with areas of open land designated "amenity areas" interspersing areas of housing arranged around cul-de-sacs. In accordance with modern practice the estate is sewered on a separate system with the stormwater runoff draining directly into the River Trent.

The Clifton Grove catchment consists of seven subcatchments defined by the "arms" of the sewers. Each subcatchment being divided into microcatchments that are drained by individual gully pots.

The whole system contains 108 gully pots receiving runoff from the surface of roadways and paved areas. Each pot normally holds about 95 litres of water resulting in a total volume of stored water of 10,240 litres, there being no other storage of water in the system. Four gully pots were selected for detailed examination of pollution parameters over an extended period and the pots chosen were situated in microcatchments which illustrated different physical features. The following catchment characteristics were considered:-

- 1 Impervious area.
- 2 Slope.
- 3 Number of houses (important in determining the degree of human activity in the catchment).

4 Presence of open land which might contribute to the runoff entering the gully pot.

5 Situation on the estate. (cul-de-sac feeder road).

Table 1 outlines the nature of gully pot catchments chosen.

Table 1

Gully Pot	Slope	Imp. Area.	No. of Houses	Open Land	Situation
1	0.044	140m <sup>2</sup>	0	No	Subsidiary Rd.
6	0.038	680m <sup>2</sup> (	4	Yes	Cul-de-sac
7	0.040	513m <sup>2</sup> (	2.5	Possibly	Feeder Road
8	0.047	270m <sup>2</sup> (	2	Little	Cul-de-sac

Sampling of the gully pot liquors was initially undertaken weekly but subsequently this was reduced to fortnightly. In addition, other gully pots were sampled during dry periods and randomly selected samples of another 25 were taken on one day to assess the variation of pollution parameters and test the significance of a single sample in representing the quality of the stored water. All samples were drawn from a depth of 110mm below the surface of the water using a 1 litre bottle attached to a rod.

Sampling of the total runoff was achieved with a Rock and Taylor Multipurpose sampling machine which could receive a maximum of 48 x 500ml samples. The machine was triggered by a float switch that simultaneously activated a constant head/ continuous injection doser upstream of the sampler. The doser discharged lithium chloride at a constant rate of 1 ml/sec and the lithium chloride surface water mixture was sampled downstream. The concentration of lithium in the stream enabled the flow rate in the sewer to be calculated. Samples were drawn from the stormwater flow over a period of 100 sec, there being a cycle time of 270 sec between samples.

RESULTS AND COMMENTS

The results of the analysis to date are presented in Tables 2 & 3. Figures 2 to 4 and 6 to 8 show graphically the temporal variation of some of the pollution parameters.

- (a) Seasonal fluctuations and dry periods
  - Seasonal variation may be clearly seen in the results. Of all the parameters investigated ammonium appeared to behave in the most predictable fashion (Fig.3). It retained consistently low values throughout the winter months, high values only being associated with summer dry

<ul> <li></li></ul>	ומטופ ל Variation and mean values of para- מאממי מטונטיליטילימיי לאט אאסוס	<pre>meter concentrations for die more observation period. (All concen- trations in mo/l)</pre>		
ca <sup>2+</sup>	272	4	30.7	30.2
Hď	12.94	6.30	8.35	0.74
<b>'</b> Ӈ	0086	1.8	142.3	578
"EON	132	0.8	7.8	14.2
+ PHN	9.13	50*0	96.0	1.49
8	11.2	0	6.1	3.2
CD	163	8.7	39.1	23.4
GOA	46	6.0	6.8	8.3
DS	17,475	35	397	1,114

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Table 3 Variation of parameter concentravariation for a randomly selected sample of 25 gully pots, samples being collected on the same day. (All concentrations in mg/l).

	SS	DS	BOD	8	+ "HN	-Eon	ĿIJ	Hđ	Ca <sup>2+</sup>	ZN <sup>a+</sup>
mmtxew	252.8	292.5	41.6	428.6	4.62	28	36	8.38	35.8	0.570
Minimum	2.8	62.5	0.25	0	0.21	2.5	9.5	7.55	17.6	0
Mean	36.0	143.2	13.7	67.3	1.39	9.6	18.2	7.87	25.7	060-0
Standard Deviation	61.8	52.6	11.4	111.2	1.03	6.3	6.5	0.20	3.7	0.132
95% Confidence Limits	64.5%	13.5%	31.3%	60.4%	26.9%	25.4%	13%	%16°C	5.3%	52.4%

6

20.6

Standard Deviation

















periods. Nitrate also showed high values during the summer (Fig.4) and was possibly the source of the high ammonium values through the action of anaerobic bacteria. It is interesting to note that in the summer of 1977 nitrate values did not decrease through this conversion mechanism, possibly due to the fresh input of nitrate from car-wash liquors.

However, in the summer of 1976 when car washing was prohibited due to the water shortage low nitrate levels were recorded. High nitrate values were also recorded during periods of winter road salting, the input being thought attributable to the impure rock salt employed. This influx of nitrate did not significantly affect ammonium values due principally to the inhibition of bacterial action by the low temperatures. The build up of ammonium over a dry period is illustrated by Fig.5.

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Values of BOD and COD (Fig.6) showed a similar but more variable pattern in comparison to that observed for ammonia, low values occurred during winter months and high peaks were coincident with summer dry periods. The inhibition of bacterial action is reflected in the BOD values observed during the winter months. BOD remained virtually constant and winter dry spells did not cause any significant build up in oxygen demand.

Dissolved solids (Fig.7) mirror in a general fashion the changes already mentioned for specific parameters. The increase in pollutants and dissolved material over dry periods may be attributed to the following events:-

- Input from sweeping and wind blown material e.g. grasscuttings and organic debris providing fresh material for 2).
- 2 Anaerobic digestion of organic matter to soluble compounds by bacterial action.
- 3 Input of soluble material from human activity i.e. car washing.
- 4 Evaporation increasing effective concentration (minor).

In winter the major event to influence dissolved solids concentration was road salting, (Fig.8). This was not only the cause of sharp increases in the values of dissolved solids, sodium and chloride but also increases were observed for nitrate, potassium and calcium. These parallel increases being thought attributable to the impure rock salt employed.

Autumn leaf fall was considered to have no significant effect upon this catchment. Tucker (6) had reported clogging of gully pots by leaves with subsequent high increases in suspended solids, BOD and COD. However, due









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to regular street sweeping and the presence of only a few deciduous trees of any size on the estate no changes in quality from this cause were identified.

A significant parameter in specifying the quality of stored water is the dissolved oxygen level. Dissolved oxygen was observed to fall to zero after a period of 4 to 7 days dry weather. This resulted in septic conditions in the gully pot followed by the development of anaerobic digestion with subsequent increases in the values of BOD, COD and dissolved solids. During hot weather the production of hydrogen sulphide was evident and the gully pots developed a pungent smell.

Day to day examination of gully pot liquor during a dry period (Fig.9) showed a decrease in the amount of suspended solids following an inverse exponential form. The period of this decrease ranged from a few hours to five days dependent upon the type of material input (fine dusts associated with building sand had the slowest settling rate). General increases in dissolved solids were also observed. Observations during a rainstorm showed a sharp increase in suspended solids due to wash off from road surfaces and the stirring of bottom sediments. The levels of dissolved solids decreased to the same value as those of the incoming rainwater and then adjusted to a new steady state dependent upon the nature of the soluble material input.

(b) Human activity and catchment size The nature and level of human activity in the microcatchment of a gully pot is thought to be a significant factor in determining the chemical composition of the pot liquor. Unusually high values of parameters that are not mirrored in measurements from other gully pots may often be ascribed to human activity in the catchment. Some of the observations which may be attributed to human influence are listed below:-

- 1 Oil washed from road surfaces and in particular driveways was often present in the gully pots as a thin film on the surface of the liquor.
- 2 Frothiness due to car-wash liquids was commonly observed. Extreme cases of this were associated with abnormal nitrate levels, and during dry periods this may have had an influence upon ammonia levels.

3 In the summer months grasscuttings were found in abundance in gully pots and gutters and their presence could contribute to BOD levels during dry periods.

4 pH values up to 12.9 were recorded, probably due to the use of garden fertilisers and other agricultural chemicals which were blown or washed into the gully





Figure 11. Variation of dissolved oxygen concentration with runoff through gully pots 1,6,7 and 8 for week prior to analysis.





pots.

5 Salting of driveways was found to add large quantities of dissolved material above and beyond that due to council road saltings.

The level of human activity might be expected to correlate well with catchment area in an urban estate i.e. the larger the catchment area the greater the likelihood of human activity affecting the gully pot. Fig. 10, a plot of mean values of COD and BOD against catchment area, shows that values of COD correlate well with catchment area i.e. human activity, whilst the values of BOD appear to be independent of area. This suggests that the polluting input due to human activity is principally not organic in nature. Dissolved oxygen (Fig.10) showed a dependence upon catchment area, however, this was less a result of the human influence but was directly related to the volume of runoff through the gully pot. A graph of total runoff through the gully pot for the week prior to analysis plothed against values of dissolved oxygen is given in Fig. 1 . The graph indicates that to keep a gully pot fully aerated a runoff volume of 6m° per week is necessary. However, to keep dissolved oxygen at a reasonable level e.g. above 6mg/l, and assuming a typical weekly summer rainfall of 6mm a catchment area of 500m would be needed. The majority of gully pot catchment areas at Clifton Grove are smaller than this, e.g. gully pot 1, which is consequently rarely fully aerated. It would seem that to reduce pollution of gully pots by increasing the dissolved oxygen level, larger catchment areas should be employed.

### (c) Council cleansing procedures

The normal council cleansing procedure was to have the estate swept by hand fortnightly and the gully pots emptied on average once every 4 months. However, during the period of this study the gullys were emptied only once. The frequency of street sweeping appeared to be sufficient to keep the gutters and the estate generally clean and tidy. Despite the fact that pots were only emptied once, no exceptional depths of settled material were noted and a frequency of cleansing every 4 to 6 months appeared to be adequate as far as solid material was concerned. However, when the gully pot liquors were at their most polluting i.e. during summer dry periods, such a frequency was inadequate to maintain the pots in a wholesome state.

#### (d) Storm runoff

Three principal types of pollutant runoff curve have been identified (Fig.12). The type of curve followed is dependent upon the nature of the material being removed

and its origin. For example a parameter whose principal origin is in gully liquors will follow a curve of type 3 in figure 12, this type of removal is typical of ammonia, calcium and potassium. Pollutants whose prime origin is on the road surface or atmosphere will follow curves of types 1 or 2. The position of the peak of these curves is determined by the removal coefficient (i.e. ease of mobilisation) of the material. For example, suspended solids have a low removal coefficient and consequently peak concentrations are observed later in the storm. Soluble matter such as salt possesses high mobility giving rise to an earlier peak concentration. Comparison of levels of parameters as found in stormwater runoff compared to parameter levels found in gully pots confirms this argument (Fig.12). During winter months nitrate values were also found to be higher in runoff than in gully pots for the storms analysed to date, examination of roof runoff revealed its principal origins to be in rainfall.

The 108 gully pots on the Clifton Grove estate contain a total volume of some 10,240 litres. Comparing this volume to the storm runoff volume of 1,045,000 litres for the catchment over a 1 hr M5 rainfall of 19mm, reveals that gully liquors contribute some 1% of the total runoff. However some 40% of daily rainfall at Clifton Grove is equal to or less than 1mm resulting in a runoff volume of which gully liquors may contribute upwards of 20% of the total runoff. Bearing in mind that the greater part of the original gully liquors are discharged in the earlier stages of runoff the significance of gully pot quality in determining the day to day polluting nature of the 'first flush' of storm runoff can be seen.

#### (e) Laboratory simulation

Initial results from a laboratory simulation of the removal of settled material from gully pots have indicated that for all but the lowest flows complete mixing of the storm water inflow and the pot liquor is established rapidly and prevails during runoff. The removal of solids by the disturbance of the bottom sediments is dependent upon:-

- 1 rate of flow of water into the gully pot;
- 2 size of the gully pot;
- 3 depth of sediment within the pot; and
- 4 mass of sediment available for release.

The typical removal of sediment is illustrated by Figure 13. The rising limb, when material is being released from the bottom sediments, can be defined by an equation of the form:

$$e = A(1 - e^{BT})$$

where c = concentration of suspended solids in the outflow

t = time

A and B are constants dependent upon flow rate.

And the falling limb, after all available material has been released by an equation of the form:-

c = Ae Bt

The mass of sediment resuspended and released in runoff is relatively small compared to the mass remaining in the bottom of the gully pot. This indicates the general efficiency of gully pots in trapping heavier particulate matter. However, the finer sediment washed over is of greater significance in determining the polluting character of storm runoff.



Time from start of inflow, minutes

Figure 13. Laboratory simulation of the removal of solid material from a gully pot under steady flow conditions. (Initial depth of material in gully pot was 50 mm). CONCLUSIONS

There are a large number of variables which determine the quality of the stored water in any one gully pot or the effect of any one source of pollution. As a result it is difficult to isolate factors and express their effect in a quantitative manner from such a short study. However, the following general conclusions may be drawn from the work to date:-

1 The most important climatic influence affecting quality of gully pot liquors is summer dry periods. These dry periods result in sharp increases in the values of BOD, COD, nitrate, anumonia and dissolved solids.

2 The loss of dissolved oxygen and the consequent anaerobic conditions is possibly the major cause of pollution.

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- 3 Over dry periods human activity in the vicinity of a gully pot causes a significant deterioration of the pot liquor quality.
- 4 Human activity leads to variation in quality between gully pots.
- 5 COD values for the liquor appear to be dependent upon human activity/catchment area. BOD appears to be unaffected.
- 6 Dissolved oxygen levels are related to catchment area. The introduction of wider spacing between gully pots would increase average dissolved oxygen levels and hence reduce pollution of runoff due to gully liquors.
- 7 Comparison of pollutant runoff curves and values of certain parameters in gully pots provides an indication of the principal origins of pollutants.
- 8 Gully pots are efficient in the removal of heavier particulates from stormwater runoff.

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### MATHEMATICAL SIMULATION OF POLLUTANT CONTRIBUTIONS TO URBAN RUNOFF FROM ROADSIDE GULLY POTS

by

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### ABSTRACT

A mathematical model to predict the contribution of pollutants from roadside gully pots to stormwater runoff is presented. The model has been used to assess the importance of the stored sediments and water in gully pots in determining the pollutant loads in stormwater runoff.

#### LIST OF SYMBOLS

#### С Concentration of material in the outflow from a gully pot, mg/1. C Initial concentration of material in suspension in a pot, mg/1. 0 Water flow rate, 1/s.

V Volume of a gully pot, 1.

Percentage of gully pot fluid mixed. P

K Rate of release of bottom sediments, mg/s.

Total mass of material available for release from the bottom M sediments, mg.

#### INTRODUCTION

Roadside gully pots form reservoirs of polluted water that may adversely affect the quality of stormwater runoff (Fletcher et al., 1978; Mance and Harman, 1978; Tucker, 1975; Sartor et al., 1974). The stored water may be particularly polluting after periods of dry weather, when anaerobic degradation of the bottom sediments produces increases in BOD and COD, through the release of soluble organic compounds, ammonium and other pollutants. The degree to which the gully liquors may contribute to the pollutant load of stormwater runoff can only be estimated by consideration of the volume and quality of the stored water on a catchment in relation to the volume and quality of the stormwater runoff. However, a quantitative assessment of the contribution may be possible through mathematical simulation of the pollutant removal process involved. This has been attempted in the following manner:-

a) Laboratory investigations into the removal of dissolved solids and settled sediments from gully pots for different inlet water flow rates,

using purpose-built apparatus and, hence, to develop and calibrate equations describing the pollutant transport process.

b) Development of a rainfall-runoff model and its integration with the pollutant transport model enabling the simulation to predict the contribution to storm discharge made by water stored in the gully pots prior to runoff.

c) Contemporaneous on-site monitoring of rainfall, runoff volume and quality and the quality of stored water in gully pots was undertaken to provide the necessary data for utilising the model.

The site investigation was conducted on a small, separately sewered, residential, suburban catchment at Clifton Grove, Nottingham, U.K. The catchment covers some 10.6ha, of which 3.7ha are impervious roof and road surfaces. The storm sewer system has a mean slope of 4.5%, contains 108 gully pots with a total volume of stored water of some 92000 litres. Research on the factors affecting the quality of water stored in gully pots and the instrumentation used on this catchment have been reported elsewhere (Fletcher et al., 1978; Pratt and Adams, 1981; Pratt and Henderson, 1981).

### LABORATORY INVESTIGATIONS

### The Removal of Dissolved Material from Gully Pots

The removal of dissolved material was investigated by dosing a gully pot, typical of the type used on the Clifton Grove catchment, with a known concentration of salt solution and, for a given inlet water flow rate, continuously monitoring the pot liquor concentration using a conductivity probe. Salt removal was characterised by an exponential decay in concentration at all inlet flow rates. The pot liquors were completely mixed with the incoming water flow for flow rates greater than 0.121/s, however, below this value 'short-circuiting' occurred and the percentage of the pot in a mixed state varied with the incoming flow rate (Fig. 1). The change in concentration of dissolved components in the gully liquor could be expressed as:-

$$dC/dt = - (QP/100V).C$$
 (1)

$$C = C_0 e^{-tQP/100V}$$
(2)

where P = 100, when  $Q \ge 0.121/s$ ; and, P = 664.0Q + 19.7, when Q < 0.121/s

### The Removal of Sediments from Gully Pots

The removal of sediment was investigated using material obtained from municipal gully emptying tankers operating in the study catchment and its environs. Prior to use, the material was sterilised, dried and sieved: large twigs, paper, etc. were removed. 2kg of dry sediments were added to the base of the test gully pot, water was added and the contents were left for two days to allow the solids to wet thoroughly. Tests were performed for inlet water flows ranging from 0.1 to 1.01/s.

The removal of sediment occurred in two fractions: material already in suspension, which obeyed the same equation as determined for the











2	Mean	Range
Rainfall, mm	3.2	1.0 - 8.6
Storm Duration, min	94	33 - 178
'First Flush' Duration, min	• 53	22 - 85
% Gully Liquors Discharged	54	6 - 87
% Contribution to Total Runoff	11	3 - 17
% Contribution to 'First Flush'	14 .	8 - 18
% 'First Flush' Volume/Total Runoff	69	28 - 94





dissolved material, and material re-suspended from the bottom sediments by the stirring action of the incoming flow. The total mass of material released from the bottom sediments, M, and the rate of release, K, were dependent upon the inlet flow rate:-

For 
$$0 \leq t \leq M/K$$
,  $dC/dt = K/V - (Q/V).C$  (3)

$$C = K/Q$$
,  $(1 - e^{-tQ/V})$  (4)

When all the available material has been released, i.e.  $M/K \le t \le \infty$ ,  $C = K/Q \cdot (1 - e^{-MQ/KV}) \cdot e^{(-tQ/V + MQ/KV)}$ 

(5)

5.

The test results showed that there was good correlation between K and the inlet flow rate (r = 0.85, Fig. 2), however, the rapid rate of release of material resulted in a limitation of the accuracy with which the time to solids exhaustion could be determined and a poorer correlation between M and the inlet flow rate resulted (r = 0.75). When K and M were defined by linear regression equations and an average value of initial suspended solids was taken, then a good simulation of the laboratory results was obtained (Fig. 3). The results showed that the gully pot was efficient in retaining trapped sediments: a maximum of only 0.2% of the bottom sediments was removed by the highest flow rate examined. The resuspended sediments were rapidly removed and recorded concentrations of 90mg/l suggested that this source of material might be of importance in the 'first flush' of storms with high inlet flow rates. However, the removal of dissolved pollutants from gully pots represented the greatest influence on the outflow quality.

Whilst the removal of dissolved solids was thought to be a good simulation of what occurs in the field, the removal of sediments probably represented the least polluting situation, as in the field, greater depths of sediment, the occurrence of bacterial degradation and scums might all result in the discharge of greater solids loads from gully pots.

#### RAINFALL-RUNOFF MODEL

The rainfall-runoff model was based upon recent research in the U.K. (Kidd,1976; Kidd,1978; Price and Kidd,1978). The computer program was written in FORTRAN V, required 9 pages of computer storage and took from 10 to 15s of CPU time in execution. It comprised the following subroutines:-

a)HYETO Subroutine formulating the rainfall hyetograph from tipping bucket raingauge time-tip count data and applying a constant proportional loss model to account for differences between the observed and predicted discharge volumes.

**b)DEPSTO** Subroutine calculating the depression storage applicable to each road sub-catchment from a general regression equation (Kidd, 1978) and formulating the net rainfall hyetographs applying to both road and roof sub-catchments.

c)LINRES Subroutine calculating the inlet hydrograph for each sub-catch-

ment from the net rainfall hyetograph using a single linear reservoir routing model. The road sub-catchment surface routing coefficients were defined by a general regression equation (Kidd, 1978): the roof surface routing coefficients were assigned empirically.

d)MUSK Subroutine calculating fixed Muskingum-Cunge coefficients for the pipe network (Price and Kidd, 1978).

e)ROUTE Subroutine for routing the inlet hydrographs through the pipe network, using the Muskingum-Cunge coefficients, to form the outlet hydrographs.

f)GULLYS Subroutine calculating the contribution from water stored in gully pots to the outlet hydrographs and pollutant concentrations and loads at the storm sewer outfall derived from gully pots. The inlet flow rates applicable to each gully were calculated from the road subcatchment inlet hydrograph divided by the number of gully pots in the sub-catchment.

Within the limitations of the available data, the model appears to give a reasonable simulation of the storm runoff hydrograph (Fig. 4). The model gave satisfactory results for the prediction of dissolved pollutants in runoff that were derived from gully pots, however, application of the model to the prediction of the contribution of solid material was dependent upon further evidence confirming that the laboratory results adequately described the field behaviour. Research is currently in progress.

Calculation of the discharge of gully liquors by the model can be used as a quantitative method of defining the duration of the 'first flush'. The 'first flush' is here defined as "that portion of a storm runoff event during which 90% of the total discharge of original gully liquors has occurred". This definition is more flexible than a fixed time (e.g. first 30min of runoff (Mance and Harman, 1978)) and more consistent with the nature of the 'first flush'.

Constituents	Contribution t Mean	to %	Total Storm Load Range %
Suspended Solids	10.6		1.9 - 22.4
Dissolved Solids	20.7	- 1	3.8 - 40.2
COD	20.5		2.9 - 49.2
BOD	13.9		2.7 - 30.4
Nitrate	15.7		2.6 - 24.6
Ammonium	31.5		9.5 - 80.9

**TABLE 2.** Percentage of Pollutant Contributionsfrom Gully Pots in 11 Storms.

### POLLUTANT CONTRIBUTIONS FROM GULLY POTS

The model indicated that the original gully liquors were rarely completely discharged. A maximum of 86% discharge was recorded for a storm of some 9mm, but 50 - 70% discharge was typical for most storms investigated (TABLE 1). At the storm sewer outfall, gully pot liquors









contributed up to 17% of the total runoff volume, being of most significance for storms of 1 to 3mm rainfall (Fig. 5). For storms less than 1mm, roof runoff predominated over road runoff in the total runoff volume, whilst for storm events producing more than 5mm rainfall, original gully pot liquors became depleted.

Pollutant contributions from gully pots were of most significance for storms of 1 to 3mm rainfall e.g. dissolved solids (Fig. 6). For the majority of storms monitored, the suspended solids contribution from gully pots was derived from material already in suspension and not from the re-suspension of bottom sediments, since inlet flow rates rarely exceeded 0.21/s. Of the other constituents investigated, significant contributions were recorded for ammonium (up to 80%), COD (up to 49%), dissolved solids (up to 40%) and for BOD (up to 30%) in the total runoff quality from original gully pot liquors (TABLE 2). Some of the highest contributions were recorded for a storm which occurred after 10 days dry weather.

The 'first flush', as defined above, comprised a high percentage of the pollutant load in the monitored events. For long duration storms, the 'first flush', although only consisting of 30 to 50% of the total storm runoff, carried between 60 and 90% of the pollutant load (TABLE 3).

TABLE 3.	Typical Polluting Characteristics of the
	'First Flush' for Long Duration Storms
	Producing More than 3mm (Mean Rainfall 5.6mm).

	Contribution of		
	'First Flush' to Total Storm, %	Gully Liquor to 'First Flush', %	
Volume	41	14	
Suspended Solids Load	68	4	
Dissolved Solids Load	58	21	
COD Load	75	-11	
Ammonium Load	69	16	

### CONCLUSIONS

Mathematical simulation of the removal of pollutants from roadside gully pots has shown that:-

a) Gully pots may be efficient in retaining trapped sediments and may not be an important source, themselves, of the solids load in stormwater runoff.

b) The removal of dissolved pollutants from gully pots was readily achieved, although the total removal of original gully liquors during a storm event occurred only infrequently. The dissolved pollutants may represent a greater threat to stormwater discharge quality.

### ACKNOWLEDGEMENTS

N. G. Henwood, formerly of Trent Polytechnic, and C. H. R. Kidd of the Institute of Hydrology, U.K., for their advice during the research reported herein.

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