Synthesis and toxicity assessment of environment friendly high yield ceria nanoparticles for biosafety

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31 Abstract

32 Different compounds at nanoscale level show more efficient behaviour because of increased surface area and optical properties. CeO₂ nanoparticles are of great importance for their unique properties. 33 34 However, the extensive release of CeO_2 nanoparticles in the environment is also a serious problem 35 that must be addressed as available data related to ceria toxicity is currently not comprehensive. The present study was aimed to evaluate the potential of CeO_2 nanoparticles in biomedical applications 36 37 and assessment of their toxicity by using mutagenic and acute *in-vivo* approaches. High yield CeO_2 38 nanoparticles with spherical morphology and with an average size of 40nm were synthesized by 39 adopting the alkaline fusion method under mild conditions. The synthesized CeO₂ nanoparticles 40 showed antibacterial activity at different concentrations (50-500 µg/mL) against E. coli. The 41 antioxidant properties of CeO₂ nanoparticles were determined, and CeO₂ nanoparticles show 42 antioxidant behaviour that may be helpful for anti-cancer and anti-inflammatory drug preparation. 43 Ames test confirms no mutagenicity at different concentrations of CeO₂ nanoparticles. Moreover, in 44 the current study CeO₂ nanoparticles showed no toxicity for aquatic life even at a concentration of 45 100 mg/L in the same way the *in-vivo* toxicology was not evident even at the highest concentration (46 ≥5000 mg/kg BW for rats), no significant change was observed in haematological and biochemical 47 parameters of control and CeO₂nanoparticles exposed rats. *In-vivo* dermal toxicity was not observed 48 in rabbits at the application of 0.5 g CeO_2 nanoparticles. These results indicate the nontoxic nature of 49 these nanomaterials. However, further experimentation is recommended to completely define the 50 toxic potential of the nanomaterials.

51 Keywords: Green synthesis; Sustainable materials; Ceria nanoparticles; Nanotoxicity; Biosafety;
52 Mutagenicity and LD₅₀.

53 **1 Introduction**

54 In nature, cerium (Ce) is present with other lanthanide elements in different minerals like alanite, 55 bastanite, cerite, monazite and samarskite but bastanite and monazite are considered important commercially. Stability in Ce⁴⁺ state is a unique feature of Ce as other lanthanide elements are 56 57 stable in the trivalent state only. Ce (placed in lanthanide series of elements in the periodic table) 58 constitutes 0.0046% of earth crust and is therefore regarded as the most abundant rare earth metals 59 [1]. The most important compound of Ce used commercially is CeO_2 [2]. Nanoparticles of CeO_2 60 are important as they are not only more efficient than non-nano sized Ce [3] but also have an 61 important use in other applications as well. For example, in catalytic reactions^[4], infrared filters 62 coatings [5], buffer layer for many superconductors [6], for plastics manufacturing, used in 63 Infrared absorbents and as sintering additives, oxidation resistant coatings. They have useful 64 applications in oxygen pumps and also in oxygen sensors [1]. Alloys coating, electrolyte/ electrode materials in fuel cells [7]. In bearing balls, glasses and electronic devices [5] as UV absorbent 65 because of their absorption at ~ 400 nm which is the strongest absorption ability for any known 66 67 oxide [8].

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The emergence of new diseases and microbial resistance against already available antibiotics demonstrates the need for novel drugs used against these pathogens and for the treatment of metabolic and other disorders [9]. Nanobiotics is relatively a new idea and different nanomaterials are available with strong potential to inhibit bacterial growth [10]. The prepared silver nanoparticles were screened for their antibacterial activity against *K. pneumonia, P. Aeruginosa, S. aureus and E. coli* [11]. CeO₂ nanoparticles also possess antibacterial activity [12-14]. Along with many other applications, CeO₂ nanoparticles were also studied in the medical field as antiinflammatory and anti-oxidant agents [15, 16] suggested the role of CeO_2 nanoparticles in the treatment of neurodegenerative disorders like trauma [17], ageing [18], Parkinson's and Alzheimer's diseases [19] and many others. The anti-inflammatory [20] and anti-angiogenic effects of CeO_2 nanoparticles were also proposed [21]. Moreover, one of the important bio applications of CeO_2 nanoparticles is their biomedical applications in treatments of ischemia reperfusion injuries [22, 23].

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83 The phenomenon that attracted the environmental biotechnologists' attention towards the rapidly 84 growing field of nanotechnology was their hazardous side effects. Many of these nanomaterials 85 were hazardous in nature and their toxic effects are spread over a wide range of species from 86 aquatic organisms to mammalians on land and serious questions were raised about their fate in the 87 environment and environmentalists showed their concern about the hazardous potential of these 88 materials [24]. The toxic effects of a nano-product could be totally different after surface 89 modification and producers and customers should keep this fact in mind before launching and 90 using a new product [25]. Some reported toxicity effects are of fullerenes, q-dots, silver 91 nanoparticles, a small contribution of carbon nanotubes, and different metals and metallic oxides 92 such as Cu [26], Zn [27] and ZnO [28], SiO₂ [29] and TiO₂ [27]. As the concerns of scientific 93 society are increasing about the toxicity of nanoparticles. Thus, different sophisticated protocols 94 and technical approaches have been developed to study the adverse effects on different organisms.

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96 Therefore, living systems, under investigation, are of different trophic levels from microbes to 97 higher organisms, plants, and animals [30]. Comprehensive data related to CeO₂ nanoparticles is 98 not present to date. On the basis of available data from different studies and EPA reports, the status

99 of these nanoparticles is still unclear [31]. Mostly reports related to CeO₂ nanoparticles are based 100 on *in-vitro* studies [32-35], on plants [36] on *Cyriocosmus* elegans(spider) [37] on green algae 101 (Pseudokirchneriella subcapitata) [38] and cyanobacterial (Anabaena CPB4337) strains 102 [39]. Therefore, in view of all the above facts, it is evident that CeO_2 nanoparticles have multiple 103 applications in different industries and have great potential to be used in more fields in the future. 104 However, excessive use and release of these nanoparticles in the environment are also subject to 105 environmental safety and their toxic status is still controversial. The present study was an effort to 106 find comprehensive results of different methods used in toxicological studies used for the 107 assessment of the toxic potential of CeO₂ nanoparticles.

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109 The toxicological evaluation of nanoparticles is normally carried out using the same assays as used 110 in toxicity testing of chemicals. A major limitation, in the use, of these methods, is the novel 111 physicochemical properties of nanomaterials that can make the results of these assays less 112 reliable[40]. However, some nano products are ecofriendly and reportedly used to mitigate 113 pollutants[41, 42]. Hence, the development of newly standardized protocols for the assessment of 114 nanoparticle toxicity is necessary. Acute fish toxicity test, In-vivo toxicity testing, Oral LD₅₀, 115 Intraperitoneal LD₅₀, Acute Dermal Irritation, Ocular LD₅₀, In-vitro toxicity testing and 116 genotoxicity testing are the most common methods used in toxicology. The present study was 117 focused to synthesize the nanoceria, their characterization, exploration of their application 118 potential for antimicrobial and anti-inflammatory drug preparation and toxicological evaluation of 119 these nanoparticles by using mutagenic, aquatic, and acute *In-vivo* toxicity tests.

120 2 Materials and methods

The detailed experimental plan of the current study is given graphically in Fig. 1. CeO₂ nanoparticles were synthesized by slightly modification of the method mentioned by [43]. A mixture of NaOH and KOH (2.51 g: 2.44 g respectively) was placed in a Teflon container and 1.1 g of Ce (NO₃)₃. 6H₂O was added. The container was placed in a vacuum oven pre-heated at 170 °C for 30 minutes. After 30 min the container was kept on a bench-top and cooled. Finally, the product washing step was done with milli-Q water thrice and then dried at 60 °C for 8 h.

127 2.1 Characterization of CeO₂ nanoparticles

128 CeO₂ nanoparticles were dispersed in deionized H₂O and characterized by using zetasizer nano
129 [44], Field Emission Scanning Electron Microscopy (FESEM), UV-visible spectrophotometry,
130 Atomic Force Microscopy [45] and X-ray Diffraction Analysis (XRD) [46].

131 **2.2** Applications of CeO₂ nanoparticles

132 2.2.1 Antimicrobial activity assessment

The antimicrobial activity of CeO₂ nanoparticles was investigated by the disc diffusion method [47]. Fresh bacterial cultures of *Bacillus subtilis* and *E. coli* were used to detect the antimicrobial activity of CeO₂ nanoparticles. Penicillin G was used as a control. Different concentrations (50, 150, 250 and 500 μ g/mL) were used to check the antimicrobial activity of different concentrations of these nanoparticles. The results were recorded the next day.

138 2.2.2 Determination of antioxidant activity

139 The antioxidant potential of synthesized CeO₂ nanoparticles was determined by a proposed method

140 of Tian et al. [48]. Different concentrations (2.5-10 mg/mL) of ceria nanoparticles were prepared

- 141 and mixed with 5ml of 0.2 M sodium phosphate buffer (pH 6.6) and 5ml of 1% potassium
- 142 ferricyanide. Mixtures were placed in an incubator preheated at 50 °C for 20 minutes. Then 5 mL

143 of 10% TCA (trichloroacetic acid) was introduced and the mixture was centrifuged at 10,000 rpm 144 for 10 minutes in refrigerated centrifuged (5 °C). The supernatant was taken and to dilute the 145 mixture, 5 mL distilled water and 1 mL of 0.1% ferric chloride was added and absorbance was 146 measured at 700 nm.

147 2.3 Mutagenicity of CeO₂ nanoparticles

For the detection of CeO_2 nanoparticles spontaneous mutation causing ability, the Ames test was performed [49]. The Mutagenicity index (M.I) was calculated using Eq. (1). A sample is considered mutagenic when M.I in the case of TA98 is 2.00 and 1.80 with TA 100 [50].

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$$Mutagenicity index (M.I) = \frac{Number of revertant colonies in test plate}{Number of revertant colonies in the negative control plate}$$
(1)

152 **2.4 Acute fish toxicity testing**

153 Cyprinus carpio L., 1758 was obtained from a Fish farm, Satyana Road, Faisalabad, Pakistan. The 154 specimens were approximately 3 months old, the average body length measured was 4.15 ± 0.75 155 cm and body weights 1.0 ± 0.25 g. After 15 days, fish were ready to be used in acute fish toxicity 156 testing and OECD 203 guidelines were followed for this purpose. Three different concentrations 157 (1, 50 and 100 mg/L) of commercially available non-nano-CeO₂ and synthesized CeO₂ 158 nanoparticles were used in this study. CeO₂ is insoluble in water so it was dissolved in dilute nitric 159 acid and pH of soln. was adjusted at 4 before its addition in water. The effect of nitric acid (used 160 as a vehicle) was also determined. Change in activity and behaviour (including loss of balance, 161 lying laterally, movement in a spiral fashion with jerks, open mouth and rapid opercular 162 movements) and death rate of fish were observed after 0, 0.5. 1.0, 2.0, 4.0 6.0, 10.0, 24, 48, 72 and 96 h of exposure. 163

164 2.5 In-vivo toxicity testing

Acute oral toxicity assessment (oral LD50) was carried out by the administration of a single dose 165 166 (of different concentrations up to 5000 mg/kg body weight of rat) were orally for the determination 167 of median lethal dose LD_{50} . LD_{50} is considered a single dose to produce mortality in 50% of 168 individuals) in rats. Guidelines of the Organization of Economic Cooperation and Development 169 (OECD) 423 were followed to assess the toxicity of CeO₂ nanoparticles. Two sets of rats in 170 triplicates were used in this study for each concentration. The rats' group administered with normal 171 saline was used as a control. For intraperitoneal LD50 the maximum dose used in this study was 172 500 mg/L (as above this concentration, soln. was unable to pass through syringe of $28G_{1/2}$ gauge) 173 was administered intra-peritoneal by following the OECD guidelines and rats were observed for a 174 period of 14 days. Finally, rats were sacrificed for biochemical and haematological studies. The 175 dermal irritation/toxicity testing was carried out in adult albino rabbits. A standard dose of 0.5 g 176 of ceria nano-powder was applied on the skin of rabbits; normal saline was used as a control. 177 Effects were observed after 1 h, 24 h and 14 days intervals.

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3 Results and discussion

The field of nanotechnology is developing very rapidly in the last few decades due to its interesting applications in multiple disciplines. On the other hand, the threat to the environment and its living biota from these nanomaterials is also of great importance and must be addressed. Different efforts have been made in this regard, but satisfactory measures still needed to be taken. The present study aimed to synthesize CeO_2 nanoparticles by different methods, to characterize these as-synthesized nanoparticles, to explore their applications and their toxicological evaluation.

185 **3.1** Structural analysis of CeO₂ nanoparticles

186 For the synthesis of CeO₂ nanoparticles, a modified alkaline fusion method proposed by Jin et al. 187 [43] was adopted and pale-yellow nano-ceria powder was obtained. Dynamic light scattering 188 (DLS) is mostly considered a suitable technique for the measurement of nanoparticles size and 189 agglomeration in solution [51]. The average size of these nanoparticles, synthesized at 170 °C for 190 30 min was 40 nm that was similar to the size reported by Jin et al. [43]. In the modern era, several 191 analytical techniques and instruments are used for the characterization of nanoparticles. For the 192 determination of average size, surface charge, surface area, hydrodynamic diameter, elemental 193 composition, agglomeration and porosity, various types of spectroscopic and microscopic 194 techniques are available [52-54]. Fig. 2 illustrates the size distribution of synthesized CeO₂ 195 nanoparticles characterized by photon correlation spectroscopy. Under UV-Visible 196 spectrophotometer, newly synthesized CeO_2 nanoparticles absorb wavelength in a range of 310-197 340 nm (Fig. 2). Similar ranges for CeO₂ nanoparticles were previously reported [55-57].

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199 The technique is focused on the whole area of the micrograph and therefore is used to create a 200 surface image of the sample and also to generate estimate images with great depth of field. This 201 UV-absorption phenomenon makes CeO₂ nanoparticles a suitable substance for their use in UV-202 filters and UV blocking cosmetics, showing maximum absorbance at a range of 310 to 330 nm, 203 these findings are in line with the findings of Sohn et al.[58]. FESEM is very useful to study 204 materials from both the qualitative and quantitative analytical aspects[59]. In the current study, 205 FESEM reveals the information that CeO_2 nanoparticles were spherical in shape and weekly 206 agglomerated (Fig. 3(a)) and AFM images showed that A 3-D image of ceria nanoparticles showed 207 the maximum height of about 85 nm (Fig. 3(b)). X-ray diffraction (XRD) is a versatile, non-208 destructive technique that is used to analyze crystalline materials qualitatively and quantitatively.

210 The technique has been used to determine the overall structure of bulk materials, including lattice 211 constants, the orientation of single and polycrystals, identification of unknown materials, stress, 212 film thickness and texture etc.[60]. Structural identification of CeO₂ nanoparticles by means of X-213 ray diffraction in the range of angle 2θ between 20° and 80° as shown in Fig. 4, the peaks were 214 obtained at (200), (202), (222), (313) and (402). It was indicated that the crystalline and cubic 215 structure of CeO₂ nanoparticles of about 85 nm in size. Furthermore, by using XRD analysis, the 216 structural arrangements of Ce and O atoms can be determined at various positions. For example, 217 Fig. 5(a) shows the hexagonal structural model of Ce and O in a unit cell. The ball and stick type 218 alternative arrangements of atoms are presented in Fig. 5(b). Fig. 5(c) shows the polyhedral present 219 in the hexagonal structure of CeO_2 . Doted surface type structural model around the ball and stick 220 style of Ce and O atomic arrangement is shown in Fig. 6(d). Further, Fig. 6(e-h) shows the 3D 221 representation of (200), (202), (222), (313) and (402) planes respectively [61-63].

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3.2 Nanoparticles as nanobiotics

223 Treatment of different diseases with the help of different naturally available sources (plant extracts 224 etc.) and laboratory synthesized chemical drugs is a common practice. The emergence of new 225 diseases and microbial resistance against already available antibiotics demonstrate the need for 226 novel drugs used against these pathogens and also for the treatment of metabolic and other 227 disorders. Nanobiotics is relatively a new idea and different nanomaterials are available with 228 strong potential [64, 65] to inhibit bacterial growth e.g; Silver monovalent ions [11], Silver oxide 229 [66], Gold [67], Copper oxide [68] and Zinc oxide [52]nanoparticles are reported strong 230 antibacterial agents against a wide range of bacterial strains e.g; Bacillus subtilis, B. Calmette-231 Guérin (BCG, used as surrogate for TB in anti TB drugs), E. coli, Salmonella typhi, Staphulococcus aureus, Psedomonas aueroginosa, Micrococcus luteus, Klebsiella pneumonia and
 Compylobacter jejuni (Food borne pathogen). Recently, it was reported that oxygen-containing
 nano-shuttles can potentially use as anti-infecting nano-robots in upcoming graphene-based nano biomedical applications [69].

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237 The use of these nanoparticles in medicines [19] is a promising idea for the prevention of diseases 238 e.g. colon cancer [59], fibrosarcoma [70], neurodegenerative disease [71] and anti-obesity [72]. 239 Cytotoxic effects of CeO₂ nanoparticles were observed on E. coli and Bacillus subtilis and results 240 showed the antibacterial activity of these nanoparticles at different concentrations (50, 150, 250 241 and 500 µg/mL) against E. coli (Fig. 6, 7). It was revealed that a gradual increase in antibacterial 242 activity with an increase in the concentration of CeO_2 nanoparticles but these effects were more 243 visible in E. coli (Fig. 6(a)). Thill et al. [73] worked on the toxicity of ceria nanoparticles on E. 244 *coli* and on the basis of findings concluded that toxic behaviour of ceria nanoparticles is dependent 245 on direct exposure of cells to these particles and in media bacterial cells may produce some 246 chemicals that inhibit nanoparticles aggregation around and side the cells and as a result, 247 prevention of growth inhibition occurs. These results suggest a possible role of CeO₂ nanoparticles 248 in the development of nanobiotics. Similar findings, related to the antibacterial activity of CeO_2 249 nanoparticles against E. coli are also previously reported.

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Furthermore, Villa et al. [14] worked on the toxicity of CeO_2 nanoparticles on *E. coli* and on the basis of findings concluded that the toxic behaviour of CeO_2 nanoparticles is dependent on direct exposure of cells to these particles and in media bacterial cells may produce some chemicals that inhibit nanoparticles aggregation around and inside the cells and in the result of it, prevention of growth inhibition occurs [14]. Pelletier *et al.* focused on bacterial toxicity and visualized the toxic effects of CeO₂ nanoparticles based on different particle sizes and growth, different media and at variable pH [13]. *B. subtilis* and *E. coli* were used as model Gram-positive and negative bacterial strain respectively while a metal-reducing Gram-negative strain, *Shewanella oneidensis*, was also used.

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261 In this study, methods like disk diffusion test, minimal inhibitory concentration (MIC) 262 determination, CFU measurement, live/dead viability assay, Microarray hybridization and analysis 263 were used, and the obtained results indicate the antibacterial ability of these nanoparticles against 264 B. subtilis and E. coli at concentrations in the range of 50 to 150 mg/L in minimal media M9. 265 Growth inhibition was not observed in LB broth media that may be because of the release of 266 bacterial compounds in media and inhibiting nanoparticle aggregation around and into the bacterial 267 cells. No effects were observed on S. oneidensis growth [13]. The antibacterial activity of 268 nanoparticles depends on different parameters like size, dispersion rate, concentration, media and 269 cell wall composition of bacteria [13, 14].

270 **3.3 Reactive oxygen species scavenging**

Along with many other applications, nanoparticles were also used in the medical field as antiinflammatory and anti-oxidant agents. These nanoparticles actually act as ROS (reactive oxygen species) scavengers and hence are of great importance in the treatment of cancer therapy and their non-toxic behaviour, even at high concentrations, make them promising substances in nanobiotics. Singh et al.[3] suggested the role of ceria nanoparticles in the treatment of neurodegenerative disorders like trauma, ageing, Parkinson's and Alzheimer's diseases and many others. The antioxidant properties of these nanoparticles make them novel nano-pharmaceutics in near future.

279 Hence, antioxidant properties of these CeO₂ nanoparticles were also explored and results 280 confirmed that the reduced ability of these nanoparticles is an indication of their antioxidant 281 potential [48]. Fig. 8 illustrates the graphical representation of the gradual increase in reduction 282 activity with a gradual increase in concentration. Antioxidant activity of CeO₂ nanoparticles was 283 also reported in other studies and they act as intracellular ROS scavengers [74] therefore it can be 284 predicted that these particles can be used for anti-cancer and anti-inflammatory drugs preparation 285 however, data related to the antioxidant potential of these nanoparticles and about toxic effects is 286 still not sufficient and extensive research is required in this area [75, 76].

287 **3.4 Mutagenicity analysis**

288 Genotoxic tests both *in-vitro* and *in-vivo* is used for the identification of compounds that cause 289 genetic damage for example DNA and chromosomal damage or gene mutations etc. Compounds 290 for which genotoxic test results are positive could be human carcinogenic and mutagenic as they 291 have the potential to cause cancer or heritable defects Food and Drug Administration Authority 292 (FDA), USA and many other regulatory agencies have some recommended series of toxicological 293 tests (including mutagenicity testing) required for the regulation of newly commercialized 294 products (e.g; pharmaceutics, food items and other chemicals) prior to their release into the market. 295 Mutagenesis is a spontaneous change in genetic material due to chemical or physical agents in the 296 next generation that is different in a heritable way from their predecessor.

297

Genotoxicity testing is useful for the detection of carcinogenic potential of different chemical
compounds. The Ames test developed from the screening and selection of numerous histidine/
tryptophan mutants that were sensitive to reverse mutation by a variety of chemical mutagens[77,

301 78]. The test was particularly designed to detect those chemical substances that can cause 302 mutations. The test has worldwide importance as an initial screening test to find out the mutagenic 303 potential of new chemicals and pharmaceutical drugs because there is a high extrapolative value 304 for rodent carcinogenicity when we obtain a mutagenic response.

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306 Ames test is one of the most commonly practiced methods used to determine the mutagenicity of 307 different chemicals. In the present study, the mutagenicity of CeO₂ nanoparticles at three different 308 concentrations (50, 150 and 500 µg/ plate) was determined and the result showed the non-309 mutagenic behaviour of CeO₂ nanoparticles. The results of CeO₂ nanoparticles are presented in 310 Table 1. A comparison of CeO₂ nanoparticles with a positive and negative control in Fig.10 clearly 311 indicates the non-mutagenic nature of these nanoparticles. To date, no studies related to the 312 evaluation of CeO₂ nanoparticles mutagenicity by Ames test is present and more work is needed 313 to evaluate the mutagenic level of these nanoparticles.

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3.5 *In-vivo* toxicological findings

315 Model organisms (mice, rats and rabbits etc.) are routinely used to assess the safety levels of food 316 and drug additives, pesticides, fertilizers and industrial chemicals. Short term and long term 317 exposures through different modes are used to determine the toxicity of the chemical compound. 318 The in-vivo studies are performed under the strict guidelines of FDA (Food and Drug 319 Administration) and other regulatory agencies (OECD or OCED). Acute Fish toxicity test was 320 performed and effects of different concentrations (1, 50 and 100 mg/L) were observed and results 321 indicate that CeO₂ nanoparticles are non-toxic at concentration $\leq 100 \text{ mg/L}$ as no mortality or 322 change in behaviour was observed during the exposure time (96 h) (Table 2). Data related to fish 323 toxicity (in-vivo) is not available.

325 However, Hoecke et al. [78] also reported the nontoxic behaviour of these nanoparticles in aquatic 326 toxicity assessment of CeO_2 nanoparticles for *Thamnocephalus platyurus*, *Daphnia magna* and 327 embryos of Danio rerio (zebra fish). Comet assay and micronucleus test may be useful in 328 determining the damage at DNA or Chromosomal level. CeO₂ nanoparticles were non-toxic even 329 at a concentration of 5000 mg/Kg when administered orally in rats. Food and water intake, weekly 330 body weight measurements and daily behaviour observation showed no difference between rats of 331 test groups to the control group (untreated). All rats were healthy and alive. No clinical or 332 behavioural changes were observed. Meanwhile, some nanomaterials are very toxic at a very low 333 dose 2000 µg/mL of nano-graphene oxide in mice [79], in the same way remarkable toxicities 334 were observed at higher concentrations of 1–100 µg/mL of cadmium based quantum dots in mice 335 [80]. Therefore, the results of current study revealed that CeO_2 nanoparticles were non-toxic. 336 Although systematic experiments are required for complete risk assessment of CeO₂ nanoparticles 337 at different dose levels.

338

339 Another approach, Intraperitoneal LD_{50} , was also used to determine the toxic status of CeO_2 340 nanoparticles. Rats were injected intraperitoneal up to a max concentration of 500mg/kg and 341 observed for mortality and behavioural changes for a time period of 24 h. All rats survived and 342 showed no difference in behavioural changes to the control group (Table 3). Comet assay, 343 micronucleus test, and blood analysis can be more helpful techniques in determining the effects of 344 these CeO₂ nanoparticles on the organism [81] reported the first *in-vivo* cytotoxic study of 345 europium-doped Gd_2O_3 (a member of lanthanide series) nanotubes in mice that were injected 346 intraperitoneally. According to Liu et al. [81], Gadolinium (Gd) is non-toxic at low and medium

347 doses but shows some changes in blood chemistry at high doses and also at higher concentrations348 its accumulation in the spleen and kidney causes damage to these tissues.

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350 Rats were sacrificed and blood was used for blood biochemistry and haematological studies. 351 Treated and control groups showed no differences in haematology response variables. All 352 parameters fall within the accepted limits of normal variation (Table 4). No significant differences 353 were observed in serum biochemical parameters in tested and control groups as all the parameters 354 fall within the accepted limits of normal variations. The values of different parameters are given 355 in Table 5. Comet assay, micronucleus test, and tissue histopathology can be more useful in 356 determining the risk assessment of these nanoparticles. To date, no details about Oral LD₅₀ of CeO₂ 357 nanoparticles are available but this study gives an idea that CeO_2 nanoparticles cause no acute 358 illness when administered orally even at high concentrations.

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360 Chen et al. [82], reported the oral LD_{50} for Copper that is 413 mg/kg and they also investigated 361 and discovered that LD₅₀ was the same for nano and micro-sized CuO. In-vivo toxicological 362 studies of different other metal oxides have been carried out and scientists have reported the toxic 363 and nontoxic behaviour of these compounds. Different in-vivo toxicological studies for orally 364 administered nanoparticles (metals and their oxides) are available for Al₂O₃, Fe₂O₃, ZnO and CuO. 365 Different nano-sized metal and their oxides can cause chromosomal aberrations, oxidative damage 366 to DNA, breaks and mutations in the DNA strand. However, due to the presence of many 367 contradictory factors in literature, it's not easy to draw any conclusion about the toxic nature of 368 these nano-materials

370 As CeO₂ nanoparticles absorb UV, therefore, they are used in UV filters and can be used in UV 371 blocking creams [55, 56]. To determine the effect of CeO₂ nanoparticles on the skin an acute 372 dermal irritation test was carried out. Rabbit skin was observed for dermal irritation after 1 h, 24 373 h and daily observations up to the end of the observation period (14 days). No significant or minor 374 effects like Oedema or Erythma were observed on the skin and daily activities of rabbits also 375 remained unaffected Fig. 11. These results suggest that the application of these CeO₂ nanoparticles 376 on the skin is not dangerous. Although no *in-vivo* studies are present in the literature to support 377 this finding however in-vitro studies on different cell lines indicate that short term exposure to 378 these nanoparticles is non-toxic [83].

379 4 Conclusion

380 In conclusion, CeO₂ nanoparticles were synthesized successfully by the alkaline diffusion method. 381 UV-visible Spectrophotometry, Field Emission Scanning Electron Microscopy, Atomic Force 382 Microscopy and X-ray Diffraction Spectroscopy revealed the spherical shape of CeO₂ 383 nanoparticles with an average size of 40 nm. This small size and higher surface area of synthesized 384 CeO₂ nanoparticles showed enhanced antibacterial (*E. coli*) and antioxidant activity. Ames test 385 confirms the non-mutagenic effects of these nanoparticles. CeO₂ nanoparticles proved nontoxic 386 even at a higher concentration by toxicology evaluation in aquatic life (fish) and results revealed 387 that the CeO₂ nanoparticles showed no toxicity for *Cyprinus carpio* even at a concentration of 100 388 mg/L. Acute in-vivo toxicity assessment indicates the non-hazardous behaviour of these 389 nanoparticles against mice, rat and rabbit and *in-vivo* toxicology was not evident even at the highest 390 concentration (\geq 5000 mg/kg BW for rats), no significant change was observed in haematological 391 and biochemical parameters of control and CeO2 nanoparticles exposed rats. In-vivo dermal 392 toxicity was not observed in rabbits at the application of 0.5 g CeO₂ nanoparticles. The current 393 study provides the baseline information regarding CeO_2 nanoparticles which will be helpful for 394 the application of these nanomaterials. Furthermore, acute, sub-acute, chronic, *in-vitro* and 395 genotoxicity testing can enhance the safety of CeO_2 nanoparticles for useful applications such as 396 their role in nano-biotics, cancer therapeutics and solid oxide fuel cells. However, similar 397 comprehensive studies are required for other nanomaterials for their risk assessment.

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List of Tables

Dose (µg/ plate)	Number of revertants/ Plate (Mean ± SD*)		
	TA98	TA100	
NC	41.23 ± 2.09	152.4 ± 3.08	
50	35.34 ± 3.1	160.37 ± 5.2	
100	$39.7 \pm 4.5 $	158.54 ± 3.8	
500	46.56 ± 2.8	165.12 ± 4.1	
PC	546.2 ± 5.1	821.20 ± 3.1	
PC	546.2 ± 5.1	821.20 :	

Table 1. The number of reverse mutants induced by CeO₂ nanoparticles to *S. typhimurium*.

*SD = standard deviation, NC = negative control (Distilled water), PC = Positive control (TA98:K₂Cr₂O₇, TA100: NaN₃)

Parameters Vehicl			Nonnano-ceria (mg/L)		CeO ₂ nanoparticles (mg/L)			
(0.	(0.1%)							
	HNO ₃	1	50	100	1	50	100	
Anxiety		-	-	-	-	-	-	-
Rate of swimming		-	-	-	-	-	-	-
Loss of balance		-	-	-	-	-	-	-
Lying laterally		-	-	-	-	-	-	-
Opercular activity		-	-	-	-	-	-	-
Movements in circular form with	jerks	-	-	-	-	-	-	-

Table 2. Impact of CeO₂ nanoparticles and nitric acid on the behavioural pattern of *Cyprinus carpio* L., 1758 (common crap) exposed to different concentrations up to 96 h.

635
636 HNO₃ = nitric acid, - (sign indicates normal behaviour) +, ++, +++ (signs indicate the levels of parameters).

637

Group		Mean body weight	Cumulative	Acute intraperitoneal LD50
		(gm)	Mortality	(mg/kg)
Test (CeO ₂ NP)	Female	198±7	0/6	> 500
Control (normal diet)	Female	185±2	0/6	> 500

Table 3. Acute Intraperitoneal LD50 of CeO2 nanoparticles.

639 Note: Data based on females in the group (n=6/group)

640

Parameters	Treated group	Control
Haemoglobin (g/ dl)	15.8 ± 0.14	14.7 ± 1.4
Haematocrite (%)	36.85 ± 0.77	36.65 ± 4.16
Total RBC count (M/ μ L)	$6.58\pm.09$	6.51 ± 0.7
M.C.V.	55.95 ± 0.56	56.42 ± 2.43
M.C.H (PG)	24 ± 0.5	22.58 ± 1.04
M.C.H.C (G/dl)	42.9 ± 0.56	40.01 ± 0.88
Total WBC count (K/ µL)	6.1 ± 1.13	9.27 ± 3.63
Neutrophiles (%)	10 ± 7.07	14.3 ± 4.2
Lymphocytes (%)	90 ± 7.07	82.3 ± 3.4
Monocytes (%)	0	1.5 ±1.2
Eosinophils (%)	0	1.83 ± 0.98
Basophils (%)	0	0
platelets count (K/ μ L)	663.5 ± 23.3	655.67 ± 70.9
ESR (mm)	5.5 ± 0.7	6.67 ± 1.6

641	Table 4. Haematological parameters of control and CeO ₂ nanoparticles (5000 mg/kg) expose	ed
642	mice.	

Blood biochemistry	Treated	Control
Blood urea (mg/dl)	21.3 ± 5.6	18 ± 4.07
Serum creatinine (mg/dl)	0.46 ± 0.05	0.4 ± 0.11
ALT (SGPT) (U/L)	60.33 ± 13. 79	57.4 ± 5.70
Blood glucose random (mg/dl)	98.33 ± 10.4	95.5 ± 13.5
Blood urea nitrogen (BUN) (mg/ dl)	10 ± 2.64	8 ± 1.80

Table 5. Clinical biochemistry evaluations of CeO₂ nanoparticles exposed mice.







Fig. 2. UV-Visible spectrophotometry showing the UV absorbance by CeO₂ nanoparticles in a range of 310-400 nm with the highest peak at 324 nm.



Fig. 3. Morphological analysis of CeO₂ nanoparticles synthesized by alkaline fusion method; (a) FESEM image and (b) AFM topographic image.







Fig. 4. X-ray diffraction pattern of ceria nanoparticles shows different peaks.





Fig. 5. Structural representation; (a) Positions of Ce atoms bonded with O atoms in a cubic unit cell, (b) Doted surface of unit cell atoms, (c) Polyhedrons in a unit cell, (d) Ce bonding with O at

- 674 interstitial positions, (e) (222) Plane, (f) (313) Plane, (g) (402) Plane and (h) All planes in one
- 675 cubic unit cell at 3.51 Å from origin.
- 676





- **Fig. 6.** Circles show the zone of inhibition against different concentrations of ceria nanoparticles in plate seeded with (a) *E. coli* (b) *B. subtilis*.



Fig. 7. Graphical representation of antibacterial activity showing a greater zone of inhibition with

687 increasing concentration (DIZ: diameter of inhibition zone).



691 Fig. 8. A line graph showing the direct association of reduction ability with an increase in the

692 concentration of CeO₂ nanoparticles (OD: Optical density).



Fig. 9. The number of reverse mutants induced by CeO₂ nanoparticles to S. typhimurium TA98 697 and TA100.



- **Fig. 10.** No significant or minor dermal irritation like edema/erythema after 14 days. Treated with
- 700 CeO₂ and control (normal saline).